

**SECOND** 

SEMESTER: 2018-2019 Course Handout (Part-II)

07-01-2019

In addition to Part I (General Handout for all courses appended to the time table) this portion gives further specific details regarding the course).

Course No.: BIO F244

**Course Title:** Instrumental Methods of Analysis **Instructor-in-charge:** TRINATH JAMMA

**Team of Instructors:** Prof. Suman Kapur, Prof. Ramakrishna Vadrevu, Dr. Trinath Jamma, Naresh Patnaik,

Pranay Amruth Maroju, Rolly Kumari

# 1. Scope & Objective of the Course:

The advent of dedicated bio-instruments and computers has facilitated an explosive progress in the instrumental methods of analysis in biology. Large number of data points, whether they are physicochemical or biological, can be collected, stored, manipulated and analyzed at a high precision with the help of modern sophisticated instruments having high sensitivity, selectivity, and extremely low detection limit. This course aims at providing a sufficient background of these instruments, their handling and application, in the field of molecular biology, structural biology and biotechnology.

#### 2. Text Book:

- **T1.** "Instrumental Methods of Analysis", Sivasankar *et. al.*, Oxford University Press, New Delhi, 1<sup>st</sup> ed., 2012.
- T2. Lab Manual for PHA C391 IMA. Mahesh et. al., 2008

#### Reference Book

- **R1**. "Principles of Instrumental Analysis", Skoog *et. al.*, Harcourt Asia, 5<sup>th</sup> ed., 2001.
- **R2** "Instrument Methods of Analysis. Williard *et al.*, CBS Publication, New Delhi, 7<sup>th</sup> edition, 1998.
- **R3** "Handbook of Analytical instruments", R.S. Khandpur, Tata Mc Graw-Hill, 2<sup>nd</sup> edition, 2006.

### 3.a) Course Plan:

No	Learning Objectives	Topic to be covered	Ref. to the Book		
1	Spectroscopy	py Characteristic of atomic and molecular spectroscopy			
			R-1		
2	Atomic Spectroscopy	Infrared Spectroscopy	T-1 (Ch. 7)		
		Atomic Absorption Spectroscopy, Flame Emission	R-1, R-3		
		Spectroscopy			
3	Molecular	Visible and Ultraviolet Spectroscopy, Fluorescence	T-1 (Ch. 8)		
	Spectroscopy	Spectroscopy	R-1, R-3		
4	Optical Spectroscopy	Polarimetry and Circular Dichroism	T-1 (Ch. 4)		
5	Electrophoresis	SDS-PAGE	T-1 (Ch. 14)		
			R-1		
6	Chromatography	High-Performance liquid Chromatography	T-1 (Ch. 13)		
			R-1, R-3		
7	Molecular Biology	ELISA, PCR	Class notes		
	Techniques				
8	Mass Spectrometry	Basic principles and applications	T-1 (Ch. 10)		

### 3. b) Lab Components:

# **Experiments**

- Exp 1: Preparation of buffer solutions and measurement of pH using a pH meter
- Exp 2: Flame photometric analysis of alkaline earth metals in biological samples
- Exp 3: Qualitative and quantitative analysis of biomolecules using UV spectroscopy

- Exp 4: Fluorescent spectroscopy; total intensity and quenching measurements
- Exp 5: Infrared spectroscopy; identification of functional groups in bio molecules
- Exp 6: Separation of proteins using SDS-polyacrylamide gel electrophoresis
- Exp 7: Identification of amino acids using TLC
- Exp 8: Separation of molecules using HPLC
- Exp 9: Measurement of molecular weight by mass spectrometry
- Exp 10: Amplification of DNA using PCR
- Exp 11: Atomic absorption spectrophotometric analysis of elements
- Exp 12: Detection of antigen using an ELISA reader
- Exp 13: Polarimeteric/ Circular Dichroic analysis of samples possessing optical activity

#### Note:

- Text book 2 will be used for experimental details
- Extra reading material will be provided to the students, if required.
- Minor changes are possible subject to availability of chemicals/ Instructors/Instruments

# 4. Evaluation Scheme:

No	Evaluation Component	Duration	Weightage	Date & Time	Nature of Component
1.	Laboratory Evaluation-1: Evaluation will be based on I <sup>st</sup> cycle experiments, punctuality, records, and participation.	-	15%		ОВ
2.	Lab Quiz-1	-	10%		СВ
3.	Midsem	1 hr	15%	15/3 1.30 -3.00 PM	СВ
4.	Laboratory Evaluation-2: Evaluation will be based on 2 <sup>nd</sup> cycle experiments, attendance, records, participation and viva.	-	15%		ОВ
5.	Lab Quiz-2	-	10%		CB
6.	Lab comprehensive examination	-	20%		OB
7.	Comprehensive Exam	2 hr	15%	11/05 FN	СВ

# 5. Guide to Writing Lab Reports:

The report must have to be written on hardbound, practical files. It should include the followings:

- a. Objective of the experiments,
- b. Theory on which the experiment is based,
- c. Steps in the experimental procedure,
- d. Results including all observations, e. Discussion and Conclusion. f. Precautions
- **6. Chamber Consultation Hours:** To be announced in the Class.
- **7. Make-up Policy:** Make-up will be granted only in the case of hospitalization and after submission of medical certificate through the proper process. *No makeup for Laboratory evaluation*.
- **8. Academic Honesty and Integrity Policy**: Academic honesty and integrity are to be maintained by all the students throughout the semester and no type of academic dishonesty is acceptable.

#### **IMPORTANT NOTE:**

- Lab Coat is mandatory for performing experiments.
- You will not be allowed to do the experiments without your lab record.
- You must enter into lab on time; in case of delay, you may lose your evaluation and marks.