# Adding biological constraints to a flux balance model

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## INTRODUCTION

A metabolic model can be converted into a condition-specific model based on the imposition of experimentally derived constraints. Constraints can be defined, for example, by imposing upper and lower flux bounds for each reaction. There are several types of constraints that can be imposed in a metabolic model and that represent specific intra- and extracellular conditions, such as biomass maintenance requirements, environmental constraints, or maximum enzyme capacities.

In general, biomass constraints [1] are added as part of a biomass reaction. In some instances, however, a cell-type (e.g. neurons) does not divide, but is only required to turn over its biomass components. This tutorial is particularly relevant for such cases. Turnover rates are commonly expressed as half-lives  $(t_{1/2})$  and represent the time required for half of the biomass precursor to be replaced [3].

Using the experimental literature, metabolite  $t_{1/2}$  were collected and converted into turnover rates ( $\lambda$ ):

$$(1) \lambda = \frac{\ln(2)}{t_{1/2}}$$

We consider a biochemical network of m molecular species and n biochemical reactions. The biochemical network is mathematically represented by a stoichiometric matrix  $S \in \mathbb{Z}^{m \times n}$ . After calculating  $\lambda$ , we integrate it into the steady-state equation:

(2) 
$$Sv = \frac{dx}{dt}$$

The steady-state flux vector v and the change in abundance over time  $\left(\frac{dx}{dt}\right)$  share the same units, with x being the abundance of the corresponding biomass precursor. Equation (2) can thus be re-written:

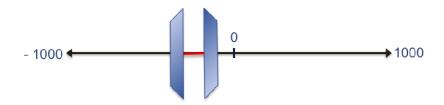
(3) 
$$Sv = \frac{dx}{dt} = \lambda x$$

## **PROCEDURE**

Unless the model is already loaded into the workspace, ReconX needs to be loaded. Here, we use Recon2.0 model for illustration, although any model can be used.

```
clear model
if ~exist('modelOrig','var')
    load('Recon2.0model.mat');
    model = Recon2model;
    model.csense(1:size(model.S,1),1)='E';
else
    model=modelOrig;
    model.csense(1:size(model.S,1),1)='E';
end
```

## 1. Environmental constraints



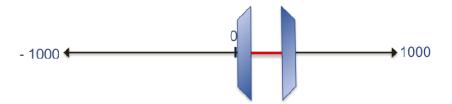
Environmental constraints are typically related to nutrient availability (e.g., glucose and oxygen). They can be defined using the function *changeRxnBounds* to set the minimal and maximal uptake and/or secretion rates possible in a specific condition. For example, in the caudate-putamen of the conscious rat, glucose consumption rate was found to range between -12.00 and -11.58  $\mu mol/gDW/hr$  [4]. Therefore, the lower bound of the glucose exchange reaction (EX\_glc(e)) can be set as follows:

```
modelConstrained = model;
modelConstrained.c = 0*modelConstrained.c; % remove any objective function
modelConstrained = changeRxnBounds(modelConstrained, 'EX_glc(e)', -12, 'l');
```

Optionally, to further constrain the model, an upper bound can also be imposed to force the model to take up between 11.58 and 12 units of glucose

```
modelConstrained = changeRxnBounds(modelConstrained, 'EX_glc(e)', -11.58, 'u');
```

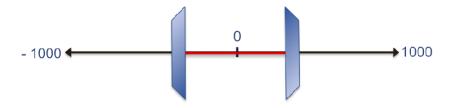
# 2. Internal enzymatic constraints



By convention, the bounds set on reaction rates in a metabolic model range from -1000 to 1000 and from 0 to 1000 for reversible and irreversible reactions, respectively [2]. Actually, the rate of a reaction is related to the activity of the enzyme catalyzing this reaction. Therefore, internal enzymatic constraints can be used to define the maximum capacity of a specified enzyme to catalyze a reaction (*Vmax*). For example, assuming that the reaction catalyzed by fructose-bisphosphate aldolase (FBA) has a *Vmax* of 128 units in our specific cell type, we can then add this constraint on the corresponding internal reaction FBA, as an upper bound.

```
modelConstrained = changeRxnBounds(modelConstrained, 'FBA', 128, 'u');
```

Optionally, if the reaction is reversible, the same constraint can be set as the lower bound, but with opposite signs.



```
modelConstrained = changeRxnBounds(modelConstrained, 'FBA', -128, 'l');
```

## 3. Constraints associated with biomass

In general, biomass constraints [1] are added as part of a biomass reaction by defining stoichiometric coefficients for each biomass precursor. For dividing cell types, the generic human biomass reaction available in Recon2 is formulated as follows:

```
printRxnFormula(modelConstrained, 'biomass_reaction');
biomass_reaction 20.6508 h2o[c] + 20.7045 atp[c] + 0.385872 glu_L[c] + 0.352607 asp_L[c] + 0.036117 gtp
```

#### 3.1 Biomass reaction

 $20.6508 \text{ h}20[c] + 20.7045 \text{ atp[c]} + 0.38587 \text{ glu\_L[c]} + 0.35261 \text{ asp\_L[c]} + 0.036117 \text{ gtp[c]} + 0.50563 \\ \text{ala\_L[c]} + 0.27942 \text{ asn\_L[c]} + 0.046571 \text{ cys\_L[c]} + 0.326 \text{ gln\_L[c]} + 0.53889 \text{ gly[c]} + 0.39253 \text{ ser\_L[c]} \\ + 0.31269 \text{ thr\_L[c]} + 0.59211 \text{ lys\_L[c]} + 0.35926 \text{ arg\_L[c]} + 0.15302 \text{ met\_L[c]} + 0.023315 \text{ pail\_hs[c]} \\ + 0.039036 \text{ ctp[c]} + 0.15446 \text{ pchol\_hs[c]} + 0.055374 \text{ pe\_hs[c]} + 0.020401 \text{ chsterol[c]} + 0.002914 \\ \text{pglyc\_hs[c]} + 0.011658 \text{ clpn\_hs[c]} + 0.009898 \text{ dgtp[n]} + 0.009442 \text{ dctp[n]} + 0.013183 \text{ datp[n]} + \\ 0.053446 \text{ utp[c]} + 0.013091 \text{ dttp[n]} + 0.27519 \text{ g6p[c]} + 0.12641 \text{ his\_L[c]} + 0.15967 \text{ tyr\_L[c]} + 0.28608 \\ \text{ile\_L[c]} + 0.54554 \text{ leu\_L[c]} + 0.013306 \text{ trp\_L[c]} + 0.25947 \text{ phe\_L[c]} + 0.41248 \text{ pro\_L[c]} + 0.005829 \\ \text{ps\_hs[c]} + 0.017486 \text{ sphmyln\_hs[c]} + 0.35261 \text{ val\_L[c]} -> 20.6508 \text{ h[c]} + 20.6508 \text{ adp[c]} + 20.6508 \text{ pi[c]} \\ \end{aligned}$ 

Any changes or adaptations can be introduced by adding a new formulation of the biomass function, using the function *addReaction*. For example, one can add the following new biomass reaction named *biomassReactionLipids*:

```
modelConstrained = addReaction(modelConstrained, 'biomasReactionLipids', '20.6508 h 20[c] + 20
test
biomasReactionLipids 20.6508 h20[c] + 20.7045 atp[c] + 0.15446 pchol_hs[c] + 0.055374 pe_hs[c] + 0.02040
```

# 4. Biomass maintenance constraints

To represent biomass maintenance (e.g. in neurons), the minimal biomass maintenance requirements can be used to set the corresponding constraints. Using the neurobiochemical literature, the degradation pathways for each biomass precursor has to be identified and the corresponding first reactions of these degradation pathways need to be mapped to ReconX. Using the fractional composition and the turnover rate of each biomass precursor, corresponding reaction rates

 $(\mu mol/gDW/hr)$  are calculated as described above and in Table 1. These reaction rates represent the minimal requirements for biomass maintenance of neurons in the human grey matter and must therefore be imposed as a lower bound on the corresponding degradation reaction(s) of the different lipids, amino acids, and nucleic acids.

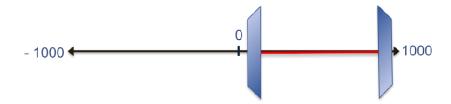


Table 1: The minimum metabolic maintenance requirement for neurons. This is a coarse-grained approximation of neuronal lipid, amino acid, and nucleic acid maintenance requirements converted into  $\mu mol/gDW/hr$ .

Recon X metabolite identifier	Metabolite name	Lower bounds (µmol/gDW/hr)
chsterol[c]	cholesterol	0.052
$pchol_hs[c]$	phosphatidylcholine	2.674
$pe_hs[c]$	phosphatidylethanolamine	2.708
$pail_hs[c]$	phosphatidylinositol	5.490
$ps_hs[c]$	phosphatidylserine	2.382
$sphmyln_hs[c]$	sphingomyelin	0.009
$clpn_hs[c]$	cardiolipin**	0.001
cmp[c]	cytidine monophosphate	0.649
amp[c]	adenosine monophosphate	0.302
gmp[c]	guanosine monophosphate	0.416
ump[c]	uridine monophosphate	0.322
$\mathrm{glu} \mathrm{L}[\mathrm{c}]$	glutamate	1.634
$asp\_L[c]$	aspartate	1.590
$ala\_L[c]$	alanine	1.531
gly[c]	glycine	1.032
$\operatorname{ser} L[c]$	serine	1.146
$\operatorname{thr}_{-}\!\!\operatorname{L}[\mathbf{c}]$	threonine	1.175
$arg_L[c]$	arginine	0.966
$phe_{-}L[c]$	phenylalanine	0.771
$pro_{-}L[c]$	proline	1.293
$\operatorname{tyr} \mathbf{L}[\mathbf{c}]$	tyrosine	0.745
val_L[c]	valine	0.940

<sup>\*\*</sup>cardiolipin is also known as diphosphatidylglycerol

# Calculation example of the minimal cholesterol maintenance requirement

## i. Identify metabolite abundance

Assuming that the specific tissue type has a total dry weight lipid composition of 39.6%. This means that there is 0.396gLipid/gDWtissue. If cholesterol has a molar composition of 31.3%, then in total there is 0.124g cholesterol per gDW of tissue.

$$\frac{31.3*39.6*1gDW}{100*100} = 0.124gDW$$

```
abundance = (31.3*39.6*1)/(100*100);
```

## ii. Calculate the molar abundance

In the experimental literature, cholesterol was also found to have a molar mass (M) of 386g/mol. Using equation (4), we can now convert the abundance (m) into molar units (n).

(4) 
$$n = \frac{m}{M}$$

```
M = 386; %g/mol
n = (abundance*1000000)/M; %micromol
```

# iii. Calculate the corresponding flux value

Finally, we know that in the brain, cholesterol has a very slow turnover and a  $t_{1/2}$  of 4320 hours. Using equation (3), we can now calculate the minimal cholesterol maintenance requirement in flux units ( $v_1$ ).

```
halfLife = 4320;
turnover = log(2)/halfLife;
v1 = n * turnover
```

The minimal cholesterol maintenance requirement was calculated to be  $0.0515 \, \mu mol/gDW/hr$  (Table 1). This value can now be used as a lower bound in the corresponding reaction.

# 4.1. Identification of degradation reactions for a biomass maintenance precursor

As previously mentioned, the degradation pathways for each biomass precursor can be identified using literature and the minimal maintenance requirements defined in section 4. can be used to constrain the first reactions of these degradation pathways. However, the identification of these reactions and the set-up of the associated constraints is not always straightforward. The following section presents the different common cases that can be encountered.

# A. Single irreversible degradation reaction

In cases where only a single irreversible degradation reaction exists for a biomass maintenance precursor, the imposition of the constraint is straightforward. For example, the major cholesterol excretion pathway in the brain involves the hydroxylation of cholesterol into the oxysterol 24-hydroxycholesterol. Only a subset of neurons express this 24-hydroxylase enzyme (P45046A1r) and it is mainly found in dendrites and somata, rather than in axons or presynaptic terminals (reviewed in [5]).

Therefore, add a lower bound ( $v_1$ , calculated above) on P45046A1r (Table 1).

```
modelConstrained = changeRxnBounds(modelConstrained, 'P45046A1r', v1, 'l');
```

# B. Single degradation reaction does not exist biochemically

A degradation reaction might not exist for a given biomass maintenance precursor. For example, the phospholipid cardiolipin is mainly present in the inner mitochondrial membrane, where it regulates the stability of the mitochondrial membrane protein complexes [6]. As part of mitochondria, cardiolipin

reaches the lysosome during macroautophagy (reviewed in [7]). It is then degraded to form the negatively charged bis(monoacylglycero)phosphate (BMP) on internal membranes.

If the corresponding demand reaction does not exist in the model, a demand reaction can be added using the function *addDemandReaction*.

```
modelConstrained = addDemandReaction(modelConstrained, 'clpn_hs[c]');

DM clpn hs[c] clpn hs[c] ->
```

Now, the constraint for cardiolipin (Table 1) can be imposed on the corresponding demand reaction

```
modelConstrained = changeRxnBounds(modelConstrained, 'DM_clpn_hs[c]', 0.001, 'l');
```

# C. Single reversible degradation reaction

In cases where the biomass precursor degradation reaction is reversible, it first needs to be split into two irreversible reactions. To this end, define the set of reversible degradation reactions (sRxns) and convert them into two irreversible reactions (i.e., sRXN\_b and sRXN\_f, respectively backward and forward reactions) using *convertToIrreversible*:

## i. Split the reversible reactions into irreversible

Select a set of reversible degradation reactions

```
sRxns = {'ASPTA'; 'GHMT2r'};
```

Copy the original model, except split a specific list of reversible degradation reactions into irreversible.

Split sRxns into irreversible reactions

```
[modelIrrev] = convertToIrreversible(modelConstrained, 'sRxns', sRxns);
```

You can check if the conversion as been done properly by searching the split reactions

```
for j=1:length(sRxns)
   if isempty(findRxnIDs(modelIrrev, [sRxns{j} '_f']))
      error('Forward reaction not found')
   end
   if isempty(findRxnIDs(modelIrrev, [sRxns{j} '_b']))
      error('Reverse reaction not found')
   end
end
```

## ii. Impose the calculated constraints

Examples are given for aspartate and glutamate (Table 1).

```
constraints = [1.590; 1.146];
```

You can also identify the new reaction names as follows:

```
rxns = setdiff(modelIrrev.rxns, modelConstrained.rxns)
```

```
'ASPTA_b'
'ASPTA_f'
'GHMT2r_b'
'GHMT2r f'
```

Using this list (rxns), manually identify the corresponding reactions that should be constrained

```
splitRxns= {'ASPTA_f'; 'GHMT2r_f'};
```

Identify the indices of these split reactions in the new model, using findRxnIDs

```
ind = findRxnIDs(modelIrrev, splitRxns);
```

Now, impose the constraints using a for loop. Note that you can easily account for experimental errors by defining a percentage error (e.g., expError = 0.25) for the constraint values.

## D. Multiple irreversible degradation reactions

In some cases, several degradation pathways may be available for one biomass precursor. For example, in the brain, phosphatidylcholine (PC) can be degraded by 3 different metabolic pathways [8]:

- PCHOLP\_hs: Phospholipase D acts on the choline/phosphate bond of PC to form choline and phosphatidic acid.
- PLA2\_2: Phospholipase A2 acts on the bond between the fatty acid and the hydroxyl group of PC to form a fatty acid (e.g. arachidonic acid or docosahexaenoic acid) and lysophosphatidylcholine.
- SMS: Ceramide and PC can also be converted to sphingomyelin by sphingomyelin synthetase.

Define the set of potential reactions associated with the degradation of PC

```
multipleRxnList={'PCHOLP_hs', 'PLA2_2', 'SMS'};
```

Make sure that all the reactions are irreversible. The lower bounds should be 0 and the upper bounds 1000.

```
modelConstrained.lb(findRxnIDs(modelConstrained, multipleRxnList));
ans =
0
0
modelConstrained.ub(findRxnIDs(modelConstrained, multipleRxnList));
```

1000

1000

1000

Constrain the weighted sum of fluxes to be above a lower bound (e.g. value of the maintenance requirement of PC in Table 1:  $d = 2.674 \ \mu mol/gDW/hr$ ). The weight for each reaction are defined in *c*.

```
c=[1,1,1];
d=2.674;
ineqSense='G';
modelConstrainedAb=constrainRxnListAboveBound(modelConstrained,multipleRxnList,c,d,ineqSense);
rxnInd = findRxnIDs(modelConstrainedAb, multipleRxnList);
```

#### Check the constraints are there

```
[nMet,nRxn]=size(modelConstrainedAb.S);
modelConstrainedAb
```

```
modelConstrainedAb =
                      S: [5063x7444 double]
                   rxns: {7444x1 cell}
                     lb: [7444x1 double]
                     ub: [7444x1 double]
                    rev: [7440x1 double]
                      c: [7444x1 double]
             rxnGeneMat: [7444x2194 double]
                  rules: {7444x1 cell}
                  genes: {2194x1 cell}
                grRules: {7444x1 cell}
             subSystems: {7444x1 cell}
               rxnNames: {7444x1 cell}
              rxnKeggID: {7440x1 cell}
    rxnConfidenceEcoIDA: {7440x1 cell}
    rxnConfidenceScores: {7440x1 cell}
             rxnsboTerm: {7440x1 cell}
          rxnReferences: {7440x1 cell}
           rxnECNumbers: {7440x1 cell}
               rxnNotes: {7440x1 cell}
                   mets: {5063x1 cell}
                      b: [5063x1 double]
               metNames: {5063x1 cell}
            metFormulas: {5063x1 cell}
              metCharge: [5063x1 double]
             metCHEBIID: {5063x1 cell}
              metKeggID: {5063x1 cell}
           metPubChemID: {5063x1 cell}
         metInchiString: {5063x1 cell}
         metHepatoNetID: {5063x1 cell}
              metEHMNID: {5063x1 cell}
            ExchRxnBool: [7440x1 logical]
              EXRxnBool: [7440x1 logical]
              DMRxnBool: [7440x1 logical]
            SinkRxnBool: [7440x1 logical]
            SIntRxnBool: [7440x1 logical]
                metHMDB: {5063x1 cell}
                modelID: 'Recon2.0model'
                 csense: [5064x1 char]
                  match: [7444x1 double]
        reversibleModel: 0
                      C: [1x7444 double]
                      d: 2.6740
```

Solve the FBA problem with added constraints  $C^*v >= d$ , with or without objective function

```
%modelConstrainedAb = changeObjective(modelConstrainedAb, 'DM_atp_c_');
FBAsolution = optimizeCbModel(modelConstrainedAb,'max',1e-6);
```

Check the values of the added fluxes

```
FBAsolution.x(rxnInd)

ans = 3x1 double

2.0509
0.2333
0.3898
```

Therefore, when you solve the FBA problem with this last constraint, the sum of flux values associated with these three reactions should be greater than the value of d

```
sum(c*FBAsolution.x(rxnInd))
ans = 2.6740
return;
```

# CRITICAL STEP: Collection of data and conversion of experimental fluxes (Timing: 4-8 weeks)

The most time consuming step when imposing constraints is the collection of required information. Depending on the available experimental literature, it can take between 4-8 weeks to retrieve the biomass composition and the turnover rates of the different biomass precursors. It is crucial to correctly convert the obtained data into the corresponding fluxes. It is recommended to first define the flux unit you wish to use. A common unit used for prokaryotic models is micromol per gramDryWeight per hour  $(\mu mol/gDW/hr)$ . However, in the experimental literature, a wide range of units is provided. Therefore, after each conversion, it is strongly recommended to double check the calculations to avoid modelling artifacts. Once all the constraints are available, it can take less than 5 minutes to impose the constraints on the corresponding reaction bounds, according to the information provided in this tutorial.

## **ANTICIPATED RESULTS**

After imposing the above constraints, we can now test the likely outcome of an optimisation problem using a constraint-based model. For example, we can take advantage of sparseFBA to identify the minimal set of essential reactions required to fulfill a certain objective function (e.g. DM\_atp\_c\_).

```
originalTest = model;
originalTest = changeObjective(originalTest , 'DM_atp_c_');
[vSparseOriginal, sparseRxnBoolOriginal, essentialRxnBoolOriginal] = sparseFBA(originalTest);
```

```
---FBA---
1000 FBA objective.
1499 reactions above epsilon = 1e-09
0.49 computation time (sec)
```

```
---Non-convex approximation---

1000 = Sparse FBA objective.

0 = ||c^T*v - f*||^2.

1499 reactions above epsilon = 1e-09

3.55 computation time (sec)

129 of these are heuristically minimal rxns.

129 of these are essential rxns.
```

1515 reactions above epsilon = 1e-09

3.98 computation time (sec) 172 of these are heuristically minimal rxns.

172 of these are essential rxns.

In the absence of constraints, the minimal set of reactions required to maximise the objective function is 129 essential reactions.

After the addition of constraints, the minimal set of reactions required is increased to 172 essential reactions. Therefore, in this example, it is useful to integrate cell-type specific constraints to further refine the minimal set of essential reactions. In most cases, constraints also allow us to restrict the feasible solution space.

## REFERENCES

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