

## 4 atoms

Point group	Shape	Example	Dihedrals
$T_d$	Regular Pyramidal	$P_4$	$\cos^{-1}\left(\frac{1}{3}\right)$
$C_{3v}$	Pyramidal	$NH_3$	$\cos^{-1}\left(\frac{\cos\theta(1-\cos\theta)}{\sin^2\theta}\right)$
$C_s$	Pyramidal	$Cl_2OS$	$\cos^{-1}\left(\frac{\cos\theta-\cos^2\varphi}{\sin^2\varphi}\right)$
$C_{2v}$	Pyramidal	$H_2Si_2$	$\cos^{-1}\left(\frac{\cos\theta-\cos^2\varphi}{\sin^2\varphi}\right)$
$C_2$	Pyramidal	$H_2O_2$	$\cos^{-1}\left(\frac{\cos\theta-\cos^2\varphi}{\sin^2\varphi}\right)$
$C_1$	Pyramidal	$H_2OS$	$\cos^{-1}\left(\frac{\cos\theta-\cos\varphi\cos\phi}{\sin\varphi\sin\phi}\right)$
$D_{3h}$	Planar	$BH_3$	0 or 180
$C_s$	Planar	$HNSi_2$	0 or 180
$D_{2h}$	Planar	$Br_2Na_2$	0 or 180
$C_{2h}$	Planar	$(HI)_2$	0 or 180
$C_{2v}$	Planar?	$O_2S_2$	0 or 180
$C_{2v}$	Planar?	$CFO_2$	0 or 180
$C_{\infty v}$	Linear	$CAgIO$	0 or 180
$D_{\infty h}$	Linear	$C_2H_2$	0 or 180
$C_{2v}$		$(Ar_2)(HCl)$	
$C_s$		$(HCl)_2$	
$C_2$		$(HF)_2$	

## 5 atoms

Point group	Shape	Example	Dihedrals
$T_d$		$O_4Os$	
$C_{3v}$		$ClH_2Si$	
$C_{3v}$		$KrO_3S$	
$C_{2v}$		$Cl_4Te$	
$C_{2v}$		$C_3H_2$	
$C_{2v}$		$H_2I_2Si$	
$C_{2v}$		$H_2O_2Si$	
$C_{2v}$		$(CO_2)(N_2)$	
$C_s(\text{syn})$		$BFH_2O$	
$C_s(\text{syn-anti})$		$CH_2O_2$	
$C_s$		$CN_4$	
$C_s$		$C_2H_3$	
$C_s$		$BF_2HO$	
$C_s$		$CH_3OH$	
$C_s$		$ClH_2NaO$	
$C_s$		$HNO_3$	
$C_s$		$H_2O_2Si$	
$C_s$	Non-planar	$CH_2N_2$	
$C_s$		$CBrF_2N$	
$C_s$		$CHClF_2$	
$C_s$		$(CO)(N_2O)$	
$C_s(\text{effective})$		$ArH_3N$	
$C_s(\text{effective})$		$(SO_2)(CO)$	
$C_s(\text{assumed})$		$H_2N_2O$	Given!
$C_{\infty v}$	Linear	$CH_3N$	
	Linear	$(CO)(HCN)$	
$D_{\infty h}$	Linear	$C_5$	
$D_{\infty h}$	Linear	$C_3Ge_2$	
$D_{\infty h}$	Linear	$C_3H_2$	
	Quadi-linear	$C_3O_2$	