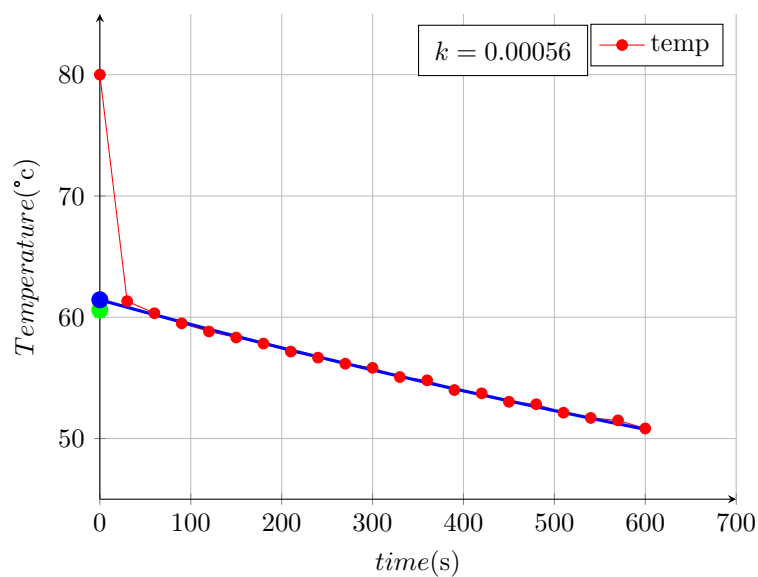


## Adding cold water after 0min

$$T(t) = S + (T_0 - S) e^{-kt}$$

$$T_{\text{eq}} = \frac{(V_1 T_1 + V_2 T_2)}{(V_1 + V_2)}$$

- Room temperature  $S = 24^\circ\text{C}$
- initial hot water temperature  $T_h = 80^\circ\text{C}$
- initial cold water temperature  $T_c = 24^\circ\text{C}$
- hot water volume = 150ml
- cold water volume = 50ml
- Equilibrium Temperature  $T_{\text{eq}} = 66.0$
- Theoretical temperature  $T_0 = 61.44$



$$T(t) = 24 + [Z - 24] e^{-0.00056t}, \quad R^2 = 0.997$$