## The Physics of Energy, Explained Simply & The Physics of Energy For Beginners

## **Thermal Energy Questions**

Equation:  $U = mc\Delta T$ 

U = Thermal Energy (J), m = Mass (kg),

c = Specific Heat Capacity (J/kg°C), ΔT = Temperature Change (°C)

1. 
$$U = 2 \text{ kg} \times 500 \text{ J/kg}^{\circ}\text{C} \times (50-20)^{\circ}\text{C} = 20000 \text{ J}$$

2. 
$$U = 1.5 \text{ kg} \times 4200 \text{ J/kg}^{\circ}\text{C} \times 30^{\circ}\text{C} = 189000.0 \text{ J}$$

3. 
$$U = 3 \text{ kg} \times 450 \text{ J/kg}^{\circ}\text{C} \times 30^{\circ}\text{C} = 40500 \text{ J}$$

4. 
$$U = 0.5 \text{ kg} \times 900 \text{ J/kg}^{\circ}\text{C} \times 15^{\circ}\text{C} = 6750.0 \text{ J}$$

5. 
$$U = 4 \text{ kg} \times 385 \text{ J/kg}^{\circ}\text{C} \times 25^{\circ}\text{C} = 38500 \text{ J}$$

6. 
$$U = 2.5 \text{ kg} \times 800 \text{ J/kg}^{\circ}\text{C} \times 12^{\circ}\text{C} = 24000.0 \text{ J}$$

7. 
$$U = 5 \text{ kg} \times 2000 \text{ J/kg}^{\circ}\text{C} \times 8^{\circ}\text{C} = 80000 \text{ J}$$

8. 
$$U = 1 \text{ kg} \times 1800 \text{ J/kg}^{\circ}\text{C} \times (30-10)^{\circ}\text{C} = 36000 \text{ J}$$

9. 
$$U = 0.8 \text{ kg} \times 380 \text{ J/kg}^{\circ}\text{C} \times 18^{\circ}\text{C} = 5472.0 \text{ J}$$

10. 
$$U = 6 \text{ kg} \times 790 \text{ J/kg}^{\circ}\text{C} \times 5^{\circ}\text{C} = 23700 \text{ J}$$

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