



**ACADEMIC UNDERGRADUATE STUDIES DIVISION**  
**SECOND SEMESTER 2019-2020**

**Course Handout Part II**

Date:

06/01/2020

In addition to part -I (general handout for all courses appended to the time table) this portion gives further specific details regarding the course.

**Course Number** CHEM F110

**Course Title** CHEMISTRY LABORATORY

**Instructor-in-Charge** N. RAJESH

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**Team of Instructors** of Amit Nag, Durba Roy, Jayanty Subbalakshmi, K. Sumithra, KVG Chandra Sekhar, Himanshu Aggarwal, R.Krishnan, Chanchal Chakravarty

**1. Course Description:** This course is based on laboratory experiments in different fields of chemistry. The course is conducted for first year students.

**2. Scope and Objective:** The main objective of this course is to educate the students with different aspects of experiments in chemistry. The students will carry out a set of experiments that will expose them to various experimental techniques like preparation of standard solution, chemical synthesis, filtrations, recrystallization, and operating of different scientific equipment for collecting data and analysis. Experiments in Organic and Inorganic chemistry will include synthesis and analysis of different compounds. Experiments in Physical chemistry will include pH metry, conductometry, spectrophotometry and chemical kinetics experiments.

**3. Text Book(s): Chemistry Laboratory Manual**

**4. Reference Book:** Reference sources for each experiment will be specified as necessary.

**5. Course Plan:** Students will perform selected experiments in Organic, Inorganic & Physical chemistry

**List of Experiments:**

**1st Cycle**

Lab 1	Synthesis and recrystallization of dibenzalacetone
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Lab 2	Determination of the concentration equilibrium constant ( $K_c$ ) of the reaction
Lab 3	Dissociation constant of a weak electrolyte by conductometry
Lab 4	Preparation of potassium oxalate complex and iron acetyl acetonate complex
Lab 5	Synthesis of glucosazone
<b>2<sup>nd</sup> Cycle</b>	
Lab 6	Determination of the activation energy of hydrogen peroxide - potassium iodide clock reaction
Lab 7	Determination of pH curve of an acid-base titration
Lab 8	Estimation of copper by Iodometry
Lab 9	Determination of the unknown strengths of a strong and a weak acid solution by conductometric titration
Lab 10	Make up for experiments
Lab 11	Lab exam (practical exam)

## 6. Evaluation:

Component of Component	Duration	Weightage (%)	Date & Time	Nature
Lab performance and Calculation and data interpretation *		70		OB
Continuous **				
Lab exam book	90 min	30		Closed

**\*Attendance and Punctuality, Laboratory Conduct, Record maintenance will be given due importance in the evaluation.**

**\*\* All the experiments are evaluated in the lab and are considered open book**

**7. Make-up policy:** Please ensure that you do not miss any experiment. Make up would be considered for very genuine reasons (*such as institute deputation outside for sports/cultural fest, hospitalization (with suitable documentary proof), marriage ceremony of own brother/sister (not cousins)*) and any other extreme

emergency situations only with prior permission which would be decided by the team of instructors.

Please note for medical reasons: Chief Warden's approval and campus doctor's certificate/prescription is necessary.

**8. Notice:** All notices concerning the course will be displayed on Chemistry Department Notice Board and/or CMS. **Lab manual, White Lab Coat with proper shoes are compulsory.**

**9. Academic Honesty and Integrity Policy:** Academic honesty and integrity are to be maintained by all the students throughout the semester and no type of academic dishonesty is acceptable.

**10. Final grading** will be done on the basis of the overall performance of a student in each of the components as listed in item no. 6. For **mid-semester grading**, progress made by a student up to that point of time would be evaluated.

**Instructor-in-Charge  
Chemistry Laboratory**

