# Useful Identities in Atomic Physics

## Nicolás Quesada Instituto de Física, Universidad de Antioquia

The quantum numbers of the energy eigenstates of the non-relativistic hydrogenoid atom are: n (energy), l (angular momentum) and m (z projection of angular momentum).  $\langle \cdot \rangle$  means average over *energy* eigenstates.

#### Scales Constants and Special Values of Hydrogenoid Wave Functions

Energies for the Coulomb Potential 
$$(V(\mathbf{r}) = -\frac{1}{4\pi\epsilon_0} \frac{Ze^2}{r})$$
 are  $E_n = -\frac{\mu}{2n^2} \left(\frac{Ze^2}{4\pi\epsilon_0\hbar}\right)^2 = -\frac{e^2}{4\pi\epsilon_0 a_0} \frac{Z^2}{2n^2} = -\frac{1}{2}\mu c^2 \frac{(Z\alpha)^2}{n^2}$ .  
Fine structure constant:  $\alpha = \frac{e^2}{4\pi\epsilon_0 \hbar c}$ . Bohr radius:  $a_0 = 4\pi\epsilon_0 \hbar^2/(\mu e^2)$ .  $|\psi_{nlm}(0)|^2 = \frac{Z^3}{\pi a_0^3 n^3} \delta_l^0 \delta_m^0$ .

### Expected Values, the Virial Theorem and the Gamma Function

Virial Theorem: If 
$$H=T(\mathbf{p})+V(\mathbf{r})$$
 and  $T(\mathbf{p})=\frac{\mathbf{p}^2}{2\mu}$  then  $2\langle T\rangle=\langle \mathbf{r}\cdot\nabla V\rangle$ .

Expectation values for the Coulomb potential: 
$$\left\langle \frac{1}{r} \right\rangle = \frac{Z}{a_0 n^2}$$
,  $\left\langle \frac{1}{r^2} \right\rangle = \frac{Z^2}{a_0^2 n^3 (l+1/2)}$ .

Recursion Relation: 
$$0 = \frac{s}{4} \left[ (2l+1)^2 - s^2 \right] \left( \frac{a_0}{Z} \right)^2 \langle r^{s-2} \rangle - (2s+1) \left( \frac{a_0}{Z} \right) \langle r^{s-1} \rangle + \frac{s+1}{n^2} \langle r^s \rangle \ .$$

$$\Gamma(z) = \int_0^\infty t^{z-1} e^{-t} \, dt \ , \ \Gamma(n+1) = n! \ , \ \Gamma(1-z) \ \Gamma(z) = \frac{\pi}{\sin{(\pi z)}} \ , \ \Gamma(z) \ \Gamma\left(z + \frac{1}{2}\right) = 2^{1-2z} \ \sqrt{\pi} \ \Gamma(2z).$$

# Spherical Harmonics, Wigner 3j Symbols and Clebsch – Gordan Coefficients

$$\int_0^{2\pi} d\phi \int_0^{\pi} \sin\theta d\theta \, \mathcal{Y}_{l_1}^{m_1}(\theta,\phi) \mathcal{Y}_{l_2}^{m_2}(\theta,\phi) \mathcal{Y}_{l_3}^{m_3}(\theta,\phi) = \sqrt{\frac{(2l_1+1)(2l_2+1)(2l_3+1)}{4\pi}} \begin{pmatrix} l_1 & l_2 & l_3 \\ 0 & 0 & 0 \end{pmatrix} \begin{pmatrix} l_1 & l_2 & l_3 \\ m_1 & m_2 & m_3 \end{pmatrix}.$$

Wigner 3-j — Clebsch–Gordan (CG) relation: 
$$\begin{pmatrix} j_1 & j_2 & j_3 \\ m_1 & m_2 & m_3 \end{pmatrix} \equiv \frac{(-1)^{j_1-j_2-m_3}}{\sqrt{2j_3+1}} \langle j_1 m_1 j_2 m_2 | j_3 - m_3 \rangle.$$

Selection rules for Wigner 3-
$$j$$
 Symbol  $\begin{pmatrix} l_1 & l_2 & L \\ m_1 & m_2 & -M \end{pmatrix}$  (they are identical to CG Selection Rules):  $-l_i \leq m_i \leq l_i, \quad m_1+m_2=M, \quad |l_1-l_2| \leqslant L \leqslant l_1+l_2 \quad l_1+l_2+L \in \mathbb{Z}$ .

Spherical components of a cartesian vector  $\vec{e} = (e_x, e_y, e_z)$ :  $e_{\pm 1} = \mp \frac{1}{\sqrt{2}} (e_x \pm i e_y)$  and  $e_0 = e_z$ .

$$\text{Ladder Operators: } \hat{L}_{\pm} \equiv \hat{L}_x \pm i \hat{L}_y. \qquad \hat{L}_{\pm} | l, m \rangle = \hbar \sqrt{l(l+1) - m(m\pm 1)} | l, m_l \pm 1 \rangle.$$

#### References

- [1] B.H. Bransden, C.J. Joachain *Physics of Atoms and Molecules*, Prentice Hall.
- [2] G.B. Arfken, H.J. Weber Mathematical Methods for Physicists, Harcourt Academic Press.
- [3] Weisstein, Eric W. "Wigner 3j-Symbol." http://mathworld.wolfram.com/Wigner3j-Symbol.html