Physics 262 Fall 2010 Exam #6 No scantron today! Name SoLUTION 3

 $p = \frac{h}{\lambda}$ If you need anything else, just ask!

Please keep your answers to a couple sig figs.

1] A wavefunction for a particle is given by:

 $\psi(x) = Ce^{-\kappa x}$, for x>0. The wavefunction is zero for x<0.

 $\kappa = 40 \text{ nm}^{-1}$. What is the value of C to properly normalize this wavefunction? (in nm^{-1/2})

$$\int_{0}^{c^{2}} e^{-2KX} dY = 0 \Rightarrow \frac{C}{2K} = 1 \qquad C = \sqrt{2K} = \sqrt{80} = 8.9 \text{ nm}^{-1}$$

2]. What is the probability (in percent) the particle will be found farther than 40 nm from x=0?

0.

3]. A particle is in the mixed wavefunction $\psi = C(4\psi_1 + 2\psi_2)$, where ψ_1 and ψ_2 are properly normalized stationary states of the potential. What is C, for proper normalization of the mixed wave?

$$160^{2} + 40^{2} = 1$$
 $c^{2} = \frac{1}{20} = 0.22$

4] If a measurement is made to determine what state this particle is in, which state is more probable?

$$(A] \psi_1$$
 B] ψ_2 C] Both states are equally probable

5] By what factor is the more probable state more probable than the less probable state?

4x

6] What is the probability (in percent) that the particle will be observed in state ψ_1 ?

80%

An infinite potential well has a different depth on its left and right sides, as shown.

Suppose the n=3 state has an energy that is 16 eV above the bottom of the well. The shelf of the well is 12 eV above the bottom.

- 7] How many "nodes" (zero crossings, not counting the ends) are there for the n=3 wavefunction?
- 8] If a free particle had the same energy as the n=3 state, in which part of the well would it have a larger momentum?

 A], eft side

 B] right side

 C] same in both sides
- 9] By what factor is the momentum larger, in the side where it is larger? $2.4 \text{ M} \, d$.

Sketch the wavefunction for the n=3 state, using its energy as the $\psi\!=\!0$ line.

- 10] A measurement of the position of a particle in the n=3 state is made. Where is the particle most likely to be found?

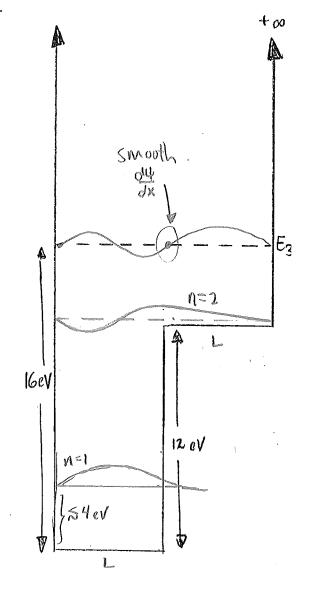
 A] left side of well

 B] right side of well

 C] both sides are equally likely
- 12] Is the actual energy of the n=1 state a little higher or lower than the answer to 1 ? A higher (B) lower C] 1 ? is exact

Sketch the wavefunction for the n=1 state using its energy as the ψ =0 line.

Extra credit: Sketch the n=2 state using its energy as the ψ =0 line. Is its energy higher or lower than the shelf?



Draw the n=2 Bohr wave on the orbit below.

13] Recall that Bohr orbits are consistent with the deBroglie wavelengths. Suppose a Roulette ball (mass 0.04 kg) moves around a roulette wheel (radius 0.3 m) once per second. In a bizarre alternate universe, this roulette ball is in the n=2 Bohr orbit. What is h?

Sketches will count for 3 points (total).

$$MV = M \cdot \frac{2\pi r}{T} = \frac{h}{\lambda}$$

$$h: M \cdot \frac{2\pi^2 r^2}{T} = 0.071 \sigma.s.$$

