Unknown Carbonate Pre-lab

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q1 \mathbf{a} • Unbalanced: $Li_2CO_3 + HCl \longrightarrow LiCl + CO_2 + H_2O$ • $\text{Li}_2\text{CO}_3 + 2 \text{HCl} \longrightarrow 2 \text{LiCl} + \text{CO}_2 + \text{H}_2\text{O}$ b • Unbalanced: $(K_2CO_3 \cdot \frac{1}{2} H_2O) + HCl \longrightarrow KCl + CO_2 + H_2O$ • $2(K_2CO_3 \cdot \frac{1}{2}H_2O) + 4HCl \longrightarrow 4KCl + 2CO_2 + 3H_2O$ \mathbf{c} • $NaHCO_3 + HCl \longrightarrow NaCl + CO_2 + H_2O$ \mathbf{d} • Unbalanced: $Na_2CO_3 + HCl \longrightarrow NaCl + CO_2 + H_2O$ • $Na_2CO_3 + 2HCl \longrightarrow 2NaCl + CO_2 + H_2O$

q2

- Unbalanced: $BaCO_3 + HCl \longrightarrow BaCl_2 + H_2O + CO_2$
- $\begin{array}{l} \bullet \ \ \, BaCO_3 + 2\,HCl \longrightarrow BaCl_2 + H_2O + CO_2 \\ \bullet \ \, 0.4440g\,BaCO_3 \ \, \cdot \ \, \frac{1\,\mathrm{mol}\,BaCO_3}{197.34\mathrm{g/mol}\,BaCO_3} \ \, \cdot \ \, \frac{1\,\mathrm{mol}\,BaCl_2}{1\,\mathrm{mol}\,BaCO_3} \\ \end{array} \, \cdot \, \frac{208.23\mathrm{g/mol}\,BaCl_2}{1\,\mathrm{mol}\,BaCl_2} \\ \end{array}$ $0.4685g\,\mathrm{BaCl}_2$