Lecture 9

'Synthesis problems' including functional group interconversions

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Reference textbook

- "Organic Chemistry" by Paula Yurkanis Bruice, 8th Edn.
- This lecture Chapter 18, in particular 18.10, 18.16, 18.17, 18.18
- We are going to cover a few examples of designing a synthesis in this lecture, but you should read the other examples in sections 18.16, 18.17, 18.18

<u>Learning Objectives – Lecture 9</u>

Section 3– Functional group interconversion, Section 6 – synthesis problems

Propose potential syntheses of aromatic target molecules from given starting materials, using the reactions covered in this lecture course

Approach synthesis problems retro synthetically (i.e. work backwards from the target molecule to commonly available materials)

Plus a lot of other LOs from the Module, now applied to a synthetic problem

Proposing a synthesis

Faced with the question – how do I synthesise THAT?

Working backwards from the product is a **RETROSYNTHESIS** refer to page 344, Bruice,

The double open ended arrow indicates the LHS is made from the RHS....





 $\mathsf{B} \implies \mathsf{A}$

'B' can be made from 'A'

 $\mathsf{A} \longrightarrow \mathsf{B}$

'A' undergoes a reaction to give 'B'

- There is usually more than one option/synthetic route that can be taken
 - for an exam there is usually more than one answer that will score 100%.
- There can be pros and cons for different routes.
- Consider number of steps, usually shorter is better.



Example 1, decide what order substituents should be installed in

What reaction could install each substituent?

In what order should these reactions be done?

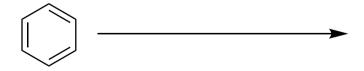
Example 1, decide what order substituents should be installed in

target compound

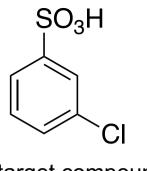
Design the retrosynthesis -

$$SO_3H$$
 CI

Write as a synthesis -



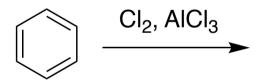
Example 1, decide what order substituents should be installed in



target compound

Consider what would happen if chlorination was done first

(.... which would not lead to the desired product)



Example 2, reaction order and functional group interconversion

What reaction **could** install each substituent?

But will direct instalment of each halogen work in this case?

Example 2, reaction order and functional group interconversion

- Maybe a functional group interconversion could be carried out....
- Whatever is installed first needs to direct the second substituent to

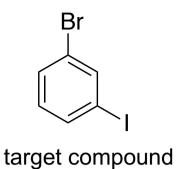
Design the retrosynthesis -

$$rac{\mathsf{Br}}{\mathsf{I}} \Longrightarrow$$

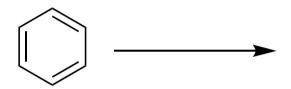
or

$$\bigcap_{\mathbf{I}}$$
 \longrightarrow

Example 2, reaction order and functional group interconversion

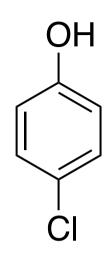


Design the synthesis, and show reagents



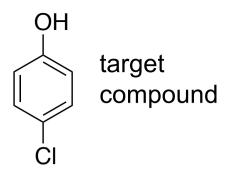
Example 3, reaction order and functional group interconversion

target compound para-chlorophenol



- What's a good way to synthesise a phenol?
- Whatever is installed first needs to direct the second substituent to

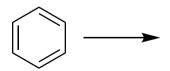
Example 3, reaction order and functional group interconversion



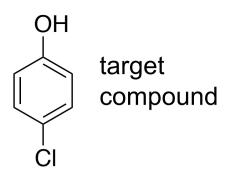
- As for most target compounds, there are many synthetic routes that could work
- To demonstrate this, let's go through a few possibilities for this target.....

Option 1 retrosynthesis

Option 1 synthesis



Example 3, reaction order and functional group interconversion

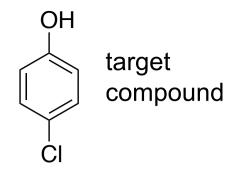


Option 2 retrosynthesis

$$\bigcup_{\mathsf{CI}}^{\mathsf{OH}} \longrightarrow$$

Option 2 synthesis

Example 3, reaction order and functional group interconversion



Option 3 retrosynthesis

$$\overset{\mathsf{OH}}{\longmapsto}$$

Write out the forwards synthesis yourself and fill in all the reagents

Proposing a synthesis – homework examples

Other targets to synthesise -

From the 2020 exam...

(c) Design a synthesis for ONE of the following compounds (i) OR (ii). Start from benzene and show the reagents you would use for each step. Assume that ortho and para products can be separated, noting when both may form. No mechanisms are required.

The End of Module 2 lectures (!)

Material revision and exam study, the text book is very useful -

We haven't covered EVERYTHING in Chapter 18, so what do you need to

know?



- Refer back to the 4 page PDF Learning objectives for the module
- The text book has some summary sections (both text and reaction summaries)
 and also some great questions (at the end of Chapter 18) to test your knowledge.

Use the past exams for CHEM202 to study – learning objectives the same

This afternoon you will be emailed a teaching evaluation for this module Please take the time to fill in – all feedback much appreciated