



User Guide

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1 Introduction to hy2Foam

hy2Foam is a newly coded open-source two-temperature computational fluid dynamics (CFD) solver that has been developed to tackle the highly complex flow physics of the hypersonic Earth atmospheric re-entry. Implemented within the OpenFOAM framework, the code has the capability to model physical phenomena relative to the high-speed chemically-reacting environment surrounding a spacecraft. The non-equilibrium conditions are treated by making the distinction between the trans-rotational and multiple vibrational-electronic energy pools. This permits to model the energy exchanges between these energy pools and to introduce a chemistry-vibration source term into the Navier-Stokes-Fourier equations. Various transport models have been implemented to determine species and mixture thermal properties at high-temperatures. There are also several new species diffusion models and boundary conditions available.

2 Installation of hy2Foam

The installation of hy2Foam is straightforward. Please make sure that the path to your OpenFOAM user directory is correct in the bash script called install.sh before execution. This script will install the new strath libraries located into the src folder, will install the hy2Foam solver located into the applications folder (as well as a few other apps) and will copy some tutorials into your run directory.

3 hy2Foam working environment

3.1 The main folder

The main working folder is no different from any other OpenFOAM solvers. The bash script *Allwelean* removes any temporary files within the main directory.

```
referenceCase
VincePC: referenceCase >ls
0 Allwclean constant gnuplot pv.foam system
VincePC: referenceCase >
```

Figure 1: Main folder contents

3.2 The θ folder

The 0 folder contains the traditional p and U files. The other mandatory files are the transrotational temperature, printed as Tt, the mass-fraction and the vibro-electronic temperature for each species. The prefix Y_{-} is omitted for mass-fractions which means that the N2 file is the mass-fraction of N_2 . The vibro-electronic temperatures are denoted by the prefix Tv_{-} in the multi vibro-electronic pool configuration (and simply Tv for the single vibro-electronic pool arrangement).

```
Terminal
VincePC: referenceCase >ls 0
        N
             N2
                  NO
                       0
                           02
                                   Τt
                                           Tv_N
                                                  Tv_N2
                                                          Tv_N0
                                                                  Tv_0
                                                                         Tv_02
                                                 Tv_N2+
include N+
            N2+
                  NO+ O+
                           02+
                                   Tv_e-
                                           Tv_N+
                                                          Tv_NO+
                                                                         Tv_02+
                                                                  Tv_0+
VincePC: referenceCase >ls 0/include/
boundaries initialConditions
VincePC: referenceCase >
```

Figure 2: The 0 folder contents

It is good practice to initialise the fields via an include statement to a dictionary and to set some of the boundary conditions by calling another dictionary. These dictionaries are regrouped into the 0/include folder. For instance, the initialConditions dictionary can be defined as follow

```
30000:
Ttr
Tve
          10000:
TveAtom
          0;
inletPressure 1.01325e5;
inletVelocity (0 0 0);
initialVelocityField (0 0 0);
Y N2
          0.4;
Y_02
          0;
Y_NO
          0;
Y_N
          0.2;
Y_0
          0;
          0.399992;
Y_N2+
Y_02+
          0;
Y_NO+
          0:
Y_N+
          0;
Y_0+
          0;
Y e-
          0.000008:
```

while the boundaries dictionary is for the case of an adiabatic heat bath

The user must set the vibro-electronic temperature for each molecule. For atoms, ions and electrons, if the electronic energy is switched on, then T_{ve} is to set to the vibro-electronic temperature of a reference species defined in the **constant** folder. In this case,

```
TveAtom = $Tve;
```

Otherwise, it is important to set T_{ve} to 0 for each of the non-vibrationally excited species composing the mixture. Several examples are given in the $\theta/default$, $\theta/5species$, and $\theta/11species$ folders.

3.3 The *constant* folder

The constant folder regroups all the dictionaries relative to the physicochemical phenomena occurring at hypersonic speed that have been implemented into hy2Foam. The mesh remains located into the polyMesh sub-folder.

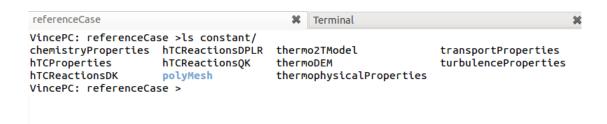


Figure 3: The *constant* folder contents

A short description of the different dictionaries is given below. It is recommended to follow the order listed here while editing the entries of the dictionaries.

- 1) thermophysical Properties: the standard OpenFOAM dictionary that prescribes which thermophysical models are going to be used and the location of some sub-dictionaries.
- 2) turbulenceProperties: the standard OpenFOAM dictionary that deals with turbulence modelling. In hy2Foam, the options are laminar and RAS2Model.
- 3a) hTCProperties: the dictionary that allows the user to activate or deactivate the high-temperature chemistry (hTC) features and to output some desired species-related fields such as number densities or vibrational degrees of freedom.
- 3b) chemistryProperties: the dictionary that defines the different chemistry inputs
- 3c) hTCReactionsDPLR / hTCReactionsQK / ...: the list of the species composing the mixture is defined in those dictionaries. It is strongly recommended not to change the order of the species in the list (molecules - ions - atoms - free electrons). Please uncomment the species and chemical reactions you may need.
- 4) *thermoDEM*: the dictionary that provides the species thermodynamic data for the different species loaded in dictionary 3c).
- 5) thermo2TModel: the dictionary that provides all the coefficients that enter into the different energy exchange processes.
- 6) transportProperties: the dictionary that defines the models used for species diffusion and the mixing rule. Some parameters such as the mean free path or the local gradientlength Knudsen number can be printed from this dictionary.

In the subsequent subsections, the main features of the dictionaries listed above are reviewed.

3.3.1 The thermophysical Properties dictionary

```
FoamFile
    version
                2.0;
                ascii:
    format
    class
                dictionary;
    location
                "constant";
    object
                thermophysicalProperties;
thermoType
{
    type
                    heRho2Thermo;
    mixture
                    reacting2Mixture;
    transport
                    BlottnerEucken;
                    decoupledEnergyModes;
                   sensible2InternalEnergy;
    energy
    equationOfState perfect2Gas;
                    advancedSpecie;
hyLight yes;
downgradeToSingleTv no;
downgradeToSingleTemperature no;
chemistryReader foam2ChemistryReader;
foamChemistryFile "$FOAM_CASE/constant/hTCReactionsDPLR";
foamChemistryThermoFile "$FOAM_CASE/constant/thermoDEM";
twoTemperatureDictFile "$FOAM_CASE/constant/thermo2TModel";
temperatureBounds
    Tlow
            200:
    Thigh
            22000;
```

The thermoType defined in the thermophysicalProperties dictionary uses the new strath thermophysical models. The number 2 indicates that the two-temperature model is employed. The user has the choice between the four transport models listed in Table 1. The thermo class decomposes the contribution of the different energy modes and the related sub-dictionary is later called via the foamChemistryThermoFile entry.

Model name	shear viscosity	thermal diffusivity
constant	constant	constant Eucken
BlottnerEucken	Blottner	Eucken
powerLawEucken	power law	Eucken
Sutherland Eucken	Sutherland	Eucken
CEA	CEA2 (NASA)	CEA2 (NASA)

Table 1: Available transport models in hy2Foam

hy2Foam is a two-temperature solver with multiple vibro-electronic temperatures. The two booleans downgradeToSingleTv and downgradeToSingleTemperature alter hy2Foam capabilities when simpler analyses can be conducted. The downgradeToSingleTv option downgrades hy2Foam to a single vibrational temperature solver. hyLight is an option that permits to run a light version of hy2Foam, thus saving time. The temperature bounds are here to make the code more robust during the transient phase. Please make sure that none of the steady-state

temperature fields are reaching these bounds.

The foamChemistryFile entry tells which set of chemical rate constants is going to be used by linking to the appropriate sub-dictionary. Available sets of chemical rate constants and their respective meanings are listed in Table 2.

Set	Meaning
DPLR	Park's rates (used in the DPLR solver), will soon be replaced
QK	Quantum Kinetics rates
DK	Dunn and Kang rates
ES	Evans and Shexnayder (1980)
J92	Jachimoski's rates (1992)
J92_fwd	Jachimoski's rates (1992), forward only
Mars94	Park's rates (1994)
Earth93	Park's rates (1993), available soon (to replace DPLR)

Table 2: Available chemical rate constant sets in hy2Foam

3.3.2 The turbulenceProperties dictionary

As stated earlier, the turbulence models (simulation Type) available in hy2Foam are laminar and RAS2Model.

If RAS2Model is chosen, then the definition of a new dictionary, RASProperties, is required

```
FoamFile
 version
       2.0;
 format
       ascii;
      dictionary;
 location
       "constant";
 object
       RASProperties;
RAS2Model
       kOmegaSST;
turbulence
       off;
printCoeffs
```

The different choices for the RAS2Model entry are laminar, kOmegaSST, kEpsilon and SpalartAllmaras.

3.3.3 The hTCProperties dictionary

The high-temperature chemistry library is the high-level strath library. The hTC2Model entry should be consistent with the turbulent model chosen in turbulenceProperties. Please turn off the active switch if the chemistry module does not need to be loaded. The vibrationalEnergy switch outputs the vibrational energy for each species as well as the electronic energy and the total energy. The molar-fractions are printed with the prefix X_{-} , the number densities with the prefix nD_{-} , the partial densities with the prefix pD_{-} , and finally the partial pressures with the prefix pP_{-} . The degrees of freedom are called $zeta_{-}$ followed by the corresponding energy mode.

```
// * * *
FoamFile
   version
             2.0;
   format
             dictionary;
   class
   location
             "constant";
             hTCProperties;
   object
hTC2Model laminar2<rho2ChemistryHTC>;
active
        true;
laminar2Coeffs
{}
chemistryOutputs
   molarFraction
                    off;
   numberDensity
                   off;
   partialDensity
                    off;
   partialPressure
                    off:
   degreesOfFreedom
                    off;
   vibrationalEnergy
                    off;
```

3.3.4 The *chemistryProperties* dictionary

hy2Foam uses finite rate chemistry. This translates into the use of the new class Euler2Implicit. Please turn on/off the chemistry switch to enable/disable chemical reactions. The initialChemicalTimeStep option should be available soon. There are two models for the chemistry-vibration coupling, the Park TTv model and the coupled vibration-dissociation-vibration (CVDV) model. More details about the coefficients of these models are given in this dictionary and omitted here for reasons of conciseness.

```
FoamFile
                2.0;
    version
    format
                ascii;
    class
                dictionary;
                "constant";
    location
                chemistryProperties;
    object
// * * *
chemistryType
    chemistrySolver
                       Euler2Implicit;
    chemistryThermo
                      rho2;
```

3.3.5 The hTCReactionsDPLR/QK/DK/... dictionaries

QK reactions should be defined as irreversible while Park (DPLR) reactions can be considered either reversible or irreversible. Printed below is an example of the implementation of one chemical reaction in the hTCReactionsDPLR dictionary

```
FoamFile
  version
           2.0;
  format
           ascii;
  class
           dictionary;
           hTCReactionsDPLR;
  object
species
  //N2
  Π2
  //NO
  //N2+
  //02+
  //NO+
  N
  Ω
  //N+
  //0+
  //e-
vibTempAssociativity (0 1 1);
/* REACTIONS NO 1 TO 15 AND 18 TO 32 -> DISSOCIATION
  REACTIONS NO 16 AND 17
                          -> EXCHANGE
  REACTION NO 33
                          -> ELECTRON IMPACT DISSOCIATION
  REACTIONS NO 34 AND 35
                         -> ELECTRON IMPACT IONISATION
  REACTIONS NO 36 TO 38
                         -> DISSOCIATIVE RECOMBINATION
  REACTIONS NO 39 TO 49
                          -> CHARGE EXCHANGE
reactions
  // Reaction no 1
  oxygenAtomicNitrogenReaction
           reversibleArrheniusReaction;
     reaction "02 + N = 20 + N";
     controlT dissociation;
           1.0e19;
     beta
            -1.5:
     Ta
            59500;
            (1e20 1e21 1e22 1e23 1e24 1e25); // local mixture number density in 1/\text{m}^3
     ni
     ΑO
            (1.8103 \ 0.91354 \ 0.64183 \ 0.55388 \ 0.52455 \ 0.50989);
            (1.9607 2.316 2.4253 2.46 2.4715 2.4773);
     A1
```

Please uncomment only the necessary species, here O_2 , N, and O. The vibTempAssociativity table associates one paticle such as an atom, ion or free-electrons to a molecule so that the vibro-electronic temperature of the atom/ion/electrons will be set the vibro-electronic temperature of this molecule. 0 can only be set to a molecule (the molecule will have its own vibro-electronic temperature). For other particles, an integer value of i refers to the i^{th} molecule as ordered in the species table.

Please then uncomment the appropriate chemical reactions, for instance Reaction no 1. The type of the reaction specifies whether the reaction is irreversible or reversible. The control T entry is the controlling temperature used in the Park TTv model for the calculation of the forward and backward rate constants. A, beta, and Ta are the standard Arrhenius coefficients. The equilibrium constant is a function of the local number density of the mixture. The lists of coefficients A0 to A4 are linearly interpolated if the local mixture number density is within the range $[1 \times 10^{20}, 1 \times 10^{25}]$ m^{-3} . If not, the interpolation is bounded.

3.3.6 The thermoDEM dictionary

This dictionary provides the species thermodynamic data for the different species loaded in hTCReactionsDPLR/QK/DK/ES. The dictionary structure is given here for the $\rm N_2$ molecule.

```
FoamFile
   version
              2.0;
   format
              ascii:
   class
              dictionary;
              "constant";
   location
              thermoDEM:
   object
                  N2
{
   specie
       nMoles
                      1:
                      28.0134;
       molWeight
       particleType
                     2;
       charge
                     0;
                      4.17e-10;
       diameter
       dissocEnergy
                     3.36e7:
                     0.74;
       omega
       noVibTemp
                      1;
       noElecLevels
                      15:
   thermodynamics
       Tlow
                          200;
                          40000;
       Thigh
       {\tt decoupledCvCoeffs}
                          (1.5 1 1 0 0 0 0);
       vibrationalList
                          (1 3371);
       electronicList
                               0
                             3 7.223157e4
                             6 8.577863e4
                               8.605027e4
                               9.535119e4
```

```
1
                              9.805636e4
                               9.968268e4
                               1.048976e5
                               1.116490e5
                               1.225836e5
                            6
                               1.248857e5
                            6 1.282476e5
                            10 1.338061e5
                            6 1.404296e5
                              1.504959e5
transport
{
    constant
    {
                    0;
        mu
    SutherlandEucken
    {
                    1.41e-6:
                    111.0;
    }
    BlottnerEucken
    {
                    2.68e-2;
        В
                    3.18e-1;
                   -1.13e1;
        C
    }
    CEA
             (200 1000 5000 15000):
        temp
        visco (
                (0.62526577 -31.779652 -1640.7983 1.7454992)
                (0.87395209 561.52222 -173948.09 -0.39335958)
                (0.88503551 909.02171 -731290.61 -0.53503838)
              ):
        kappa (
                (0.85439436 105.73224 -12347.848 0.47793128)
                (0.88407146 133.57293 -11429.64
                                                  0.24417019)
                (2.4176185 8047.7749 3105580.2 -14.517761)
    }
```

Inside the specie sub-dictionary, new entries have been added. The particle Type gives the type of the particle: electron (0), atom (1), molecule (2), atom (3). The charge entry indicates the charge of particle: atom and molecule (0), ion (1), electron (-1). The diameter of the particle and its temperature coefficient of viscosity, omega, are also specified. The dissociation energy of the particle, named dissocEnergy, is given in particular for use in the CVDV model. The last two entries indicate the number of vibrational modes (noVibTemp) and electronic levels (noElecLevels) of the particle. Further detail about vibrational and electronic energy modes are then given into the thermodynamics sub-directory thanks to the vibrationalList and electronicList lists. These two lists contain a sequence of scalars grouped by pair: the degeneracy and the associated characteristic temperature for a given mode or level.

The following decoupledCvCoeffs list has seven elements. Their meaning is detailed in Table 3.

```
decoupledCvCoeffs ( 1.5 1 1 0 0 0 0 );
```

Element number	Meaning
1	translational coefficient
2	rotational coefficient
3	vibrational coefficient
4	electronic coefficient
5	electron coefficient
6	chemical enthalpy at 298.15 K
7	chemical entropy at 298.15 K (unused)

Table 3: The decoupledCvCoeffs list

Hence, the scalar number 4 should be set to 0 to deactivate the electronic energy mode in a simulation. The same can be repeated for the scalar number 3 to have vibrationless nitrogen.

The transport sub-dictionary specifies some constants depending on the transport model that has been chosen in the thermophysicalProperties dictionary. mu is used with the constant transport model so that

```
constant
{
    mu    0;
}
```

allows the user to run inviscid calculations. As and Ts are the standard coefficients to determine the viscosity from the Sutherland's law while coefficients A, B, and C coefficients appear in Blottner's formula. CEA stands for the CEA2 (Chemical Equilibrium with Applications) library from NASA Glenn Research Center.

3.3.7 The thermo2TModel dictionary

```
FoamFile
    version
                2.0;
                ascii:
    format
    class
                dictionary;
    location
                "constant";
                thermo2TModel;
    object
thermalRelaxationModels
    VT
    {
        {\tt relaxationType}
                               LandauTellerVT:
        model
                               MillikanWhitePark;
        fullCoeffsForm
                               on:
        overwriteDefault
                               on;
        speciesDependent
                               on;
        collidingPair
                               on;
    VV
                               noVVEnergyTransfer;
        relaxationType
        model
                               Knab:
        overwriteDefault
                               on;
        speciesDependent
                               on:
        collidingPair
                               on;
    he
    {
```

The different models available for vibrational-translational (V-T), vibrational-vibrational (V-V), heavy-particle - electron (H-E), and electron-vibrational (e-V) energy exchanges are listed in Tables 4, 5, 6, and 7. In the VT sub-dictionary, the fullCoeffsForm entry refers to the tabulated Millikan-White coefficients (using Aij and Bij) when switched on, and to the original formulation otherwise (using preAij and preMij). If overwriteDefault is on, then the coefficients will be read from a sub-dictionary located in thermo2TModel. The coefficients can be species dependent and even colliding-pair dependent.

relaxation Type entry	Meaning
no VT Energy Transfer	disables the V-T energy exchange
Landau Teller VT	Landau-Teller equation
model entry	Meaning
Millikan White	Millikan and White relaxation time correlation
MillikanWhitePark	introduces the Park's correction
LeMANSMWP	see first paper: mixture number density
SSH	Schwartz-Slawsky-Herzfeld theory (use with caution)

Table 4: Available V-T energy exchange models in hy2Foam

relaxation Type entry	Meaning
no VVEnergy Transfer	disables the V-V energy exchange
KnabVV	Knab's equation
model entry	Meaning
Knab	Knab's equation

Table 5: Available V-V energy exchange models in hy2Foam

relaxation Type entry	Meaning
noHEEnergyTransfer	disables the H-E energy exchange
Appleton Bray	Appleton and Bray (1964)

Table 6: Available H-E energy exchange models in hy2Foam

relaxation Type entry	Meaning
noe V Energy Transfer	disables the e-V energy exchange
Lee Landau Tellere V	Lee (1984)
model entry	Meaning
Bourdon Vervisch	Bourdon and Vervisch (1997)

Table 7: Available e-V energy exchange models in hy2Foam

3.3.8 The transportProperties dictionary

This dictionary is composed of two sections. The first one concerns the quantities used to gauge the degree of rarefaction of a gas. The characteristic length of the case scenario should be entered here. There are two mean free path models (mfpModel) available: variableHardSphere and hardSphere. The rarefied parameters can be printed from this dictionary. KnGLL is the local gradient-length Knudsen number defined as the maximum of the three local gradient-length Knudsen numbers based on the density, velocity, and temperature fields. Turning the $writeKnGLL_components$ switch on and the three KnGLL numbers will be printed.

The second section of the transportProperties dictionary completes the information already given in thermophysicalProperties and thermoDEM about the different transport models.

```
FoamFile
   version
               2.0;
   format.
               ascii:
   class
               dictionary;
   location
               "constant";
   object
              transportProperties;
rarefiedParameters
   {\tt computeFieldAndBoundaries}
                                 false;
   compute MfpBoundaries
                                 true;
   mfpModel
                                 variableHardSphere;
   writeMfpSpecies
                                 off:
   writeMfpMixture
                                 off;
                                 1e-5:
   characteristicLength
   writeKn_overall
                                 off;
   writeKnGLL.
                                 off:
   writeKnGLL_components
                                 off;
transportModels
   {\tt viscosityAndThermalConductivityModels}
                                         "see thermophysicalProperties";
   mixingRule
                                molar:
   writeViscositySpecies
   writeViscosityMixture
                                 off;
   writeThermalConducSpecies
                                 off:
   writeThermalConducMixture
                                 off;
   writeHeatFluxes
                                 off;
   writeWallHeatFlux
                                 on;
   multiSpeciesTransport
                                 LewisNumber;
   binaryDiffusivityModel
                                 noBinaryDiffusivityModel;
```

To this date, there are three mixing rule implemented in hy2Foam: a simple and generally unsuitable molar average using the keyword molar, Wilke, and ArmalySutton mixing rules. The viscosity and thermal conductivity can be printed, either for each species or for the mixture. The transportModels sub-dictionary also prescribes the multi-species transport model used in hy2Foam. The different possibilities are listed in Table 8.

multiSpeciesTransport entry	Meaning
noSpeciesDiffusion	disables species diffusion
Lewis Number	uses the Lewis number
modified Lewis Number	see Gollan's PhD thesis
Fick	generalisation of Fick's binary formula for a mixture
binary Diffusivity Model entry	Meaning
no Binary Diffusivity Model	no binary diffusivity model
constant	constant binary diffusion coefficients (used with Fick only)
Hirschfelder	Hirschfelder model (used with Fick only)
Stephani	see K. A. Stephani's PhD thesis (used with Fick only)
Gupta	available soon, see Gupta 1989 (used with Fick only)

Table 8: Available multi-species transport models in hy2Foam

3.4 The *system* folder

A few modifications have been made to the system folder. They are located in fvSchemes and fvSolution. Local time stepping can be switched on by simply selecting the localEuler rDeltaT temporal scheme in place of the traditional Euler scheme. Three divergence scheme entries have been added.

```
referenceCase

VincePC: referenceCase > ls system/
controlDict decomposeParDict fvSchemes fvSolution
VincePC: referenceCase >
```

Figure 4: The **system** folder

```
FoamFile
           2.0;
  version
  format
           ascii:
  class
           dictionary;
  location
           "system";
           fvSchemes:
  object
fluxScheme
               Kurganov;
ddtSchemes
               Euler; // Euler or localEuler rDeltaT
  default.
```

4 New boundary conditions

4.1 Velocity field

$4.1.1 \quad rampInlet$

The time-dependent rampInlet boundary condition is aiming at reducing the very steep initial velocity gradients and thus falicitating the convergence of hypervelocity simulations. Its implementation is given hereafter

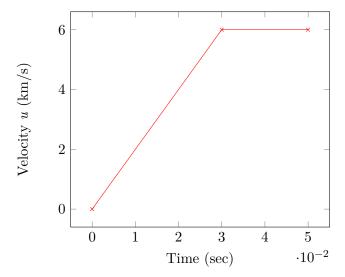
Using the rampInlet boundary type, the inlet velocity vector is defined as follow

$$U(t) = (u(t), v(t), w(t))^{T}$$

$$= \left(amplitude \times refValue - offset \odot refValue\right)$$

$$\times min(t/tRamp, 1) + offset \odot refValue$$
(1)

where \odot is the Hadamard product.



$4.1.2 \quad non Eq Maxwell Slip U$

The original maxwellSlipU velocity boundary condition implemented in rhoCentralFoam is modified and takes the following form in hy2Foam

```
wall
{
                            nonEqMaxwellSlipU;
        type
                             (0 0 0);
        Uwall
        refValue
                             uniform (0 0 0);
        accommodationCoeff 1.0;
        valueFraction
                            uniform 1.0;
        thermalCreep
                             off;
        curvature
                             off;
                             uniform $Uwall;
        value
}
```

4.2 Temperature field

$4.2.1 \quad non Eq Smoluchowski Jump T$

The original smoluchowskiJumpT temperature boundary condition implemented in rhoCentralFoam is modified and takes the following forms in hy2Foam

```
wall
                nonEqSmoluchowskiJumpT;
    type
                uniform 800.0;
    Twall
    accommodationCoeff 1.0;
    value
                $Twall;
}
or
wall
{
    type
                nonEqSmoluchowskiJumpTv; or nonEqSmoluchowskiJumpTvMix;
                uniform 800.0;
    Twall
    accommodationCoeff 1.0;
    value
                $Twall;
}
```

These formulations are using the overall free path boundary field.

