

# A universal standard archive file for adsorption data

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## Abstract

The rise in interest of new advanced adsorbents for application in promising areas has followed a lack of detailed and open reporting of experimental or even computational studies. This report details the specification of a new standard adsorption information file (AIF) inspired by the ubiquitous crystallographic information file (CIF) and based on the self-defining text archive and retrieval (STAR) procedure, also used to represent biological nuclear magnetic resonance experiments (NMR-STAR). The AIF is a flexible, general and easily extended free-format archive file and is readily human and machine readable — simple to edit using a basic text editor or parse for database curation. This format represents the first steps to an open adsorption data format. An open format sorely needed for the electronic transmission of adsorption data between laboratories, journals and larger databases in the effort to increase open science in the field of porous materials.

## Introduction

It is important in many branches of science for there to exist a uniform and flexible method of archiving and exchanging data. The data presented in research contributions must be readily accessible and easily verifiable. Key to scientific discovery and innovation the use of good data management practices can provide significant knowledge through the reuse of original data by the community after the data publication process.<sup>1</sup> Unfortunately, the existing ecosystem surrounding scholarly data publication often prevent the community extracting maximum benefit from published research. This lead to the foundational principles discussed by Wilkinson et al.— Findability, Accessibility, Interoperability, and Reusability— that serve as a guideline to to maximize the added-value gained by digital publishing.<sup>1</sup>

Porous solids are of important scientific and industrial interest owing to their impressive ability to interact with all manner of atoms, ions and molecules throughout the entirety of the solid, not only the surface. This has led to many exciting applications of , traditional applications of porous materials thus involve ion exchange, adsorption (for separation) and catalysis, and many of these benefit from the high order that can be achieved in solids such as zeolites.

Measuring the interaction of gas with a material is crucial to understanding and comparing the plethora of different porous materials. The quantity of gas adsorbed is measured in any convenient units, but for the presentation of the data, the International Union of Pure and Applied Chemistry (IUPAC) has recommended that the amount adsorbed should be expressed in moles per gram of outgassed adsorbent. Moreover, if possible, the composition of the adsorbent should be specified and its surface characterized. To facilitate the comparison of adsorption data, it is recommended that adsorption isotherms are displayed in a graphical form with the amount adsorbed (preferably presented using  $\text{mol g}^{-1}$ ) plotted against the equilibrium relative pressure ( $p/p_0$ ), where  $p_0$  is the saturation pressure of the pure adsorptive at the operational temperature, or against absolute pressure, when the temperature is above the critical temperature of the adsorptive. If the adsorption measurements

are made under conditions where the gas phase deviates appreciably from ideality (e.g., at high pressure), it is desirable that the isotherms should be presented in terms of gas fugacity rather than pressure.

Rapid advances in computer simulations, coupled with the expansion of fruitful local, national and international networks, have fueled the need for a standardization of adsorption data. The variety and relative inflexibility of existing adsorption data formats can inhibit the effective use of reported data. A general, flexible, rapidly extensible and universal file format protocol is essential to further accelerate the development of porous materials. As with other standard formats, this must be machine-independent and portable so that accessibility to data items is independent of their point of origin. It must allow new data items to be incorporated without the need to modify existing files.

## **AIF structure and syntax**

The information contained within an AIF is arranged in a strict structural arrangement. Alternative data storage structures were considered, during the development of the AIF, such as the comma-separated-value (CSV) file. The key, however, is that a universal data format must also include a well-structured record of metadata associated with the adsorption experiment. Without the metadata of, containing at the very least, what gas or temperature was used during the experiment the resulting adsorption data is meaningless. Metadata is also important in the development of databases. This is crucial to identify identical or related data, understand the evolution of databases over time, and also permit further examination for specific data items. As a result the STAR data structure was adopted, which is famously used in the crystallographic information file (CIF) and biological nuclear magnetic resonance experiments (NMR-STAR). An important property of the STAR format is that its syntax is defined by a few simple rules using an object/relational model. The data stored in a STAR file is self-descriptive as the underpinning data structure consists of tag-value pairs or data

item. This data item is the combination a data value with the data name, an identifying unique tag. This approach is flexible and able to represent. nested and repetitive data, which are particularly important for the collection of pressure and adsorbed amount that comprise an adsorption isotherm experiment.

The data structure of an AIF is comprised of comment lines and data blocks, which contains the required metadata and quantities related to the adsorption experiment, separated for adsorption and desorption (Figure 1). Files can contain any number of data blocks to represent multiple experiments. Fundamentally, a data block is a data cell containing a sequence of data items, data loops. It starts with a data block code statement (`data_example`) where a block code is a unique identifying code with in a file. A data block is closed by another either another data block code statement or the end of file. Comment lines can be placed anywhere in the file and are identified by a hash (#) symbol. As these comment lines do not necessarily represent structured data, which is able to be easily parsed, it is not recommended to solely include important experimental conditions in this way. Single values or strings can be included within a data block using the data name and data value pairs. Data values are preceded by an identifying data name that is identify by an underscore character, for example `_temperature`. A data loop structure consists of a loop statement (`loop_`) followed by a list of data names and then a list of data, each of which contains data values that matches the data names in the list of data names. Though data loops can be nested at any level this is not necessary for the current implementation. Importantly, data values that follow the nested data declarations must be in exact multiples of the number of data names (which are referred to as loop packets). The data loops, as used here, are terminated by a new data item, a new data loop or an end of file. These data loops can thus straightforwardly represent the data recorded at each point of an adsorption and desorption experiment, for example the pressure and quantities adsorbed.

The above data structures of the STAR format prove ideal for describing the detailed quantities relating to an adsorption experiment. An entire experiment can be described

```

nk_DUT-6_LP_N2_114pkt.aif
1 # generated using raw2aif comment
2 data_raw2aiftest
3 _exptl_operator Nicole experimental
4 _exptl_date '2009-09-17 00:00:00'
5 _exptl_instrument Autosorb
6 _exptl_adsorptive Nitrogen
7 _exptl_temperature 77.3
8 _exptl_sample_mass 0.0339
9 _sample_id nk_DUT-6_LP_N2_114pkt sample
10 _sample_material_id dut-6
11 loop_ adsorption
12 _adsorp_pressure
13 _adsorp_p0
14 _adsorp_amount
15 0.269367243408 101860.98004799998 0.006484305926579284
16 0.27275763204 101860.98004799998 0.008217577878132952
17 0.273485572344 101860.98004799998 0.009928476539499517
18 loop_ desorption
19 _desorp_pressure
20 _desorp_p0
21 _desorp_amount
22 97113.50270639999 101860.98004799998 57.919842028778376
23 90734.559156 101860.98004799998 57.83008565285129
24 85586.58132479999 101860.98004799998 57.74993663094573

```

Figure 1: The structure of a typical adsorption information file (AIF) with the different data areas representing metadata (experimental and sample information) and the adsorption and desorption data.

within a single data block. Initially, data name and data value pairs can be used to identify quantities relating to the whole experiment such as the date and time, temperature and adsorptive. The data points for the adsorption and desorption branch can then be detailed in separate data loops. For each point, quantities such as the absolute pressure, relative pressure and amount adsorbed can be detailed. This leads us to the general data structure for the AIF (Figure 1).

## AIF dictionary

Each data item within a data block is identified using a unique data name. The currently accepted AIF data names are listed and defined in Table 1. These data items form a first version of a core dictionary, which are intended for primarily reporting primary physisorption data for gas on porous materials. As a result the dictionary contains information about crucial experimental parameters, basic sample characterization, adsorption and desorption measures, each distinguished by a data name prefix (`_exptl_`, `_sample_`, `_adsorp_` and `_desorp_`).

The goal of the experimental data names is to capture the information required to ac-

Table 1: Core dictionary of data items

data name	description
<code>_exptl_operator</code>	name of the person who ran the experiment (string)
<code>_exptl_date</code>	date and time of the experiment (datetime string representation)
<code>_exptl_instrument</code>	instrument id used for the experiment (string)
<code>_exptl_adsorptive</code>	name of the adsorptive (string)
<code>_exptl_temperature</code>	temperature of the experiment (float)
<code>_exptl_sample_mass</code>	mass of the sample (float, gram)
<code>_sample_id</code>	unique identifying code used by the operator (string)
<code>_sample_material_name</code>	designated name for the material (string)
<code>_adsorp_pressure</code>	pressure of the adsorption measurement (float, pascal)
<code>_adsorp_p0</code>	saturation pressure of the adsorption measurement (float, pascal)
<code>_adsorp_amount</code>	amount adsorbed during the adsorption measurement (float, mol kg <sup>-1</sup> )
<code>_desorp_pressure</code>	pressure of the desorption measurement (float, pascal)
<code>_desorp_p0</code>	saturation pressure of the desorption measurement (float, pascal)
<code>_desorp_amount</code>	amount adsorbed during the desorption measurement (float, mol kg <sup>-1</sup> )

curately portray the results of experiment, which at the very minimum are the temperature and adsorptive. As metadata should include information that may aid in identifying possible errors this also includes the mass of sample used in the experiment. Additionally, data is included relative to the date, time and person responsible for the measurement to produce a general picture of the provenance of the experiment.

Details of the sample are recorded for comparison and to uniquely identify the sample that was used in the experiment. Simply, this includes the sample identity code (`_sample_id`), a laboratory code used to record and represent the sample and a string relating to the general designation given to the material (`_sample_material_name`).

For each data point measured during an adsorption experiments data can be stored that relates to the equilibrated pressure, the saturation pressure and the amount adsorbed. Importantly, these should be stored in units of pascal and mol kg<sup>-1</sup> to follow the IUPAC recommendations. The saturation pressure can be recorded at each data point as we have discovered that some instruments will calculate and report this value for every point.

The above features serve as an example of the data that can be archived in a AIF and can be extended in future versions. For example, the intention of this initial version is to provide digital archive of the presented primary data of an adsorption experiment, the isotherm. As a result, it describes only basic information that usually exported from adsorption instruments and a few convenient metadata items. The flexibility of this approach can be extended in future versions to include more detailed parameters not commonly presented, such as dead volume measurements, or combined with CIF data names to archive in situ measurements.

## Creating and using the AIF

A simple python 3.7 routine was produced during the development of the AIF to convert the plain text data output from Quantachrome and Belsorb instruments, used in our lab and many others. Specifically, this has been written to parse raw analysis text files exported by Quantachrome software and the raw data files (.DAT) and csv files exported by Belsorb software. At this moment we are unable to read binary data files that are saved by Quatachrome instruments (.qps).

This simple program will parse the output data to find data items detailed in the AIF dictionary. These data items are converted to the appropriate units and the isotherm is split into the adsorption and desorption parts. Subsequently an AIF is produced using the pymatgen CIF routines. This demonstrated that routines and methods developed for treating CIF files can be used in this application to adsorption data. Furthermore, produced AIF files can also be read using existing CIF methods, such as the `gemmi` python module, which we have used to demonstrate how an isotherm can be displayed from an AIF using a few lines. This is a very simple implementation and it is hoped that future versions, following the cooperation of several manufactures, could result in to routines to parse the information directly from binary data used by the instruments.

To illustrate the importance of a universal data format for adsorption data we can com-

147 pare data contained within an AIF to data digitized from a figure in a manuscript, for  
 148 the same experiment. The porous material DUT-6 (also referred to as MOF-206) is a  
 149 metal-organic framework produced from zinc nitrate and two different different ligands; 2,6-  
 150 naphthalenedicarboxylic acid and benzene-1,3,5-tribenzoic acid.<sup>2</sup> DUT-6 has mesopores with  
 151 dodecahedral geometry and smaller cages that produce an extremely porous material with  
 152 large adsorption capacities for *n*-butane (up to 1.1 g g<sup>-1</sup> at 293 K), methane (230 mg g<sup>-1</sup> at  
 153 100 bar and 298 K) and hydrogen (60 mg g<sup>-1</sup> at 50 bar and 77 K). The nitrogen adsorption  
 154 isotherm at 77 K was reported in the original manuscript using relative pressure and a linear  
 155 axis (Figure /reffgr:example). From this plot the isotherm can be classified as type Ib clearly  
 156 exhibiting characteristics of a hierarchal pore system with micropore and mesopores. Clearly  
 157 there is a distinct step  $p/p_0 = 0.17$ , however, from this plot it is not possible to correctly  
 158 identify low pressure features. The low pressure regions is important for understanding the  
 159 process of initial pore filling or for independent application of analysis methods, such as  
 160 the Brunauer–Emmett–Teller (BET) method. The reported isotherm was digitized using  
 161 WebPlotDigitizer, which uses a time-consuming but powerful point-picking methodology.<sup>3</sup>  
 162 In this methodology each point in the graph is identified by clicking on the image area and  
 163 the point’s location in on-screen pixels is then converted to data valued based on an axes  
 164 calibration. The digitized isotherm and the data from the experiment, when plotted on  
 165 semi-log axes, clearly shows important low pressure information is lost when this informa-  
 166 tion is simply included as a figure in a manuscript (Figure /reffgr:example). To prevent  
 167 this information loss and to allow further analysis of reported adsorption experiments by  
 168 other researchers adsorption data should be archived and shared using the a universal and  
 169 complete data format like that of the AIF.

## 170 Summary and perspective

171 The



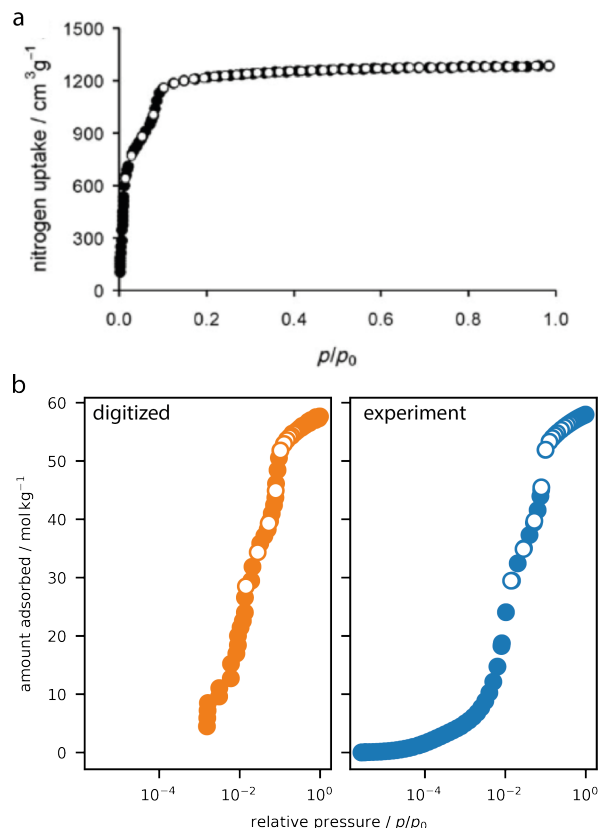


Figure 2: The reported nitrogen adsorption isotherm of DUT-6 recorded at 77 K (a). A digitized representation of this reported isotherm was achieved using a point-picking method (b) and compared to the experimental data as contained within an AIF (c).

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## 183 Graphical TOC Entry

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