

CLASSROOM CONTACT PROGRAMME

(Academic Session: 2024 - 2025)

LEADER & ACHIEVER COURSE

PHASE: MLA, MAZA, MAZB, MAZC, MAZD, MAZL, MAZN, MAZO, MAAX, MAAY, MAPA, MAPB, LAKSHYA

TARGET: PRE MEDICAL 2025

Test Type: MAJOR Test Pattern: NEET (UG)

TEST DATE: 06-03-2025

ANSWER KEY																														
Q.	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30
A.	1	1	3	3	1	2	1	2	4	4	3	3	4	2	2	1	3	3	1	2	2	3	4	2	2	1	1	1	1	2
Q.	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60
A.	1	3	1	3	2	1	3	2	3	3	4	3	2	2	1	3	4	2	4	1	3	2	3	2	3	4	1	1	2	1
Q.	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90
A.	4	2	3	1	4	3	4	3	4	2	1	2	4	3	2	2	3	1	3	2	1	3	4	3	4	4	2	2	3	2
Q.	91	92	93	94	95	96	97	98	99	100	101	102	103	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118	119	120
A.	3	3	1	4	3	2	3	4	2	3	1	1	2	2	3	1	4	4	4	2	2	3	4	4	3	1	4	1	2	2
Q.	121	122	123	124	125	126	127	128	129	130	131	132	133	134	135	136	137	138	139	140	141	142	143	144	145	146	147	148	149	150
A.	3	1	2	3	1	3	4	3	2	2	1	2	3	3	2	3	2	4	3	3	3	4	3	2	1	3	4	1	4	2
Q.	151	152	153	154	155	156	157	158	159	160	161	162	163	164	165	166	167	168	169	170	171	172	173	174	175	176	177	178	179	180
Α.	2	3	1	3	4	3	2	1	1	2	3	1	3	3	2	2	3	3	1	4	3	3	3	3	2	1	1	3	1	3

HINT - SHEET

- 1. Ans (1) NCERT-XI, Pg # 150
- 2. Ans (1) NCERT, Pg. # 140, 141
- 3. Ans (3) NCERT XI, Page No. 134, 135
- 4. Ans (3)
 NCERT XI Page No. 136, 140, 147, 150
- 5. Ans (1) NCERT-XI, Pg. No. - 145
- 6. Ans (2) XI NCERT Pg. # 143
- 7. Ans (1) NCERT, Pg. # 135
- 8. Ans (2) NCERT, Pg. # 146

- 9. Ans (4) NCERT, Pg. # 147
- 10. Ans (4) NCERT, Pg # 142
- 11. Ans (3) NCERT, Pg # 142
- 12. Ans (3) NCERT, Pg. # 143
- 13. Ans (4) NCERT Pg. # 142, 143
- **14. Ans (2)** NCERT, Pg. # 140
- 15. Ans (2) NCERT Pg. # 139
- **16. Ans (1)** NCERT, Pg. # 150

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- **17. Ans (3)** NCERT, Pg. # 145, 146
- **18. Ans (3)** NCERT, Pg # 137, 146
- **19. Ans (1)** NCERT Pg. # 157, 160
- **20. Ans (2)** NCERT, Pg. # 159, 160
- **21. Ans (2)** NCERT Pg. # 155, 158
- 22. Ans (3) NCERT XI Page No. 157, 162, 163
- 23. Ans (4) NCERT XI Pg # 157
- **24. Ans (2)** NCERT, Pg. # 154, 157
- **25. Ans (2)** NCERT XI, Pg. # 157
- **26. Ans (1)** NCERT-XI, Pg. # 158
- **27. Ans (1)** NCERT XI Page No. 161
- 28. Ans (1) NCERT Pg. # 162
- **29. Ans (1)** NCERT-XII, Pg. # 156
- **31. Ans (1)** NCERT XI Pg # 159
- 32. Ans (3) NCERT Pg. # 164
- **33. Ans (1)** NCERT Pg. # 162
- **34. Ans (3)** NCERT Pg. # 158
- 35. Ans (2) NCERT Pg. # 161

- **36. Ans (1)** XI NCERT Pg # 175, 176, 177, 178
- **37. Ans (3)** NCERT-XI, Pg # 177
- **38. Ans (2)** NCERT XI Pg. # 166, 171
- **39. Ans (3)** NCERT-XI, Pg # 177
- **40. Ans (3)** NCERT-XI, Pg. # 174, 175
- **41. Ans (4)** NCERT, Pg. # 175
- **42. Ans (3)** NCERT Pg. # 167
- **43. Ans (2)** NCERT Pg. # 168, 169
- **44. Ans (2)** NCERT Pg. # 170
- **45. Ans (1)** NCERT Pg. # 172
- **46. Ans (3)** NCERT XI Pg. # 234
- **47. Ans (4)** NCERT 316
- **48. Ans (2)** NCERT Page No. 319
- **49. Ans (4)** NCERT XI Pg. No. 232
- **50. Ans (1)** NCERT XI Pg. No. # 320, 321
- **51. Ans (3)** NCERT-XI, Pg. # 321, Para-3
- **52. Ans (2)** NCERT-XI, Pg. # 107
- 53. Ans (3) NCERT XI Pg#321

54. Ans (2)

NCERT XI Pg. # 247

55. Ans (3)

NCERT -XI, Pg. # 245

56. Ans (4)

NCERT-XI, Pg. # 242

57. Ans (1)

NCERT Pg. No. # 240

58. Ans (1)

NCERT XI, Pg. No. 339, Para - 22.4

59. Ans (2)

NCERT XI Page No. # 336

60. Ans (1)

NCERT-XI, Pg. # 242

61. Ans (4)

NCERT Pg. # 333

62. Ans (2)

NCERT Pg # 332

63. Ans (3)

NCERT, Pg # 334,336,337

64. Ans (1)

MSH help in dispersion of melanin while ACTH stimulate adrenal cortex for the secretion of mineralo-cocorticoides and glucocorticoides.

65. Ans (4)

NCERT Page#336

66. Ans (3)

NCERT Page#338.

67. Ans (4)

NCERT Page#335.

68. Ans (3)

NCERT-XI Pg#227

69. Ans (4)

NCERT (XI) Pg. # 311, 312

70. Ans (2)

NCERT-XI, Pg No. 306

71. Ans (1)

NCERT XI (E)Pg.# 219

72. Ans (2)

NCERT Page No 227

73. Ans (4)

NCERT (XIth) Pg. # 227

74. Ans (3)

NCERT XI Pg # 224

75. Ans (2)

NCERT XI - Page No. 221

76. Ans (2)

NCERT XI Page No. 220

77. Ans (3)

NCERT Pg. # 236

Broca's area — motor speech area

Wernicke's area – Language comprehension area Somaesthetic area – Somatic sensation like touch, pain, pressure, temperature

78. Ans (1)

NCERT XI Pg. No. # 233

79. Ans (3)

NCERT-XI, Pg. # 236

80. Ans (2)

NCERT XI Pg. No. 232

81. Ans (1)

NCERT XI Page No. # 247-248

82. Ans (3)

NCERT Pg # 244, 245, 246

83. Ans (4)

NCERT Pg. No. # 242-243

84. Ans (3)

NCERT Page # 244

85. Ans (4)

NCERT XI Page # 334 (II Para)

86. Ans (4)

NCERT Page-335/337/342(E)

87. Ans (2)

NCERT XI Pg # 311

88. Ans (2)

NCERT (XI) Pg. # 222

89. Ans (3)

Module, Pg. # 163

90. Ans (2)

NCERT (XI) Pg. # 221

91. Ans (3)

According to FLOT:-

$$Q = W + \Delta U$$

In Cyclic process :- $\Delta U = 0$

- \therefore Q = W = -2PV
- ∴ Heat Rejected = 2 PV
- 92. Ans (3)

$$(C_{V})_{mix} = \frac{n_{1}(C_{V})_{1} + n_{2}(C_{V})_{2}}{n_{1} + n_{2}}$$

$$= \frac{2\left(\frac{5R}{2}\right) + 8\left(\frac{3R}{2}\right)}{2 + 8}$$

$$= \frac{17R}{10} = 1.7 R$$

93. Ans (1)

By second law of thermodynamics (SLOT)

94. Ans (4)

$$\frac{R}{C_P + C_V} = \frac{1}{6} \Rightarrow C_P + C_V = 6R$$

$$C_{\rm P} - C_{\rm V} = R$$

$$\therefore C_P = \frac{7}{2}R, C_V = \frac{5}{2}R$$

$$\gamma = \frac{C_P}{C_V} = 1 + \frac{2}{R}$$

$$\Rightarrow$$
 f = 5

95. Ans (3)

$$\frac{\Delta L}{L} = \alpha \Delta T$$
$$= 12 \times 10^{-6} \times 50$$

$$= 6 \times 10^{-4}$$

Strain will be negative, as rod is in compressed

state.

96. Ans (2)

$$\therefore PV = nRT$$

$$P = \left(\frac{nR}{V}\right)T \Rightarrow \left(\frac{mR}{MV}\right)T$$

- $\therefore slope = \frac{mR}{MV}$
- : m is doubled and volume is halved.
- ∴ slope becomes 4 times.
- 97. Ans (3)

$$P = \frac{\rho RT}{M_w}$$

Slope =
$$\frac{\rho R}{M_w}$$

98. Ans (4)

> As per Kirchoff's law a body can emit only those radiations at high temperature which it has absorbed at low temperature.

99. Ans (2)

> A cooking pot should have low specific heat so that it requires less heat in heating upto a particular temperature & consume less fuel or gas.

Its thermal conductivity should be high so that it cooks food in less time.

100. Ans (3)

Q_R = Heat Required to melt all ice

$$= ms_{ice} \Delta T + mL_f$$

$$=200 \times 0.5 \times 5 + 200 \times 80$$

= 16500 calorie

 Q_S = Heat supplied by hot water = $ms_w \Delta T = 500 \times 1 \times 25$

Since, $Q_S < Q_R$ all ice will not melt. Temperature of mixture

101. Ans (1)

$$\lambda m_1 T_1 = \lambda m_2 T_2$$

$$4 \times 900 = \lambda m_2 \times 1200$$

$$\lambda m_2 = \frac{4 \times 900}{1200} = 3 \mu m$$

102. Ans (1)

Black colour absorbs maximum at the time of heating so emits maximum at the time of cooling.

103. Ans (2)

A/C to Prevost - theory.

104. Ans (2)

$$\frac{80-64}{5} = K(72-T_0)$$

$$16 = 5K(72 - T_0) \dots (1)$$

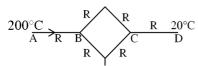
$$\frac{64-52}{50}$$
 = K (58 – T₀)

$$12 = 5K(58 - T_0)...(2)$$

Eq.
$$(1) \div (2)$$

$$\frac{4}{3} = \frac{72 - T_0}{58 - T_0} \Rightarrow T_0 = 16^{\circ} \text{C}$$

105. Ans (3)



$$\cdot \cdot \cdot R_{aa} = 3R$$

$$i = \frac{\Delta T}{R_{eq.}} = \frac{180}{3R} = \frac{60}{R}$$

For AB :-
$$i = \frac{200 - T_B}{P}$$

For AB :- i =
$$\frac{200 - T_B}{R}$$

 $\frac{60}{R} = \frac{200 - T_B}{R}$
 $T_B = 140^{\circ}C$

106. Ans (1)

$$T = 2\pi \sqrt{\frac{m}{k}}$$

$$T' = 2\pi \sqrt{\frac{m}{nk}} = \frac{T}{\sqrt{n}}$$

107. Ans (4)

resultant amplitude = $\sqrt{(4)^2 + (4)^2}$

108. Ans (4)

$$T = 2\pi \sqrt{\frac{\ell}{g}} = 2\pi \sqrt{\frac{\ell r^2}{GM}} = 2\pi r \sqrt{\frac{\ell}{GM}}$$
$$\frac{T_2}{T_1} = \frac{r_2}{r_1} = \frac{(R+R)}{R} = \frac{2}{1}$$

109. Ans (4)

$$T.E = \frac{1}{2}ka^2$$

$$T.E \propto a^2$$

$$\omega = \frac{2\pi}{T} = \frac{2\pi}{8} = \frac{\pi}{4}$$

$$\Delta t = 4 - 2 = 2$$

$$\Delta \phi = \omega \, \Delta t = \frac{\pi}{4} \times 2 = \frac{\pi}{2}$$

111. Ans (2)

$$a = -\omega^2 x$$

when x = 0 then a = 0

112. Ans (3)

$$V_{\text{max}} = a\omega$$

$$\therefore 100 = 10\omega, \omega = 10$$

$$v = \omega \sqrt{a^2 - x^2}$$

$$\therefore 50 = 10\sqrt{(10)^2 - x^2}$$

$$\therefore 25 = 100 - x^2$$

$$\therefore x^2 = 75$$

$$x = 5\sqrt{3}$$
cm

113. Ans (4)

Periodic motion repeats itself after a certain fixed time interval.

114. Ans (4)

S.H.M is a periodic motion,

velocity is maximum at mean position in SHM, acceleration is directly proportional to the displacement in S.H.M,

115. Ans (3)

$$a_{max} = V_{max}$$

$$a\omega^2=a\omega$$

$$\omega = 1, \frac{2\pi}{T} = 1$$

$$T = 6.28 \text{ sec}$$

116. Ans (1)

$$v = n\lambda = 2 \times 5 = 10 \text{ cm/s}$$

117. Ans (4)

$$x = A \sin \omega t$$
 $v = A \omega \cos \omega t = v_0 \cos \omega t$

$$a = -v_0 \omega \sin \omega t = -\omega^2 A \sin \omega t = -\omega^2 x$$

$$a = -v_0 \omega \sqrt{1 - \cos^2 \omega t}$$

$$=-v_0\omega\sqrt{1-\frac{v^2}{v_0^2}}=-\omega\sqrt{v_0^2-v^2}$$

a-x graph is straight line passing through origin.

a-v graph is neither straight line nor a parabola.

v-t graph is of either sine or cosine function

118. Ans (1)

The velocity of particle executing S.H.M at its extreme position is zero, hence its momentum is also zero.

119. Ans (2)

For fundamental mode $\frac{\lambda}{2} = 100 \text{ cm}$, $\lambda = 200 \text{ cm}$ $V = n\lambda = 330 \times \frac{200}{100} = 660 \text{ m/s}$

120. Ans (2)

Path difference = $\pi r - 2r = (2n - 1) \frac{\lambda}{2}$ {for minima}

$$(3.14 - 2)r = \frac{\lambda}{2}$$
 (For smallest radius n = 1)

$$r = \frac{\lambda}{2 \times 1.14} = \frac{0.342}{2 \times 1.14} = 0.15 \text{ m}$$

121. Ans (3)

$$3 \times \frac{V}{4\ell_C} = 2 \times \frac{V}{2\ell_0}$$
$$\frac{\ell_C}{\ell_0} = \frac{3}{4}$$

122. Ans (1)

$$A_x = 4 \sin\left(\frac{\pi x}{15}\right)$$

At x = 5 cm,
$$A_x = 4 \sin\left(\frac{\pi}{15} \times 5\right) = 4 \sin\frac{\pi}{3}$$

 $A_x = 4 \times \frac{\sqrt{3}}{2} = 2\sqrt{3} \text{ cm}$

123. Ans (2)

 $f_0 = 50 \text{ Hz}, f_1 = 3f_0, f_2 = 5f_0 \Rightarrow COP$ only odd harmonics

124. Ans (3)

Beat period $T = \frac{1}{n_1 \sim n_2} = \frac{1}{384 - 380} = \frac{1}{4} \text{sec.}$ Hence minimum time interval between maxima and minima $t = \frac{T}{2} = \frac{1}{8} \text{sec.}$

125. Ans (1)

$$L_2 - L_1 = \Delta L = 10 \log \frac{P_2}{P_1} = 10 \log \frac{600}{30}$$

 $\Delta L = 10 \log 20 \approx 13 \text{ dB}$

126. Ans (3)

$$V_S = \sqrt{\frac{E}{\rho}}$$

E: Coefficient of elasticity

ρ : Density of medium

128. Ans (3)

$$\omega = 60 \implies 2\pi n = 60 \implies n = \frac{30}{\pi} \text{ Hz}$$

$$V = \frac{\omega}{K} = \frac{60}{2} = 30 \text{ m/s}$$

$$K = \frac{2\pi}{\lambda} = 2 \implies \lambda = \pi \text{ metre}$$

129. Ans (2)

$$A^{2} = a_{1}^{2} + a_{2}^{2} + 2a_{1}a_{2}\cos\theta$$

$$a^{2} = a^{2} + a^{2} + 2a^{2}\cos\theta$$

$$\cos\theta = -\frac{1}{2}$$

$$\theta = \frac{2\pi}{3}$$

130. Ans (2)

131. Ans (1)

NCERT Pg. # 285 $y = 0.02 \sin(x + 30 t)$ $\omega = 30, K = 1$ $V = \frac{\omega}{K} = \frac{30}{1} = 30 \text{m/s}$ $\therefore V = \sqrt{\frac{T}{m}}$ $\therefore T = \text{mV}^2 = (30)^2 \times 10^{-4} = 9 \times 10^{-2}$

132. Ans (2)

T = 0.09 N

$$n_0 = \frac{V}{4\ell} = \frac{330}{4 \times 0.15} = 550 Hz$$

133. Ans (3)

$$\ell_2 = 3 \, \ell_1 = 3 \times 16 = 48 \text{ cm}$$

134. Ans (3)

 $A = A_0 e^{-\gamma t}$

135. Ans (2)

 $V_{\rm rms} \propto \sqrt{T}$ & it is independent of pressure.

136. Ans (3)

NCERT 12th Page No. # 317

143. Ans (3)

NCERT-XII Pg#348

Fact

148. Ans (1)

NCERT-XII, Pg. # 345

153. Ans (1)

NCERT (XI) Pg # 344, 3rd para

157. Ans (2)

$$\begin{array}{cccc} \mathrm{CH_3-CH-Br} & \xrightarrow{\quad \mathrm{Na} \quad } \mathrm{CH_3-CH-CH-CH_3} \\ \mathrm{CH_3} & \mathrm{CH_3} & \mathrm{CH_3} \end{array}$$

160. Ans (2)

$$2CH_4 + O_2 \xrightarrow{MO_2O_3} HCHO$$

161. Ans (3)

$$\bigcirc + \operatorname{Br}_2 \xrightarrow{\operatorname{AlCl}_3} \bigcirc$$

(Aromatic electrophilic substitution)

$$HC \equiv CH \xrightarrow{HBr} CH_2 = CH - Br$$
(Electrophilic addition)

$$CH_4 + Br_2 \xrightarrow{hv} CH_3 - Br + HBr$$
(Free radical substitution)

CH₃-CH=CH₂+HBr
$$\xrightarrow{(C_6H_5COO)_2}$$
 CH₂-CH-CH₂-Br (Free radical addition)

162. Ans (1)

E-2 1-phenylpropene.

164. Ans (3)

C-C-C-C-C
$$\xrightarrow{O_3}$$
 C-C-C-C-C + C-C-H

C-C O

NCERT (XIth) Part II, Pg. # 397

OR

$$\begin{array}{c} \text{CH}_3\text{-CH}_2\text{-C-CH}_2\text{-CH}_3\\ \text{Oll}\\ \text{Oll}\\ \text{OCH-CH}_3\\ \\ \text{(i) O}_3\text{/ (ii) H}_2\text{O/Zn}\\ \\ \text{CH}_3\text{-CH}_2\text{-C-CH}_2\text{-CH}_3\text{+CH}_3\text{-CHO}\\ \\ \text{O} \end{array}$$

166. Ans (2)

$$\begin{array}{c|c}
\hline
-CH \neq CH - CH_3 \xrightarrow{KMnO_4} \\
\hline
-CHO + CH_3CHO
\\
\hline
-COOH CH_3-COOH
\\
in absence of Zn)
\end{array}$$

169. Ans (1)

Fact Based

171. Ans (3)

Cummene followed by FRSR mechanism.

172. Ans (3)

Aniline do not show FCR.

173. Ans (3)

A = Benzene B = Friedel craft reaction

175. Ans (2)

$$\% N = \frac{1.4}{w} \times NV = \frac{1.4}{0.1} \times 4 = 56\%$$

178. Ans (3)

Silver acetylides are obtained by passing 1-alkyne in the ammonical solution of silver nitrate (tollen's reagent)