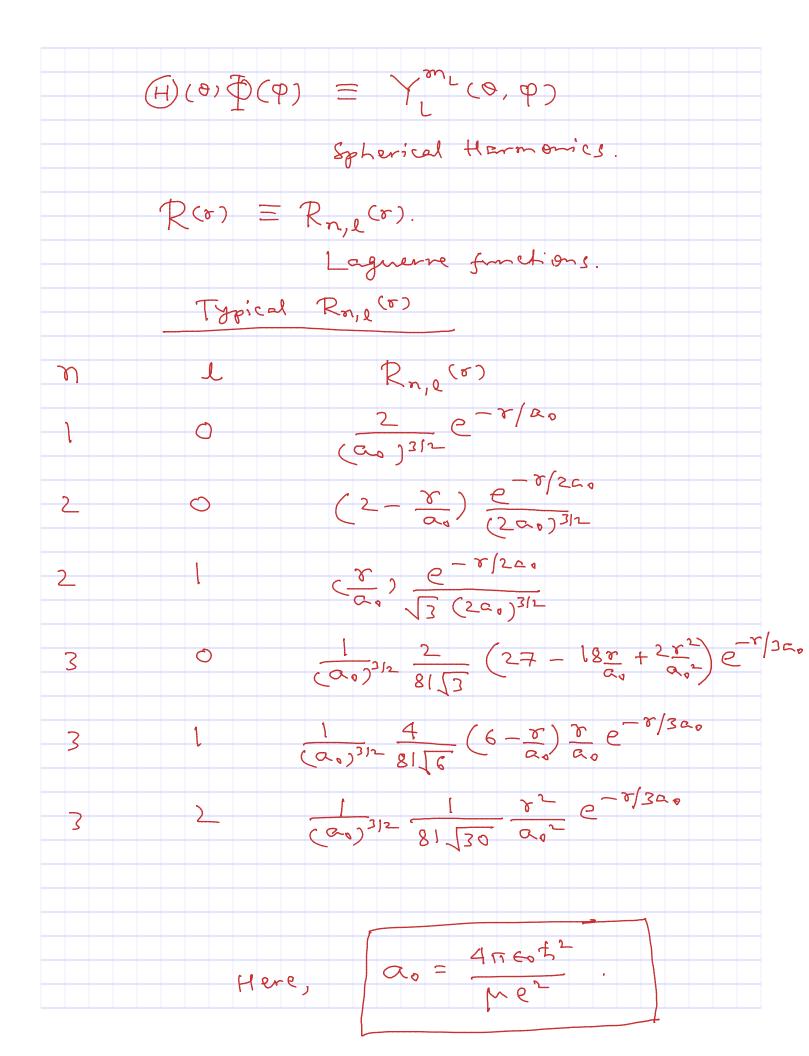
Hydrogen atom Schrödingen time independent equalion $-\frac{t^2}{2\mu} \nabla^2 \psi + V \psi = E \psi.$ $\Rightarrow \nabla^2 \psi + \frac{2h}{t^2} (E - V) \psi = 0.$ For Hydrogen dom, $V(\sigma) = -\frac{Z}{x}$. (in SI with, Z = [e] -) Useful to write Tim spherical-polar condinates. & Y = Y(r, e, 9). · 1 2 2 2 4 1 2 (5, 0 34) + 2r (E-v) 4=0. Use \((r,0,9) = R(r) (0) \(\beta(0)\), substitute in above equation, multiply by Thin's and rearrange $=) - \frac{3i^20}{R} \frac{3}{3r} \left(r^2 \frac{3}{3r} R\right) - \frac{2}{5} r^2 \frac{3i^20}{5} \left(E - V\right)$ - 200 30 (8:00 30) = 1 3-0.3. L.H.s. is a function of the o R.H.S. is a function of of alone Lat both sides equal —on 2 (a negative desinite questity)

 $\frac{d^2\Phi}{d\rho^2} = -m^2\Phi$ Rearranging (wing I), =) [] (r 2R) + 2mr2(E-v) $= \frac{8!^{2}}{3!^{2}} - \frac{1}{3!} \frac{30}{3!} (8!^{2} + \frac{30}{3!})$ L.H.s. is a function of or, while, R.H.S. is a function of O. Thus, both sides equal a constant. Let that constant be l(l+1) (inspired by the existing standard differential egue o for Laguerre & Legendre polynomiels) $\frac{1}{r^2} \frac{\partial}{\partial r} r^2 \frac{\partial R}{\partial r} + \frac{2r}{L^2} \left[E - V - \frac{l(l+1)}{2r} \right] R = 0.$ $2\left(\frac{1}{5^{2}} + \frac{1}{20}\left(\frac{1}{5^{2}} + \frac{1}{20}\right) + \left(\frac{1}{20}\left(\frac{1}{5}\right) + \frac{1}{20}\right) + \frac{1}{20} = 0.$



Y"(0, 9) Typical Ym, (0, p) 0 1 3 Caro 7-1/3 sinde tiq 1 5 (3 cas 20 - 1) 7 1 15 8:00 coso e + ip 1 15 Sin 20 e ± 2; \$ +2 $\Phi(\varphi + 2\pi) = \Phi(\varphi).$ Note:

Typical case
$$1 = 0, m_0 = 0$$
Radial equation:
$$\frac{d^2R}{dr^2} + \frac{2}{r} \frac{dR}{dr} + \frac{Lr}{dr} \left(E + \frac{e^2}{4\pi \epsilon_0 r}\right)^{R=0}$$

$$\frac{1}{r} \frac{1}{r} \frac$$