

## SBML Model Report

### Model name: “Thomsen1988- \_AdenylateCyclase\_Inhibition”



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## 1 General Overview

This is a document in SBML Level 2 Version 1 format. This model was created by Enuo He<sup>1</sup> at September 26<sup>th</sup> 2006 at 6:47 p. m. and last time modified at February thirteenth 2014 at 3:49 p. m. Table 1 shows an overview of the quantities of all components of this model.

Table 1: Number of components in this model, which are described in the following sections.

Element	Quantity	Element	Quantity
compartment types	0	compartments	1
species types	0	species	10
events	0	constraints	0
reactions	6	function definitions	0
global parameters	0	unit definitions	0
rules	0	initial assignments	0

## Model Notes

This model was created according to the paper *Inhibition of Adenylate Cyclase Is Mediated by the High Affinity Conformation of the alpha2-Adrenergic Receptor* published in 1988.

The figure4 (steady state curve) in the paper has been simulated having the same plot with Copasi 4.0.19 (development) and roadRunner(online).Because the initial concentration of R and D were not given in the paper ,so we gave it 1e-9 Mol/L and 1e-8 Mol/L respectively.

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Pay attention that the simulations of steady state concentration of species in arbitrary units are shown for figure4 and figure6 in the paper.

## 2 Unit Definitions

This is an overview of five unit definitions which are all predefined by SBML and not mentioned in the model.

### 2.1 Unit substance

**Notes** Mole is the predefined SBML unit for substance.

**Definition** mol

### 2.2 Unit volume

**Notes** Litre is the predefined SBML unit for volume.

**Definition** l

### 2.3 Unit area

**Notes** Square metre is the predefined SBML unit for area since SBML Level 2 Version 1.

**Definition** m<sup>2</sup>

### 2.4 Unit length

**Notes** Metre is the predefined SBML unit for length since SBML Level 2 Version 1.

**Definition** m

### 2.5 Unit time

**Notes** Second is the predefined SBML unit for time.

**Definition** s

## 3 Compartment

This model contains one compartment.

Table 2: Properties of all compartments.

Id	Name	SBO	Spatial Dimensions	Size	Unit	Constant	Outside
cell	cell		3	1	litre	<input checked="" type="checkbox"/>	

### 3.1 Compartment `cell`

This is a three dimensional compartment with a constant size of one litre.

**Name** `cell`

## 4 Species

This model contains ten species. Section 6 provides further details and the derived rates of change of each species.

Table 3: Properties of each species.

Id	Name	Compartment	Derived Unit	Constant	Boundary Condition
agonist	D	cell	$\text{mol} \cdot \text{l}^{-1}$	$\square$	$\square$
DR	DR	cell	$\text{mol} \cdot \text{l}^{-1}$	$\square$	$\square$
DRG_GDP	DRG_GDP	cell	$\text{mol} \cdot \text{l}^{-1}$	$\square$	$\square$
DRG	DRG	cell	$\text{mol} \cdot \text{l}^{-1}$	$\square$	$\square$
GDP	GDP	cell	$\text{mol} \cdot \text{l}^{-1}$	$\square$	$\square$
DRG_GTP	DRG_GTP	cell	$\text{mol} \cdot \text{l}^{-1}$	$\square$	$\square$
GTP	GTP	cell	$\text{mol} \cdot \text{l}^{-1}$	$\square$	$\square$
Recptor	R	cell	$\text{mol} \cdot \text{l}^{-1}$	$\square$	$\square$
G_GDP	G_GDP	cell	$\text{mol} \cdot \text{l}^{-1}$	$\square$	$\square$
G_GTP	G_GTP	cell	$\text{mol} \cdot \text{l}^{-1}$	$\square$	$\square$

## 5 Reactions

This model contains six reactions. All reactions are listed in the following table and are subsequently described in detail. If a reaction is affected by a modifier, the identifier of this species is written above the reaction arrow.

Table 4: Overview of all reactions

Nº	Id	Name	Reaction Equation	SBO
1	Forming	Forming DR	$\text{agonist} + \text{Recptor} \rightleftharpoons \text{DR}$	
2	reaction_1	DR binding with G_GDP produc DRG_GDP	$\text{DR} + \text{G\_GDP} \rightleftharpoons \text{DRG\_GDP}$	
3	reaction_2	GDP releasing	$\text{DRG\_GDP} \rightleftharpoons \text{GDP} + \text{DRG}$	
4	reaction_3	GTP binding with DRG	$\text{DRG} + \text{GTP} \rightleftharpoons \text{DRG\_GTP}$	
5	reaction_4	DRG_GTP dissociates(G protein activation)	$\text{DRG\_GTP} \longrightarrow \text{G\_GTP} + \text{DR}$	
6	reaction_5	G protein inactivated	$\text{G\_GTP} \longrightarrow \text{G\_GDP}$	

## 5.1 Reaction Forming

This is a reversible reaction of two reactants forming one product.

**Name** Forming DR

### Reaction equation



### Reactants

Table 5: Properties of each reactant.

Id	Name	SBO
agonist	D	
Recptor	R	

### Product

Table 6: Properties of each product.

Id	Name	SBO
DR	DR	

### Kinetic Law

**Derived unit** contains undeclared units

$$v_1 = \text{vol}(\text{cell}) \cdot (k_1 \cdot [\text{agonist}] \cdot [\text{Recptor}] - k_7 \cdot [\text{DR}]) \quad (2)$$

Table 7: Properties of each parameter.

Id	Name	SBO	Value	Unit	Constant
k1			5000000.0		<input checked="" type="checkbox"/>
k7	k7		0.5		<input checked="" type="checkbox"/>

## 5.2 Reaction reaction\_1

This is a reversible reaction of two reactants forming one product.

**Name** DR binding with G\_GDP produc DRG\_GDP

### Reaction equation



### Reactants

Table 8: Properties of each reactant.

Id	Name	SBO
DR	DR	
G_GDP	G_GDP	

### Product

Table 9: Properties of each product.

Id	Name	SBO
DRG_GDP	DRG_GDP	

### Kinetic Law

**Derived unit** contains undeclared units

$$v_2 = \text{vol}(\text{cell}) \cdot (k_2 \cdot [\text{DR}] \cdot [\text{G\_GDP}] - k_8 \cdot [\text{DRG\_GDP}]) \quad (4)$$

Table 10: Properties of each parameter.

Id	Name	SBO	Value	Unit	Constant
k2			10 <sup>8</sup>		<input checked="" type="checkbox"/>
k8	k8		0.100		<input checked="" type="checkbox"/>

## 5.3 Reaction [reaction\\_2](#)

This is a reversible reaction of one reactant forming two products.

**Name** GDP releasing

### Reaction equation



## Reactant

Table 11: Properties of each reactant.

Id	Name	SBO
DRG_GDP	DRG_GDP	

## Products

Table 12: Properties of each product.

Id	Name	SBO
GDP	GDP	
DRG	DRG	

## Kinetic Law

**Derived unit** contains undeclared units

$$v_3 = \text{vol}(\text{cell}) \cdot (k_3 \cdot [\text{DRG\_GDP}] - k_9 \cdot [\text{DRG}] \cdot [\text{GDP}]) \quad (6)$$

Table 13: Properties of each parameter.

Id	Name	SBO	Value	Unit	Constant
k3			0.1		<input checked="" type="checkbox"/>
k9	k9		100000.0		<input checked="" type="checkbox"/>

### 5.4 Reaction `reaction_3`

This is a reversible reaction of two reactants forming one product.

**Name** GTP binding with DRG

#### Reaction equation



## Reactants



Table 14: Properties of each reactant.

Id	Name	SBO
DRG	DRG	
GTP	GTP	

## Product

Table 15: Properties of each product.

Id	Name	SBO
DRG_GTP	DRG_GTP	

## Kinetic Law

**Derived unit** contains undeclared units

$$v_4 = \text{vol}(\text{cell}) \cdot (k_4 \cdot [\text{DRG}] \cdot [\text{GTP}] - k_{10} \cdot [\text{DRG\_GTP}]) \quad (8)$$

Table 16: Properties of each parameter.

Id	Name	SBO	Value	Unit	Constant
k4			$10^7$		<input checked="" type="checkbox"/>
k10	k10		0.100		<input checked="" type="checkbox"/>

## 5.5 Reaction `reaction_4`

This is an irreversible reaction of one reactant forming two products.

**Name** DRG\_GTP dissociates(G protein activation)

## Reaction equation



## Reactant

Table 17: Properties of each reactant.

Id	Name	SBO
DRG_GTP	DRG_GTP	

## Products

Table 18: Properties of each product.

Id	Name	SBO
G_GTP	G_GTP	
DR	DR	

## Kinetic Law

**Derived unit** contains undeclared units

$$v_5 = \text{vol}(\text{cell}) \cdot k_5 \cdot [\text{DRG\_GTP}] \quad (10)$$

Table 19: Properties of each parameter.

Id	Name	SBO	Value	Unit	Constant
k5			0.05		<input checked="" type="checkbox"/>

## 5.6 Reaction `reaction_5`

This is an irreversible reaction of one reactant forming one product.

**Name** G protein inactivated

## Reaction equation



## Reactant

Table 20: Properties of each reactant.

Id	Name	SBO
G_GTP	G_GTP	

## Product

Table 21: Properties of each product.

Id	Name	SBO
G_GDP	G_GDP	

### Kinetic Law

**Derived unit** contains undeclared units

$$v_6 = \text{vol}(\text{cell}) \cdot k_6 \cdot [\text{G\_GTP}] \quad (12)$$

Table 22: Properties of each parameter.

Id	Name	SBO	Value	Unit	Constant
k6			0.1		<input checked="" type="checkbox"/>

## 6 Derived Rate Equations

When interpreted as an ordinary differential equation framework, this model implies the following set of equations for the rates of change of each species.

Identifiers for kinetic laws highlighted in gray cannot be verified to evaluate to units of SBML substance per time. As a result, some SBML interpreters may not be able to verify the consistency of the units on quantities in the model. Please check if

- parameters without an unit definition are involved or
- volume correction is necessary because the `hasOnlySubstanceUnits` flag may be set to `false` and `spacialDimensions`  $> 0$  for certain species.

### 6.1 Species `agonist`

**Name** D

**Initial concentration**  $10^{-8} \text{ mol} \cdot \text{l}^{-1}$

This species takes part in one reaction (as a reactant in `Forming`).

$$\frac{d}{dt} \text{agonist} = -v_1 \quad (13)$$

### 6.2 Species `DR`

**Name** DR

**Initial concentration**  $0 \text{ mol} \cdot \text{l}^{-1}$

This species takes part in three reactions (as a reactant in [reaction\\_1](#) and as a product in [Forming, reaction\\_4](#)).

$$\frac{d}{dt}\text{DR} = v_1 + v_5 - v_2 \quad (14)$$

### 6.3 Species [DRG\\_GDP](#)

**Name** DRG\_GDP

**Initial concentration**  $0 \text{ mol} \cdot \text{l}^{-1}$

This species takes part in two reactions (as a reactant in [reaction\\_2](#) and as a product in [reaction\\_1](#)).

$$\frac{d}{dt}\text{DRG\_GDP} = v_2 - v_3 \quad (15)$$

### 6.4 Species [DRG](#)

**Name** DRG

**Initial concentration**  $0 \text{ mol} \cdot \text{l}^{-1}$

This species takes part in two reactions (as a reactant in [reaction\\_3](#) and as a product in [reaction\\_2](#)).

$$\frac{d}{dt}\text{DRG} = v_3 - v_4 \quad (16)$$

### 6.5 Species [GDP](#)

**Name** GDP

**Initial concentration**  $0 \text{ mol} \cdot \text{l}^{-1}$

This species takes part in one reaction (as a product in [reaction\\_2](#)).

$$\frac{d}{dt}\text{GDP} = v_3 \quad (17)$$

### 6.6 Species [DRG\\_GTP](#)

**Name** DRG\_GTP

**Initial concentration**  $0 \text{ mol} \cdot \text{l}^{-1}$

This species takes part in two reactions (as a reactant in [reaction\\_4](#) and as a product in [reaction\\_3](#)).

$$\frac{d}{dt}\text{DRG\_GTP} = v_4 - v_5 \quad (18)$$

## 6.7 Species GTP

**Name** GTP

**Initial concentration**  $10^{-5} \text{ mol} \cdot \text{l}^{-1}$

This species takes part in one reaction (as a reactant in [reaction\\_3](#)).

$$\frac{d}{dt} \text{GTP} = -v_4 \quad (19)$$

## 6.8 Species Recptor

**Name** R

**Initial concentration**  $10^{-9} \text{ mol} \cdot \text{l}^{-1}$

This species takes part in one reaction (as a reactant in [Forming](#)).

$$\frac{d}{dt} \text{Recptor} = -v_1 \quad (20)$$

## 6.9 Species G\_GDP

**Name** G\_GDP

**Initial concentration**  $10^{-9} \text{ mol} \cdot \text{l}^{-1}$

This species takes part in two reactions (as a reactant in [reaction\\_1](#) and as a product in [reaction\\_5](#)).

$$\frac{d}{dt} \text{G\_GDP} = v_6 - v_2 \quad (21)$$

## 6.10 Species G\_GTP

**Name** G\_GTP

**Initial concentration**  $0 \text{ mol} \cdot \text{l}^{-1}$

This species takes part in two reactions (as a reactant in [reaction\\_5](#) and as a product in [reaction\\_4](#)).

$$\frac{d}{dt} \text{G\_GTP} = v_5 - v_6 \quad (22)$$

SBML<sup>2</sup>TeX was developed by Andreas Dräger<sup>a</sup>, Hannes Planatscher<sup>a</sup>, Dieudonné M Wouamba<sup>a</sup>, Adrian Schröder<sup>a</sup>, Michael Hucka<sup>b</sup>, Lukas Endler<sup>c</sup>, Martin Golebiewski<sup>d</sup> and Andreas Zell<sup>a</sup>. Please see <http://www.ra.cs.uni-tuebingen.de/software/SBML2LaTeX> for more information.

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