

Reviewing,

$$\begin{cases} \frac{d}{dt} u = -\frac{\partial \Phi}{\partial x} \bigg|_p + f v + \cancel{F_x} \\ \frac{d}{dt} v = -\frac{\partial \Phi}{\partial y} \bigg|_p - f u + \cancel{F_y} \end{cases}$$

prediction eqs. through
conservation

take curl
⇕

$$\frac{d\zeta}{dt} = 0 + \dots$$

vortical flow
advects vorticity

$$v \frac{df}{dy} = v \beta$$

$$\frac{d(\beta + f)}{dt} = 0$$

← same, but also
Rossby waves
- westward phase speed
- eastward group speed
- proportional to L^2

descriptive (no time
derivative)

$$0 = -\frac{\partial \Phi}{\partial x} \bigg|_p + f v_g$$

$$0 = -\frac{\partial \Phi}{\partial y} \bigg|_p - f u_g$$

not assumed exactly... it
defines this

geostrophic part of the
total wind field.

"80%" - good approx.

$$\frac{dw}{dt} = -\frac{\partial p}{\partial z} \bigg|_p + g \approx 0$$

hydrostatic approx.

Our prediction through cons. eqs.
project continues!

$$\frac{d}{dt}(\zeta + f) = 0 - (\zeta + f) \cdot \left(\frac{\partial u}{\partial x} + \frac{\partial v}{\partial y} \right) + \text{terms involving } w \text{ or } \omega \text{ (vertical motion)}$$

$= \text{Div}_h V$
 $= (\nabla_h \cdot V)$
 small @ 500mb



$$\frac{d}{dt}(P \cdot V) = 0 + c(\zeta + f) \cdot \frac{\partial}{\partial z}(Q) + D \cdot (\text{curl of friction})$$

transport only

heating

friction

driver

dumper

But what is P.V?
A: It involves Θ "potential temperature"
 $\frac{d\Theta}{dt} = 0!$ ← this again!

shapes of the flow

Potential temperature: the T a parcel of air would have if it were compressed adiabatically to 1000mb = 10⁵ Pa.
 Comes from 1st Law of Thermo: energy is conserved.

$$\frac{d}{dt}(\text{internal energy}) = \underbrace{\left(\text{work done on env.} \right)}_{\substack{\text{volume} \\ \text{for unit} \\ \text{mass}}} + Q \quad \begin{matrix} \text{internal} + \text{external} \\ \text{energy added (zero for} \\ \text{adiabatic processes)} \end{matrix}$$

$$\frac{d}{dt} \left(\underset{\substack{\uparrow \\ \text{(J/kg)/K}}}{C_v T} \right) = P \frac{dv}{dt} + Q$$

ideal gas law
 (equation of state)
 $p\alpha = RT$
 $\alpha = \frac{1}{\rho} = \frac{V}{M}$
 ("specific volume")

$$\frac{d}{dt} (C_v T) = P \frac{d\alpha}{dt} + Q$$

$$\frac{d}{dt} (C_p T) = -\alpha \frac{dp}{dt} + Q \quad C_p \equiv (C_v + R)$$

take d/dt.

$$\frac{d}{dt} (p\alpha) = \frac{d}{dt} (RT) = R \frac{dT}{dt}$$

$$\left(p \frac{d\alpha}{dt} \right) + \alpha \frac{dp}{dt} = R \frac{dT}{dt}$$

Recap of 1st Law in form we like:

$C_p = C_v + R$
 $= \text{const}$

$$\frac{d}{dT}(G_T) = \alpha \frac{dP}{dT} + Q$$

$$\alpha = \frac{P}{RT} \quad \text{Ideal gas law again}$$

$$C_p \frac{dT}{dT} = \left(\frac{RT}{P} \right) \frac{dP}{dT} + Q$$

Trick: divide by T!

$$\frac{d}{dT}(\text{something}) = 0 + Q \dots \text{how to get here?}$$

$$C_p \left(\frac{1}{T} \frac{dT}{dT} \right) = R \left(\frac{1}{P} \frac{dP}{dT} \right) + \frac{Q}{T}$$

$$C_p \frac{d}{dT}(\ln T) = R \frac{d}{dT}(\ln P) + \frac{Q}{T}$$

$$\boxed{C_p \frac{d}{dT}(\Theta) = 0 + Q \left(\frac{P_0}{P} \right)^{R/C_p}}$$

$$\Theta = T \left(\frac{P_0}{P} \right)^{R/C_p}$$

note:

$$\left(\frac{P_0}{P} \right)^{R/C_p} = \frac{\Theta}{T}$$