

1. Eq<sup>n</sup>s with recycling

$$\frac{dR_s}{dt} = -k_f L R_s + k_r R_s^* - k_e R_s + V_s + k_{rec}(R_i^o) \quad - (1)$$

$$\frac{dR_s^*}{dt} = k_f L R_s + (-k_r R_s^*) - k_e^* R_s^* + k_{rec} R_i^{o*} \quad - (2)$$

$$\frac{dR_i^{oT}}{dt} = k_e R_s + k_e^* R_s^* - k_{deg} R_i^{oT} - k_{rec} R_i^{oT} \quad - (3)$$

$$\frac{dR_i^{o*}}{dt} = k_e^* R_s^* - k_{deg} R_i^{o*} - k_{rec} R_i^{o*} \quad - (4)$$

@ S.S.

Adding (1), (2) and (3) we get

$$V_s = k_{deg} R_i^{oT}$$

From (3) we get  $k_{deg} R_i^{oT} = k_e R_s + k_e^* R_s^* - k_{rec} R_i^{oT}$

$$\therefore \boxed{V_s = k_e R_s + k_e^* R_s^* - k_{rec} R_i^{oT}}$$

$$\Rightarrow \boxed{R_s = \frac{V_s - k_e^* R_s^* + k_{rec} R_i^{oT}}{k_e}}$$

Substituting above value of  $R_s$  in (2) we get

$$k_f L \left[ \frac{V_s - k_e^* R_s^* + k_{rec} R_i^{oT}}{k_e} \right] - k_r R_s^* - k_e R_s^* + k_{rec} R_i^{o*} = 0$$

$$\Rightarrow R_s^* \left[ \frac{k_f L k_e^*}{k_e} + k_r + k_e^* \right] = \frac{k_f L V_s}{k_e} + k_{rec} \left[ \frac{k_f L}{k_e} (R_i^{oT}) + R_i^{o*} \right]$$

$$\Rightarrow R_s^* = \frac{k_f L V_s}{k_e(k_r + k_e^*)} \left[ \frac{1}{1 + \frac{k_f L k_e^*}{k_e(k_r + k_e^*)}} \right] + k_{rec} \left[ \frac{\frac{k_f L R_i^{oT}}{k_e(k_r + k_e^*)}}{1 + \frac{k_f L k_e^*}{k_e(k_r + k_e^*)}} + \frac{\frac{R_i^{o*}}{\frac{k_f L k_e^*}{k_e(k_r + k_e^*)} + 1}}{\frac{k_f L k_e^*}{k_e(k_r + k_e^*)} + 1} \right]$$

$$\boxed{R_s^* = \left( \frac{K_{ss} L}{1 + K_{ss} L} \right) \left[ \frac{V_s}{k_e^*} + \frac{k_{rec} R_i^{oT}}{k_e^*} \right] + \frac{k_{rec} R_i^{o*}}{(k_e^* + k_r) \left[ \frac{1}{K_{ss} L + 1} \right]}}$$

Now, from (4)  $(k_{rec} + k_{deg}) R_i^* = k_e^* R_s^* \rightarrow k_{rec}$

$$\Rightarrow R_i^* = \frac{k_e^* R_s^*}{(k_{deg} + k_{rec})}$$

Total active receptor

$$R_{total}^* = R_s^* + R_i^*$$

$$= R_s^* \left[ 1 + \frac{k_e^*}{(k_{deg} + k_{rec})} \right]$$

$$= \left[ \left( \frac{k_{ss} L}{k_{ss} L + 1} \right) \left[ \frac{V_s}{k_e^*} + \frac{k_{rec} R_i^{*T}}{k_e^*} \right] + \frac{k_{rec} R_i^{*T}}{(k_e^* + k_{rec})} \left( \frac{1}{k_{ss} L + 1} \right) \right] \left[ 1 + \frac{k_e^*}{k_{deg} + k_{rec}} \right]$$

$$R_{total}^* = \left[ \frac{1}{k_e^*} + \frac{1}{(k_{deg} + k_{rec})} \right] \left[ \left( \frac{k_{ss} L}{k_{ss} L + 1} \right) [V_s + k_{rec} R_i^{*T}] + \frac{k_{rec} R_i^{*T} k_e^*}{(k_e^* + k_{rec})} \left( \frac{1}{k_{ss} L + 1} \right) \right]$$

Maximum total active receptor conc<sup>n</sup> is obtained when ligand conc<sup>n</sup> is very high i.e.  $L \gg 1 \Rightarrow k_{ss} L \gg 1$

$$\therefore R_{total}^{*max} = \left( \frac{1}{k_e^*} + \frac{1}{(k_{deg} + k_{rec})} \right) [V_s + k_{rec} R_i^{*T}]$$

$\therefore$  recycling increases amount of total active receptors

$$\therefore \text{for no recycle case } R_{total}^{*max} = \left( \frac{1}{k_e^*} + \frac{1}{k_{deg}} \right) [V_s]$$

$$2. (a) \quad \frac{dCa}{dt} = -d_a Ca + \frac{k_{oa} + k_a Ca^2}{1 + Ca^2 + Ca^2}$$

$$\frac{dCa}{dt} = -c_k + \frac{k_{or} + k_r Ca^2}{1 + Ca^2}$$

A is an activator of A and an activator of R.

R is an inhibitor of A and has no effect on R.

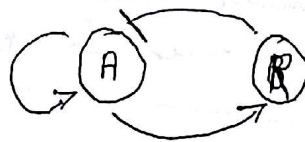
$d_a$  is the first order degradation constant of A.



2.(b) The fixed point is unstable, as soon as the point is encountered the system changes rapidly.

2.(d) Initially when the conc<sup>n</sup> of A is low, production of A gets activated and increases with increasing conc<sup>n</sup> of A. As A increases, concentration of R increases too. When the concentration of R reaches a certain concentration, it starts inhibiting the production of A. As R keeps increasing, A keeps decreasing and finally at very high R there is a sharp decrease in amount of A which brings the system back to its original state as A can no longer activate R. Same cycle continues in a loop and hence the system represents an oscillator. The black line traces the path of a point which goes around in a loop.

The arrows represent the same observation.



$$3.(a) \quad \frac{du}{dt} = \frac{\kappa}{1+v^n} - u = f(u, v) \quad (1)$$

$$\frac{dv}{dt} = \frac{\alpha}{1+u^n} - v = g(u, v) \quad (2)$$

3.(a) (i) v acts as a repressor for u  
u acts as a repressor for v

(ii)  $\kappa$  gives the effective rate of synthesis

(iii) n = cooperativity of repression

(iv) +1 is the degradation rate const. for both repressors.

3 (b) 1 steady state solution exists for  $n=1$

3 steady state solutions exist for  $n=2$

Increasing degree of cooperativity increases number of steady state sol's.

3 (c) For  $n=1$  1 steady state sol<sup>n</sup> exists. It is a stable sol<sup>n</sup> since all arrows direct to that point. It is a sink

For  $n=2, 3$  steady state sol's exist.

The point at around  $(2.5, 2.5)$  is a saddle point

The points near the axis are unstable spirals.

≠ Increase in  $n$  introduces instability.

3 (d) 
$$\overline{J} = \begin{bmatrix} -1 & -n\alpha u_s^{n+1}(1-u_s^n)^{-2} \\ -n\alpha(1+u_s^n)u_s^{n-1} & -1 \end{bmatrix}$$

For a centre  $u_s = u_s^*$

$$\therefore \overline{J} = \begin{bmatrix} -1 & -n\alpha u_s^{n+1}(1-u_s^n)^{-2} \\ -n\alpha u_s^{n+1}(1-u_s^n)^{-2} & -1 \end{bmatrix}$$

$$\text{tr}(\overline{J}) = -2 \quad \det(\overline{J}) = 1 - [n\alpha u_s^{n+1}(1-u_s^n)^{-2}]^2$$

Eigen values at centre point are given by.

$$\lambda_{\pm} = \frac{\text{tr}(\overline{J}) \pm \sqrt{\text{tr}(\overline{J})^2 - 4\det(\overline{J})}}{2}$$



$$\lambda_{\pm} = \frac{-2 \pm \sqrt{4 - 4[n^2 \alpha^2 u_s^{n-1} (1-u_s^n)^{-2} + 1]}}{2}$$

$$\Rightarrow \lambda_{\pm} = -1 \pm \sqrt{1 \pm (n \alpha u_s^{n-1} (1-u_s^n)^{-2})^2}$$

$$\Rightarrow \lambda_{\pm} = -1 \pm n \alpha u_s^{n-1} (1-u_s^n)^{-2}$$

When  $n=1$ ,  $\lambda_{\pm} = -1 \pm \alpha u_s^0 (1-u_s)^{-2} = -1 \pm \alpha (1-u_s)^{-2}$

$n=2$ ,  $\lambda_{\pm} = -1 \pm 2 \alpha u_s (1-u_s^2)^{-2}$

For  $u_s \ll 1$ ,  $n=1$  is unstable,  $n=2$  is stable,  $u_s \gg 1$ ;  $n=1$  is stable,  $n=2$  is unstable

For  $u=v=2.5$

3 (d)

At  $n=1$ , Eigen values are  $-1.8163, -0.1837$   
Stable

$n=2$ , Eigen values are  $-5.0816, 3.0816$   
Unstable.

At same conc<sup>n</sup>, increasing cooperativity makes the system unstable.

3 (e) For  $u=v=2.5$

$n=1$ ,  $K_{crit} = 12.25$

$n=2$ ,  $K_{crit} = 2.45$