

1. What is isomerism?
 - a. The study of atoms and molecules
 - b. The study of the physical and chemical properties of matter
 - c. The study of the arrangement of atoms in molecules
 - d. The study of the way atoms bond to form molecules
2. Which of the following is NOT an isomer?
 - a. Ethanol and dimethyl ether
 - b. Butane and isobutane
 - c. Pentane and isopentane
 - d. Propane and propene
3. Which of the following is an example of constitutional isomerism?
 - a. Ethanol and dimethyl ether
 - b. Butane and isobutane
 - c. Pentane and isopentane
 - d. Propane and propene
4. Which of the following is an example of stereoisomerism?
 - a. Ethanol and dimethyl ether
 - b. Butane and isobutane
 - c. Pentane and isopentane
 - d. Propane and propene
5. Which of the following is an example of cis-trans isomerism?
 - a. Ethanol and dimethyl ether
 - b. Butane and isobutane
 - c. Pentane and isopentane
 - d. Propane and propene
6. Which of the following is an example of enantiomerism?
 - a. Ethanol and dimethyl ether
 - b. Butane and isobutane
 - c. Pentane and isopentane
 - d. Propane and propene
7. Which of the following is an example of diastereomerism?
 - a. Ethanol and dimethyl ether
 - b. Butane and isobutane
 - c. Pentane and isopentane
 - d. Propane and propene
8. Which of the following is an example of conformational isomerism?
 - a. Ethanol and dimethyl ether
 - b. Butane and isobutane
 - c. Pentane and isopentane
 - d. Propane and propene
9. Which of the following is an example of metamerism?
 - a. Ethanol and dimethyl ether
 - b. Butane and isobutane
 - c. Pentane and isopentane
 - d. Propane and propene
10. Which of the following is an example of tautomerism?
 - a. Ethanol and dimethyl ether
 - b. Butane and isobutane
 - c. Pentane and isopentane

d. Propane and propene

Answer Key:

1. c
2. d
3. b
4. c
5. d
6. b
7. c
8. d
9. b
10. c