



**Birla Institute of Technology & Science, Pilani**  
Hyderabad Campus

**ACADEMIC UNDERGRADUATE STUDIES DIVISION**

**Second Semester 2022-23**

**Course Handout Part II**

Date: 14/03/2023

In addition to part -I (general handout for all courses appended to the time table) this portion gives further specific details regarding the course.

<b>Course Number</b>	CHEM F110
<b>Course Title</b>	CHEMISTRY LABORATORY
<b>Instructor-in-Charge</b>	Durba Roy
<b>Team of Instructors</b>	Dr. Arijit Mukherjee, Prof. Balaji Gopalan, Dr. Himanshu Aggarwal, Prof. J. Subbalakshmi, Prof. K. Sumithra, Prof. N. Rajesh, Dr. Nilanjan Dey, Prof. G. Ramakrishnan, Dr. Tanmay Chatterjee, Prof. Sounak Roy

**1. Course Description:** This course is based on laboratory experiments in different fields of chemistry. The course is conducted for first year students.

**2. Scope and Objective:** The main objective of this course is to educate the students with different aspects of experiments in chemistry. The students will carry out a set of experiments that will expose them to various experimental techniques like preparation of standard solution, chemical synthesis, filtrations, recrystallization, and operating of different scientific equipment for collecting data and analysis. Experiments in Organic and Inorganic chemistry will include synthesis and analysis of different compounds. Experiments in Physical chemistry will include pH metry, conductometry, spectrophotometry and chemical kinetics experiments.

**3. Text Book(s):** Chemistry Laboratory Manual

**4. Reference Book:** Reference sources for each experiment will be specified as necessary.

**5. Course Plan:** Students will perform selected experiments in Organic, Inorganic & Physical chemistry

**List of Experiments:**

**1st Cycle**

Lab 1	Synthesis and recrystallization of dibenzalacetone
Lab 2	Determination of the concentration equilibrium constant ( $K_c$ ) of the reaction
Lab 3	Dissociation constant of a weak electrolyte by conductometry

Lab 4a	Preparation of potassium oxalate complex
Lab 4b	Preparation of iron acetyl acetonate complex
Lab 5	Synthesis of glucosazone
<b>2<sup>nd</sup> Cycle</b>	
Lab 6	Determination of the activation energy of hydrogen peroxide - potassium iodide clock reaction
Lab 7	Determination of pH curve of an acid-base titration
Lab 8	Estimation of copper by Iodometry
Lab 9	Determination of the unknown strengths of a strong and a weak acid solution by conductometric titration
Lab 10	Make up for experiments for both first and second cycles.
Lab 11	Lab exam (practical exam)

#### 6. Evaluation:

Component	Duration	Weightage (%)	Date & Time
Lab performance and Calculation and data interpretation *		70	Open-Book, Continuous**
Lab exam	90 min	30	Practical examination

\***Attendance and Punctuality, Laboratory Conduct, Record maintenance** will be given due importance in the evaluation.

\*\* All the experiments are evaluated in the lab and are considered open book

#### 7. **Make-up policy: Please ensure that you do not miss any experiment. Only 1 make-up will be given for the entire course as per the following:**

Make up would be considered for very **genuine reasons (such as institute deputation outside for sports/cultural fest, hospitalization (with suitable documentary proof), and any other extreme emergency situations only with prior permission** which would be decided by the team of instructors. **Please note for medical reasons: Chief Warden's approval and campus doctor's certificate/prescription are necessary.**

#### 8. **Notice:** All notices concerning the course will be displayed on Chemistry Department Notice Board and/or CMS. **Lab manual, White Lab Coat with proper shoes covering the entire feet are compulsory.**

#### 9. **Academic Integrity Policy:** It is expected that in compliance with institute rules and regulations, academic integrity should be adhered to in all the evaluation components. No type of academic dishonesty is acceptable and malpractice in any form will have serious implications.

#### 10. **Final grading** will be done on the basis of the overall performance of a student in each of the components as listed in item no. 6. For **mid-semester grading**, progress made by a student up to that point of time will be evaluated.

**Instructor-in-Charge**  
**Chemistry Laboratory**