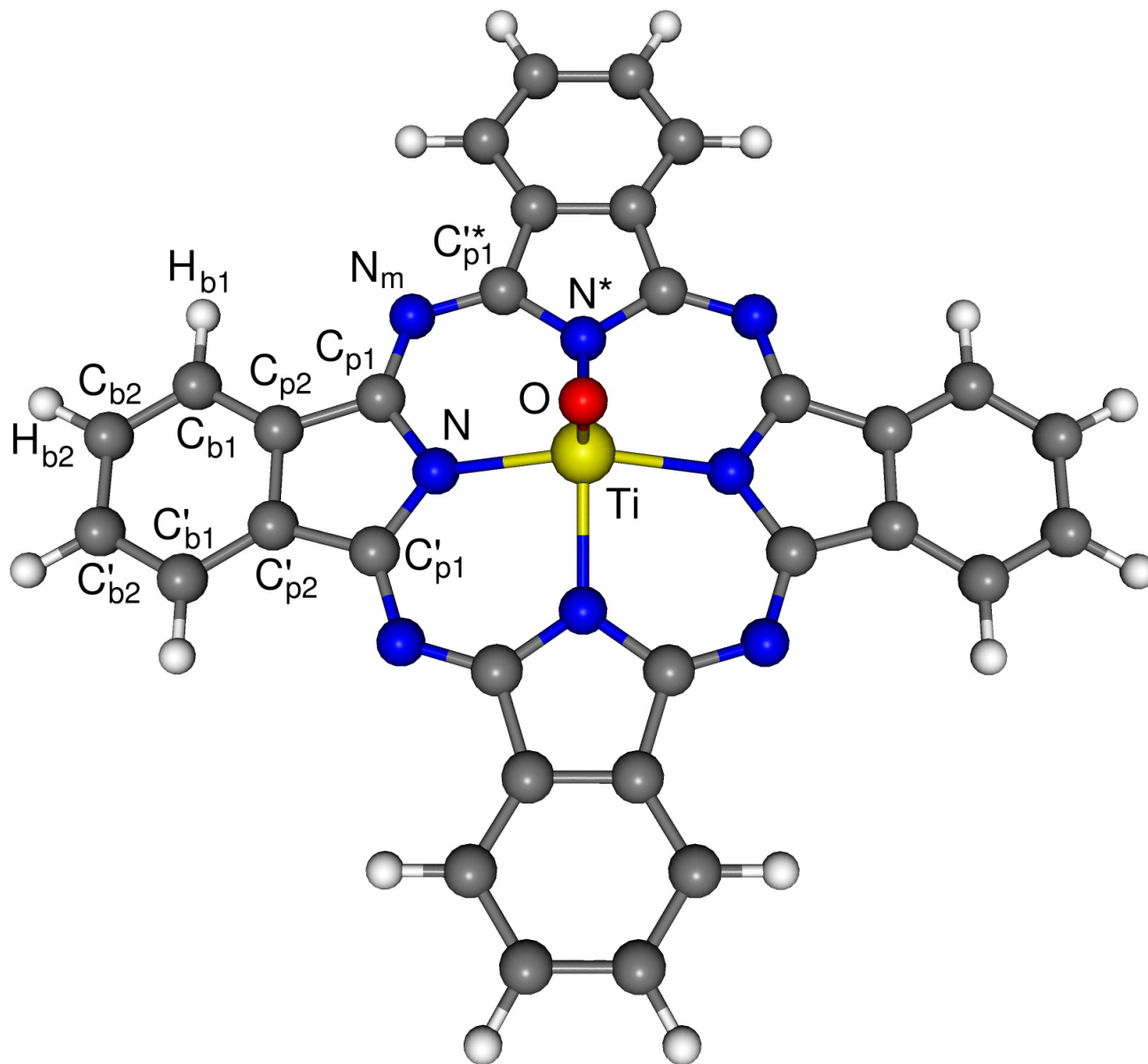


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Oxotitanium and dihalogenotitanium porphyrins and  
phthalocyanines: a density functional theory and gas  
electron diffraction study

(TiOPc,  $\text{TiX}_2\text{P}$  and  $\text{TiX}_2\text{Pc}$ , where P = porphyrin,  
Pc = phthalocyanine, X = F, Cl, Br, I)



TiOPc molecule ( $C_{4v}$  symmetry) with atom designations

## X-Ray crystallographic data

1. W. Hillar, J. Strähle, W. Kovel, and M. Hanack, *Z. Kristallogr.*, 1982, **159**, 173.

Oxotitanium phthalocyanine: polymorphic in the solid state, exists in several crystal modifications, mainly I ( $\beta$ -TiOPc), II ( $\alpha$ -TiOPc) and Y, as well as in amorphous form.

Phase I: monoclinic structure in space group  $P2_1/c$ .

Phase II: triclinic in space group  $P\bar{1}$ .

## Density functional theory calculations

2. A. V. Zakharov and G. V. Girichev, *J. Mol. Struct.: THEOCHEM*, 2008, **851**, 183.

Oxotitanium porphyrin, octamethylporphyrin, porphyrazine and phthalocyanine: equilibrium structures of  $C_{4v}$  symmetry with convex macrocycles.

## Computational

### Combination of basis sets 1 (used in study [1])

**Ti:**  $8s7p6d1f/6s5p3d1f$  basis set and RECP (M. Dolg, U. Wedig, H. Stoll, H. Preuss, *J. Chem. Phys.*, 1987, **86**, 866)

**C, N, H, O:**  $10s6p/5s3p$  basis set (T. H. Dunning, *J. Chem. Phys.*, 1971, **55**, 716) with one polarisation  $d$  function (C, N, F) or  $p$  function (H)

### Combination of basis sets 2

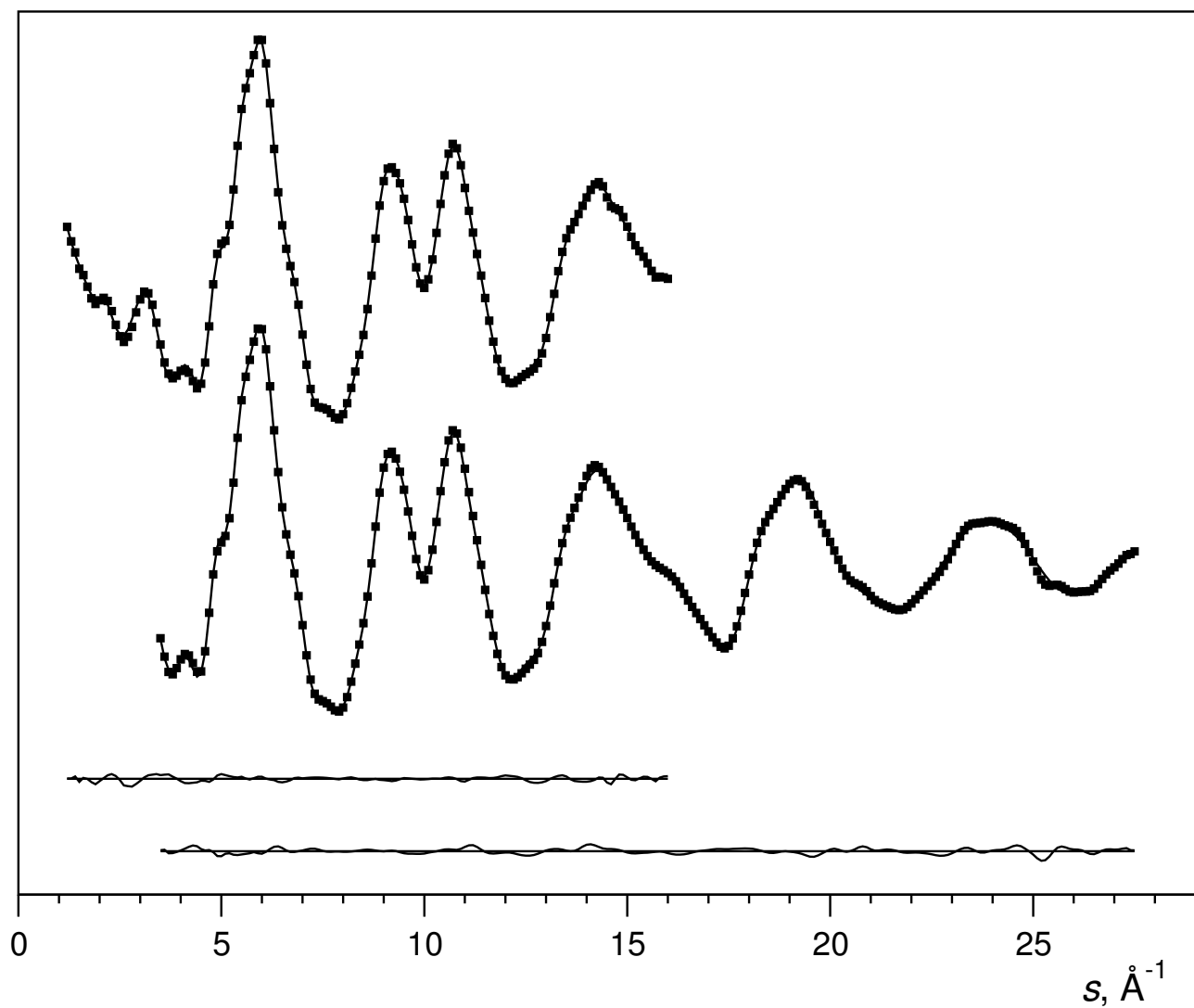
**Ti:**  $8s7p6d2f1g/6s5p3d2f1g$  basis set and RECP (M. Dolg, U. Wedig, H. Stoll, H. Preuss, *J. Chem. Phys.*, 1987, **86**, 866; J. M. L. Martin and A. Sundermann, *J. Chem. Phys.*, 2001, **114**, 3408)

**C, N, H, O:** cc-pVTZ basis sets (J. T. H. Dunning, *J. Chem. Phys.*, 1989, **90**, 1007)

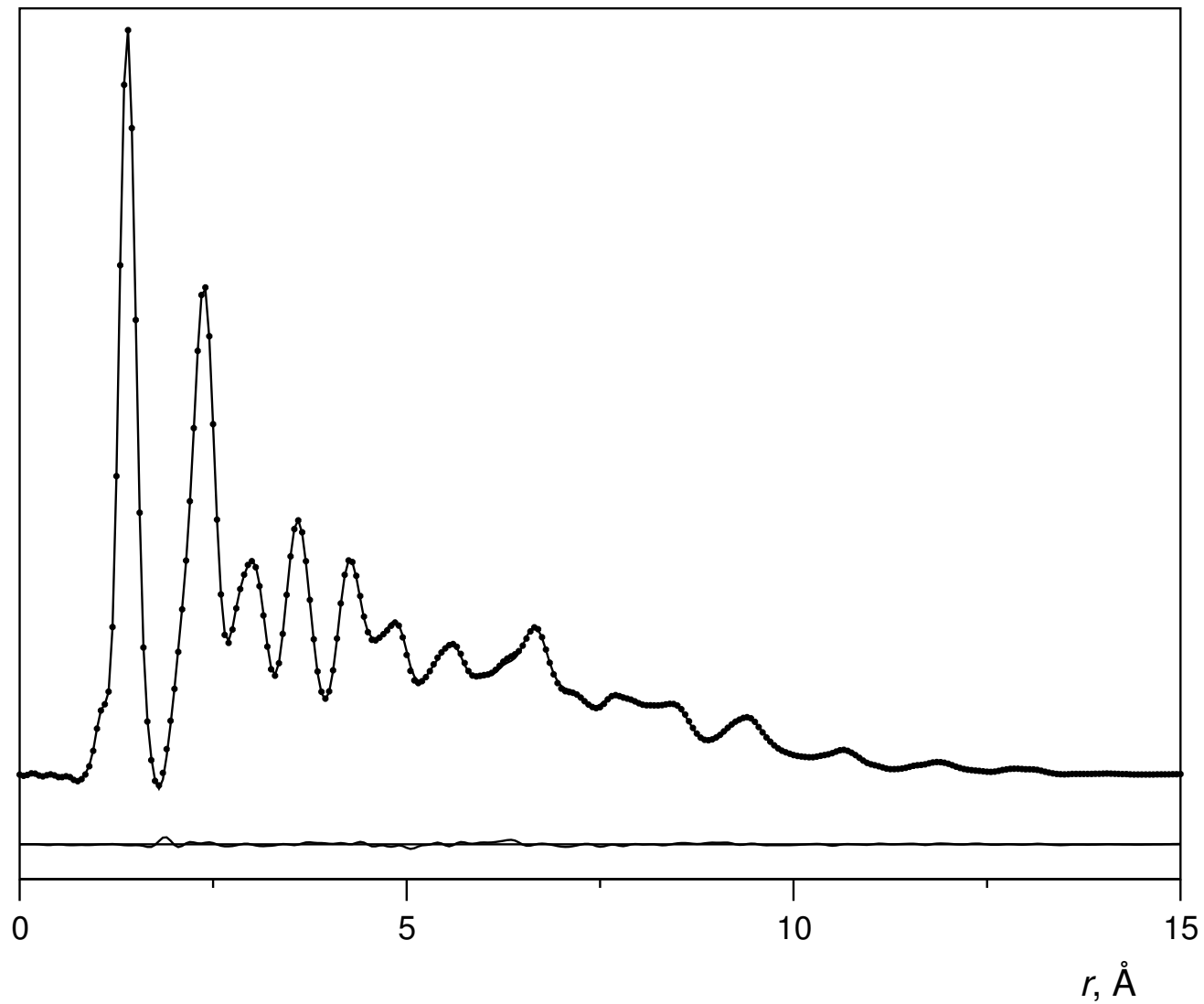
### Combination of basis sets 3

**Ti:** cc-pVTZ-NR (N. B. Balabanov and K. A. Peterson, *J. Chem. Phys.*, 2005, **123**, 064107)

**C, N, H, O:** cc-pVTZ basis sets



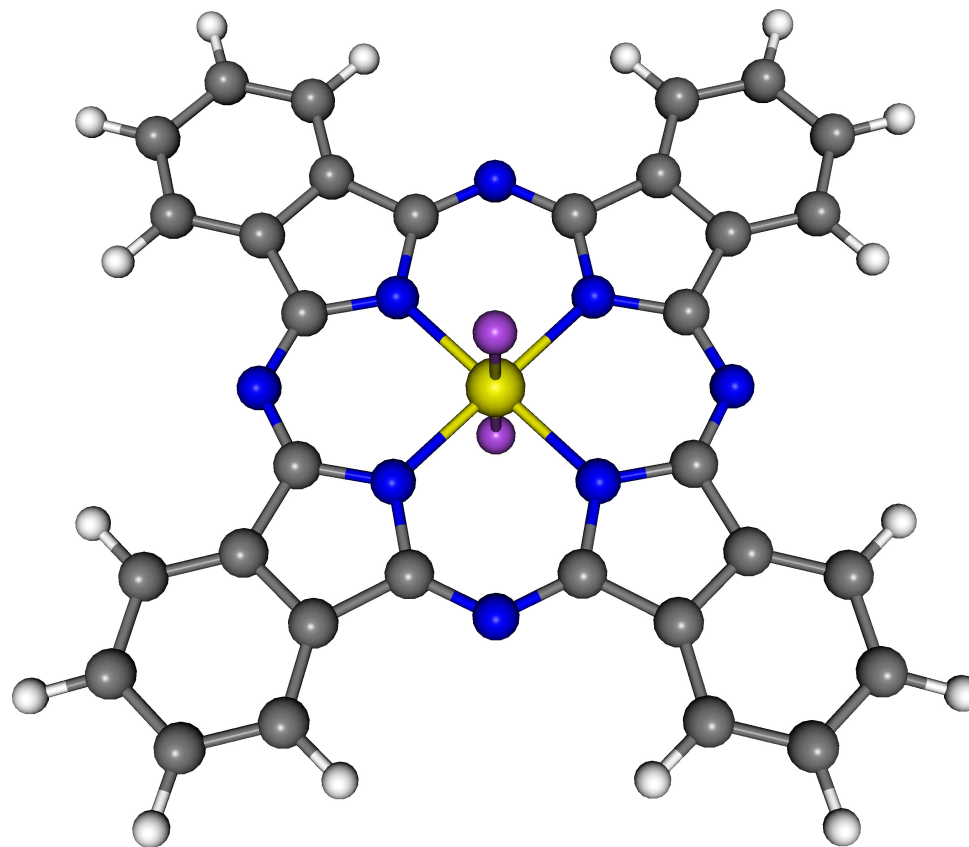
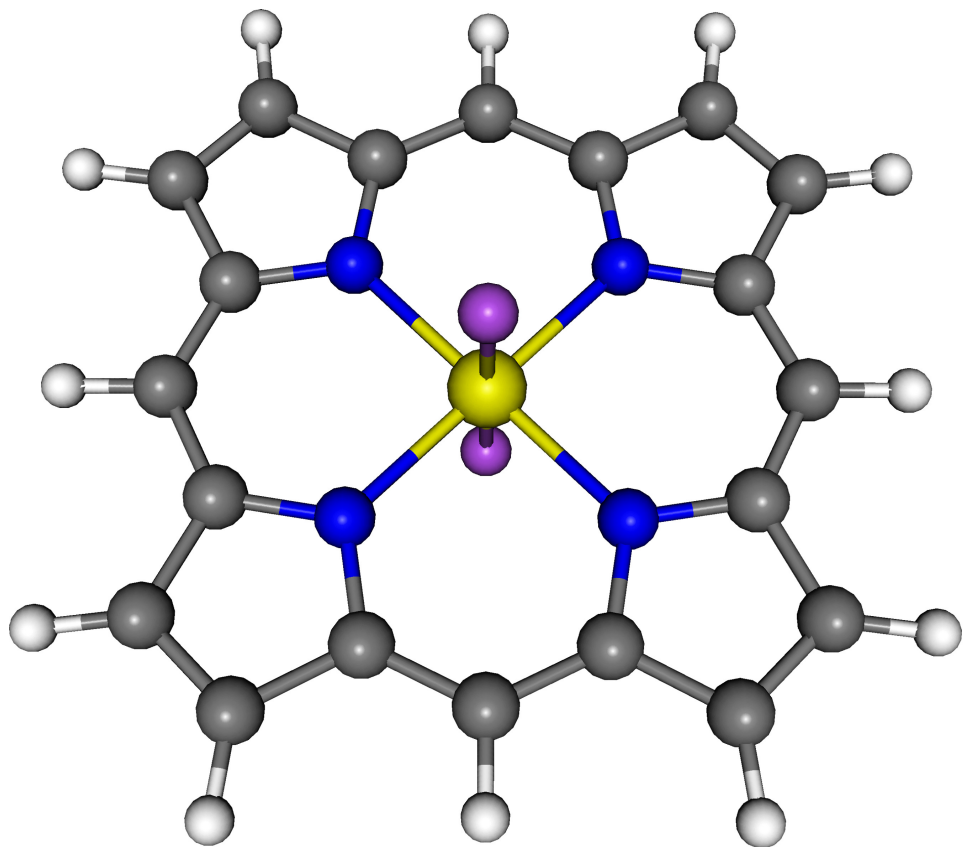
Average experimental (dots) and theoretical (full lines) molecular intensity curves for TiOPc from the long (upper curve) and short (lower curve) camera distances and difference curves (below).



Experimental (dots) and theoretical (full line) radial distribution curves for TiOPc and difference curve (below).

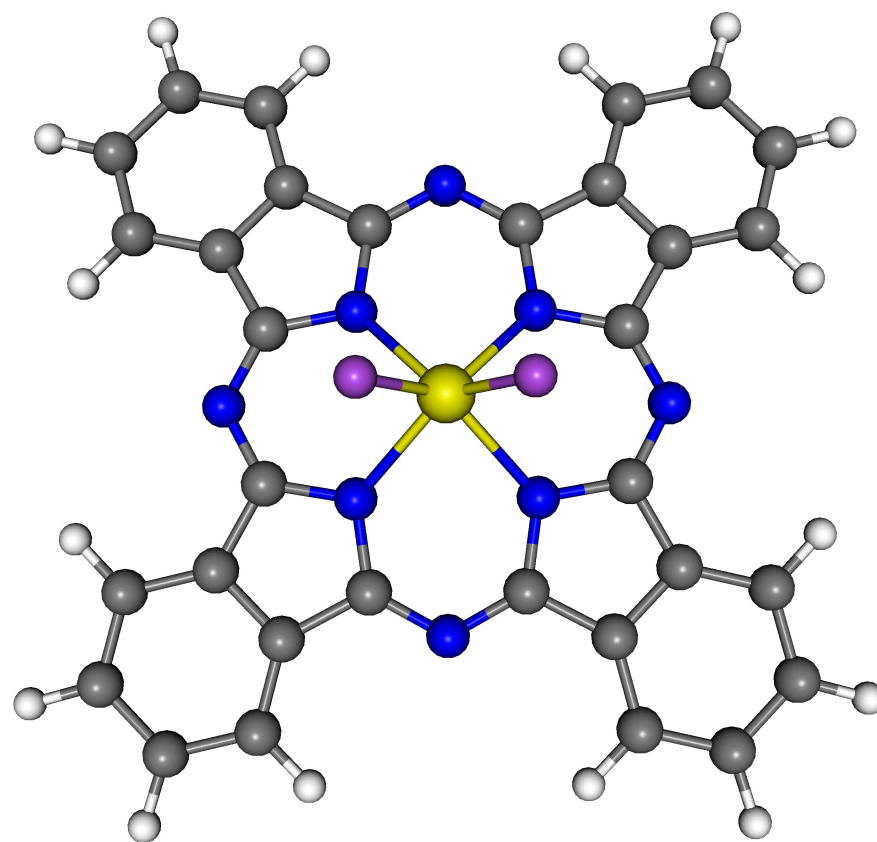
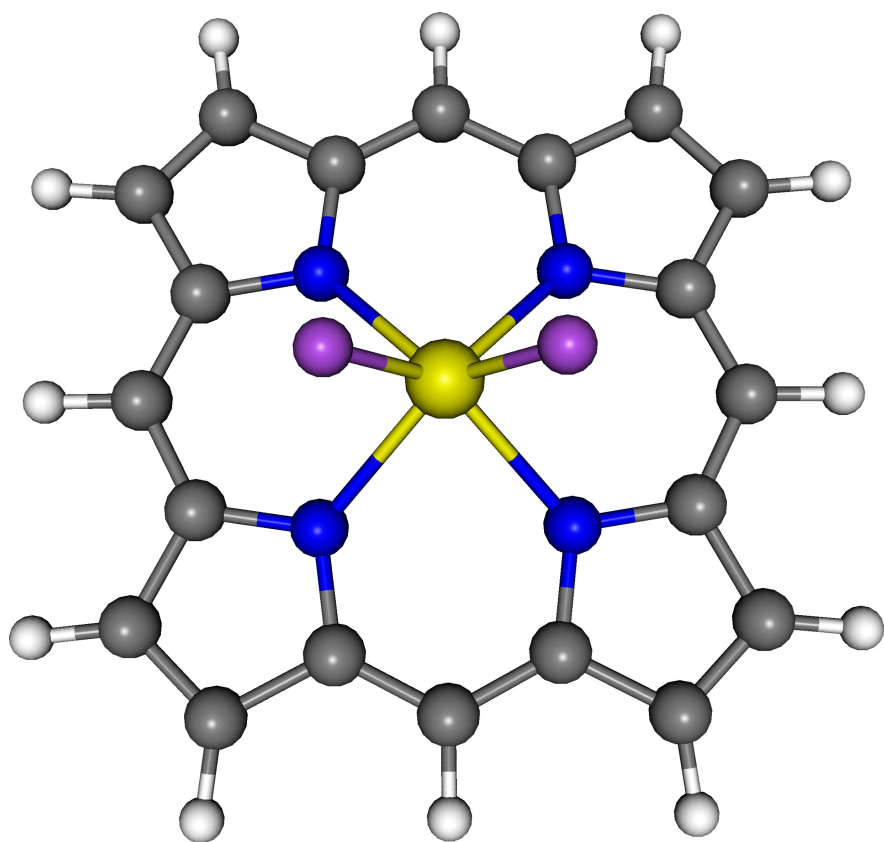
# Structural parameters of oxotitanium phthalocyanine

	Theoretical calculations			ED	X-ray [1]	
	1 [2]	2	3		$\alpha$	$\beta$
Distances, Å						
$r(\text{Ti-O})$	1.617	1.617	1.619	1.595(10)	1.650(4)	1.626(7)
$r(\text{Ti-N})$	2.087	2.087	2.094	2.089(7)	2.066	2.066
$r(\text{N-C}_{p1})$	1.376	1.373	1.373	1.382(8)	1.376	1.378
$r(\text{N}_m\text{-C}_{p1})$	1.323	1.322	1.322	1.326(8)	1.330	1.326
$r(\text{C}_{p1}\text{-C}_{p2})$	1.456	1.454	1.454	1.449(9)	1.454	1.459
$r(\text{C}_{p2}\text{-C}'_{p2})$	1.406	1.404	1.404	1.400(21)	1.409	1.402
$r(\text{C}_{p2}\text{-C}_{b1})$	1.393	1.391	1.391	1.407(9)	1.383	1.387
$r(\text{C}_{b1}\text{-C}_{b2})$	1.391	1.388	1.388	1.405(9)	1.391	1.383
$r(\text{C}_{b2}\text{-C}'_{b2})$	1.406	1.403	1.403	1.388(24)	1.413	1.403
$z(\text{Ti})\text{-}z(\text{N})$	0.649	0.653	0.667	0.641	0.625	0.639
Valence and dihedral angles, degrees						
$\alpha(\text{O-Ti-N})$	108.1	108.2	108.6	107.9(8)	107.6	107.9
$\alpha(\text{Ti-N-C}_{p1})$	124.6	124.6	124.6	124.3(7)		
$\alpha(\text{N-C}_{p1}\text{-N}_m)$	127.5	127.4	127.5	126.4(8)		
$\gamma(\text{O-Ti-N-C}_{p1})$	81.9	81.9	81.9	80(3)		
$\gamma(\text{Ti-N-C}_{p1}\text{-N}_m)$	17.7	17.7	17.9	21(7)		

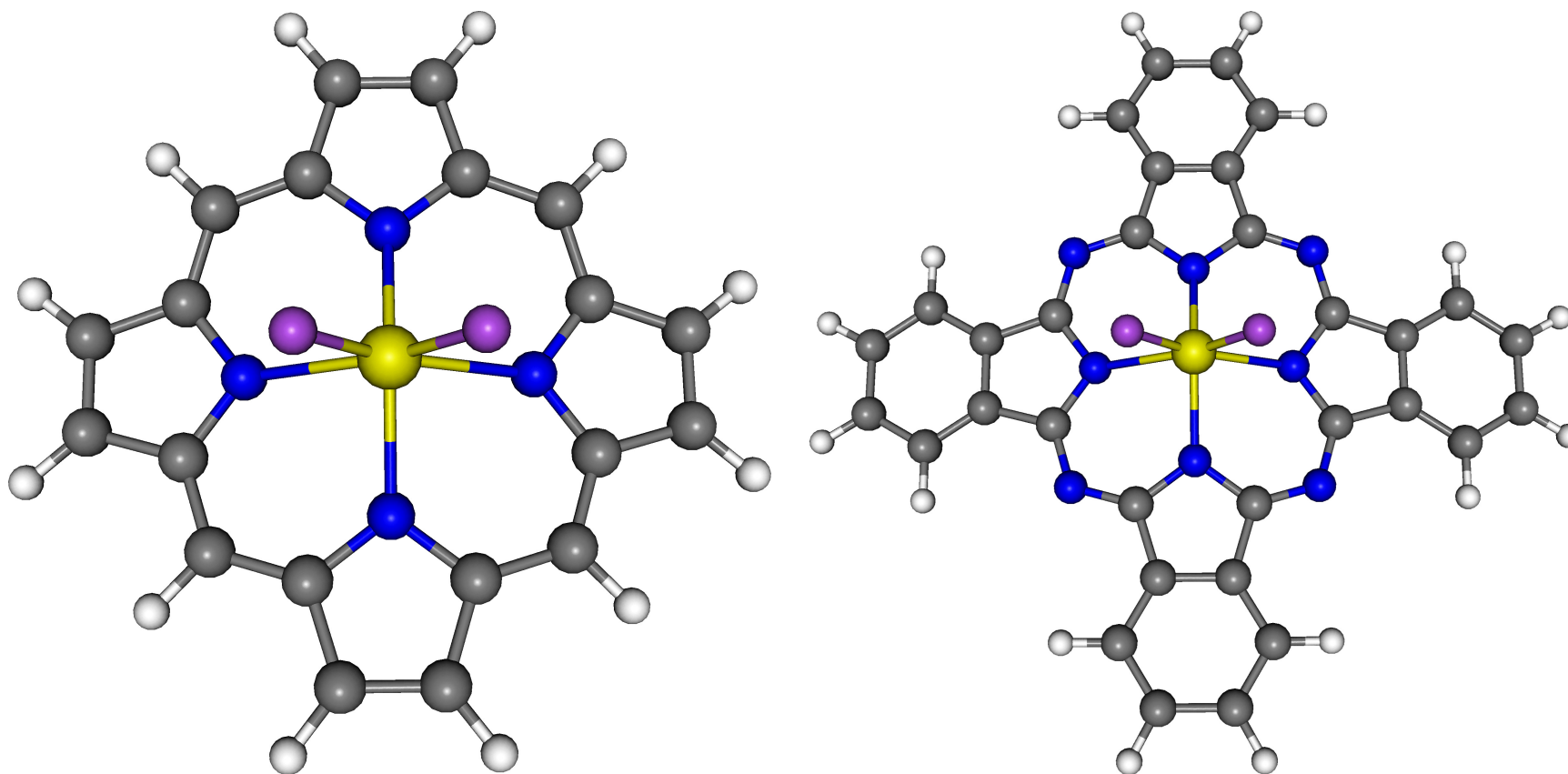


TiF<sub>2</sub>P and TiF<sub>2</sub>Pc, *trans* isomers ( $D_{4h}$  symmetry)





TiF<sub>2</sub>P and TiF<sub>2</sub>Pc, *cis* isomers ( $C_{2v}$  symmetry), halogens in the same plane with meso atoms



$\text{TiF}_2\text{P}$  and  $\text{TiF}_2\text{Pc}$ , *cis* isomers ( $\text{C}_{2v}$  symmetry), halogens in the same plane with central nitrogen atoms

## X-Ray crystallographic data

**TiX<sub>2</sub>TPP** (X = F, Cl, Br; TPP - tetraphenylporphyrin) – *trans* isomers:

**TiF<sub>2</sub>TPP:** J.-C. Marchon, J.-M. Latour, A. Grand, M. Belakhovsky, M. Loos, J. Goulon, *Inorg. Chem.*, 1990, **29**, 57;

**TiCl<sub>2</sub>TPP, TiBr<sub>2</sub>TPP:** A. N. Christensen, A. Grand, M. S. Lehmann, D. E. Cox, *Acta Chem. Scand.*, 1990, **44**, 103;

**TiBr<sub>2</sub>TPP:** C. Lecomte, J. Protas, J.-C. Marchon, M. Nakajima, *Acta Crystallogr., Sect. B: Struct. Crystallogr. Cryst. Chem.*, 1978, **34**, 2856.

**TiCl<sub>2</sub>Pc** – *cis* isomer:

V. L. Goedken, G. Dessy, C. Ercolani, V. Fares, L. Gastaldi, *Inorg. Chem.*, 1985, **24**, 991.

## Computational

**Ti:**  $8s7p6d1f/6s5p3d1f$  basis set and RECP (M. Dolg, U. Wedig, H. Stoll, H. Preuss, *J. Chem. Phys.*, 1987, **86**, 866)

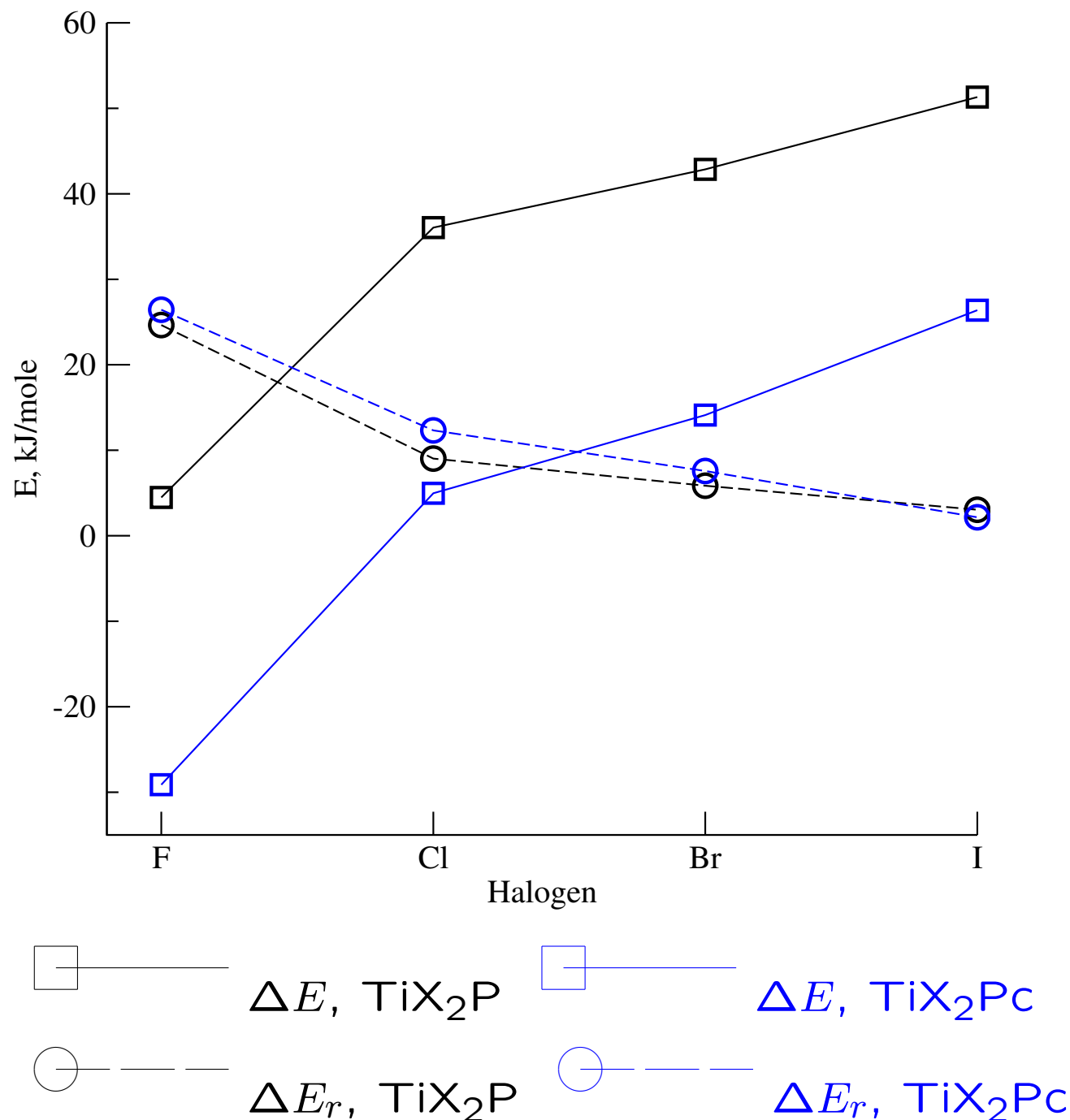
**C, N, H, F:**  $10s6p/5s3p$  basis set (T. H. Dunning, *J. Chem. Phys.*, 1971, **55**, 716) with one polarisation  $d$  function (C, N, F) or  $p$  function (H) and one diffuse  $p$  function (F only)

**Cl:**  $12s9p/6s5p$  basis set (A. D. McLean, G. S. Chandler, *J. Chem. Phys.*, 1980, **72**, 5639) with one polarisation  $d$  function and one diffuse  $p$  function

**Br, I:**  $14s10p2d1f/3s3p2d1f$  (Br) or  $14s10p3d1f/3s3p2d1f$  (I) basis sets and RECP (J. M. L. Martin, A. Sundermann, *J. Chem. Phys.*, 2001, **114**, 3408) with one diffuse  $p$  function

Energy differences between *trans* isomers ( $D_{4h}$ ) and *cis* isomers ( $C_{2v}$ )

	$\Delta E$ , kJ/mol
TiF <sub>2</sub> P	4.48
TiCl <sub>2</sub> P	36.04
TiBr <sub>2</sub> P	42.86
TiI <sub>2</sub> P	51.31
TiF <sub>2</sub> Pc	-29.13
TiCl <sub>2</sub> Pc	4.97
TiBr <sub>2</sub> Pc	14.11
TiI <sub>2</sub> Pc	26.39
$\Delta E = E(C_{2v}) - E(D_{4h})$	



Selected structural parameters of  $\text{TiX}_2\text{P}$  molecules

	F ( $D_{4h}$ )	F ( $C_{2v}$ )	Cl ( $D_{4h}$ )	Cl ( $C_{2v}$ )	Br ( $D_{4h}$ )	I ( $D_{4h}$ )
Distances, Å						
$r(\text{Ti-X})$	1.813	1.799	2.315	2.327	2.483	2.746
$r(\text{Ti-N})$	2.080	2.169	2.071	2.158	2.070	2.065
$r(\text{N-C}_{p1})$	1.370	1.376	1.372	1.378	1.373	1.375
$r(\text{C}_m\text{-C}_{p1})$	1.397	1.384	1.395	1.383	1.395	1.394
$r(\text{C}_{p1}\text{-C}_{p2})$	1.443	1.438	1.440	1.436	1.440	1.438
$r(\text{C}_{p2}\text{-C}'_{p2})$	1.363	1.359	1.364	1.359	1.364	1.364
$r(\text{C}_m\text{-H}_m)$	1.083	1.082	1.083	1.082	1.083	1.083
$r(\text{C}_{p2}\text{-H}_p)$	1.079	1.079	1.079	1.079	1.079	1.079
Valence angles, degrees						
$\alpha(\text{Ti-N-C}_{p1})$	125.9	128.8	126.0	125.9	126.0	126.1
$\alpha(\text{X-Ti-X})$		85.0		82.9		

# Selected structural parameters of TiX<sub>2</sub>Pc molecules

	F ( <i>D</i> <sub>4h</sub> )	F ( <i>C</i> <sub>2v</sub> )	Cl ( <i>D</i> <sub>4h</sub> )	Cl ( <i>C</i> <sub>2v</sub> )	Br ( <i>D</i> <sub>4h</sub> )	I ( <i>D</i> <sub>4h</sub> )
Distances, Å						
<i>r</i> (Ti-X)	1.820	1.800	2.325	2.319	2.492	2.751
<i>r</i> (Ti-N)	2.029	2.125	2.021	2.119	2.019	2.015
<i>r</i> (N-C <sub><i>p</i>1</sub> )	1.372	1.377	1.375	1.380	1.376	1.378
<i>r</i> (C <sub><i>m</i></sub> -C <sub><i>p</i>1</sub> )	1.328	1.319	1.326	1.317	1.326	1.325
<i>r</i> (C <sub><i>p</i>1</sub> -C <sub><i>p</i>2</sub> )	1.459	1.452	1.457	1.450	1.456	1.455
<i>r</i> (C <sub><i>p</i>2</sub> -C' <sub><i>p</i>2</sub> )	1.412	1.402	1.411	1.401	1.412	1.412
<i>r</i> (C <sub><i>p</i>2</sub> -C <sub><i>b</i>1</sub> )	1.392	1.394	1.393	1.394	1.393	1.393
<i>r</i> (C <sub><i>b</i>1</sub> -C <sub><i>b</i>2</sub> )	1.391	1.390	1.391	1.389	1.391	1.390
<i>r</i> (C <sub><i>b</i>2</sub> -C' <sub><i>b</i>2</sub> )	1.405	1.407	1.405	1.407	1.405	1.406
<i>r</i> (C <sub><i>b</i>1</sub> -H <sub><i>b</i>1</sub> )	1.082	1.082	1.082	1.082	1.082	1.082
<i>r</i> (C <sub><i>b</i>2</sub> -H <sub><i>b</i>2</sub> )	1.083	1.083	1.083	1.083	1.083	1.083
Valence angles, degrees						
α(Ti-N-C <sub><i>p</i>1</sub> )	124.4	127.3	124.6	125.8	124.6	124.7
α(X-Ti-X)		85.9		83.7		