

**Question 1:**

Convert the following quantities to  $\frac{\text{m}}{\text{s}}$  (Remember to refer to the conversion table)

1.  $6004 \frac{\text{ft}}{\text{h}}$

2.  $312300 \frac{\text{cm}}{\text{h}}$

3.  $5 \frac{\text{km}}{\text{h}}$

4.  $10^3 \frac{\text{mi}}{\text{h}}$

5.  $566 \frac{\text{in}}{\text{min}}$

**Question 2:**

At the University of Waterloo, students may begin to feel nervous if during an exam, someone manages to complete it after 5 minutes. Lets say the fastest problem solver in the exam room is student  $X$ , who solves problems at a rate of  $120 \frac{\text{problems}}{\text{h}}$ . Determine whether or not the students in the exam room will feel nervous or not, completely justify your answer.

**Question 3:**

Daniel has recently ran into a potentially lucrative opportunity, he happened to come across 60 carrots of gold. He wants to know how many coffees he can order. He knows the following information,

- 1 carrot of gold = 0.5 brits
- 1 brit = 6000 USD
- 1 USD = 1.25 CAD
- 1 coffee = 2 CAD

Help him determine the number of coffees he can order.

**Question 4:**

**(CHALLENGE WARNING)** A mechanical engineering student over at the University of Waterloo wants to know the amount of energy he will need in order to weld a 6 rods of steel. He knows that each rod of steel has a density of  $650 \text{ kg/m}^3$  and a Calorific Value of  $6 \text{ kWh/kg}$ . Determine the amount of energy **(In Kilo Joules)** 6 rods of steel will require. Make note of the following,

- 1 rod of steel has a volume of  $100 \text{ ft}^3$
- $1 \text{ BTU} = 2.931 \times 10^{-4} \text{ kWh}$
- $1 \text{ J} = 9.4782 \times 10^{-4} \text{ BTU}$
- $1 \text{ kJ} = 1000 \text{ J}$