

- In the Bohr's model of the H atom, give an expression for:
  - the electron speed as a function of the orbit radius.
  - the total energy as a function of the orbit radius.

(a)

We recall these two equations

$$\rightarrow r = r_0 n^2 \quad (1)$$

$$\rightarrow V_R = \alpha c \frac{1}{n} \quad (2)$$

Using (1) we get  $n$  in function of the radius

$$\frac{r}{r_0} = n^2 \longrightarrow n = \sqrt{\frac{r}{r_0}} \quad (3)$$

Now we substitute (3) in (1)

$$V_R = \alpha c \frac{1}{n} = \alpha c \sqrt{\frac{r_0}{r}} \longrightarrow V_R = \alpha c \sqrt{\frac{r_0}{r}} \quad (4)$$

(b)

We recall the next equation :

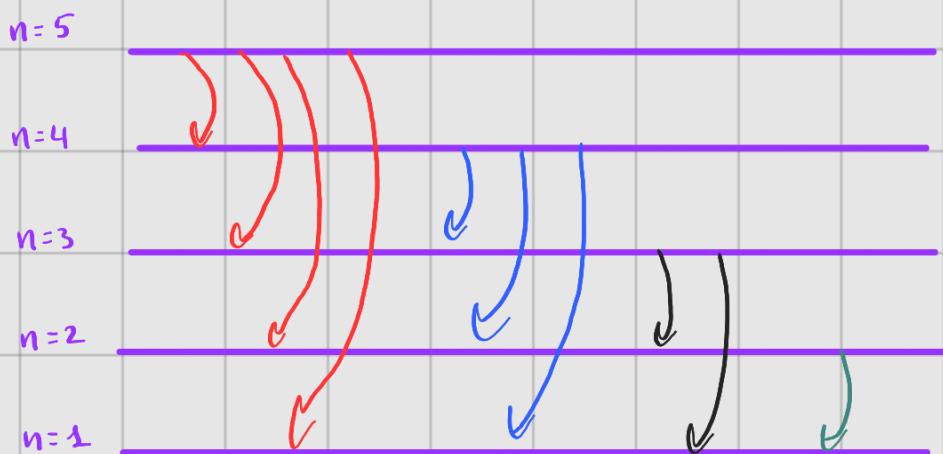
$$\rightarrow E = - \frac{R}{n^2} \quad (5)$$

We substitute (3) in (5)

$$E = - \frac{R}{n^2} = - R \cdot \frac{r_0}{r} \longrightarrow E = - R \frac{r_0}{r} \quad (5)$$

2. How many different photons could be emitted upon a transition from the  $n=5$  down to the fundamental  $n=1$  of an H atom. Compute the exact frequency for one of the transitions.

There are 10 different photons that can be emitted when we start from level  $n=5$  to the level  $n=1$ . We can see these transitions in the next graphic.



10 transitions

$$\text{In general: } \# \text{transitions} = \frac{n(n-1)}{2}$$

The frequency of transition  $n=4$  to  $n=2$

$$E_n = -\frac{R}{n^2}$$

$$\hookrightarrow E_4 = -\frac{R}{16}$$

$$\hookrightarrow E_2 = -\frac{R}{4}$$

$$E_4 - E_2 = h\nu_{42} \rightarrow \nu_{42} = \frac{1}{h} (E_4 - E_2) = \frac{1}{h} \left[ -\frac{R}{16} - \left( -\frac{R}{4} \right) \right] = \frac{3}{16} \frac{R}{h} = \\ = \frac{3}{16} \cdot \frac{13 \text{ eV}}{6.64 \cdot 10^{-34} \text{ J} \cdot \text{s}} \cdot \frac{1,602 \cdot 10^{-19} \text{ J}}{1 \text{ eV}} = 5,88 \cdot 10^{14} \text{ s}^{-1}$$

$\nu_{42} = 5,88 \cdot 10^{14} \text{ Hz}$

3. A commercial (green) laser pointer has a max/output power  $P \leq 1 \text{ mW}$  and a beam spot  $w_o = 1.1 \text{ mm}$ . If  $\lambda = 532 \text{ nm}$  is the light source wavelength, compute the (1.) pointer photon's flux and (2.) the number of photons emitted in 10 sec when you purposely cover half of the exit hole with your finger. (Neglect divergence).

(1) First we determine the intensity

$$P = I \cdot S \quad \text{we assume that the spot has a circular shape}$$

$$S = \pi r^2$$



$$r = \frac{w_o}{2}$$

$$I = \frac{P}{S} = \frac{P}{\pi (w_o/2)^2} = \frac{4 P}{\pi w_o^2} = \frac{4 \cdot 1 \cdot 10^{-3} \text{ W}}{\pi \cdot (1,1)^2 \cdot 10^{-6} \text{ m}^2} = 1052,66 \text{ W/m}^2$$

We calculate the flux like the number of photons per area per second

$$\Phi = \frac{I}{h\nu} = \frac{I \cdot \lambda}{hc} = 1052,66 \frac{\text{W}}{\text{m}^2} \cdot \frac{532 \cdot 10^{-9} \text{ m}}{6,64 \cdot 10^{-34} \text{ J} \cdot \text{s} \cdot 3 \cdot 10^8 \frac{\text{m}}{\text{s}}} = 2,81 \cdot 10^{21} \frac{1}{\text{m}^2 \cdot \text{s}}$$

$$\boxed{\Phi = 2,81 \cdot 10^{21} \text{ m}^{-2} \cdot \text{s}^{-1}}$$

(2) If we cover the half of area:

$$A' = A/2 = \frac{\pi}{8} w_o^2 = 0,475 \cdot 10^{-6} \text{ m}^2$$

And the number of photons  $N$ :

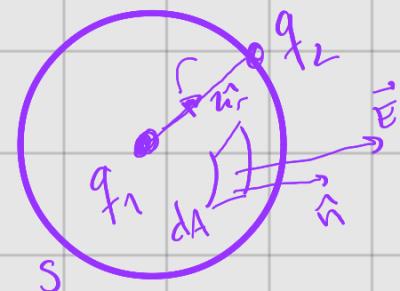
$$N = \Phi \cdot A' \cdot t = 2,81 \cdot 10^{21} \frac{1}{\text{m}^2 \cdot \text{s}} \cdot 0,475 \cdot 10^{-6} \text{ m}^2 \cdot 10 \text{ s} = 1,34 \cdot 10^{16} \text{ photons}$$

$$\boxed{N = 1,34 \cdot 10^{16} \text{ photons}}$$

4.

- a. Give the potential energy of a charge  $q_2$  located at  $\mathbf{r}_2$  due to the presence of a charge  $q_1$  located at  $\mathbf{r}_1$ .
- b. Derive the force exerted on the charge  $q_2$  as due to the charge  $q_1$ .
- c. Discuss the two charge possibilities.
- d. Apply the above results to the case of an electron placed at a distance  $r$  from a nucleus of a H atom (Bohr's model).
- e. Redo part d. for a nucleus of charge  $Ze$ .

(a) Using Gauss' Law for a  $q_1$  charge



$$\oint_S \vec{E} \cdot d\vec{S} = \frac{q_{\text{int}}}{\epsilon_0}$$

$$\oint_S \vec{E} \cdot \hat{n} dS = \frac{q_1}{\epsilon_0} \rightarrow E \oint_S ds = \frac{q_1}{\epsilon_0} \rightarrow E 4\pi r^2 = \frac{q_1}{\epsilon_0}$$

$$\rightarrow \vec{E} = \frac{1}{4\pi\epsilon_0} \frac{q_1}{r^2} \hat{u}_r$$

Scalar potential

$$\vec{E} \cdot \hat{n} = E \cos\theta = E_n = E$$

$$\theta = 0^\circ$$

$$\vec{E}(r) = -\nabla \phi(r) = -\frac{d}{dr} \phi(r)$$

$$\Rightarrow \int_{\infty}^{\phi(r)} d\phi(r) = - \int_{\infty}^r \vec{E}(r) dr \rightarrow \phi(r) - \phi(\infty) = -\frac{q_1}{4\pi\epsilon_0} \int_{\infty}^r r^{-2} dr$$

$$\Rightarrow \phi(r) = -\frac{q_1}{4\pi\epsilon_0} (-1) \left[ \frac{1}{r} \right]_{\infty}^r = \frac{q_1}{4\pi\epsilon_0} \frac{1}{r}$$

Potential energy at  $q_2$ :  $V = q_2 \phi(r) \rightarrow$

$$V(r) = \frac{q_1 q_2}{4\pi\epsilon_0} \frac{1}{r}$$

(b) For the force at  $q_2$  due to  $q_1$ :  $\vec{F}(r) = q_2 \cdot \vec{E}(r)$

$$\boxed{\vec{F}(r) = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r^2} \hat{u}_r}$$

(c) The two possible scenarios are

1) The charges repel each other

$$q_1 = q_2 = e \quad \text{or} \quad q_1 = q_2 = -e$$

$$\vec{F}(r) = \frac{1}{4\pi\epsilon_0} \frac{e^2}{r^2} \hat{u}_r$$



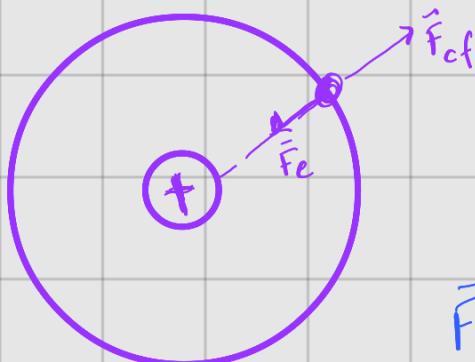
2) The charges attract each other

$$q_1 = e, q_2 = -e \quad \text{or} \quad q_1 = -e, q_2 = e$$

$$\vec{F}(r) = - \frac{1}{4\pi\epsilon_0} \frac{e^2}{r^2} \hat{u}_r$$



(d) Applying the results to Bohr's model



$$U_e(r) = - \frac{e^2}{4\pi\epsilon_0} \cdot \frac{1}{r}$$

$$\vec{F}_e(r) = - \frac{e^2}{4\pi\epsilon_0} \frac{1}{r^2}$$

$$L = n \hbar$$

$$L = m v r \rightarrow v = \frac{L}{mr}$$

$$\vec{F}_{cf} = m v^2 r \downarrow w = \frac{v}{r} \Rightarrow \vec{F}_{cf} = m \frac{v^2}{r}$$

$$\text{Force balance} \rightarrow \sum F = 0 \rightarrow \vec{F}_e + \vec{F}_{cf} = - \frac{e^2}{4\pi\epsilon_0} \frac{1}{r^2} + m \frac{v^2}{r} = 0$$

$$\Rightarrow - \frac{e^2}{4\pi\epsilon_0} \frac{1}{r^2} + m \frac{L^2}{m^2 r^3} = 0 \Rightarrow \frac{e^2}{4\pi\epsilon_0} = \frac{L^2}{mr} \Rightarrow r = \frac{L^2}{e^2 m} 4\pi\epsilon_0$$

$$\Rightarrow r = \frac{n^2 \hbar^2}{e^2 m} 4\pi\epsilon_0 = 4\pi\epsilon_0 r_0 n^2 \Rightarrow r = 4\pi\epsilon_0 r_0 n^2$$

Now we apply the expression of  $r$  in  $F_e$  and  $U_e$

$$\bar{F}_{e_n} = -\frac{e^2}{(4n\epsilon_0)^3 r_0^2} \frac{1}{n^4} \hat{u}_r$$

$$V_{e_n} = -\frac{e^2}{(4n\epsilon_0)^2 r_0} \frac{1}{n^2}$$

④ When we use  $q_1 = Ze$

$$r = 4n\epsilon_0 \frac{r_0}{2} n^2 \Rightarrow$$

$$\bar{F}_{e_n} = -\frac{z^3 e^2}{(4n\epsilon_0)^3 r_0^2} \frac{1}{n^4} \hat{u}_r$$

$$V_{e_n} = -\frac{z^2 e^2}{(4n\epsilon_0)^2 r_0} \frac{1}{n^2}$$

**Extra.**

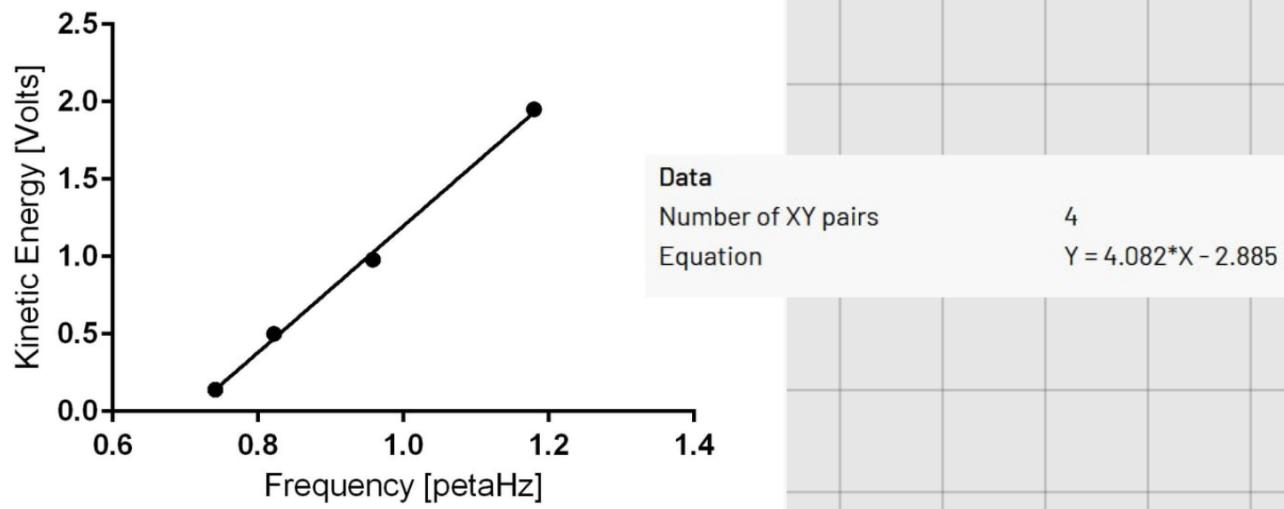
In a photoelectric experiment Ca is used as photocathode and the following values of stopping potential  $V_s$  vs. wavelength  $\lambda$  are measured:

$\lambda, \text{\AA}$	2536	3132	3650	4047
$\nu, \text{Hz} \times 10^{15}$	1.18	0.958	0.822	0.741
$V_s, \text{V}$	1.95	0.98	0.50	0.14

Calculate the Planck constant  $\hbar$  and the work function  $\Phi$ .

Using an online tool to compute the linear regression we obtain:

### Linear Regression



Now we have to compare with the next equation

$$E_K = h\nu - \phi$$

So, comparing both equations:

$$\Rightarrow h = 4,082 \cdot 10^{-15} \text{ V} \cdot \text{s} = 4,082 \cdot 10^{-15} \frac{\text{J}}{\text{C}} \cdot \text{s} = \frac{4,082 \cdot 10^{-15}}{6,24 \cdot 10^{18}} \text{ J.s} \quad [3.5]$$

$$h = 6,54 \cdot 10^{-34} \text{ J.s} \rightarrow \hbar = \frac{h}{2\pi} \Rightarrow \boxed{\hbar = 1,041 \cdot 10^{-34} \text{ J.s}}$$

$$\boxed{\phi = 2,885 \text{ V}}$$