Birla Institute of Technology & Science, Pilani
Hyderabad Campus

FIRSTSEMESTER 2020-2021

Course Handout (Part - II)

Date: 17.08.2020

In addition to part-I (General Handout for all courses appended to the time table) this portion gives further specific details regarding the course.

Course No. : CHEM G552
Course Title : Advanced Inorganic Chemistry
Instructor-in-charge : Dr. R. Krishnan

1. Course Description:
This course deals with various aspects on the coordination chemistry, reactions, kinetics and mechanism; advanced organometallic chemistry, bonding models in inorganic chemistry, inorganics chains, rings, cges and clusters; group theory and its applications to crystal field theory, molecular orbital theory and spectroscopy (electronic and vibrational); inorganic chemistry in biological systems.

2. Scope and Objective of the Course: Theories of coordination chemistry, electronic spectroscopy and magnetism of complexes, organometallic chemistry. Bioinorganic, medicinal and environmental inorganic chemistry.

3. Text Book:T1. " Inorganic Chemistry"Huheey J. E., Keiter, Ellen A., Keiter, Richard L.,Medhi, O.K.;4th ed., Pearson.
T2. . I. Bertini, H. B. Gray, S. J. Lippard, J. S. Valentine, "Bioinorganic Chemistry", Viva, 1998.
Reference Books:R1. "Concise Inorganic Chemistry", Lee, J.D. 5thEdition,Wiley, India Edition.
R2 "Inorganic Chemistry", Shriver, D.F.; Atkins, P.W.; Overton T. L., Rourke, J. P., Weller, M. T., Armstrong, F. A. 4thedition, Oxford.
R3 "Concepts & Models of Inorganic Chemistry" B. Douglas, D. McDaniel and J. Alexander 3rdEdn , wiley India.
R4. E. Ochiai, "Bioinorganic Chemistry: A Survey", Academic Press, 2008.

4. Course Plan:

Lecture No.	Learning Objectives	Topics to be covered	Chapter in the Text Book
1	Coordination chemistry	Bonding VB theory applied to coordination compounds	**T1**: 12.1-12.7

2-3	Structure Nomenclature	Structure – Nomenclature, Coordination numbers 1, 2, 3, 4, 5, 6, 7. Generalization about coordination numbers Isomerism: Linkage and other types of isomerism Chelate effect	T1:Chapter 15: 461-492
4-6	Crystal Field Theory (CFT)	Crystal field splitting; d orbitals in different crystal fields; applications of CFT	T1:Chapter 14: 428-444
7-8	Molecular orbital theory	Molecular orbital theory	T1:Chapter 14: 444-459
9-11	Electronic spectra of complexes, Magnetic properties of complexes	Electronic spectra of complexes, Magnetic properties of complexes	T1:Chapter 15: 461-492
12-15	Reactions of coordinated complexes	Reactions – Nucleophilic substitution reactions, Kinetics, Mechanisms	T1:Chapter 17 (542-569) and Lecture notes
16-20	Organometallic chemistry	The 18-electron rule Metal-carbonyl complexes Nitrosyl complexes Dinitrogens Alkyls Carbenes, Carbynes, Carbides Alkenes Alkynes Metallocenes Catalysis by organometallic compounds Stereo chemically non-rigid molecules	T1:Chapter 18 and Lecture notes
21	Bioinorganic chemistry	Introduction to Bioinorganic Chemistry	T2
22-25	Metal ion storage	Metal ion storage, transport and biomineralization	T2(Ch 1)
26-28	Oxygen carriers	Oxygen carriers: Iron and copper in biological systems	T2(Ch 4)
29-30	Oxygenases	Oxygenation reaction: iron and copper	T2(Ch 5)
31-32	Electron transfer	Electron transfer and redox processes in biological systems	T2(Ch 6)

33-35	Metal-sulfur proteins	Metal-sulfur proteins and metalloenzymes	T2(Ch 7)
36	Photosynthesis	Photosynthesis and artificial photosynthetic models	T2(Ch 7)
37	Vitamin B_{12}	Cobalt in biological systems	T2(Ch 2)
38-39	Metal in medicine	Medicinal inorganic chemistry	T2(Ch 9)
40	Metal-Nucleic acid	Metal-Nucleic acid interactions	T2(Ch 8)
41-42	Environmental inorganic chemistry	Toxicity of metal ions and environmental bioinorganic chemistry	R4

5. Evaluation scheme:

Component	Duration	Weightage (%)	**Date and Time**	Remarks
Test 1	30 min	15		**Open Book**
Test 2	30 min	15		**Open Book**
Test 3	30 min	15		**Open Book**
* Seminars		20	Continuous	**Open Book**
Comprehensive Examination	120 min	35	03/12 FN	**Open Book**

***** There will be **two seminars on advanced research topics.**

6. Chamber consultancy hour: Friday 11:00 AM to 12:00 PM.

7. Notices: Notices concerning the course will be displayed on the Chemistry Group notice board and/or CMS.

8. Academic Honesty and Integrity Policy: Academic honesty and integrity are to be maintained by all the students throughout the semester and no type of academic dishonesty is acceptable.

9. Make-up-policy: May be granted only for genuine cases.

Instructor-in-charge

R. Krishnan