Exercises

—,	Filling the blanks
1.	If the operator \hat{R} is a self-conjugate operator, it should satisfy the relationship
	, and the eigenvalues of self-conjugate operators are
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2.	The wavefunction should satisfy the three requirements:, and
3.	ψ (x_1 , y_1 , z_1 , z_2 , z_2) ² represents
4.	For the following functions, (A) $\cos kx$ (B) e^{-bx} (C) e^{-ikx} (D) e^{-kx^2} ,are the
	eigenfunctions of $\frac{d}{dx}$, andare the eigenfunctions of $\frac{d^2}{dx^2}$.
5.	The angular distribution of d_{z^2} orbital is $d_{z^2} = \frac{1}{4} \sqrt{\frac{5}{\pi}} (3\cos^2\theta - 1)$, then the angle(s) for the
	angular nodal plane is(are), and the angle(s) for the extremum
	is(are)
6.	For the electronic configuration p^2 , all the corresponding spectrum terms are
	, among which the ground-state spectrum term is
_	Short answer questions
	Please write the electronic configurations of the ground-state O_2 , O_2^+ , O_2^- , O_2^{2-} , and tell if
	th of them is paramagnetic or diamagnetic, calculate their bond orders, and list the order of
the	O-O bond strength.
三、	Calculations
1.	The wavefunction of 1s orbital of H is $\psi_{1s} = \left(\frac{1}{\pi a_0^3}\right)^{\frac{1}{2}} \exp\left(-\frac{r}{a_0}\right)$, please calculate the position

(r) corresponding to the maximum of the radial distribution function.