Monday Reading Assessment: Unit 0, Coulomb's Law and E-fields

Prof. Jordan C. Hanson

February 3, 2020

1 Memory Bank

- $\vec{F}_{12} = kq_1q_2/r^2 \ \hat{r}$... Coulomb force.
- $k = 8.988 \times 10^9 \text{ N C}^{-2} \text{ m}^2$... The constant of proportionality in Coulomb force.
- $\vec{F} = q\vec{E}$... Definition of an electric field.
- $1e = -1.602 \times 10^{-19}$ Coulombs ... Charge of an electron.
- $1p = +1.602 \times 10^{-19}$ Coulombs ... Charge of a proton.

2 Coulomb Force

1. Suppose you have a charge of +1 nC, or 10^{-9} Coulombs, separated from another identical charge. (a) What will be the force between them? (b) Will the charges accelerate toward each other or away from each other?

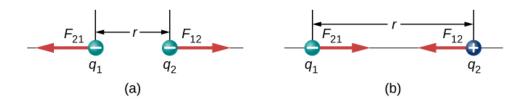


Figure 1: With the Coulomb force, like charges repel each other, and opposite charges attract.

- 2. How many protons are required to create a total charge of +1 nC, or 10^{-9} Coulombs?
- 3. If two charges are a distance r apart, and then r is tripled, by what factor does the force decrease?
 - A: 3
 - B: 6
 - C: 9
 - D: 12