plate. The plate loses its charge due to the photoelectric effect. Discuss how the rate of loss of charge from the plate depends on the frequency and (a) intensity of the incident radiation. In your answer you should explain why: the plate loses its charge the photoelectric effect occurs only for frequencies greater than a particular value the rate of loss of charge increases with intensity for radiation above that particular value of frequency.

An isolated metal plate is given a negative charge. Electromagnetic radiation is incident on the

1.

(6)

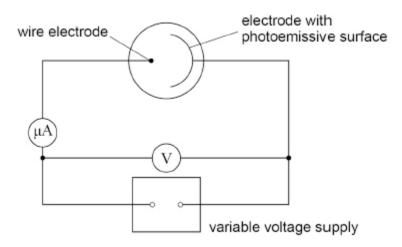
Charged particles are emitted from the metal plate with a maximum kinetic energy of $1.1\ eV$ when radiation of frequency $1.2\times 10^{15}\ Hz$ is incident on the plate.

Calculate, in eV, the work function of the metal.

(Total 9 marks)

Figure 1 shows an arrangement used to investigate the photoelectric effect. 2.

Figure 1



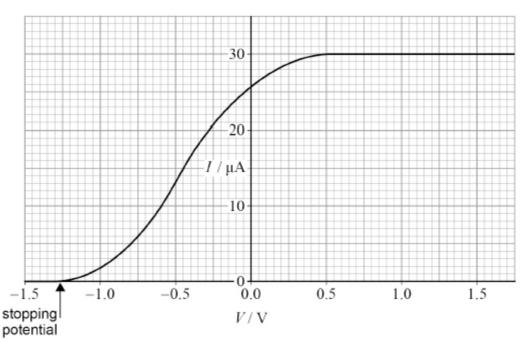
A current is measured on the microammeter only when electromagnetic radiation with a frequency greater than a certain value is incident on the photoemissive surface.

(a)	Explain why the frequency of the electromagnetic radiation must be greater than a certain value.

The apparatus in **Figure 1** is used with a monochromatic light source of constant intensity. Measurements are made to investigate how the current I in the microammeter varies with positive and negative values of the potential difference V of the variable voltage supply.

The **Figure 2** shows how the results of the investigation can be used to find the stopping potential.

Figure 2



(2)

0)	when the current is a maximum.	
	number of photoelectrons per second =	
:)	Explain why I reaches a constant value for positive values of V .	
l)	Explain why I decreases as the value of V becomes more negative.	
,	Explain why I decreased as the value of V becomes more negative.	
		
		

value of the work function. The sou	_		_	t:_1
Discuss the effect that this change	in surface has on t	the value of the	stopping potent	tial.
				-
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notoelectric effect can be demonst ertain metals, with ultraviolet (UV)	light and showing	that the plate los	harged plate, m	nade
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ertain metals, with ultraviolet (UV) Explain why, when ultraviolet light	light and showing	that the plate los	harged plate, m	nade

(b) Threshold frequency and work function are important ideas in the study of the photoelectric effect.

Tables 1 and **2** summarise the work functions of three metals and photon energies of three UV light sources.

Table 1

Metal	Work function / eV
Zinc	4.3
Iron	4.5
Copper	4.7

Table 2

Light source	Photon energy / eV
1	4.0
2	4.4
3	5.0

Discuss the combinations of metal and UV light source that could best be used to demonstrate the idea of threshold frequency and the idea of work function.		

(6)

(c)	Calculate the maximum kinetic eilluminated with ultraviolet light.	energy, in J, of the electrons emitted from a zinc plate when	1
	work function of zinc = 4.3 eV		
	frequency of ultraviolet light = 1.	2 × 10 ¹⁵ Hz	
		maximum kinetic energy J	(3)
(d)	Explain why your answer is a m	aximum.	
			(1)
		(Total 12	marks)
	oelectrons are released when mo ent on a metal surface.	pnochromatic light with a photon energy of $4.2 \times 10^{-19} \mathrm{J}$ is	
The	work function of the surface is 2.4	ŀeV.	
Wha	t is the maximum speed of the ph	notoelectrons as they leave the surface?	
A	$1.3 \times 10^6 \mathrm{m\ s^{-1}}$	0	
В	$6.3 \times 10^5 \text{ m s}^{-1}$	0	
С	$2.8 \times 10^5 \mathrm{m\ s^{-1}}$	0	
D	$2.0 \times 10^5 \mathrm{m\ s^{-1}}$	0	
		(Total 1	1 mark)

4.

5.	Monochromatic light of frequency f is incident on a metal surface in a vacuum. Photoelectrons are emitted from the surface.				
	The p	photoelectric current \emph{I} is measured.			
	The r	magnitude of the stopping potential	$V_{ m s}$ is then measured.		
	f is in	creased without changing the rate a	at which photons arrive at the meta	I surface.	
	What poter	are the new measurements of the patial?	photoelectric current and the magn	itude of the s	stopping
		Photoelectric current	Magnitude of the stopping potential		
	Α	I	V_{s}	0	
	В	I	> V _s	0	
	С	>1	V_{s}	0	
	D	> I	> V _s	0	
				l	(Total 1 mark)
6.	In ph	otoelectricity, $V_{ m s}$ is the stopping pote	ential.		
	What	quantity is $eV_{ m s}$?			
	Α	energy of an incident photon	0		
	В	maximum kinetic energy of a photo	pelectron		

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C

D

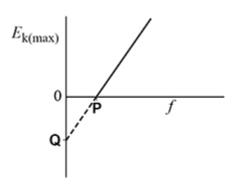
work function

threshold frequency \mathbf{x} the Planck constant

7.

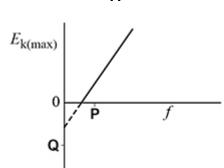
The graph shows how the maximum kinetic energy $Ek_{(max)}$ of photoelectrons emitted from a metal surface varies with the frequency f of the incident radiation.

 ${f P}$ is the intercept on the f axis. ${f Q}$ is the intercept on the $Ek_{(max)}$ axis.

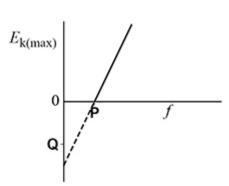


Which graph shows the variation of $\mathit{Ek}_{(\mathit{max})}$ with f for a metal with a greater work function?

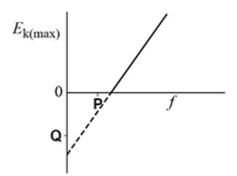
Α



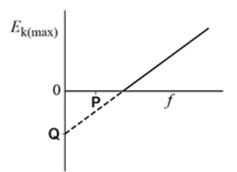
В



С



D



- Α Ο
- В
- С
- D O

8.	Monochromatic light with a photon energy of 4.1×10^{-19} J is incident on a metal surface. The maximum speed of the photoelectrons released is 4.2×10^5 m s ⁻¹ .
	What is the work function of the metal?

A
$$2.5 \times 10^{-19} \text{ J}$$

B
$$3.3 \times 10^{-19} \text{ J}$$

C
$$4.1 \times 10^{-19} \text{ J}$$

D
$$4.9 \times 10^{-19} \text{ J}$$

(Total 1 mark)

A photon has energy of 1 x 10¹⁸ eV.

An object of mass 0.03 kg has kinetic energy equa

An object of mass 0.03 kg has kinetic energy equal to the energy of the photon.

What is the speed of the object?

D
$$30 \text{ m s}^{-1}$$

(Total 1 mark)

Photons of energy 1.0×10^{-18} J are incident on a metal surface and cause the emission of electrons from the metal surface.

Which statement about the emitted electrons is correct?

A They each have a kinetic energy of
$$1.0 \times 10^{-18}$$
 J.

B They each have a kinetic energy that is a multiple of
$$1.0 \times 10^{-18}$$
 J.

C Their mean kinetic energy is
$$1.0 \times 10^{-18}$$
 J.

D The kinetic energy of each must be less than
$$1.0 \times 10^{-18}$$
 J.



11.

Light of frequency 2.0 \times 10¹⁵ Hz is incident on a metal surface. The work function of the metal is 4.6 \times 10⁻¹⁹ J.

Which statement is correct?

- A No photoelectrons are released.
- **B** Photoelectrons are released with a maximum kinetic energy of 3.1×10^{-19} J.
- **C** Photoelectrons are released with a maximum kinetic energy of 8.7×10^{-19} J.
- **D** Photoelectrons are released with a maximum kinetic energy of 18×10^{-19} J.