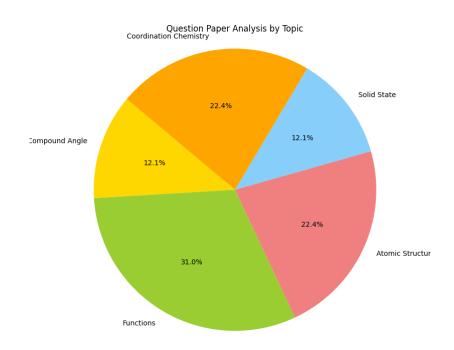
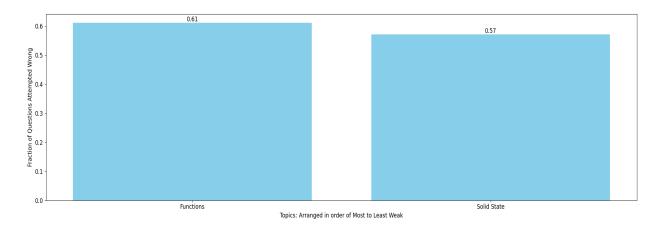
Riddhim Verma Total MLAssist - Personalised DPP

Question Paper Analysis:



Weak Topic Analysis:



Practice Questions:

Functions:

- A function f: R \rightarrow R is such that $f\left(\frac{1-x}{1+x}\right) = x$ for all $x \neq -1$. Prove the following. 7.
 - (a) f(f(x)) = x

(b) $f(1/x) = -f(x), x \neq 0$ (c) f(-x-2) = -f(x) - 2

- (a) Let $P(x) = x^6 + ax^5 + bx^4 + cx^3 + dx^2 + ex + f$ be a polynomial such that 1. P(1) = 1; P(2) = 2; P(3) = 3; P(4) = 4; P(5) = 5 and P(6) = 6 then find the value of P(7).
 - (b) Let a and b be real numbers and let f(x) = asin x + b ³√x + 4, ∀x ∈ R. If $f(\log_{10}(\log_3 10)) = 5$ then find the value of $f(\log_{10}(\log_{10} 3))$.
- Find the number of integer in the range of the function, 1.

$$f(x) = \sqrt{\sin \frac{\pi x}{2}} + \sqrt{16 - x^2} + \sqrt{x} + \log_2(x(x - 2))$$

Daily Work Sheet-2

INGLE CORRECT TYPE

- Which of the following statements are incorrect? I. If f(x) and g(x) are one to one then f(x) + g(x) is also one to one.
 - II. If f(x) and g(x) are one-one then $f(x) \cdot g(x)$ is also one-one.
 - III. If f(x) is odd then it is necessarily one to one.
 - (A) I and II only

(B) II and III only

(C) III and I only

(D) I, II and III

- 18. The number of functions f from {1,2,3,20} onto {1,2,3, ...,20} such that f(k) is a multiple of 3, [JEE - Main 2019] whenever k is a multiple of 4, is
 - (A) (15)! × 6!

- (B) $5^6 \times 15$ (C) $5! \times 6!$ (D) $6^5 \times (15)!$

- 5. If $f(x) + 2f(\frac{1}{x}) = 3x, x \neq 0$ and $S = \{x \in R: f(x) = f(-x)\}$; then S: [JEE Main 2016]
 - (A) contains exactly one element.
 - (B) contains exactly two elements.
 - (C) contains more than two elements
 - (D) is an empty set.

Solid State:

- The interstitial hole is called tetrahedral because
 - (A) It is formed by four spheres.
 - (B) Partly same and partly different.
 - (C) It is formed by four spheres the centres of which form a regular tetrahedron.
 - (D) None of the above three.
- Calculate the density of diamond from the fact that it has face centered cubic structure with two
- 19. A cubic solid is made up of two elements A and B. Atoms B are at the corners of the cube and A at the body centre. What is the formula of compound?
- 34. Ammonium chloride crystallizes in a body centred cubic lattice with edge length of unit cell of 390 pm. If the size of chloride ion is180 pm, the size of ammonium ion would be:

[Jee-Main (online)-12]

- (A) 158 pm
- (B) 174 pm
- (C) 142 pm
- (D) 126 pm
- How many of the following are covalent network solids?

C. D. C. C. Diamand Z. CO. C. AM. C.C. CO.