

Heat equation

heat flows $\underline{H} = -K \underline{\nabla} T(\underline{x}, t)$ (Fourier's law)

Change in internal energy (without work) must come from heat Q :

$$\Delta Q = C_S \Delta T$$

or $Q(t) = \int d^3x C_S T(\underline{x}, t)$

Energy conservation: heat flow = change of energy inside

$$\frac{\partial Q}{\partial t} = \oint dA \cdot \underline{H} \quad (\text{heat flowing through boundary per unit time})$$

$$\frac{\partial}{\partial t} \int d^3x C_S T(\underline{x}, t) = - \int d^3x \underline{\nabla} \cdot (-K \underline{\nabla} T) \quad \left(\int_V d^3x \underline{\nabla} \cdot \underline{\phi} = \oint_{\partial V} \underline{n} \cdot \underline{\phi} dA \right)$$

↑
divergence theorem

$$\frac{\partial}{\partial t} C_S T(\underline{x}, t) = -(-K \underline{\nabla}^2 T(\underline{x}, t))$$

$$\frac{\partial T}{\partial t} = + \frac{K}{C_S} \underline{\nabla}^2 T(\underline{x}, t)$$

Diffusion equations are wide-spread:

- heat
- particle diffusion & Brownian motion
- finance (Black-Scholes)
- Schrödinger equation