Kursplan för Kvantfysik (1FA521) för F3 ht 2014, period 1&2

Föreläsningar:

Peter Oppeneer (Å13237, 471 3748, peter.oppeneer@physics.uu.se)

Lektioner:

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Laborationer:

Filip Heijkensköld (Å11264, 471 3529, filip.heijkenskold@physics.uu.se) Anna Shepidchenko (Å13240, 471 5876, anna.shepidchenko@physics.uu.se) Jonathan Chico (Å13209, 471 3580, Jonathan.chico@physics.uu.se)

Hemsida: http://studentportalen.uu.se/

Web-registrering, kursmaterial att hämta hem, tips och länkar.

Schema: http://se.timeedit.net/web/uu/db1/schema/

Kursböcker:

P.A. Tipler and R.A. Llewellyn, *Modern Physics*, 5th ed., Freeman and Company, New York, 2008 (ISBN 0-7167-7550-6)

D.J. Griffiths, *Introduction to Quantum Mechanics*, 2nd ed., Pearson Int., NJ (ISBN 0-13-191175-9)

S. Gasiorowicz, *Quantum Physics*, 3rd ed., Wiley, New York (ISBN 0-471-42945-7)

C. Nordling och J. Österman, *Physics Handbook*, Studentlitteratur, 1999 eller senare.

Laborationer:

Optical spectroscopy of hydrogen

Photoelectric effect

X-ray spectroscopy >> with small oral presentation in English (=> 1 bonus point)

Tentamen: October 13 and December 12, 2014. Required is 40% to pass (13 out of 30 points max.)

Kvantfysik, 10hp, föreläsnings innehåll:

- 1. Introduction, background of Quantum Physics, principle of quantization, black body radiation
- 2. Photoelectric effect, Compton scattering (duality of waves-particles), Rutherford and Bohr atomic models
- 3. De Broglie wave length, experimental confirmation, uncertainty relations (simple form)
- 4. Schrödinger equation, time dep., and time independent; wave function, probability interpretation
- 5. Operators, expectation values
- 6. Infinite square well
- 7. Scattering from potential barriers, quantum tunelling
- 8. The harmonic oscillator (full solution)
- 9. Extensions of the Schrödinger equation to more dimensions
- 10. Formalism of quantum mechanics, bras and kets, Heisenberg uncertainty
- 11. The hydrogen atom I
- 12. The hydrogen atom II
- 13. Angular momentum
- 14. Spin, addition of angular momenta
- 15. Many electrons, Pauli principle, shell structure of atoms, the Periodic Table
- 16. Formalism, harmonic oscillator with ladder operators
- 17. Nondegenerate perturbation theory
- 18. Fine structure; identical particles, fermions and bosons
- 19. Di-atomic molecules and bonding