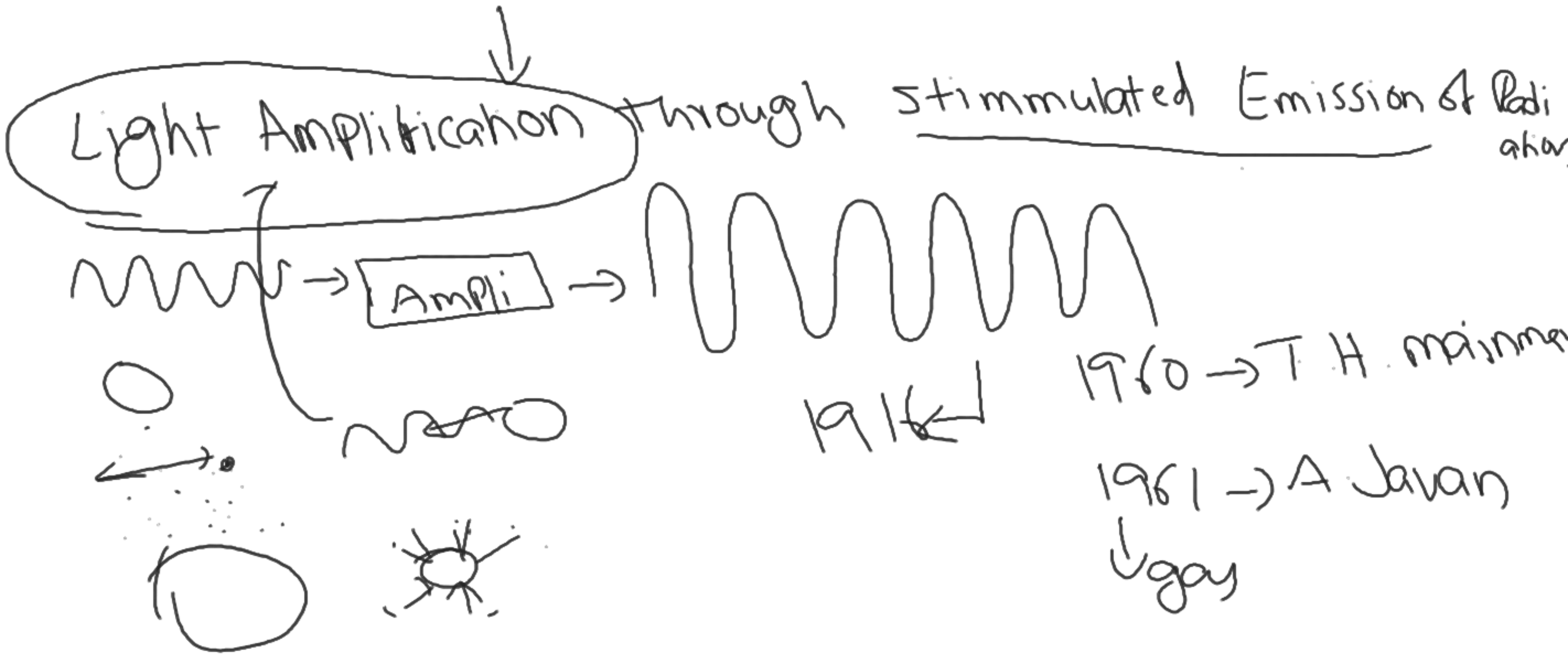
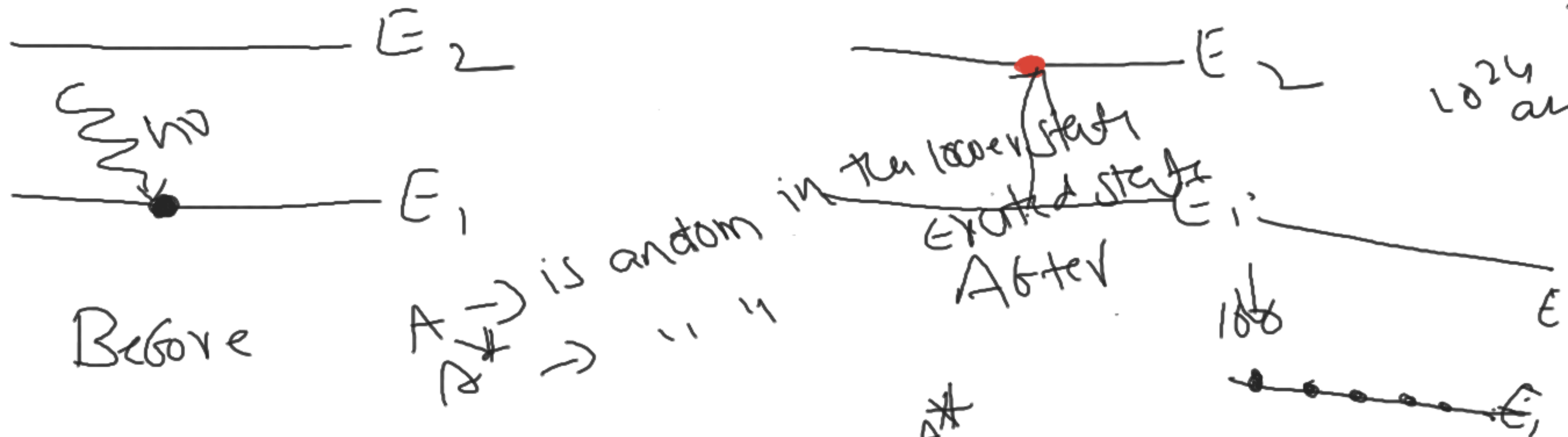


# LASER





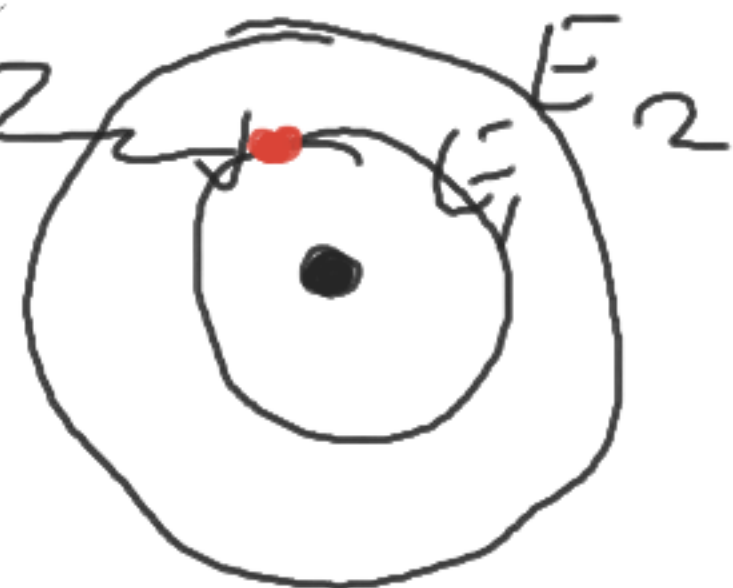
Beğove



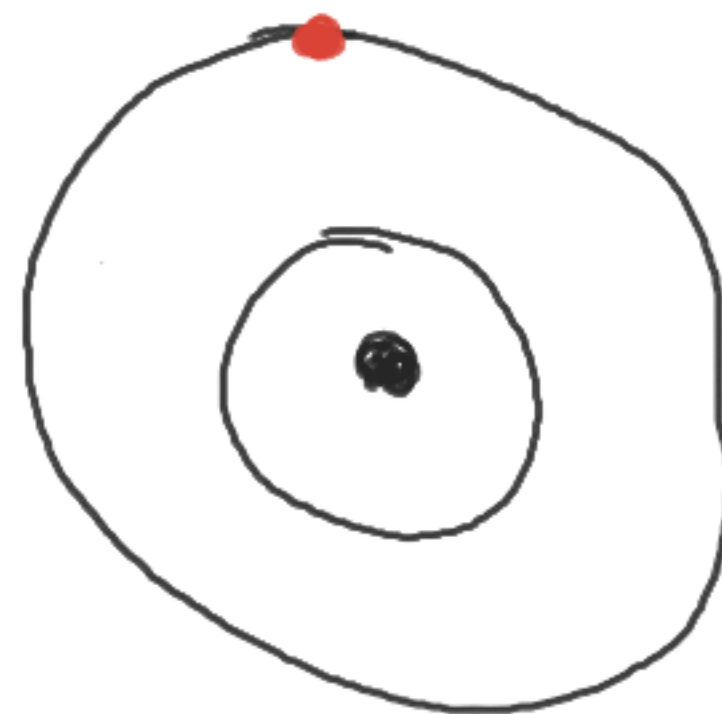
$E_1$   
 $A \rightarrow$  is random in the lower state  
 $\rightarrow$  " " excited state  
 $\rightarrow$  " " After

$$h\nu = E_2 - E_1 \quad A + h\nu \rightarrow A^*$$

## Absorption :-



Before



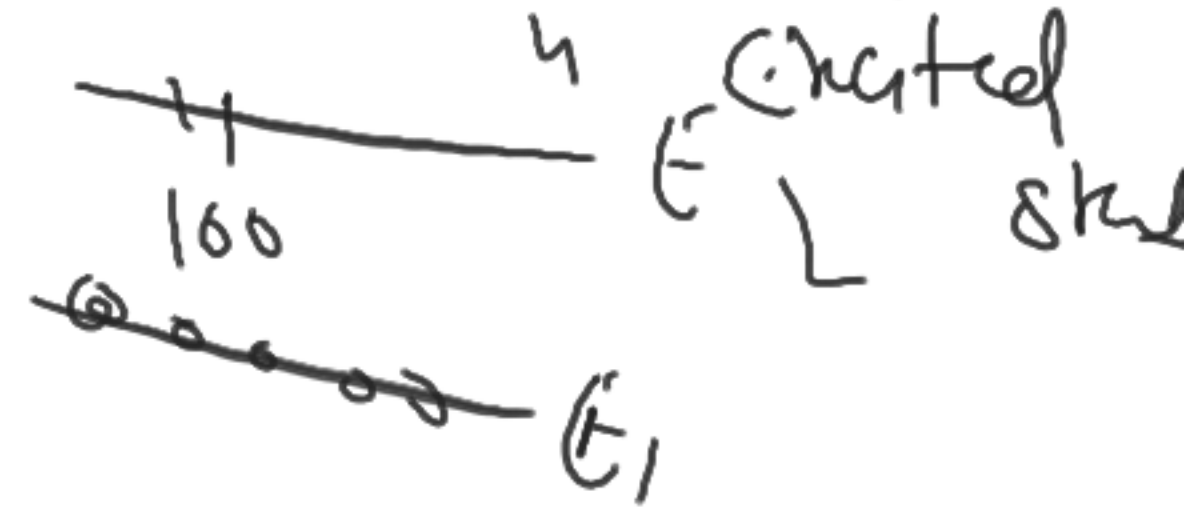
## Agter

The Probability that an absorption transition occurs is proportional to the photon density  $P(\nu)$

$$P_{12} \propto P(\nu)$$

$N_1$  = no. of atoms at ground state  
 $N_2$  = no. of atoms in excited state

$$P_{12} = B_{12} P(\nu)$$

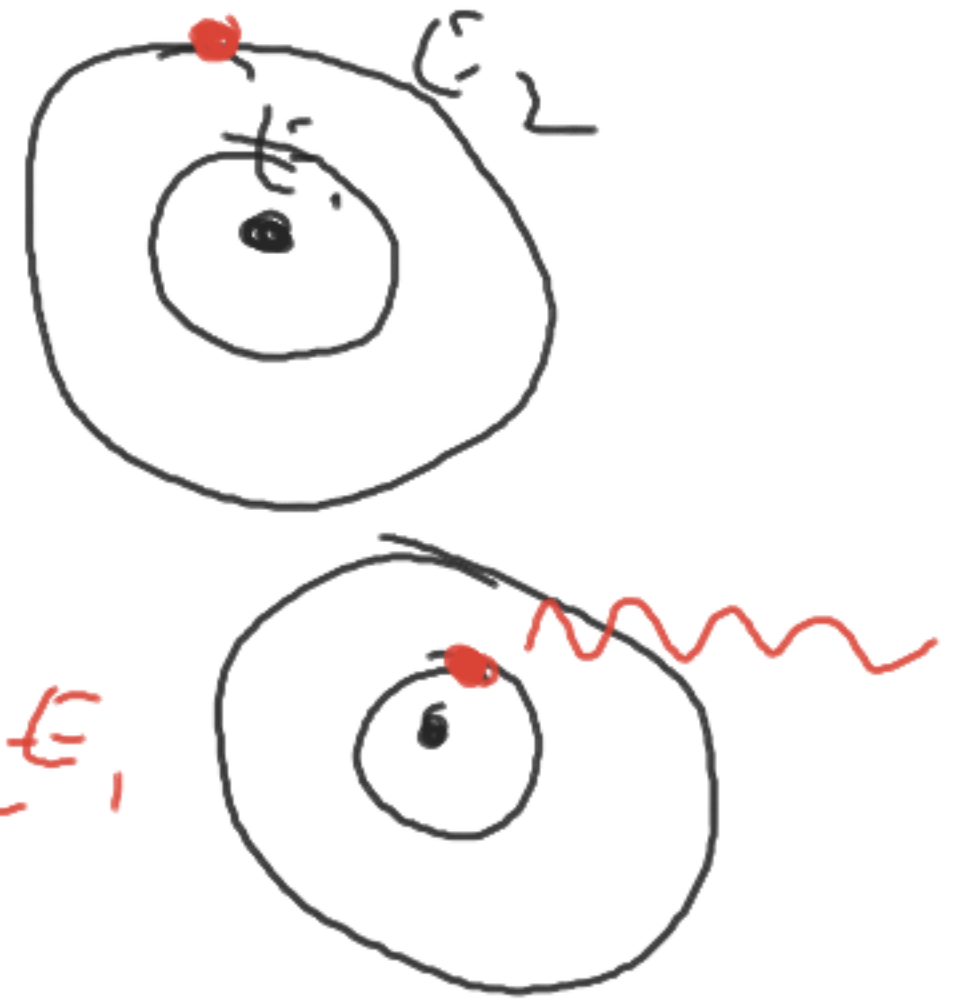
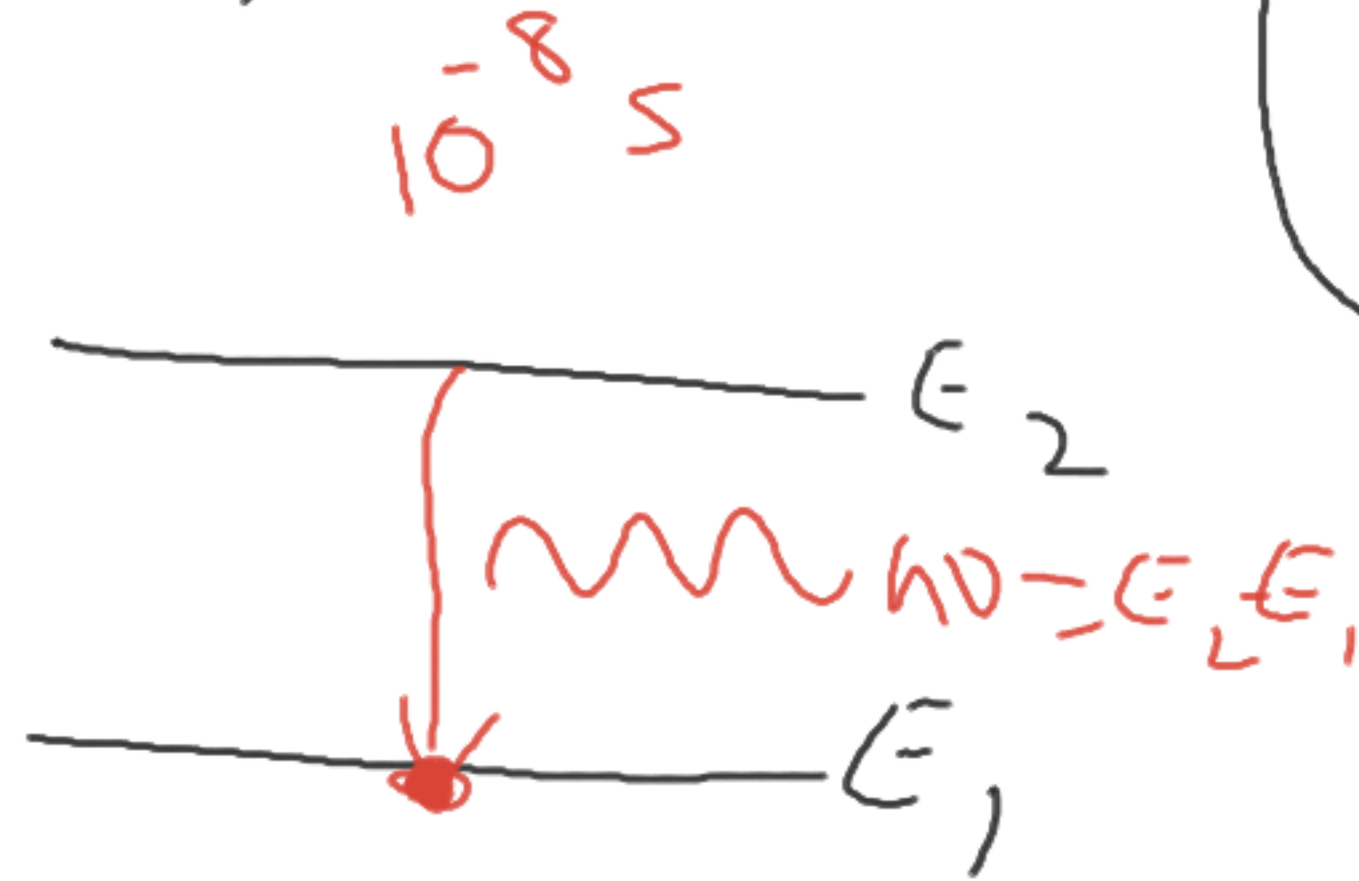
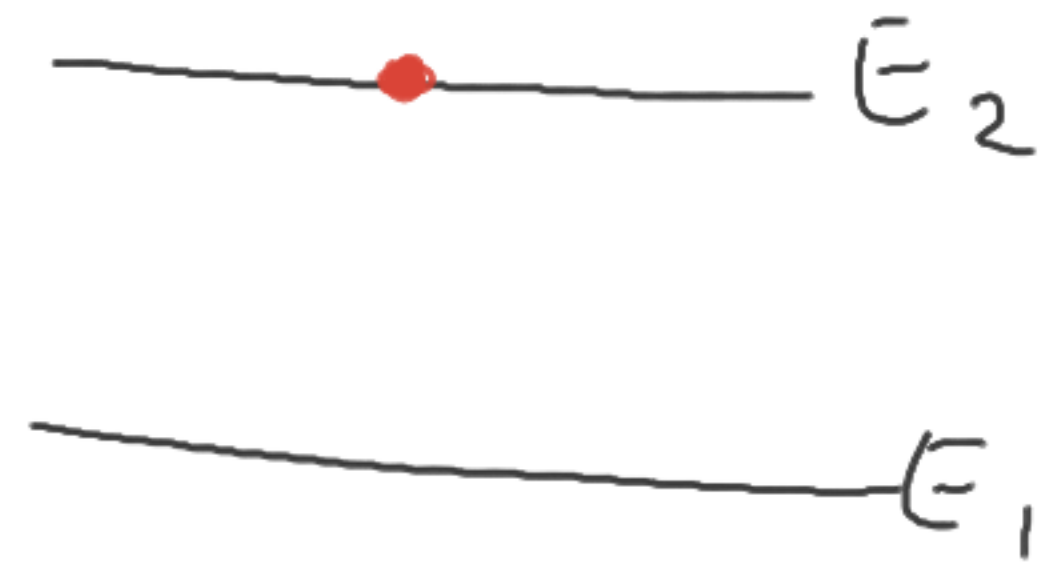


$B_{12}$  is constant of proportionality.  $B_{12}$  is known as the Einstein coefficient for absorption

The total no. of atoms  $N_{ab}$  excited during  $\Delta t$  time

$$N_{ab} = B_{12} N_1 P(\nu) \Delta t \quad \text{--- (1)}$$

## 2. Spontaneous Emission



$P_{21} = A_{21} \rightarrow$  Einstein coefficient for spontaneous emission

$\frac{1}{A_{21}}$  is a measure of the lifetime of the upper state against spontaneous transition to the lower state

$$N_{sp} = A_{21} N_2 \Delta t \quad (2)$$

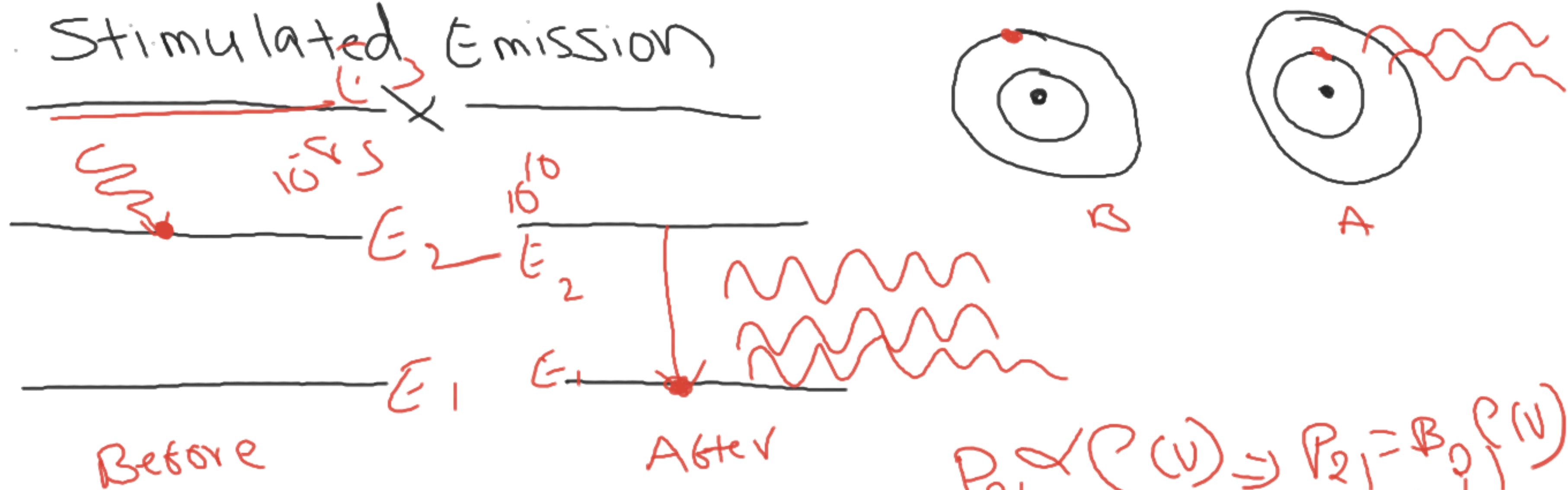


$10^{24}$

1. The instant of transition
2. Direction of the emission of the photon
3. Phase of the photon
4. Polarization state of the photon



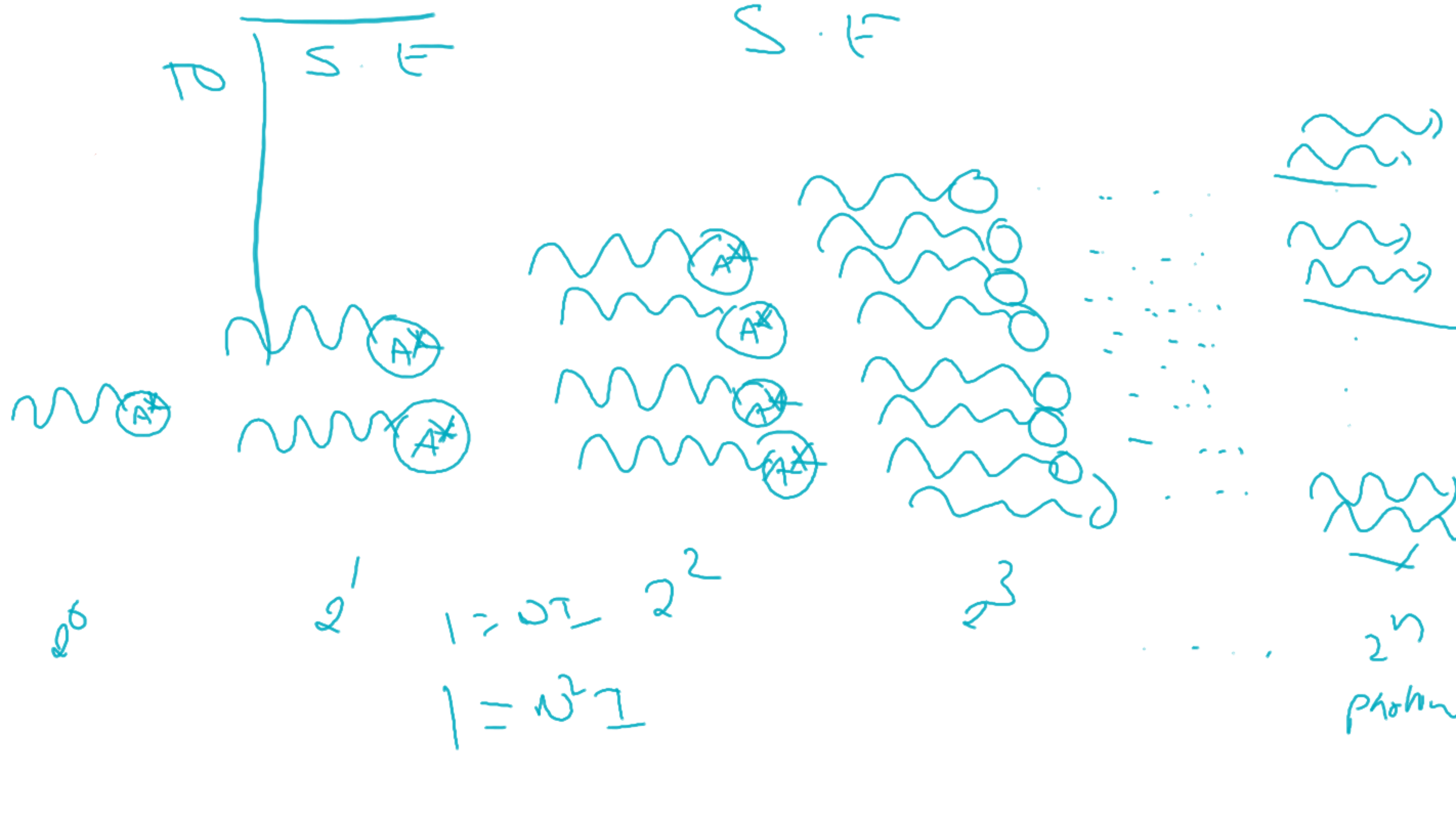
### 3. Stimulated Emission



$P_{21} \propto P(\nu) \Rightarrow P_{21} = B_{21} P(\nu)$   
 $B_{21} \rightarrow$  Einstein coefficient for stimulated emission

$$N_{St} = B_{21} N_2 P(\nu) \Delta t \quad (3)$$

multiplication of stimulated photon



20

$2^1$

$1 = \sigma_1 - 2^2$

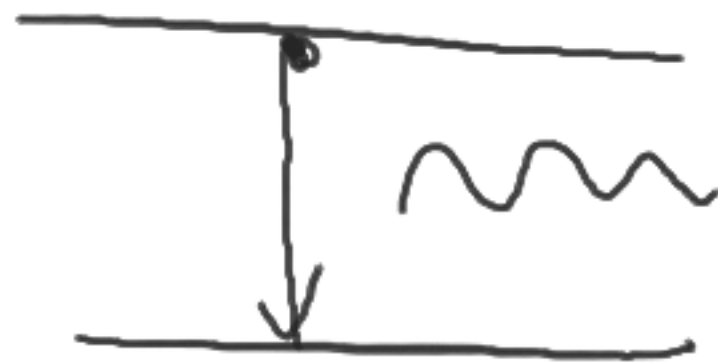
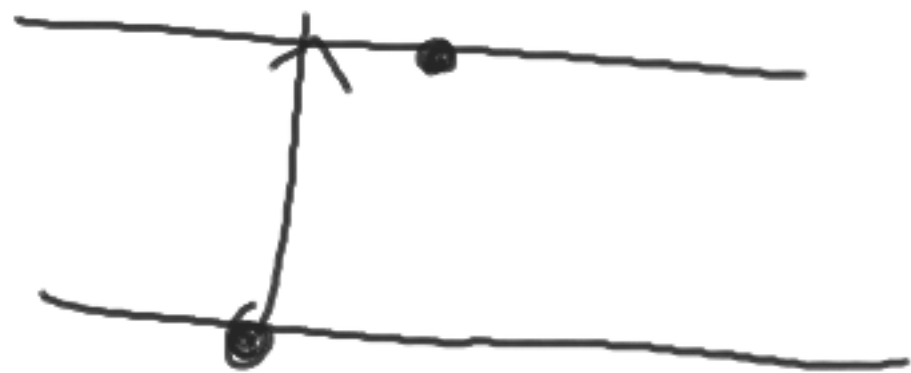
$1 = v^2 \tau$

$2^3$

$2^n$

photon

# Steady State Condition



30



0

10<sup>24</sup>

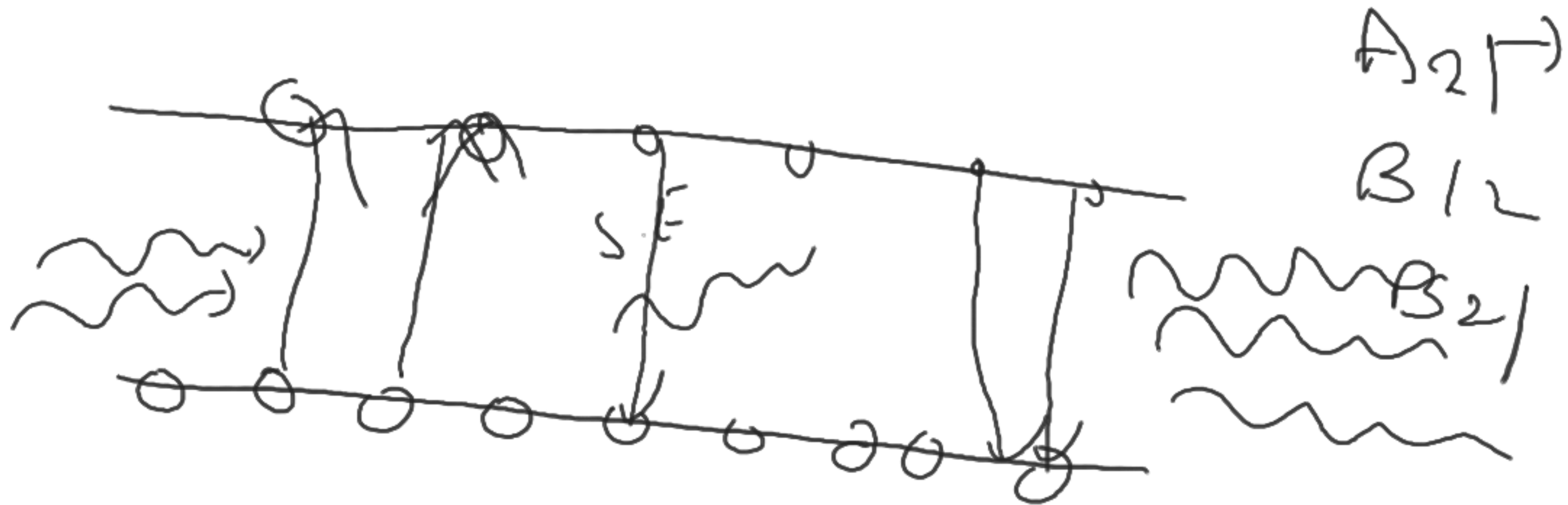
20



$$N_{ab} = N_{sp} + N_{st}$$

$$B_{12} N_1 \rho(\nu) \Delta t = A_{21} N_2 \Delta t + B_{21} N_2 \rho(\nu) \Delta t$$

$$B_{12} N_1 \rho(\nu) = A_{21} N_2 + B_{21} N_2 \rho(\nu)$$



1. The atom and the radiation are in Thermal Equilibrium
2. The radiation is identical with B.B radiation & consistent with Planck's radiation law for any value of  $T$
3. The population density  $N_1$  &  $N_2$  at the lower and upper  $E.L$  respectively constant in time and are distributed according to the Boltzmann law in energy level

$$\begin{array}{c}
 N_2 \text{ ————— } E_2 \\
 N_1 \text{ ————— } E_1
 \end{array}
 \quad
 \frac{N_2}{N_1} = e^{\frac{-(E_2 - E_1)/kT}{-h\nu/kT}}$$

$$\left( \frac{N_1}{N_2} = e^{h\nu/kT} \right)$$

$$N_{ab} = N_{sp} + N_{st}$$

$$B_{12}P(V)N_1 = A_{21}N_2 + B_{21}N_2P(V)$$

$$P(V)[B_{12}N_1 - B_{21}N_2] = A_{21}N_2$$

$$P(V) = \frac{A_{21}N_2}{[B_{12}N_1 - B_{21}N_2]}$$

dividing  $B_{12}N_1$

$$P(V) = \frac{A_{21}/B_{12}}{\frac{N_1}{N_2} - \frac{B_{21}}{B_{12}}}$$

$$\frac{N_1}{N_2} = e^{h\nu/KT}$$

$$\rho(\nu) = \frac{A_{21}/B_{12}}{\left[ e^{h\nu/KT} - \frac{B_{21}}{B_{12}} \right]}$$

$$\rho(\nu) = \frac{A_{21}}{B_{12}} \left[ \frac{1}{e^{h\nu/KT} - \frac{B_{21}}{B_{12}}} \right]$$

~~$$\frac{A_{21}}{B_{12}} =$$~~

$$C(\nu) = \frac{8\pi h\nu^3}{c^3} \cdot \frac{1}{e^{h\nu/KT} - 1}$$

~~$$C(\nu) = \frac{8\pi h\nu^3}{c^3} \left[ \frac{1}{e^{h\nu/KT}} \right]$$~~