Don't Let Spills Stain Your Evening

STAIN AWAY

The stain remover that is safe for your dinner table, dinner guests, and dinner ambiance

Verbotomy

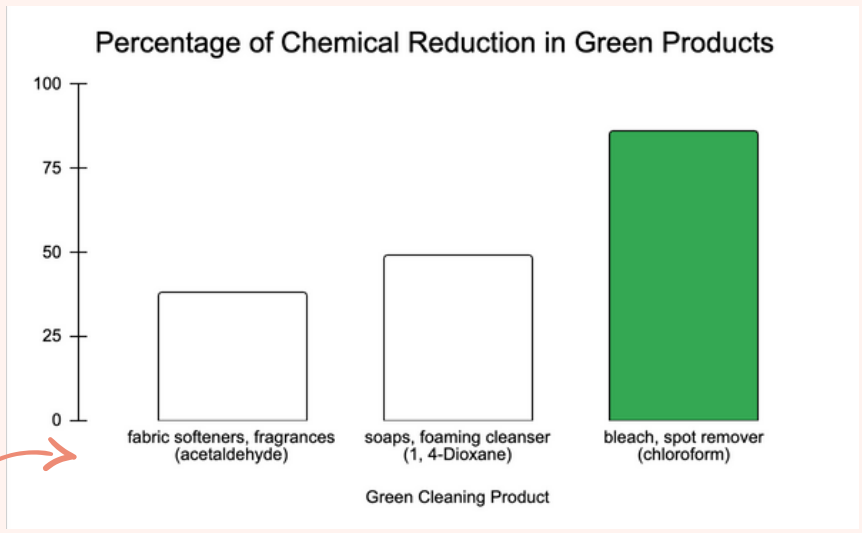
Are you nervous about a knocked over drink ruining your tablecloth but tired of searching for safe and easy cleaning products?

DID YOU KNOW?

Other cleaning products might remove stains, but they can also add toxic chemicals to your dinner table and the air around you. Health experts warn that common cleaning toxins can irritate the skin, eyes, and respiratory system, and many companies disguise these toxins as "dyes" and fragrances". Stain Away has no added chemicals or artificial ingredients and is made from NaCl minerals found right here in the U.S. so it is the perfect way to skip the toxins.

DID YOU KNOW?

If you do not remove stains from your tablecloth quickly, the substance spilled will fuse to the cloth, forming a chemical bond.



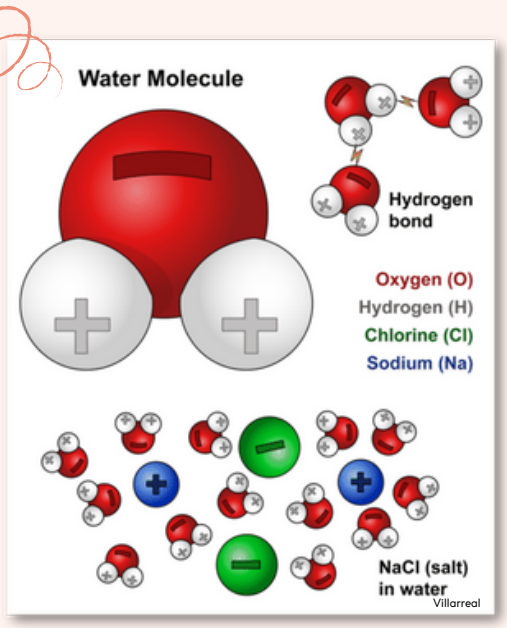
Green stain removers have the highest reduction of chemicals in comparison to their non-green counterparts than other cleaning supplies

DISSOLVING IN WATER

NaCl is already a part of your body! Both in the body and in the exterior world, NaCl dissolves in water, making it hydrophilic. While both NaCl and H2O have positive and negative changes, H2O is a covalent bond because electrons are shared equally between the atoms. NaCl is an ionic bond because the electrons are not shared equally between the atoms. Because the ionic bonds of NaCl are not as strong as the covalent bonds of H2O, salt dissolves in water.

Stains Bond

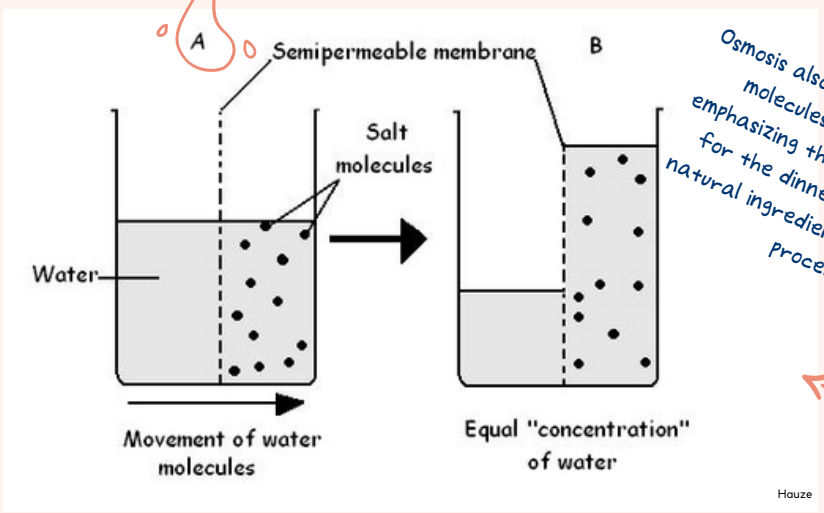
Like chemical bonds, hydrogen bonds are also present between tablecloth fabric and molecules in the spilled substance. Hydrogen bonds are bonds between atoms of the same molecule, most often between water and water. Water molecules have positively charged atoms and negative charged oxygen atoms. This means the positive hydrogens on one water molecule can bond to the negative oxygen atoms on another water molecule.



This figure shows the molecular formula of water. Water molecules separate the Na from the Cl, making the salt dissolve

DIFFUSION

When you (or a guest) spills something on your tablecloth, the molecules of the spill are free to flow. This free movement of molecules is known as diffusion. During diffusion, molecules move from areas where they are highly concentrated to areas where they are less concentrated. So when you change the concentration of molecules in an area, the molecules will move. Osmosis is a specific type of diffusion where the diffusing molecules are water. Water will move from an area where it is highly concentrated to an area where it is less concentrated. This entails that an area with lots of other dissolved particles will have fewer water molecules, making it a place that water wants to go



This figure depicts adding NaCl to an area makes that area have a high concentration of NaCl and a low concentration of water. Water will then move to the area with higher NaCl concentration because the surrounding cells have more dissolved particles. Pouring NaCl on a stain changes the concentrations of the spill, moving water towards the NaCl.

Osmosis also occurs within the water molecules in the human body, emphasizing that Stain Away is safe for the dinner table with its all natural ingredients and all natural processes.

Get 20% off your first 16 oz container for only $11.29
For more information on this product, contact biomedpharmawustl@gmail.com