



**Department of Humanities and Sciences**  
**National Institute of Technology Goa**  
**Farmagudi, Ponda, Goa - 403 401**

**Subject: Material Science**  
**Course Code: PH150**

**Minor-I**

**Time: 60 minutes**  
**Max Marks: 10**

**Answerer all the questions**

1. Calculate the interplanar spacing for a (321) plane in a simple cubic lattice whose lattice constant is  $4.2 \times 10^{-10}$  m. 1 M
2. A NaCl crystal is used as a diffraction grating with X-rays. For the  $d_{121}$  spacing of the chloride ions, the angle of diffraction  $2\theta$  is 60. If the lattice constant of the crystal is 0.73 nm, what is the wavelength of X-rays? 1 M
3. In a crystal whose primitives are 1.1 Å, 1.2 Å and 1.8 Å. A plane (111) cuts an intercept of 1.4 Å along the X-axis. Find the lengths of intercepts along the Y and Z axes 1 M
4. Draw the following planes in a cubic unit cell (0 1 1), (1 0 1) and (1 1 1). 1 M
5. Copper has sc structure of atomic radius 0.1278 nm. Calculate the interplanar spacing for (1 2 1) plane. 1 M
6. Determine the Coulomb interaction energy for a NaCl. Given that the distance between oppositely charged ions is 2.6 Å. 1 M
7. What are point defects? Explain, in detail, the different types of point defects with suitable sketches. 1.5 M
8. Define the terms coordination number, atomic radius, and packing density. Calculate the above factors for a simple cubic, body centred cubic and face centred cubic crystals 1.5 M