Introduction to Nuclear Magnetic Resonance Spectroscopy

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1.1. Introduction to Magnetic Resonance

- Magnetic resonance (MR) is a phenomenon of resonant energy absorption by a system of nuclei (and electrons).
- Nuclear magnetic resonance (NMR) results from the intrinsic magnetic moment of the nuclei of some atoms. Magnetic moments of electrons are exploited in electron spin resonance.
- Magnetic resonance (MR) generally involves placing a sample in a strong magnetic field (to generate polarisation at a fixed resonant frequency) and detecting signals produced following application of pulsed radio-frequency electromagnetic fields (RF pulses).
- MR is a very powerful method for studying the structure of materials: used in physics, chemistry, biology, medicine etc.

1.2. Applications of NMR

 NMR spectroscopy is used for chemical analysis and for molecular structure determination

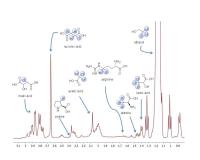


Fig.1: ¹H NMR spectrum of a sample of Spanish wine (http://www.unirioja.es/gsoe/NMR.htm)

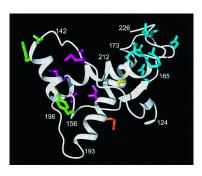


Fig.2: NMR-derived structure of a prion http://www.pnas.org/content/94/14/7281.full

1.2. Applications of NMR

 NMR relaxometry can be used to monitor molecular environment

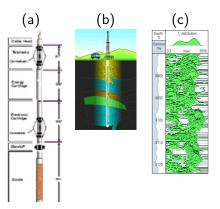


Fig.3: (a) NMR-logging probe, (b) Schematic positioning of the probe in a well, (c) T_2 -relaxation profile along the bore. Sources: 1) Allenet al. Oilfield review, Autumn 2000; 2) Coates, Xiao NMR Logging Principles and Applications, Hulliburton

1.2. Applications of NMR

NMR forms the basis for magnetic resonance imaging (MRI)

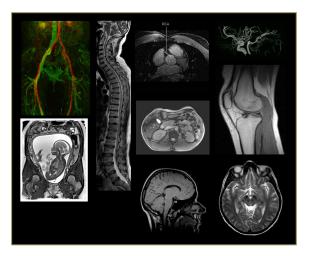


Fig.4: Example magnetic resonance images of blood vessel (in legs), fetus in utero, spine, heart, abdomen, head, blood vessels (in brain), knee, brain (courtesy of Prof. Richard Bowtell)

• Consider charges moving in a limited volume. The position of a charge $\mathbf{e_n}$ will be given by a vector $\mathbf{r_n}$ and its velocity by $\mathbf{v_n}$. The overall magnetic moment of such a system is defined as:

$$\mathbf{M} = \frac{1}{2} \sum_{n} e_{n} \mathbf{r_{n}} \times \mathbf{v_{n}} \tag{1}$$

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 If all the charges and masses are the same, then M can be rewritten as:

$$\mathbf{M} = \frac{e}{2m} \sum_{\mathbf{n}} m \mathbf{r_n} \times \mathbf{v_n} = \gamma \mathbf{L}, \tag{2}$$

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• Gyromagnetic ratio(or magnetogyric):

$$\gamma = \frac{e}{2m} \tag{4}$$

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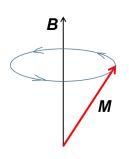
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• Now using equation 2 we can obtain the equation describing the motion of vector **M**:

$$\frac{d\mathbf{M}}{dt} = \gamma \mathbf{M} \times \mathbf{B} \tag{7}$$

• In a uniform magnetic field directed along z-axis $\mathbf{B} = (0, 0, B_0)$, the equation for individual components of \mathbf{M} follow the equations:



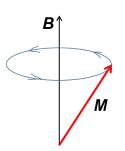
$$\frac{dM_x}{dt} = \omega_L M_y$$

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• A solution to this system of differential equations with initial values of $M_x(0)$, $M_y(0)$, $M_z(0)$ has the following form:

$$M_{x}(t) = M_{x}(0)\cos(\omega_{L}t) + M_{y}(0)\sin(\omega_{L}t)$$

$$M_{y}(t) = -M_{y}(0)\sin(\omega_{L}t) + M_{y}(0)\cos(\omega_{L}t) \qquad (9)$$

$$M_{z}(t) = M_{z}(0)$$

• In quantum mechanics physical quantities A are represented by their operators \hat{A} . The mechanical angular momentum is replaced by its corresponding operator:

$$\mathbf{L} = \sum_{n} \mathbf{p_n} \times \mathbf{r_n} \longleftrightarrow \hat{\mathbf{L}} = \frac{1}{\hbar} \sum_{n} \hat{\mathbf{r}}_{n} \times \hat{\mathbf{p}}_{n} = -i \sum_{n} \hat{\mathbf{r}}_{n} \times \nabla_{n}$$
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Angular momentum operator properties. Commutation:

$$[\hat{L}_y, \hat{L}_z] = i\hat{L}_x, [\hat{L}_z, \hat{L}_x] = i\hat{L}_y, [\hat{L}_x, \hat{L}_y] = i\hat{L}_z$$
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Angular momentum squared, and its commutation properties:

$$\hat{L}^2 = \hat{L}_x^2 + \hat{L}_y^2 + \hat{L}_z^2 \tag{12}$$

$$[\hat{L}^2, \hat{L}_x^2] = [\hat{L}^2, \hat{L}_y^2] = [\hat{L}^2, \hat{L}_z^2] = 0$$
 (13)

• Eigenfunctions of both \hat{L}^2 and \hat{L}_z operators can be characterized by integer quantum numbers I and m respectively. These eigen functions will be denoted as $|Im\rangle$. Their eigenvalues are:

$$\hat{L}_z|Im\rangle = m|Im\rangle \tag{14}$$

$$\hat{L}^2|\mathit{Im}\rangle = \mathit{I}(\mathit{I}+1)|\mathit{Im}\rangle \tag{15}$$

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• Another useful operator are raising and lowering operators:

$$\hat{L}_{+} = \hat{L}_{x} + i\hat{L}_{y}, \hat{L}_{-} = \hat{L}_{x} - i\hat{L}_{y}$$

$$\langle Im|\hat{L}_{+}|I(m-1)\rangle = \langle I(m-1)|\hat{L}_{-}|Im\rangle = \sqrt{(I+m)(I-m+1)}$$
(17)

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Bohr magneton:

$$\beta = \frac{e\hbar}{2m} \approx 9.27 \cdot 10^{-24} \text{J T}^{-1} \tag{19}$$

Free electron and many nuclei have spin.

These equations describe a precession of a vector **M** around the direction of the external magnetic field **B** with the frequency ω_0 as shown schematically in Fig.??A. Such motion is called "Larmor precession" and $\omega_0 = \gamma B_0$ is called "Larmor frequency". Of course, this primitive classical picture serves only as an illustration to the actual behaviour of magnetic moments placed into a magnetic field. However, a more rigorous description using quantum mechanics for an ensemble of magnetic moments provides a similar answer. Larmor precession of an overall magnetic moment is a real effect, and as we will see later, it is essential for acquiring of magnetic resonance spectra.

Now we'll briefly sketch the basic quantum mechanical description of a magnetic moment in a magnetic field. The angular momentum operator $\hat{\mathbf{L}}$ (which could be an orbital or spin angular momentum) is proportional to the magnetic moment operator as:

$$\hat{\boldsymbol{\mu}} = \gamma \hbar \hat{\mathbf{L}},\tag{20}$$

The Hamiltonian of a system can be written by analogy with expression 5 as:

$$\hat{H} = -\hat{\mu}\mathbf{B} = -\gamma\hbar\hat{l}_{r}B_{0} = -\mu\bar{l}_{0}\hat{l}_{r} \stackrel{\text{def}}{=} + \frac{1}{2}(2\dagger) \stackrel{\text{def}}{=} + \frac$$