# Useful Identities in Atomic Physics

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#### **Notation and Conventions**

The quantum numbers of the energy eigenstates of the non-relativistic hydrogen atom are: n (energy), l (angular momentum) and  $m_l$  (z projection of angular momentum). Otherwise stated  $\langle \cdot \rangle$  means average over energy eigenstates.

#### Scales Constants and Special Values of Hydrogenoid Wave Functions

Energies for the Coulomb Potential 
$$(V(\mathbf{r}) = -\frac{1}{4\pi\epsilon_0} \frac{Ze^2}{r})$$
 are  $E_n = -\frac{m}{2n^2} \left(\frac{Ze^2}{4\pi\epsilon_0 \hbar}\right)^2 = -\frac{e^2}{4\pi\epsilon_0 a_0} \frac{Z^2}{2n^2} = -\frac{1}{2}mc^2 \frac{(Z\alpha)^2}{n^2}$ 

Fine structure constant:  $\alpha = \frac{e^2}{4\pi\epsilon_0\hbar c}$ . Bohr radius:  $a_0 = 4\pi\epsilon_0\hbar^2/(me^2)$ .  $|\psi_{nlm}(0)|^2 = \frac{Z^3}{\pi a_0^3 n^3} \delta_l^0 \delta_m^0$ 

#### Expected Values, the Virial Theorem and the Gamma Function

Virial Theorem: If  $H = T(\mathbf{p}) + V(\mathbf{r})$  and  $T(\mathbf{p}) = \mathbf{p}^2/2m$  then  $2\langle T \rangle = \langle \mathbf{r} \cdot \nabla V \rangle$ 

Some expectation values:  $\langle\frac{1}{r}\rangle=\frac{Z}{a_0n^2}$  ,  $\langle\frac{1}{r^2}\rangle=\frac{Z^2}{a_0^2n^3(l+1/2)}$ 

Recursion Relation:  $0 = \frac{s}{4} \left[ (2l+1)^2 - s^2 \right] a_0^2 \langle r^{s-2} \rangle - (2s+1) a_0 \langle r^{s-1} \rangle + \frac{s+1}{n^2} \langle r^s \rangle$ 

Definition:  $\Gamma(z) = \int_0^\infty t^{z-1} e^{-t} dt$ .

Identities:  $\Gamma(n+1)=n!$ ,  $\Gamma(1-z)$   $\Gamma(z)=\frac{\pi}{\sin{(\pi z)}}$ ,  $\Gamma(z)$   $\Gamma\left(z+\frac{1}{2}\right)=2^{1-2z}$   $\sqrt{\pi}$   $\Gamma(2z)$ .

## Integrals of 3 Spherical Harmonics, Wigner 3j Symbols and Clebsch Gordan Coefficients

$$\int_0^{2\pi} d\phi \int_0^\pi \sin\theta d\theta \mathcal{Y}_{l_1}^{m_1}(\theta,\phi) \mathcal{Y}_{l_2}^{m_2}(\theta,\phi) \mathcal{Y}_{l_3}^{m_3}(\theta,\phi) = \sqrt{\frac{(2l_1+1)(2l_2+1)(2l_3+1)}{4\pi}} \begin{pmatrix} l_1 & l_2 & l_3 \\ 0 & 0 & 0 \end{pmatrix} \begin{pmatrix} l_1 & l_2 & l_3 \\ m_1 & m_2 & m_3 \end{pmatrix}$$

 $\text{Wigner 3-j Clebsch Gordan relation: } \begin{pmatrix} j_1 & j_2 & j_3 \\ m_1 & m_2 & m_3 \end{pmatrix} \equiv \frac{(-1)^{j_1-j_2-m_3}}{\sqrt{2j_3+1}} \langle j_1 m_1 j_2 m_2 | j_3 - m_3 \rangle.$ 

Selection rules for Wigner 3-j Symbol  $\begin{pmatrix} l_1 & l_2 & L \\ m_1 & m_2 & -M \end{pmatrix}$  (they are identical to C-G Selection Rules):  $m_i \in \{-l_i, -l_i+1, ..., l_i-1, l_i\}$   $m_1+m_2=M$   $|l_1-l_2| \leqslant L \leqslant l_1+l_2$   $l_1+l_2+L \in \mathbb{N}$ 

Spherical components of a cartesian vector  $\vec{e} = (e_x, e_y, e_z)$ :  $e_{\pm 1} = \mp \frac{1}{\sqrt{2}} (e_x \pm i e_y)$  and  $e_0 = e_z$ 

Ladder Operators:  $\hat{L}_{\pm} = \hat{L}_x \pm i\hat{L}_y$ .  $\hat{L}_{\pm}|l,m_l\rangle = \hbar\sqrt{l(l+1) - m_l(m_l \pm 1)}|l,m_l \pm 1\rangle$ 

### References

- [1] B.H. Bransden, C.J. Joachain *Physics of Atoms and Molecules*, Prentice Hall.
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- [3] Weisstein, Eric W. "Wigner 3j-Symbol." From MathWorld-A Wolfram Web Resource. http://mathworld.wolfram.com/Wigner3j-Symbol.html