

# First Law of Thermo

Energy added  $Q = \Delta \text{internal energy} + \text{work done in expanding}$

$$Q = C_v \frac{dT}{dt} + p \frac{d\alpha}{dt}$$

$\alpha = \frac{1}{\rho} = \text{specific volume}$

$p\alpha = RT$  ideal gas law

$$Q = \underbrace{C_v + R}_{C_p} \frac{dT}{dt} - \frac{RT}{p} \frac{dp}{dt}$$

$$p \frac{d\alpha}{dt} + \alpha \frac{dp}{dt} = R \frac{dT}{dt}$$

We desire one term on the RHS, not two! How?

Divide by T:

the  
entropy  
route

$$\frac{Q}{T} = C_p \left( \frac{1}{T} \frac{dT}{dt} \right) - R \left( \frac{1}{p} \frac{dp}{dt} \right)$$

$$= C_p \frac{d}{dt}(\ln T) - R \frac{d}{dt}(\ln p) = \frac{d}{dt} (C_p \ln \Theta) = \frac{d}{dt} (S)$$

$\Theta = T \left( \frac{p_0}{p} \right)^{\frac{R}{C_p}}$  potential temperature

entropy

Is there another trick?



$$\frac{dp}{dz} = -\rho g$$

$$Q = C_p \frac{dT}{dt} - \alpha \frac{dp}{dt}$$

want to combine these.  
For a hydrostatic atmosphere.

$$dp = -\rho g dz$$

$$Q = C_p \frac{dT}{dt} + g \frac{dz}{dt} = \frac{d}{dt} (C_p T + g z) \equiv \frac{d}{dt} (s)$$

↑  
constant
↑  
constant
↑  
dry static energy

- dry static energy is conserved when  $Q=0$  (adiabatic flow)
  - just like  $\Theta$ !
- At a given height  $z$  (related to  $p$ ), it measures  $C_p T$  or  $T$  enthalpy
  - and thus parcel buoyancy  $b = g \left( \frac{\rho_{\text{parcel}} - \rho_{\text{env}}}{\rho_{\text{env}}} \right)$
- A layer of constant  $s$  has lapse rate:
 

$\frac{\partial s}{\partial z} = 0 = C_p \frac{\partial T}{\partial z} + g \Rightarrow \frac{\partial T}{\partial z} = -\frac{g}{C_p} = \frac{-10K}{km}$ 

↑  
fractional density difference

Suppose  $Q$  is condensation only.

if  $Q = -L \frac{dq_v}{dt}$  only, then

$$-L \frac{dq_v}{dt} = \frac{d}{dt}(C_p T + g z) \quad \text{or (Pump Roll: Conservation!)}$$

$$\frac{d}{dt} \left( \underbrace{C_p T}_{\text{enthalpy}} + \underbrace{g z}_{\text{potential energy}} + \underbrace{L q_v}_{\text{latent heat}} \right) = 0$$

dry static energy

moist static energy  $h$

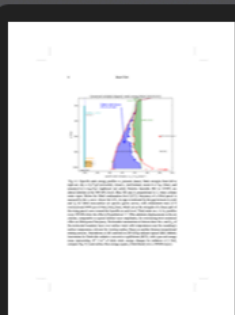
$q_v$  is  $\frac{K_{\text{vapor}}}{K_{\text{air}}}$  (specific humidity)

sometimes just  $q$

$h$  is conserved

- for changes in  $z$  (vertical motion)
- even with condensation
- conserved when  $\delta$  is like  $\Theta_e$  because  $q$  is.

Connection to buoyancy: If  $T_{\text{parcel}} > T_{\text{env}}$  and parcel is saturated, then  $h_{\text{parcel}} > h_{\text{sat}}(T_{\text{env}})$



3



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