SBML Model Report

Model name: "Aguda1999_CellCycle"



May 6, 2016

1 General Overview

This is a document in SBML Level 2 Version 1 format. This model was created by Harish Dharuri¹ at April third 2008 at 1:23 p.m. and last time modified at April thirteenth 2015 at 1:56 p.m. Table 1 shows an overview of the quantities of all components of this model.

Table 1: Number of components in this model, which are described in the following sections.

Element	Quantity	Element	Quantity
compartment types	0	compartments	1
species types	0	species	11
events	0	constraints	0
reactions	27	function definitions	3
global parameters	32	unit definitions	0
rules	0	initial assignments	0

Model Notes

The model reproduces the time profiles of p27, E2F and aE/cdk2 as depicted in Figure 5 c of the paper. Model was simulated on MathSBML.

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2 Unit Definitions

This is an overview of five unit definitions which are all predefined by SBML and not mentioned in the model.

2.1 Unit substance

Notes Mole is the predefined SBML unit for substance.

Definition mol

2.2 Unit volume

Notes Litre is the predefined SBML unit for volume.

Definition 1

2.3 Unit area

Notes Square metre is the predefined SBML unit for area since SBML Level 2 Version 1.

Definition m²

2.4 Unit length

Notes Metre is the predefined SBML unit for length since SBML Level 2 Version 1.

Definition m

2.5 Unit time

Notes Second is the predefined SBML unit for time.

Definition s

3 Compartment

This model contains one compartment.

Table 2: Properties of all compartments.

Id	Name	SBO	Spatial Dimensions	Size	Unit	Constant	Outside
cell_1	cell		3	1	litre	Ø	

3.1 Compartment cell_1

This is a three dimensional compartment with a constant size of one litre.

Name cell

4 Species

This model contains eleven species. Section 8 provides further details and the derived rates of change of each species.

Table 3: Properties of each species.

Id	Name	Compartment	Derived Unit	Constant	Boundary Condi- tion
Y3_1	pRB_E2F	cell_1	mol		\Box
Y4_1	E2F	cell_1	mol	\Box	\Box
$Y11_{-}1$	pRB_P	$\mathtt{cell}_{\mathtt{-}}\mathtt{1}$	mol	\Box	\Box
$Y2_{-}1$	i_cyclinE_CDK2	$\mathtt{cell}_{\mathtt{-}}\mathtt{1}$	mol	\Box	\Box
Y1_1	a_cyclinE_CDK2	$\mathtt{cell}_{\mathtt{-}}\mathtt{1}$	mol	\Box	\Box
Y5_1	pRB	$\mathtt{cell}_{\mathtt{-}}\mathtt{1}$	mol	\Box	\Box
Y6_1	CycD_CDK4	cell_1	mol		\Box
Y7_1	p27	cell_1	mol		\Box
Y8_1	cycE_CDK2_p27	cell_1	mol		\Box
Y10_1	p16	cell_1	mol		\Box
Y9_1	cycD_CDK4_p27	\mathtt{cell}_1	mol	\Box	

5 Parameters

This model contains 32 global parameters.

Table 4: Properties of each parameter.

Id	Name	SBO Value Unit	Constant
k1p_1	k1p	0.500	✓
$k1pp_{-}1$	k1pp	0.500	$ \overline{\mathscr{A}} $
$k1_{-}1$	k1	0.100	\square
k2_1	k2	0.100	\square
${\tt kminus2_1}$	kminus2	1.000	
k3_1	k3	1.420	
$k3p_{-}1$	k3p	0.000	
$k4_{-}1$	k4	10^{-6}	
${\tt kminus4_1}$	kminus4	0.016	$ \overline{\mathscr{A}} $
$kminus1_{-}1$	kminus1	0.001	$ \overline{\mathscr{A}} $
$k5_1$	k5	0.020	
$k6_{-}1$	k6	0.018	
${\tt kminus6_1}$	kminus6	5.000	
$k7_{-}1$	k7	10^{-5}	$ \overline{\checkmark} $
k8_1	k8	2.000	
k9_1	k9	2.000	
K10_1	K10	0.035	
$k17_{-}1$	k17	3.500	\square
k18_1	k18	10^{-5}	
k19_1	k19	0.050	
$k20_{-}1$	k20	0.010	$ \overline{\mathscr{A}} $
k21_1	k21	0.100	$ \overline{\mathscr{A}} $
k22_1	k22	0.001	\square
k23_1	k23	0.200	
k24_1	k24	0.100	\square
k25_1	k25	0.010	\square
$k25p_{-}1$	k25p	0.020	
k26_1	k26	0.010	$\overline{\checkmark}$
$k26p_{-}1$	k26p	0.100	$\overline{\checkmark}$
k27_1	k27	0.010	$\overline{\checkmark}$
k28_1	k28	0.010	$\overline{\checkmark}$
k29_1	k29	0.001	$\overline{\checkmark}$

6 Function definitions

This is an overview of three function definitions.

6.1 Function definition Mass_Action_1_1

Name Mass_Action_1

Arguments k1, S1

Mathematical Expression

 $k1 \cdot S1$ (1)

6.2 Function definition Mass_Action_2_1

Name Mass_Action_2

 $\textbf{Arguments} \ k1, S1, S2$

Mathematical Expression

 $k1 \cdot S1 \cdot S2$ (2)

6.3 Function definition Mass_Action_0_1

Name Mass_Action_0

Argument k1

Mathematical Expression

k1 (3)

7 Reactions

This model contains 27 reactions. All reactions are listed in the following table and are subsequently described in detail. If a reaction is affected by a modifier, the identifier of this species is written above the reaction arrow.

Table 5: Overview of all reactions

$N_{\bar{0}}$	Id Name	Reaction Equation	SBO
1	_1 pRB/E2F complex deassociation	$Y3_{-1} \xrightarrow{Y3_{-1}, Y1_{-1}, Y6_{-1}, Y9_{-1}} Y4_{-1} + Y11_{-1}$	
2	pRBE2FcomplexformaRB6E2F complex formation _1	$Y5_{-1} + Y4_{-1} \longrightarrow Y3_{-1}$	
3	cycECDK2activationycE/CDK2 activation	$Y2_{-}1 \longrightarrow Y1_{-}1$	
4	cycECDK2deactivatdyxE/CDK2 deactivation	$Y1_{-}1 \longrightarrow Y2_{-}1$	
5	cycECDK2synthesisicycE/CDK2 synthesis	$\emptyset \xrightarrow{\mathbf{Y4}_{-1}} \mathbf{Y2}_{-1}$	
6	E2Fsynthesis_1 E2F synthesis	$\emptyset \longrightarrow Y4_{-}1$	
7	E2Fdegradation E2F degradation	$Y4_{-}1 \longrightarrow \emptyset$	
8	cycECDK2degradatimyeE/CDK2 degradation	$Y2_{-}1 \longrightarrow \emptyset$	
9	cycDCDK4synthesiseycD/CDK4 synthesis	$\emptyset \longrightarrow Y6_{-}1$	
10	cycDCDK4degradati cyc D/CDK4 degradation	$Y6_{-}1 \longrightarrow \emptyset$	
11	p27synthesis_1 p27 synthesis	$\emptyset \longrightarrow Y7_{-}1$	
12	p27degradationvia pŷ∂⊠egKΩl ation via cycE/CDK2 _1	$Y7_{-}1 \xrightarrow{Y1_{-}1} \emptyset$	

N⁰	Id Name	Reaction Equation	SBO
13	cycECDK2p27complexyfcEdftKadtp27 complex formation	$Y11 + Y71 \longrightarrow Y81$	
14	cycECDK2p27deasso cycFt/CdnK 2/p27 deassociation	$Y8_{-}1 \longrightarrow Y1_{-}1 + Y7_{-}1$	
15	cycDCDK4p16complexyfcDyfGDKydyp16 complex formation	$Y61 + Y101 \longrightarrow \emptyset$	
16	E2FautostimulatioE2F auto stimulation _1	$\emptyset \longrightarrow Y4_{-}1$	
17	cycDCDK4p27complexyfcDyfGDbKoddp27 complex formation	$Y71 + Y61 \longrightarrow Y91$	
18	cycDCDK4p27complexyddakSDKidp17xomplex deassociation	$Y9_{-}1 \longrightarrow Y7_{-}1 + Y6_{-}1$	
19	acycECDK2degradatacycE/CDK2 degradation	$Y1_{-}1 \longrightarrow \emptyset$	
20	p27degradation- p27 degradation	$Y7_{-}1 \longrightarrow \emptyset$	
21	_20 p16 synthesis	$\emptyset \longrightarrow Y10_{-}1$	
22	p16degradation p16 degradation $_{-1}$	$Y10_{-}1 \longrightarrow \emptyset$	
23	p16synthesisinhib pt6dbyptRB is inhibited by pRB_1	$\emptyset \xrightarrow{\mathbf{Y5}_1} \mathbf{Y10}_1$	
24	pRBpdephosphorila pR&ns ynthesis inhibited by p16 _1	$\emptyset \xrightarrow{Y10_1} Y5_1$	
25	pRBsynthesis_1 pRB synthesis	$\emptyset \longrightarrow Y5_{-}1$	
26	pRBdegradation- pRB degradation	$Y5_{-}1 \longrightarrow \emptyset$	
27	pRBpdephosphoryla pRda p dephosphorylation _1	Y11 ₋ 1 → Y5 ₋ 1	

7.1 Reaction _1

This is an irreversible reaction of one reactant forming two products influenced by four modifiers

Name pRB/E2F complex deassociation

Reaction equation

$$Y3_{-}1 \xrightarrow{Y3_{-}1, Y1_{-}1, Y6_{-}1, Y9_{-}1} Y4_{-}1 + Y11_{-}1$$
 (4)

Reactant

Table 6: Properties of each reactant.

Id	Name	SBO
Y3_1	pRB_E2F	

Modifiers

Table 7: Properties of each modifier.

Id	Name	SBO
Y3_1	pRB_E2F	
$Y1_{-}1$	a_cyclinE_CDK2	
$Y6_{-}1$	CycD_CDK4	
$Y9_{-}1$	cycD_CDK4_p27	

Products

Table 8: Properties of each product.

Id	Name	SBO
Y4_1	E2F	
$Y11_{-}1$	pRB_P	

Kinetic Law

Derived unit contains undeclared units

$$v_1 = k1p_{-1} \cdot Y6_{-1} \cdot Y3_{-1} + k1pp_{-1} \cdot Y9_{-1} \cdot Y3_{-1} + k1_{-1} \cdot Y1_{-1} \cdot Y3_{-1}$$
 (5)

7.2 Reaction pRBE2Fcomplexformation_1

This is an irreversible reaction of two reactants forming one product.

Name pRB/E2F complex formation

Reaction equation

$$Y5_{-}1 + Y4_{-}1 \longrightarrow Y3_{-}1 \tag{6}$$

Reactants

Table 9: Properties of each reactant.

Id	Name	SBO
Y5_1		
$Y4_{-}1$	E2F	

Product

Table 10: Properties of each product.

	•	
Id	Name	SBO
Y3_1	pRB_E2F	

Kinetic Law

Derived unit contains undeclared units

$$v_2 = \text{Mass_Action_2_1} (\text{kminus1_1}, \text{Y5_1}, \text{Y4_1})$$
 (7)

Mass_Action_2_1 (k1, S1, S2) =
$$k1 \cdot S1 \cdot S2$$
 (8)

7.3 Reaction cycECDK2activation_1

This is an irreversible reaction of one reactant forming one product.

Name cycE/CDK2 activation

Reaction equation

$$Y2_1 \longrightarrow Y1_1 \tag{9}$$

Reactant

Table 11: Properties of each reactant.

Id	Name	SBO
Y2_1	i_cyclinE_CDK2	

Product

Table 12: Properties of each product.

Id	Name	SBO
Y1_1	a_cyclinE_CDK2	

Kinetic Law

Derived unit contains undeclared units

$$v_3 = k2_1 \cdot Y_{1-1} \cdot Y_{2-1} \tag{10}$$

7.4 Reaction cycECDK2deactivation_1

This is an irreversible reaction of one reactant forming one product.

Name cycE/CDK2 deactivation

Reaction equation

$$Y1_{-}1 \longrightarrow Y2_{-}1$$
 (11)

Reactant

Table 13: Properties of each reactant.

Id	Name	SBO
Y1_1	a_cyclinE_CDK2	

Product

Table 14: Properties of each product.

Id	Name	SBO
Y2_1	i_cyclinE_CDK2	

Kinetic Law

Derived unit contains undeclared units

$$v_4 = \text{kminus} 2 \cdot 1 \cdot Y 1 \cdot 1 \tag{12}$$

7.5 Reaction cycECDK2synthesis_1

This is an irreversible reaction of no reactant forming one product influenced by one modifier.

Name icycE/CDK2 synthesis

Reaction equation

$$\emptyset \xrightarrow{Y4_1} Y2_1 \tag{13}$$

Modifier

Table 15: Properties of each modifier.

Id	Name	SBO
Y4_1	E2F	

Product

Table 16: Properties of each product.

Id	Name	SBO
Y2_1	i_cyclinE_CDK2	

Kinetic Law

Derived unit not available

$$v_5 = k_3 - 1 \cdot Y_4 - 1 + k_3 p_1$$
 (14)

7.6 Reaction E2Fsynthesis_1

This is an irreversible reaction of no reactant forming one product.

Name E2F synthesis

Reaction equation

$$\emptyset \longrightarrow Y4_{-}1 \tag{15}$$

Product

Table 17: Properties of each product.

Id	Name	SBO
Y4_1	E2F	

Kinetic Law

Derived unit not available

$$v_6 = k4_-1$$
 (16)

7.7 Reaction E2Fdegradation_1

This is an irreversible reaction of one reactant forming no product.

Name E2F degradation

Reaction equation

$$Y4_{-}1 \longrightarrow \emptyset \tag{17}$$

Reactant

Table 18: Properties of each reactant.

Id	Name	SBO
Y4_1	E2F	

Kinetic Law

Derived unit contains undeclared units

$$v_7 = \text{Mass_Action_1_1} (\text{kminus4_1}, \text{Y4_1})$$
 (18)

Mass_Action_1_1
$$(k1, S1) = k1 \cdot S1$$
 (19)

7.8 Reaction cycECDK2degradation_1

This is an irreversible reaction of one reactant forming no product.

Name icycE/CDK2 degradation

Reaction equation

$$Y2_{-}1 \longrightarrow \emptyset \tag{20}$$

Reactant

Table 19: Properties of each reactant.

Id	Name	SBO
Y2_1	i_cyclinE_CDK2	

Kinetic Law

Derived unit contains undeclared units

$$v_8 = \text{Mass_Action_1_1} (k5_1, Y2_1)$$
 (21)

$$Mass_Action_1_1(k1,S1) = k1 \cdot S1 \tag{22}$$

7.9 Reaction cycDCDK4synthesis_1

This is an irreversible reaction of no reactant forming one product.

Name cycD/CDK4 synthesis

Reaction equation

$$\emptyset \longrightarrow Y6_{-1} \tag{23}$$

Product

Table 20: Properties of each product.

Id	Name	SBO
Y6_1	CycD_CDK4	

Kinetic Law

Derived unit not available

$$v_9 = \text{Mass_Action_0_1} (k6_1) \tag{24}$$

$$Mass_Action_0_1 (k1) = k1$$
 (25)

7.10 Reaction cycDCDK4degradation_1

This is an irreversible reaction of one reactant forming no product.

Name cycD/CDK4 degradation

Reaction equation

$$Y6_{-1} \longrightarrow \emptyset$$
 (26)

Reactant

Table 21: Properties of each reactant.

Id	Name	SBO
Y6_1	CycD_CDK4	

Kinetic Law

Derived unit contains undeclared units

$$v_{10} = \text{Mass_Action_1_1} (\text{kminus6_1}, \text{Y6_1})$$
 (27)

7.11 Reaction p27synthesis_1

This is an irreversible reaction of no reactant forming one product.

Name p27 synthesis

Reaction equation

$$\emptyset \longrightarrow Y7_{-}1 \tag{29}$$

Product

Table 22: Properties of each product.

Id	Name	SBO
Y7_1	p27	

Kinetic Law

Derived unit not available

$$v_{11} = \text{Mass_Action_0_1} (k7_1)$$
 (30)

$$Mass_Action_0_1 (k1) = k1$$
 (31)

7.12 Reaction p27degradationviacycECDK2_1

This is an irreversible reaction of one reactant forming no product influenced by one modifier.

Name p27 degradation via cycE/CDK2

Reaction equation

$$Y7_{-1} \xrightarrow{Y1_{-1}} \emptyset \tag{32}$$

Reactant

Table 23: Properties of each reactant.

Id	Name	SBO
Y7_1	p27	

Modifier

Table 24: Properties of each modifier.

Id	Name	SBO
Y1_1	a_cyclinE_CDK2	

Kinetic Law

Derived unit contains undeclared units

$$v_{12} = k8_{-1} \cdot Y7_{-1} \cdot Y1_{-1} \tag{33}$$

7.13 Reaction cycECDK2p27complexformation_1

This is an irreversible reaction of two reactants forming one product.

Name cycE/CDK2/p27 complex formation

Reaction equation

$$Y1_{-}1 + Y7_{-}1 \longrightarrow Y8_{-}1 \tag{34}$$

Reactants

Table 25: Properties of each reactant.

Id	Name	SBO
Y1_1	a_cyclinE_CDK2	
Y7_1	p27	

Product

Table 26: Properties of each product.

Id	Name	SBO
Y8_1	cycE_CDK2_p27	

Kinetic Law

Derived unit contains undeclared units

$$v_{13} = \text{Mass_Action_2_1} (k9_1, Y1_1, Y7_1)$$
 (35)

Mass_Action_2_1 (k1, S1, S2) =
$$k1 \cdot S1 \cdot S2$$
 (36)

7.14 Reaction cycECDK2p27deassociation_1

This is an irreversible reaction of one reactant forming two products.

Name cycE/CDK2/p27 deassociation

Reaction equation

$$Y8_{-}1 \longrightarrow Y1_{-}1 + Y7_{-}1 \tag{37}$$

Reactant

Table 27: Properties of each reactant.

Id	Name	SBO
Y8_1	cycE_CDK2_p27	

Products

Table 28: Properties of each product.

Id	Name	SBO
Y1_1	a_cyclinE_CDK2	
Y7_1	p27	

Kinetic Law

Derived unit contains undeclared units

$$v_{14} = \text{Mass_Action_1_1} (K10_1, Y8_1)$$
 (38)

7.15 Reaction cycDCDK4p16complexformation_1

This is an irreversible reaction of two reactants forming no product.

Name cycD/CDK4/p16 complex formation

Reaction equation

$$Y6_{-}1 + Y10_{-}1 \longrightarrow \emptyset \tag{40}$$

Reactants

Table 29: Properties of each reactant.

Id	Name	SBO
Y6_1	CycD_CDK4	
$Y10_{-}1$	p16	

Kinetic Law

Derived unit contains undeclared units

$$v_{15} = \text{Mass_Action_2_1} (k17_1, Y6_1, Y10_1)$$
 (41)

Mass_Action_2_1 (k1, S1, S2) =
$$k1 \cdot S1 \cdot S2$$
 (42)

7.16 Reaction E2Fautostimulation_1

This is an irreversible reaction of no reactant forming one product.

Name E2F auto stimulation

Reaction equation

$$\emptyset \longrightarrow Y4_{-}1 \tag{43}$$

Product

Table 30: Properties of each product.

Id	Name	SBO
Y4_1	E2F	

Kinetic Law

Derived unit contains undeclared units

$$v_{16} = k18_{-}1 \cdot Y4_{-}1 \tag{44}$$

7.17 Reaction cycDCDK4p27complexformation_1

This is an irreversible reaction of two reactants forming one product.

Name cycD/CDK4/p27 complex formation

Reaction equation

$$Y7_{-}1 + Y6_{-}1 \longrightarrow Y9_{-}1 \tag{45}$$

Reactants

Table 31: Properties of each reactant.

Id	Name	SBO
Y7_1	p27	
$Y6_{-}1$	CycD_CDK4	

Product

Table 32: Properties of each product.

Id	Name	SBO
Y9_1	cycD_CDK4_p27	

Kinetic Law

Derived unit contains undeclared units

$$v_{17} = \text{Mass_Action_2_1} (k19_1, Y7_1, Y6_1)$$
 (46)

Mass_Action_2_1 (k1, S1, S2) =
$$k1 \cdot S1 \cdot S2$$
 (47)

7.18 Reaction cycDCDK4p27complexdeassociation_1

This is an irreversible reaction of one reactant forming two products.

Name cycD/CDK4/p27 complex deassociation

Reaction equation

$$Y9_{-}1 \longrightarrow Y7_{-}1 + Y6_{-}1 \tag{48}$$

Reactant

Table 33: Properties of each reactant.

Id	Name	SBO
Y9_1	cycD_CDK4_p27	

Products

Table 34: Properties of each product.

Id	Name	SBO
Y7_1	p27	
$Y6_{-}1$	CycD_CDK4	

Kinetic Law

Derived unit contains undeclared units

$$v_{18} = \text{Mass_Action_1_1} (k20_1, Y9_1)$$
 (49)

7.19 Reaction acycECDK2degradation_1

This is an irreversible reaction of one reactant forming no product.

Name acycE/CDK2 degradation

Reaction equation

$$Y1_{-}1 \longrightarrow \emptyset$$
 (51)

Reactant

Table 35: Properties of each reactant.

Id	Name	SBO
Y1_1	a_cyclinE_CDK2	

Kinetic Law

Derived unit contains undeclared units

$$v_{19} = k21_{-1} \cdot Y1_{-1} \cdot Y1_{-1} \tag{52}$$

7.20 Reaction p27degradation_1

This is an irreversible reaction of one reactant forming no product.

Name p27 degradation

Reaction equation

$$Y7_{-}1 \longrightarrow \emptyset$$
 (53)

Reactant

Table 36: Properties of each reactant.

Id	Name	SBO
Y7_1	p27	

Kinetic Law

Derived unit contains undeclared units

$$v_{20} = \text{Mass_Action_1_1} (k22_1, Y7_1)$$
 (54)

Mass_Action_1_1
$$(k1, S1) = k1 \cdot S1$$
 (55)

7.21 Reaction _20

This is an irreversible reaction of no reactant forming one product.

Name p16 synthesis

Reaction equation

$$\emptyset \longrightarrow Y10_{-1}$$
 (56)

Product

Table 37: Properties of each product.

Id	Name	SBO
Y10_1	p16	

Kinetic Law

Derived unit not available

$$v_{21} = \text{Mass_Action_0_1} (k23_1)$$
 (57)

$$Mass_Action_0_1(k1) = k1$$
 (58)

7.22 Reaction p16degradation_1

This is an irreversible reaction of one reactant forming no product.

Name p16 degradation

Reaction equation

$$Y10_{-}1 \longrightarrow \emptyset$$
 (59)

Reactant

Table 38: Properties of each reactant.

Id	Name	SBO
Y10_1	p16	

Kinetic Law

Derived unit contains undeclared units

$$v_{22} = \text{Mass_Action_1_1} (k24_1, Y10_1)$$
 (60)

7.23 Reaction p16synthesisinhibitedbypRB_1

This is an irreversible reaction of no reactant forming one product influenced by one modifier.

Name p16 synthesis inhibited by pRB

Reaction equation

$$\emptyset \xrightarrow{Y5_1} Y10_1 \tag{62}$$

Modifier

Table 39: Properties of each modifier.

Id	Name	SBO
Y5_1	pRB	

Product

Table 40: Properties of each product.

Id	Name	SBO
Y10_1	p16	

Id	Name	SBO

Kinetic Law

Derived unit contains undeclared units

$$v_{23} = \frac{k25_{-1}}{1 + k25p_{-1} \cdot Y5_{-1}} \tag{63}$$

7.24 Reaction pRBpdephosphorilation_1

This is an irreversible reaction of no reactant forming one product influenced by one modifier.

Name pRB synthesis inhibited by p16

Reaction equation

$$\emptyset \xrightarrow{Y10_1} Y5_1 \tag{64}$$

Modifier

Table 41: Properties of each modifier.

Id	Name	SBO
Y10_1	p16	

Product

Table 42: Properties of each product.

Id	Name	SBO
Y5_1	pRB	

Kinetic Law

Derived unit contains undeclared units

$$v_{24} = \frac{k26_{-1}}{1 + k26p_{-1} \cdot Y10_{-1}} \tag{65}$$

7.25 Reaction pRBsynthesis_1

This is an irreversible reaction of no reactant forming one product.

Name pRB synthesis

Reaction equation

$$\emptyset \longrightarrow Y5_{-1}$$
 (66)

Product

Table 43: Properties of each product.

Id	Name	SBO
Y5_1	pRB	

Kinetic Law

Derived unit not available

$$v_{25} = \text{Mass_Action_0_1} (k27_1)$$
 (67)

$$Mass_Action_0_1 (k1) = k1$$
(68)

7.26 Reaction pRBdegradation_1

This is an irreversible reaction of one reactant forming no product.

Name pRB degradation

Reaction equation

$$Y5_{-}1 \longrightarrow \emptyset \tag{69}$$

Reactant

Table 44: Properties of each reactant.

Id	Name	SBO
Y5_1	pRB	

Kinetic Law

Derived unit contains undeclared units

$$v_{26} = \text{Mass_Action_1_1} (k28_1, Y5_1)$$
 (70)

7.27 Reaction pRBpdephosphorylation_1

This is an irreversible reaction of one reactant forming one product.

Name pRB-p dephosphorylation

Reaction equation

$$Y11_{-}1 \longrightarrow Y5_{-}1$$
 (72)

Reactant

Table 45: Properties of each reactant.

Id	Name	SBO
Y11_1	pRB_P	

Product

Table 46: Properties of each product.

Id	Name	SBO
Y5_1	pRB	

Kinetic Law

Derived unit contains undeclared units

$$v_{27} = \text{Mass_Action_1_1} (k29_1, Y11_1)$$
 (73)

$$Mass_Action_1_1 (k1, S1) = k1 \cdot S1$$

$$(74)$$

8 Derived Rate Equations

When interpreted as an ordinary differential equation framework, this model implies the following set of equations for the rates of change of each species.

Identifiers for kinetic laws highlighted in gray cannot be verified to evaluate to units of SBML substance per time. As a result, some SBML interpreters may not be able to verify the consistency of the units on quantities in the model. Please check if

• parameters without an unit definition are involved or

• volume correction is necessary because the hasOnlySubstanceUnits flag may be set to false and spacialDimensions > 0 for certain species.

8.1 Species Y3_1

Name pRB_E2F

Initial amount 1.95 mol

This species takes part in three reactions (as a reactant in _1 and as a product in pRBE2Fcomplexformation_1 and as a modifier in _1).

$$\frac{d}{dt}Y3_{-1} = v_2 - v_1 \tag{75}$$

8.2 Species Y4_1

Name E2F

Initial amount 0 mol

This species takes part in six reactions (as a reactant in pRBE2Fcomplexformation_1, E2Fdegradation_1 and as a product in _1, E2Fsynthesis_1, E2Fautostimulation_1 and as a modifier in cycECDK2synthesis_1).

$$\frac{\mathrm{d}}{\mathrm{d}t}Y4_{-}1 = v_1 + v_6 + v_{16} - v_2 - v_7 \tag{76}$$

8.3 Species Y11_1

Name pRB_P

Initial amount 0.01 mol

This species takes part in two reactions (as a reactant in pRBpdephosphorylation_1 and as a product in _1).

$$\frac{d}{dt}Y11_{-}1 = |v_1| - |v_{27}| \tag{77}$$

8.4 Species Y2_1

Name i_cyclinE_CDK2

Initial amount 0.01 mol

This species takes part in four reactions (as a reactant in cycECDK2activation_1, cycECDK2degradation_1 and as a product in cycECDK2deactivation_1, cycECDK2synthesis_1).

$$\frac{\mathrm{d}}{\mathrm{d}t}Y2_{-}1 = |v_4| + |v_5| - |v_3| - |v_8| \tag{78}$$

8.5 Species Y1_1

Name a_cyclinE_CDK2

Initial amount 0 mol

This species takes part in seven reactions (as a reactant in cycECDK2deactivation_1, cycECDK2p27complexforma_1, acycECDK2degradation_1 and as a product in cycECDK2activation_1, cycECDK2p27deassociation_1 and as a modifier in _1, p27degradationviacycECDK2_1).

$$\frac{\mathrm{d}}{\mathrm{d}t} Y 1_{-}1 = |v_3| + |v_{14}| - |v_4| - |v_{13}| - |v_{19}| \tag{79}$$

8.6 Species Y5_1

Name pRB

Initial amount 0.05 mol

This species takes part in six reactions (as a reactant in pRBE2Fcomplexformation_1, pRBdegradation_1 and as a product in pRBpdephosphorilation_1, pRBsynthesis_1, pRBpdephosphorylation_1 and as a modifier in p16synthesisinhibitedbypRB_1).

$$\frac{\mathrm{d}}{\mathrm{d}t}Y5_{-1} = v_{24} + v_{25} + v_{27} - v_2 - v_{26} \tag{80}$$

8.7 Species Y6_1

Name CycD_CDK4

Initial amount 0 mol

This species takes part in six reactions (as a reactant in cycDCDK4degradation_1, cycDCDK4p16complexformation_1, cycDCDK4p27complexformation_1 and as a product in cycDCDK4synthesis_1, cycDCDK4p27complexdeass_1 and as a modifier in _1).

$$\frac{\mathrm{d}}{\mathrm{d}t} Y 6_{-1} = v_9 + v_{18} - v_{10} - v_{15} - v_{17} \tag{81}$$

8.8 Species Y7_1

Name p27

Initial amount 15 mol

This species takes part in seven reactions (as a reactant in p27degradationviacycECDK2_1, cycECDK2p27complexformation_1, cycDCDK4p27complexformation_1, p27degradation_1 and as a product in p27synthesis_1, cycECDK2p27deassociation_1, cycDCDK4p27complexdeassociation_1).

$$\frac{\mathrm{d}}{\mathrm{d}t}Y7_{-}1 = |v_{11}| + |v_{14}| + |v_{18}| - |v_{12}| - |v_{13}| - |v_{17}| - |v_{20}|$$
(82)

8.9 Species Y8_1

Name cycE_CDK2_p27

Initial amount 1 mol

This species takes part in two reactions (as a reactant in cycECDK2p27deassociation_1 and as a product in cycECDK2p27complexformation_1).

$$\frac{d}{dt}Y8_{-1} = v_{13} - v_{14} \tag{83}$$

8.10 Species Y10_1

Name p16

Initial amount 5 mol

This species takes part in five reactions (as a reactant in cycDCDK4p16complexformation_1, p16degradation_1 and as a product in _20, p16synthesisinhibitedbypRB_1 and as a modifier in pRBpdephosphorilation_1).

$$\frac{\mathrm{d}}{\mathrm{d}t}Y10_{-}1 = v_{21} + v_{23} - v_{15} - v_{22} \tag{84}$$

8.11 Species Y9_1

Name cycD_CDK4_p27

Initial amount 0 mol

This species takes part in three reactions (as a reactant in cycDCDK4p27complexdeassociation—1 and as a product in cycDCDK4p27complexformation_1 and as a modifier in _1).

$$\frac{d}{dt}Y9_{-1} = v_{17} - v_{18} \tag{85}$$

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