

## Second Semester 2022-23 Course Handout Part II

Date:

13/03/2023

In addition to part -I (general handout for all courses appended to the time table) this portion gives further specific details regarding the course.

**Course Number** CHEM F110

**Course Title** CHEMISTRY LABORATORY

**Instructor-in-Charg** Durba Roy

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**Team** of Dr. Arijit Mukherjee, Prof. Balaji Gopalan, Dr. Himanshu

Instructors Aggarwal, Prof. J. Subbalakshmi, Prof. K. Sumithra, Prof. N.

Rajesh, Dr. Nilanjan Dey, Prof. G. Ramakrishnan, Dr. Tanmay

Chatterjee, Prof. Sounak Roy

- **1. Course Description**: This course is based on laboratory experiments in different fields of chemistry. The course is conducted for first year students.
- 2. Scope and Objective: The main objective of this course is to educate the students with different aspects of experiments in chemistry. The students will carry out a set of experiments that will expose them to various experimental techniques like preparation of standard solution, chemical synthesis, filtrations, recrystallization, and operating of different scientific equipment for collecting data and analysis. Experiments in Organic and Inorganic chemistry will include synthesis and analysis of different compounds. Experiments in Physical chemistry will include pH metry, conductometry, spectrophotometry and chemical kinetics experiments.
- **3. Text Book(s):** Chemistry Laboratory Manual
- 4. Reference Book: Reference sources for each experiment will be specified as necessary.
- **5. Course Plan:** Students will perform selected experiments in Organic, Inorganic & Physical chemistry

## **List of Experiments:**

## 1st Cycle

Lab 1	Synthesis and recrystallization of dibenzalacetone			
Lab 2	Determination of the concentration equilibrium constant (K <sub>c</sub> ) of the reaction			
Lab 3	Dissociation constant of a weak electrolyte by conductometry			
Lab	Preparation of potassium oxalate complex			
4a				
Lab	Preparation of iron acetyl acetonate complex			
4b				
Lab 5	Synthesis of glucosazone			
2 <sup>nd</sup> Cycle				
Lab 6	Determination of the activation energy of hydrogen peroxide - potassium iodide			
	clock			
	reaction			
Lab 7	Determination of pH curve of an acid-base titration			
Lab 8	Estimation of copper by Iodometry			
Lab 9	Determination of the unknown strengths of a strong and a weak acid solution by			
	conductometric titration			
Lab	Make up for experiments for both first and second cycles.			
10				
Lab	Lab exam (practical exam)			
11				

## 6. Evaluation:

o. Evaluation.					
Component	Duration	Weightage (%)	Date & Time		
Lab performance ar	nd				
Calculation and data	a interpretation *	70	Open-		
Book, Continuous**					
Lab exam	90 min	30	Practical		
examination					

<sup>\*</sup>Attendance and Punctuality, Laboratory Conduct, Record maintenance will be given due importance in the evaluation.

7. Make-up policy: Please ensure that you do not miss any experiment. Only 1 make-up will be given for the entire course as per the following:

<sup>\*\*</sup> All the experiments are evaluated in the lab and are considered open book

Make up would be considered for very **genuine reasons** (**such as institute deputation outside for sports/cultural fest, hospitalization (with suitable documentary proof)**, and any other **extreme emergency situations only with prior permission** which would be decided by the team of instructors. **Please note for medical reasons**: **Chief Warden's approval and campus doctor's certificate/prescription are necessary**.

- **8.** Notice: All notices concerning the course will be displayed on Chemistry Department Notice Board and/or CMS. Lab manual, White Lab Coat with proper shoes covering the entire feet are compulsory.
- **9. Academic Honesty and Integrity Policy:** Academic honesty and integrity are to be maintained by all the students throughout the semester and no type of academic dishonesty is acceptable.
- **10. Final grading** will be done on the basis of the overall performance of a student in each of the components as listed in item no. 6. For **mid-semester grading**, progress made by a student up to that point of time will be evaluated.

Instructor-in-Charge Chemistry Laboratory