

## Set #1

1. In the Bohr's model of the H atom, give an expression for:
  - a. the electron speed as a function of the orbit radius.
  - b. the total energy as a function of the orbit radius.
2. How many different photons could be emitted upon a transition from the  $n=5$  down to the fundamental  $n=1$  of an H atom. Compute the exact frequency for one of the transitions.
3. A commercial (green) laser pointer has a max/output power  $P \leq 1 \text{ mW}$  and a beam spot  $w_o = 1.1 \text{ mm}$ . If  $\lambda = 532 \text{ nm}$  is the light source wavelength, compute the (1.) pointer photon's flux and (2.) the number of photons emitted in 10 sec when you purposely cover half of the exit hole with your finger. (Neglect divergence).
4.
  - a. Give the potential energy of a charge  $q_2$  located at  $\mathbf{r}_2$  due to the presence of a charge  $q_1$  located at  $\mathbf{r}_1$ .
  - b. Derive the force exerted on the charge  $q_2$  as due to the charge  $q_1$ .
  - c. Discuss the two charge possibilities.
  - d. Apply the above results to the case of an electron placed at a distance  $r$  from a nucleus of a H atom (Bohr's model).
  - e. Redo part d. for a nucleus of charge  $Ze$ .

**Extra.**

In a photoelectric experiment Ca is used as photocathode and the following values of stopping potential  $V_s$  vs. wavelength  $\lambda$  are measured:

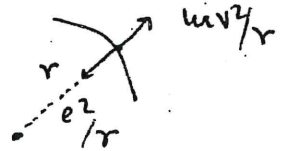
$\lambda, \text{\AA}$	2536	3132	3650	4047
$\nu, \text{Hz} \times 10^{15}$	1.18	0.958	0.822	0.741
$V_s, \text{V}$	1.95	0.98	0.50	0.14

Calculate the Planck constant  $\hbar$  and the work function  $\Phi$ .

1. In the Bohr's model of the hydrogen atom, give an expression of:

- the electron speed as a function of the orbit radius.
- the total energy as a function of the orbit radius.
- the fraction of the potential energy that contributes to the total energy.

1. a.) In ogni punto dell'orbita la forza di Coulomb deve essere uguale alla forza centripeta:  $\frac{e^2}{r^2} = \frac{mv^2}{r} \Rightarrow v^2 = \frac{e^2}{mr}$



b. En. potenziale:  $U = -\frac{e^2}{r}$  ( $U = 9,92$   $F = \frac{9,92}{r^2}$ )

En. cinetica:  $K = \frac{1}{2}mv^2 = \frac{1}{2}m \frac{e^2}{mr} = \frac{e^2}{2r}$


$E = \text{En. Totale} = U + K = \frac{e^2}{2r} - \frac{e^2}{r} = -\frac{1}{2} \frac{e^2}{r}$

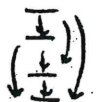
c.  $E = \frac{1}{2} \left( -\frac{e^2}{r} \right)$

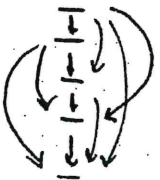
How many different photons could be emitted upon a transition from the  $n=5$  down to the fundamental  $n=1$  of an hydrogen atom. Show the frequencies.

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Da  $n=2 \rightarrow n=1$   un foto

Da  $n=3 \rightarrow n=1$   3 diverse fotoni

Da  $n=4 \rightarrow n=1$   6 diverse fotoni

Da  $n=5 \rightarrow n=1$   10 diverse fotoni

In generale, da  $n$  al fondamentale ( $n=1$ ) uno ha  
fotoni emessi.

$$\boxed{\frac{n(n-1)}{2}} \text{ diverse}$$

• Transizione  $n=5 \rightarrow n=4$

$$E_n = -\frac{R}{n^2} \quad E_5 - E_4 = -\frac{R}{25} + \frac{R}{16} = R \left( \frac{1}{16} - \frac{1}{25} \right) = 13.6 \text{ eV} \left( \frac{1}{16} - \frac{1}{25} \right) = 0.306 \text{ eV}$$
$$= h\nu = \frac{hc}{\lambda_{54}}$$

$$\lambda_{54} = \frac{0.306 \text{ eV}}{12.4 \times 10^3 \text{ eV } \text{\AA}} = \Rightarrow \lambda_{54} \approx 40523 \text{ \AA} \approx 4 \mu\text{m}$$

Transizione  $n=5 \rightarrow n=3$

$$E_5 - E_3 = R \left( \frac{1}{9} - \frac{1}{25} \right) = 0.967 \text{ eV}$$
$$= \frac{hc}{\lambda_{54}}$$

$$\lambda_{54} = hc / 0.967 \text{ eV} = \frac{12.4 \times 10^3 \text{ eV } \text{\AA}}{0.967 \text{ eV}} = \dots \approx 1.28 \mu\text{m}$$

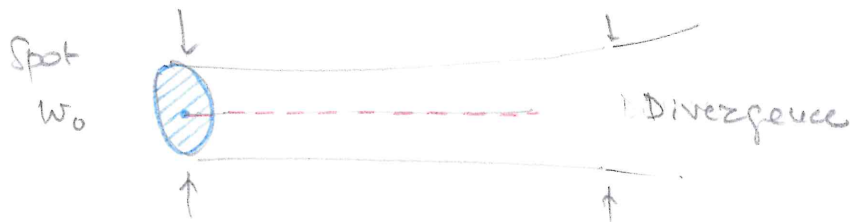
Transizione  $n=5 \rightarrow n=2$

$$E_5 - E_2 = R \left( \frac{1}{4} - \frac{1}{25} \right) = 2.85 \text{ eV}$$
$$= \frac{hc}{\lambda_{52}}$$

$$\lambda_{52} = hc / 2.85 \text{ eV} = \frac{12.4 \times 10^3 \text{ eV } \text{\AA}}{2.85 \text{ eV}} = \dots \approx 0.43 \mu\text{m}$$

A commercial (green) laser pointer has a max/output power  $P \leq 1 \text{ mW}$  and a beam spot  $w_0 = 1.1 \text{ mm}$ . If  $\lambda = 532 \text{ nm}$  is the light source wavelength, compute the (1.) pointer photon's flux and (2.) the number of photons emitted in 10 sec when you purposely cover half of the exit hole with your finger. (Neglect divergence).

$$1.) \text{ Intensity} = \frac{\text{Energy}}{\text{Surface} \times \text{time}} = \frac{\text{Power}}{\text{Surface}} = \frac{1 \text{ mW}}{\pi \left( \frac{1.1 \text{ mm}}{2} \right)^2} = \frac{10^{-3} \text{ J}}{\pi \left( \frac{1.1}{2} \right)^2 \times 10^{-5} \text{ m}^2}$$



$$= 1.05 \times 10^3 \left( \frac{\text{J}}{\text{m}^2 \text{ sec}} \right) \quad (\text{Flux})$$

$$2.) \text{ Energy across } \frac{1}{2} \text{ beam spot: } 1.05 \times 10^3 \frac{\text{J}}{\text{m}^2 \text{ sec}} \times \underbrace{\frac{1}{2} \left( \frac{1.1}{2} \text{ mm} \right)^2 \pi \times 10 \text{ sec}}_{\frac{1}{2} \text{ beam spot surface}}$$

$$= [\text{J}] \frac{1.05 \times 10^3}{\text{m}^2 \text{ sec}} \times \frac{1}{2} \left( \frac{1.1}{2} \right)^2 \pi \times 10^{-6} \text{ m}^2 \times 10 \text{ sec}$$

$$\approx 5 \times 10^{-3} (\text{J})$$

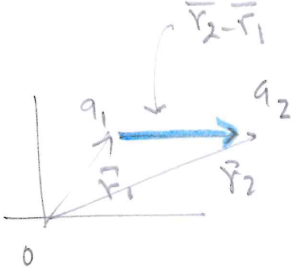
$$\text{Energy of 1 photon: } E = h\nu = \frac{hc}{\lambda} = \frac{12.41 \times 10^3 \text{ eV} \cdot \text{\AA}}{532 \text{ nm}} \approx [\text{eV}] 2.3 \times 10^3$$

$$\approx [\text{J}] 3.7 \times 10^{-22}$$

$$N_{\text{photons}} = \frac{5 \times 10^{-3} (\text{J})}{3.7 \times 10^{-22} (\text{J})} \approx 10^{19}$$

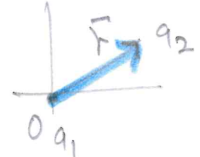
5. a. Give the potential energy of a charge  $q_2$  located at  $\mathbf{r}_2$  due to the presence of a charge  $q_1$  located at  $\mathbf{r}_1$ .  
 b. Derive the force exerted on the charge  $q_2$  as due to the charge  $q_1$ .  
 c. Discuss the two charge possibilities.  
 d. Apply the above results to the case of an electron placed at a distance  $r$  from a nucleus of a H atom (Bohr's model).  
 e. Redo part d. for a nucleus of charge  $Ze$ .

a.)



$$\phi = \frac{q_1 q_2}{|\mathbf{r}_1 - \mathbf{r}_2|} \quad \{q_1, q_2\} = \text{charges}$$

Translation:  $\mathbf{r}_1 = 0 \quad \mathbf{r}_2 \equiv \mathbf{r}$

$$\phi = \frac{q_1 q_2}{|\mathbf{r}|}$$


b.)  $\vec{F} = -\vec{\nabla} \phi$

$$= -\vec{\nabla} \left( \frac{q_1 q_2}{|\mathbf{r}|} \right) = -q_1 q_2 \vec{\nabla} \left( \frac{1}{|\mathbf{r}|} \right)$$

$\vec{r} = x\hat{i} + y\hat{j} + z\hat{k}$   
 $\hookrightarrow$  position vector

$$\vec{\nabla} \frac{1}{|\mathbf{r}|} = \vec{\nabla} \frac{1}{\sqrt{x^2 + y^2 + z^2}} = \hat{i} \frac{\partial}{\partial x} \frac{1}{\sqrt{}} + \hat{j} \frac{\partial}{\partial y} \frac{1}{\sqrt{}} + \hat{k} \frac{\partial}{\partial z} \frac{1}{\sqrt{}}$$

$$\frac{\partial}{\partial x} (x^2 + y^2 + z^2)^{-1/2} = -\frac{1}{2} \frac{2x}{(x^2 + \dots)^{3/2}} = -\frac{x}{|\mathbf{r}|^3} \hat{i}$$

Similarly for  $y, z$ :

$$\vec{F} = -q_1 q_2 \left( -\frac{x\hat{i}}{|\mathbf{r}|^3} - \frac{y\hat{j}}{|\mathbf{r}|^3} - \frac{z\hat{k}}{|\mathbf{r}|^3} \right) = +q_1 q_2 \left( \frac{\vec{r}}{|\mathbf{r}|^3} \right) \left[ +q_1 q_2 \frac{\vec{r}_2 - \vec{r}_1}{|\vec{r}_2 - \vec{r}_1|} \right]$$

General case

c.) If  $q_1 = +1e$   $q_2 = -1e$   $\vec{F}_{12} = -1e^2 \frac{\vec{r}}{|\mathbf{r}|^3}$  force is opposite to  $\vec{r}$  (Attractive)

If  $q_1 = +1e$   $q_2 = +1e$   $\vec{F}_{12} = +1e^2 \frac{\vec{r}}{|\mathbf{r}|^3}$  " " along  $\vec{r}$  (Repulsive)

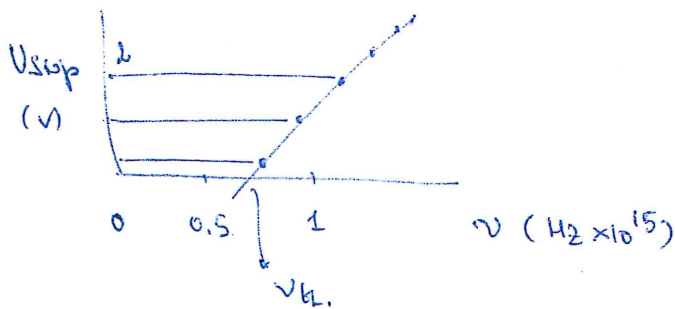
d.)  $\phi = -\frac{1e^2}{|\mathbf{r}|}$   $\vec{F} = -1e^2 \frac{\vec{r}}{|\mathbf{r}|^3}$  (Attractive)

e.)  $q_1 = +Ze$   $q_2 = -e$   $\phi = -\frac{Ze^2}{|\mathbf{r}|}$   $\vec{F} = -Ze^2 \frac{\vec{r}}{|\mathbf{r}|^3}$

In a photoelectric experiment Ca is used as photocathode, and the following values of stopping potential vs. wavelength are measured:

$\lambda, \text{\AA}$	2536	3132	3650	4047
$\nu, \text{Hz} \times 10^{15}$	1.18	0.958	0.822	0.741
$V_s, \text{V}$	1.95	0.98	0.50	0.14

Calculate the Planck constant  $h$  and the work function  $\Phi$ .



$\lambda$	2536	3650	4047	( $\text{\AA}$ )
$\nu$	1.18	0.82	0.74	( $10^{15} \text{Hz}$ )
$V_s$	1.95	0.50	0.14	(V)

$$\left. \begin{aligned} h\nu &= E + \phi \\ E &= e V_{\text{stop}} \end{aligned} \right\} \quad e V_{\text{stop}} = h\nu - \phi \quad V_{\text{stop}} = \left(\frac{h}{e}\right)\nu - \frac{\phi}{e}$$

$$\text{Slope} = \frac{(1.95 - 0.14) \text{V}}{(1.18 - 0.74) 10^{15} \text{Hz}} \equiv \frac{h}{e}$$

$$h = \frac{e (1.95 - 0.14) \text{V}}{(1.18 - 0.74) 10^{15} \text{Hz}} = \frac{(1.6 \times 10^{-19} \text{C}) (1.81 \text{V})}{0.44 \times 10^{15} \text{Hz}} \approx 6.59 \times 10^{-34} \text{J/Hz}^{-1}$$

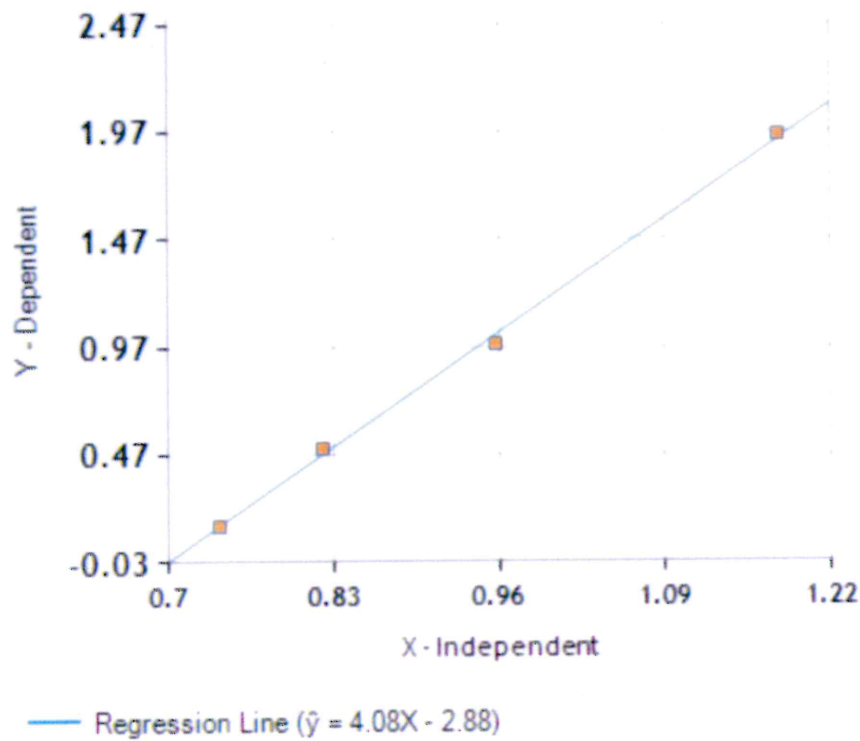
"Planck's const."

$$\text{If } V_{\text{stop}} = 0 \Rightarrow \frac{\phi}{e} = \frac{h}{e} \nu_{\text{th}} \quad \nu_{\text{th}} \approx 0.7 \times 10^{15} \text{Hz}$$

$$\phi \approx 6.6 \times 10^{-34} \text{J/Hz} \times 0.7 \times 10^{15} \text{Hz} = 4.62 \times 10^{-19} \text{J} \approx 2.9 \text{eV}$$



using best fit:



$$E_k = eV_{\text{stop}} = h\nu - \Phi \quad \left[ \begin{array}{l} \text{V} \\ [\text{V}] \times 10^{15} \text{ V} \end{array} \right] \quad V_{\text{stop}} = + \frac{h}{e} \cdot \nu - \frac{\Phi}{e} \equiv \left[ \frac{h}{e} 10^{15} \right] [\text{V}] - \frac{\Phi}{e}$$

Then:  $\frac{h}{e} \cdot 10^{15} = 4.08 \left[ \frac{\text{V}}{\text{V}} \right] \Rightarrow h = 4.08 \times 10^{-15} \text{ e} \left[ \frac{\text{V}}{\text{V}} \right] = 4.08 \times 10^{-15} \times 1.6 \times 10^{-19} \text{ C} \cdot \text{V} = 6.54 \times 10^{-34} \text{ J} \cdot \text{sec.}$

$$\frac{\Phi}{e} = 2.8 \text{ V} \Rightarrow \Phi = 2.8 \text{ eV}$$

↓

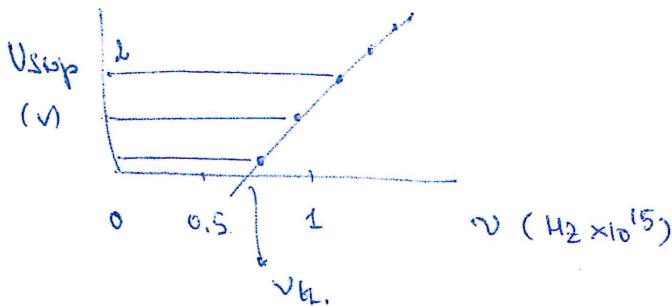
$$2.8 \text{ V} \cdot 1.6 \times 10^{-19} \text{ C} = 4.5 \times 10^{-19} \text{ J}$$



In a photoelectric experiment Ca is used as photocathode, and the following values of stopping potential vs. wavelength are measured:

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$V_s$	1.95	0.50	0.14	(V)

$$\left. \begin{array}{l} h\nu = E + \phi \\ E = e V_{\text{stop}} \end{array} \right\} \quad e V_{\text{stop}} = h\nu - \phi \quad V_{\text{stop}} = \left(\frac{h}{e}\right)\nu - \frac{\phi}{e}$$

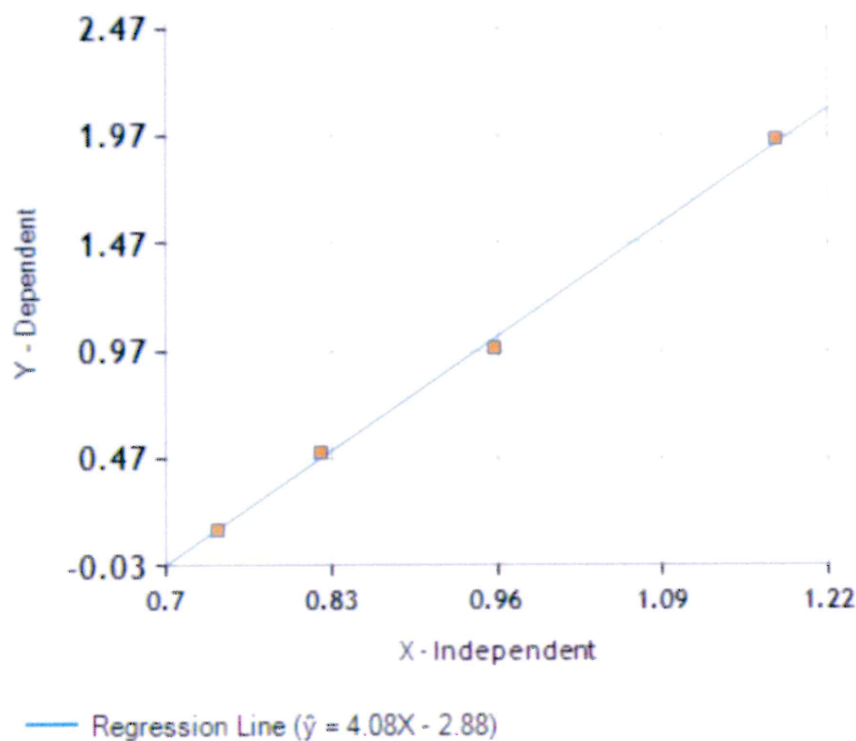
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$$h = \frac{e (1.95 - 0.14) \text{V}}{(1.18 - 0.74) \times 10^{15} \text{Hz}} = \frac{(1.6 \times 10^{-19} \text{C}) (1.81 \text{V})}{0.44 \times 10^{15} \text{Hz}} \approx 6.59 \times 10^{-34} \text{J/Hz} \quad \text{"Planck's const."}$$

$$\text{If } V_{\text{stop}} = 0 \Rightarrow \frac{\phi}{e} = \frac{h}{e} \nu_0 \quad \nu_0 \approx 0.7 \times 10^{15} \text{Hz}$$

$$\phi \approx 6.6 \times 10^{-34} \text{J/Hz} \times 0.7 \times 10^{15} \text{Hz} = 4.62 \times 10^{-19} \text{J} \approx 2.9 \text{eV}$$

using best fit:



$$E_k = eV_{\text{stop}} = h\nu - \Phi \quad \left[ \begin{array}{l} \nearrow V \\ \nearrow [\nu] \times 10^{15} \text{ sec}^{-1} \end{array} \right] \quad V_{\text{stop}} = + \frac{h}{e} \cdot \nu - \frac{\Phi}{e} \equiv \left[ \frac{h}{e} 10^{15} \right] [\nu] - \frac{\Phi}{e}$$

Then:  $\frac{h}{e} \cdot 10^{15} = 4.08 \left[ \frac{V}{\text{sec}^{-1}} \right] \Rightarrow h = 4.08 \times 10^{-15} e \left[ \frac{V}{\text{sec}^{-1}} \right] = 4.08 \times 10^{-15} \times 1.6 \times 10^{-19} \frac{\text{C} \cdot \text{V}}{\text{sec}^{-1}}$   
 $= 6.54 \times 10^{-34} \text{ J} \cdot \text{sec}.$

$$\frac{\Phi}{e} = 2.8 \text{ V} \Rightarrow \Phi = 2.8 \text{ eV}$$

↓

$$2.8 \text{ V} \cdot 1.6 \times 10^{-19} \text{ C} = 4.5 \times 10^{-19} \text{ J}$$