

# **BITSAT**

## **Solved Paper 2011**

### **Instructions**

1. There are 150 questions in all. The number of questions in each part is as follows

**Subjects**

Part I (Physics)

Part II (Chemistry)

Part III

(a) English Proficiency

(b) Logical Reasoning

Part IV (Mathematics)

**No. of Questions**

1–40

41–80

81–95

96–105

106–150

2. All questions are multiple choice questions with four options, only one being correct.  
3. Each correct answer fetches 3 marks while incorrect answer fetches –1 mark.

## **Part I**

### **Physics**

1. Suppose the gravitational force varies inversely as the  $n$ th power of distance. Then the time period of a planet in circular orbit of radius  $R$  around the sun will be proportional to
- (a)  $R^{(n+1)/2}$  (b)  $R^{(n-1)/2}$   
(c)  $R^n$  (d)  $R^{(n-2)/2}$
2. Two wires are made of the same material and have the same volume. However wire 1 has cross-sectional area  $A$  and wire 2 has cross-sectional area  $3A$ . If length of wire 1 increased by  $\Delta x$  on applying force  $F$ , how much force is needed to stretch wire 2 by the same amount?
- (a)  $4F$  (b)  $6F$   
(c)  $9F$  (d)  $F$
3. The satellite of mass  $m$  revolving in a circular orbit of radius  $r$  around the earth has kinetic energy  $E$ . Then its angular momentum will be
- (a)  $\sqrt{\frac{E}{mr^2}}$  (b)  $\frac{E}{2mr^2}$   
(c)  $\sqrt{2Emr^2}$  (d)  $\sqrt{2Emr}$
4. A galvanometer of resistance  $100\Omega$  gives full scale deflection with  $0.01$  A current. How much resistance should be connected in parallel to convert it into an ammeter of range  $10$  A?
- (a)  $0.100\Omega$  (b)  $1.00\Omega$   
(c)  $10.00\Omega$  (d)  $100.00\Omega$
5. A car is moving on a circular road of diameter  $50$  m with a speed of  $5$  m/s. It is suddenly accelerated at a rate of  $1$  m/s<sup>2</sup>. If the mass of the car is  $500$  kg, then the net force acting on the car is
- (a)  $5$  N (b)  $1000$  N  
(c)  $500\sqrt{2}$  N (d)  $\frac{500}{\sqrt{2}}$  N

6. Hard X-rays for the study of fractures in bones should have a minimum wavelength of  $10^{-11}$  m. The accelerating voltage for electrons in X-ray machine should be  
 (a)  $< 124$  kV  
 (b)  $> 124$  kV  
 (c) between 60 kV and 70 kV  
 (d)  $= 100$  kV
7. Natural length of a spring is 60 cm and its spring constant is 4000 N/m. A mass of 20 kg is hung from it. The extension produced in the spring is (Take  $g = 9.8 \text{ m/s}^2$ )  
 (a) 4.9 cm (b) 0.49 cm  
 (c) 9.4 cm (d) 0.94 cm
8. A point source of light is placed 4 m below the surface of water of refractive index  $\frac{5}{3}$ . The minimum diameter of a disc, which should be placed over the source, on the surface of water to cut-off all light coming out of water is  
 (a) infinite (b) 6 m  
 (c) 4 m (d) 3 m
9. What is the maximum acceleration of the particle doing the SHM?  
 $y = 2 \sin \left[ \frac{\pi t}{2} + \phi \right]$ , where 2 is in cm  
 (a)  $\frac{\pi}{2} \text{ cm/s}^2$  (b)  $\frac{\pi^2}{2} \text{ cm/s}^2$
- (a) 10 days (b) 20 days  
 (c) 40 days (d) None of these
13. The velocity of efflux of a liquid through an orifice in the bottom of the tank does not depend upon  
 (a) size of orifice  
 (b) height of liquid  
 (c) acceleration due to gravity  
 (d) density of liquid
14. A neutron with velocity  $v$  strikes a stationary deuterium atom, its KE changes by a factor of  
 (a)  $\frac{15}{16}$  (b)  $\frac{1}{2}$   
 (c)  $\frac{2}{1}$  (d) None of these
15. The Poisson's ratio of a material is 0.5. If a force is applied to a wire of this material, there is a decrease in the cross-sectional area by 4%. The percentage increase in the length is  
 (a) 1% (b) 2%  
 (c) 2.5% (d) 4%
16. Lenz's law of electromagnetic induction corresponds to the  
 (a) law of conservation of charge  
 (b) law of conservation of energy  
 (c) law of conservation of momentum  
 (d) law of conservation of angular



19. The force constant of a spring gun is 50 N/m. If a ball of 20 g be shoot by the gun so, that its spring is compressed by 10 cm, the velocity of the ball is  
 (a) 5 m/s (b) 15 m/s  
 (c) 25 m/s (d) 20 m/s
20. 1 g of water (volume  $1 \text{ cm}^3$ ) becomes  $1671 \text{ cm}^3$  of steam when boiled at a pressure of 1 atm. The latent heat of vapourisation is 540 cal/g, then the external work done is  
 ( $1 \text{ atm} = 1.013 \times 10^5 \text{ N/m}^2$ )  
 (a) 499.7 J (b) 40.3 J  
 (c) 169.2 J (d) 128.57 J
21. A cube has a side of length  $1.2 \times 10^{-2} \text{ m}$ . Calculate its volume.  
 (a)  $1.7 \times 10^{-6} \text{ m}^3$  (b)  $1.73 \times 10^{-6} \text{ m}^3$   
 (c)  $1.70 \times 10^{-6} \text{ m}^3$  (d)  $1.732 \times 10^{-6} \text{ m}^3$
22. A ball is dropped from height  $h$  and another from  $2h$ . The ratio of time taken by the two balls to reach the ground is  
 (a)  $1 : \sqrt{2}$  (b)  $\sqrt{2} : 1$   
 (c)  $2 : 1$  (d)  $1 : 2$
23. The linear momentum  $p$  of a body moving in one dimension varies with time  $t$  according to the equation  $p = a + bt^2$ , where  $a$  and  $b$  are positive constant. The net force acting on the body is  
 (a) a constant  
 (b) proportional to  $t^2$   
 (c) inversely proportional to  $t$   
 (d) proportional to  $t$
24. Which of the following is not an example of perfectly inelastic collision?  
 (a) A bullet fired into a block, if bullet gets embedded into block  
 (b) Capture of an electron by an atom  
 (c) A man jumping onto a moving boat  
 (d) A ball bearing striking another ball bearing
25. If a new planet is discovered rotating around sun with the orbital radius double that of the earth, then what will be its time period? (in earth's days)  
 (a) 1032 (b) 1023  
 (c) 1024 (d) 1043
26. If density of earth increases 4 times and its radius becomes half of what it is, our weight will  
 (a) be 4 times its present value  
 (b) be doubled  
 (c) remain same  
 (d) be halved
27. The magnitude of electric field intensity  $E$ , such that an electron placed in it would experience an electrical force equal to its weight, is given by  
 (a)  $mge$  (b)  $\frac{mg}{e}$   
 (c)  $\frac{e}{mg}$  (d)  $\frac{e^2}{m^2}g$
28. The work done in placing a charge of  $8 \times 10^{-18} \text{ C}$  on a capacitor of capacity  $100 \mu\text{F}$  is  
 (a)  $32 \times 10^{-32} \text{ J}$  (b)  $16 \times 10^{-32} \text{ J}$   
 (c)  $3.1 \times 10^{-26} \text{ J}$  (d)  $4 \times 10^{-10} \text{ J}$
29. A steady current flow in a metallic conductor of non-uniform cross-section. The quantity/quantities remaining constant along the whole length of the conductor is/are  
 (a) current, electric field and drift speed  
 (b) drift speed only  
 (c) current and drift speed  
 (d) current only
30. A galvanometer of  $50 \Omega$  resistance has 25 divisions. A current of  $4 \times 10^{-4} \text{ A}$  gives a deflection of one division. To convert this galvanometer into a voltmeter having a range of 25 V, it should be connected with a resistance of  
 (a)  $2500 \Omega$  as a shunt  
 (b)  $2950 \Omega$  as in shunt  
 (c)  $2550 \Omega$  in series  
 (d)  $2450 \Omega$  in series
31. The cyclotron frequency of an electron gyrating in a magnetic field of  $1 \text{ T}$  is approximately  
 (a) 28 MHz  
 (b) 280 MHz  
 (c) 2.8 GHz  
 (d) 28 GHz

32. If  $\mathbf{M}$  is magnetic moment and  $\mathbf{B}$  is the magnetic field, then the torque is given by  
 (a)  $\mathbf{M} \cdot \mathbf{B}$  (b)  $\frac{|\mathbf{M}|}{|\mathbf{B}|}$   
 (c)  $\mathbf{M} \times \mathbf{B}$  (d)  $|\mathbf{M}||\mathbf{B}|$
33. A coil of inductance  $L$  is carrying a steady current  $I$  what is the nature of its stored energy?  
 (a) Magnetic  
 (b) Electrical  
 (c) Both magnetic and electrical  
 (d) Heat
34. Energy conversion in a photoelectric cell takes place from  
 (a) chemical to electrical  
 (b) magnetic to electrical  
 (c) optical to electrical  
 (d) mechanical to electrical
35. If the ionisation potential of helium atom is 24.6 V, the energy required to ionise it will be  
 (a) 24.6 eV (b) 24.6 V  
 (c) 13.6 V (d) 13.6 V
36. Fast neutrons can easily be slowed down by  
 (a) the use of lead shielding  
 (b) passing them through water  
 (c) elastic collision with heavy nuclei  
 (d) applying a strong electric field
37. A film projector magnifies a  $100 \text{ cm}^2$  film strip on a screen. If the linear magnification is 4, the area of the magnified film on the screen is  
 (a)  $1600 \text{ cm}^2$  (b)  $400 \text{ cm}^2$   
 (c)  $800 \text{ cm}^2$  (d)  $6400 \text{ cm}^2$
38. If  $v_m$  is the speed of sound in moist air and  $v_d$  is the speed of sound in dry air under identical conditions of pressure and temperature, then  
 (a)  $v_m > v_d$  (b)  $v_m < v_d$   
 (c)  $v_m = v_d$  (d)  $v_m \cdot v_d = 1$
39. A hot and a cold body are kept in vacuum separated from each other. Which of the following cause decrease in temperature of the hot body?  
 (a) Radiation  
 (b) Convection  
 (c) Conduction  
 (d) Temperature remains unchanged
40. An ideal refrigerator has a freezer at a temperature of  $-13^\circ\text{C}$ . The coefficient of performance of the engine is 5. The temperature of the air (to which heat is rejected) will be  
 (a)  $325^\circ\text{C}$  (b) 325 K  
 (c)  $39^\circ\text{C}$  (d)  $320^\circ\text{C}$

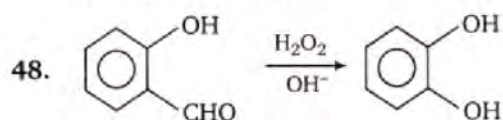
## Part II

### Chemistry

41. The mutual heat of neutralisation of 40 g NaOH and 60 g  $\text{CH}_3\text{COOH}$  will be  
 (a) 57.1 kJ  
 (b) less than 57.1 kJ  
 (c) more than 57.1 kJ  
 (d) 13.7 kJ
42. Which has the smallest size?  
 (a)  $\text{Al}^{3+}$  (b)  $\text{Mg}^{2+}$   
 (c)  $\text{P}^{5+}$  (d)  $\text{Na}^+$
43. The treatment of benzene with *iso*-butene in the presence of sulphuric acid gives  
 (a) *iso*-butylbenzene (b) *tert*-butylbenzene  
 (c) *n*-butylbenzene (d) no reaction
44. Toluene on reaction with N-bromo-succinimide gives  
 (a) *p*-bromomethylbenzene  
 (b) *o*-bromomethylbenzene  
 (c) phenyl bromomethane  
 (d) *m*-bromomethylbenzene
45. Pinacolone is  
 (a) 2, 3-dimethyl-2, 3-butanediol  
 (b) 3, 3-dimethyl-2-butanone  
 (c) 1-phenyl-2-propanone  
 (d) 1, 1-diphenyl-1, 2-ethandiol
46. A synthetic rubber which is resistant to the action of oils, gasoline and other solvents is  
 (a) buna-S (b) polyisoprene  
 (c) neoprene (d) polystyrene



47. Ozone depletion over Antarctica is due to the  
 (a) formation of chlorine nitrate ( $\text{ClONO}_2$ )  
 (b) formation of  $\text{HCl}$   
 (c) formation of  $\text{HOCl}$  and  $\text{Cl}_2$  which are converted back into reactive  $\text{Cl}$  atoms  
 (d) None of the above



This reaction is called

- (a) Reimer-Tiemann reaction  
 (b) Liebermann's nitroso reaction  
 (c) Dakin reaction  
 (d) Leader-Manase reaction
49. Which anion is the weakest base?  
 (a)  $\text{C}_2\text{H}_5\text{O}^-$  (b)  $\text{NO}_3^-$   
 (c)  $\text{F}^-$  (d)  $\text{CH}_3\text{COO}^-$
50.  $K_b$  for water is  $0.52 \text{ K/m}$ . Then  $0.1 \text{ m}$  solution of  $\text{NaCl}$  will boil approximately at  
 (a)  $100.52^\circ\text{C}$  (b)  $100.052^\circ\text{C}$   
 (c)  $101.04^\circ\text{C}$  (d)  $100.104^\circ\text{C}$
51. One mole of  $\text{P}_2\text{O}_5$  undergoes hydrolysis as  

$$\text{P}_2\text{O}_5 + \text{H}_2\text{O} \longrightarrow \text{H}_3\text{PO}_4$$

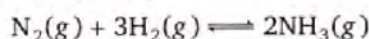
The normality of the phosphoric acid formed is (The volume of solution is  $1 \text{ L}$ .)

- (a) 2 (b) 12  
 (c) 24 (d) 4
52.  $1 \text{ L}$  of a gas is at a pressure of  $10^{-6} \text{ mm}$  of  $\text{Hg}$  at  $25^\circ\text{C}$ . How many molecules are present in the vessel?  
 (a)  $3.2 \times 10^6$  (b)  $3.2 \times 10^{13}$   
 (c)  $3.2 \times 10^{10}$  (d)  $3 \times 10^4$
53. Which of the following has the largest de-Broglie wavelength, given that all have equal velocity?  
 (a)  $\text{CO}_2$  molecule (b)  $\text{NH}_3$  molecule  
 (c) Electron (d) Proton
54.  $1 \text{ g}$  of  $\text{U-235}$  is converted into  $\text{UF}_6$ . The radioactivity of  $\text{UF}_6$  thus obtained is  
 (a) zero  
 (b) less than that of  $1 \text{ g}$  of  $\text{U-235}$   
 (c) more than that of  $1 \text{ g}$  of  $\text{U-235}$   
 (d) same as that of  $1 \text{ g}$  of  $\text{U-235}$

55. In which of the following molecules  $\text{S}$  atom does not assume  $sp^3$  hybridisation?

- (a)  $\text{SO}_4^{2-}$  (b)  $\text{SF}_4$   
 (c)  $\text{SF}_2$  (d)  $\text{S}_8$

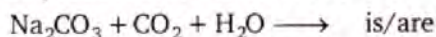
56. For the reaction,



the units of  $K$  are

- (a)  $\text{L mol}^{-1}$  (b)  $\text{L}^2 \text{ mol}^{-2}$   
 (c)  $\text{mol L}^{-1}$  (d) No units
57. A sulphuric acid solution has  $\text{pH} = 3$ . Its normality is  
 (a)  $1/1000$  (b)  $1/200$   
 (c)  $1/2000$  (d)  $1/100$
58. The oxidation number of  $\text{N}$  and  $\text{Cl}$  in  $\text{NOClO}_4$  respectively are  
 (a)  $+2$  and  $+7$  (b)  $+3$  and  $+7$   
 (c)  $-3$  and  $+5$  (d)  $+2$  and  $-7$
59. Pyrolusite is a/an  
 (a) oxide ore (b) sulphide ore  
 (c) carbide ore (d) Not an ore
60. When potassium ferrocyanide crystals are heated with conc.  $\text{H}_2\text{SO}_4$ , the gas evolved is  
 (a)  $\text{SO}_2$  (b)  $\text{NH}_3$   
 (c)  $\text{CO}_2$  (d)  $\text{CO}$

61. The product/s of the reaction,



- (a)  $2\text{NaOH} + \text{CO}_2$  (b)  $\text{Na}_2\text{CO}_3 + \text{H}_2\text{CO}_3$   
 (c)  $2\text{NaHCO}_3$  (d) None of these
62. Which among the following is likely to show geometrical isomerism?  
 (a)  $\text{CH}_3\text{CH}=\text{NOH}$   
 (b)  $\text{CH}_3\text{CH}=\text{CH}_2$   
 (c)  $\text{CH}_2=\text{CH}-\text{CH}=\text{CCl}_2$   
 (d)  $\text{CH}_3\text{C}(\text{Cl})=\text{C}(\text{CH}_3)_2$
63. A fuel has the same knocking property as a mixture of  $70\%$  *iso*-octane (2, 2, 4-trimethylpentane) and  $30\%$  *n*-heptane by volume. The octane number of the fuel is  
 (a) 100 (b) 70  
 (c) 50 (d) 40
64. Sodium carbonate reacts with  $\text{SO}_2$  in aqueous medium to give  
 (a)  $\text{NaHSO}_3$  (b)  $\text{Na}_2\text{S}_2\text{O}_3$   
 (c)  $\text{NaHSO}_4$  (d)  $\text{Na}_2\text{SO}_4$

65. For a given reaction  $t_{1/2} = 1/ka$ . The order of this reaction is  
 (a) 0 (b) 1  
 (c) 2 (d) 3
66. Which of the following compounds will react with two moles of  $\text{CH}_3\text{MgBr}$ ?  
 (a)  $\text{C}_2\text{H}_5\text{COOH}$  (b)  $\text{CH}_3\text{COOH}$   
 (c)  $\text{CH}_3\text{C}\equiv\text{CCH}_3$  (d)  $\text{HC}\equiv\text{C}-\text{CH}_2\text{OH}$
67. The number of polypeptide chains present in a molecule of haemoglobin is  
 (a) four (b) one  
 (c) two (d) three
68. The pentose sugar in DNA and RNA has the  
 (a) open chain structure  
 (b) pyranose structure  
 (c) furanose structure  
 (d) All of the above
69. Which of the following is an artificial edible colour?  
 (a) Saffron (b) Carotene  
 (c) Tetrazine (d) Melamine
70. The number of unpaired electrons in nickel carbonyl is  
 (a) zero (b) one  
 (c) four (d) five
71. The time taken for 90% of a first order reaction to complete is approximately  
 (a) 1.1 times that of half-life  
 (b) 2.2 times that of half-life  
 (c) 3.3 times that of half-life  
 (d) 4.4 times that of half-life
72. The pH of a 0.01 M HCN solution for which  $pK_a$  is 4 is  
 (a) 0.47 (b) 1.2  
 (c) 3.0 (d) 4.0
73. Which of the following does not contain any coordinate bond?  
 (a)  $\text{H}_3\text{O}^+$  (b)  $\text{BF}_4^-$   
 (c)  $\text{HF}_2^-$  (d)  $\text{NH}_4^+$
74. If  $E$  is the energy of the combining atomic orbitals,  $E_1$  and  $E_2$  are the energies of the bonding and anti-bonding molecular orbitals formed, then  
 (a)  $E - E_1 > E_2 - E$   
 (b)  $E - E_1 < E_2 - E$   
 (c)  $E - E_1 = E_2 - E$   
 (d) Any one of the above is possible
75. The equilibrium constant ( $K$ ) for the reaction  $\text{Cu}(s) + 2\text{Ag}^+(aq) \rightarrow \text{Cu}^{2+}(aq) + 2\text{Ag}(s)$  will be [Given,  $E^\circ_{\text{cell}} = 0.46 \text{ V}$ ]  
 (a)  $K_c = \text{Antilog } 15.6$  (b)  $K_c = \text{Antilog } 2.5$   
 (c)  $K_c = \text{Antilog } 1.5$  (d)  $K_c = \text{Antilog } 12.2$
76.  $E^\circ$  for  $\text{Fe}/\text{Fe}^{2+}$  is  $+0.44 \text{ V}$  and  $E^\circ$  for  $\text{Cu}/\text{Cu}^{2+}$  is  $-0.32 \text{ V}$ . Then, in the cell  
 (a) Cu oxidises  $\text{Fe}^{2+}$  ion  
 (b)  $\text{Cu}^{2+}$  oxidises iron  
 (c) Cu reduces  $\text{Fe}^{2+}$  ion  
 (d)  $\text{Cu}^{2+}$  ion reduces Fe
77. Which of the following carbon atoms is most electronegative?  

$$\begin{array}{ccccc} \text{III} & & \text{II} & & \text{I} \\ & \text{CH}_3 & - & \text{CH}_2 & - & \text{C} \equiv \text{CH} \end{array}$$
  
 (a) I  
 (b) II  
 (c) III  
 (d) All are equally electronegative
78. The reaction/method that does not give an alkane is  
 (a) catalytic hydrogenation of alkenes  
 (b) hydrolysis of alkylmagnesium bromide  
 (c) Kolbe's electrolytic method  
 (d) dehydrohalogenation of an alkyl halide
79. Which of the following will yield a mixture of 2-chlorobutene and 3-chlorobutene on treatment with HCl?  
 (a)  $\text{CH}_2=\text{C}=\text{CH}-\text{CH}_3$   
 (b)  $\text{H}_2\text{C}=\text{C}-\underset{\text{CH}_3}{\text{CH}}=\text{CH}_2$   
 (c)  $\text{CH}_2=\text{CH}-\text{CH}=\text{CH}_2$   
 (d)  $\text{HC}\equiv\text{C}-\text{CH}=\text{CH}_2$
80. The well known urinary antiseptic urotropine is formed when formaldehyde reacts with  
 (a)  $\text{NH}_2\text{OH}$  (b)  $\text{NH}_3$   
 (c)  $\text{NH}_2 \cdot \text{NH}_2$  (d)  $\text{C}_6\text{H}_5\text{NH} \cdot \text{NH}_2$



96. Which one number is wrong in the given series?

5, 10, 17, 24, 37

- (a) 10 (b) 17  
(c) 24 (d) 37

97. Find the next two letters in the given series.

E F H K O ?

- (a) T, Z (b) Z, T  
(c) S, Z (d) T, Y

98. If MONKEY is coded as NNOJFX, what will be the code for TARGET?

- (a) ZUSFFS (b) SFFSZU  
(c) UZSFSF (d) UZSFFS

99. Among six friends L, M, N, P, Q and S, each having a different height, N is shorter than Q and P but taller than M. S is shorter than only L. Which of the following represents the tallest among six friends?

- (a) P  
(b) Q  
(c) L  
(d) Cannot be determined

100. Manick is fourteenth from the right end in a row of 40 boys. What is his position from the left end?

- (a) 24<sup>th</sup> (b) 25<sup>th</sup>  
(c) 26<sup>th</sup> (d) 27<sup>th</sup>

101. The missing number in the given figure is



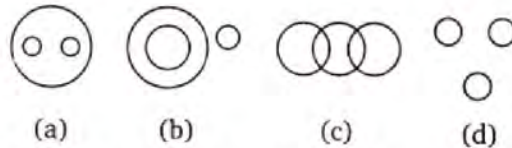
- (a) 13 (b) 15  
(c) 17 (d) 19

102. Select the combination of numbers so that the letters arranged will form a meaningful word.

H N R C A B  
1 2 3 4 5 6

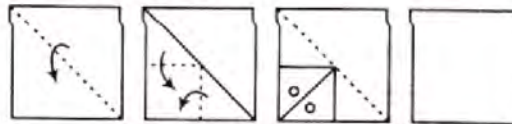
- (a) 2, 5, 3, 4, 1, 6 (b) 3, 5, 6, 4, 1, 2  
(c) 4, 1, 5, 6, 2, 3 (d) 6, 3, 5, 2, 4, 1

103. Which of the given Venn diagrams out of (a), (b), (c) or (d) correctly represents the relationship among the following classes? Rose, Flower, Lotus

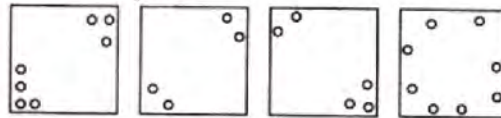


104. A piece of paper is folded and a cut is made as shown below. From the given responses indicate how it will appear when opened?

Question figures

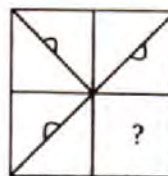


Answer figures



- (a) (b) (c) (d)
105. Which answer figure will complete the question figure?

Question figure



Answer figures



- (a) (b) (c) (d)

## Part IV

### Mathematics

106. The equation of the normal to the circle  $x^2 + y^2 = a^2$  at point  $(x', y')$  will be  
 (a)  $x'y - xy' = 0$  (b)  $xx' - yy' = 0$   
 (c)  $x'y + xy' = 0$  (d)  $xx' + yy' = 0$
107. Equation of the bisector of the acute angle between lines  $3x + 4y + 5 = 0$  and  $12x - 5y - 7 = 0$  is  
 (a)  $21x + 77y + 100 = 0$   
 (b)  $99x - 27y + 30 = 0$   
 (c)  $99x + 27y + 30 = 0$   
 (d)  $21x - 77y - 100 = 0$
108. If  $z = \cos \theta + i \sin \theta$ , then the value of  $z^n + \frac{1}{z^n}$  will be  
 (a)  $\sin 2n\theta$  (b)  $2 \sin n\theta$   
 (c)  $2 \cos n\theta$  (d)  $\cos 2n\theta$
109. If  $\alpha$  and  $\beta$  are the roots of the equation  $x^2 - 2x + 4 = 0$ , then the value of  $\alpha^n + \beta^n$  will be  
 (a)  $i2^{n+1} \sin(n\pi/3)$  (b)  $2^{n+1} \cos(n\pi/3)$   
 (c)  $i2^{n-1} \sin(n\pi/3)$  (d)  $2^{n-1} \cos(n\pi/3)$
110. If  $A = \begin{bmatrix} 3 & 1 \\ -1 & 2 \end{bmatrix}$  and  $I = \begin{bmatrix} 1 & 0 \\ 0 & 1 \end{bmatrix}$ , then the correct statement is  
 (a)  $A^2 + 5A - 7I = O$   
 (b)  $-A^2 + 5A + 7I = O$   
 (c)  $A^2 - 5A + 7I = O$   
 (d)  $A^2 + 5A + 7I = O$
111. The value of the determinant  $\begin{vmatrix} a-b-c & 2a & 2a \\ 2b & b-c-a & 2b \\ 2c & 2c & c-a-b \end{vmatrix}$  will be  
 (a)  $(a-b-c)(a^2 + b^2 + c^2)$   
 (b)  $(a+b+c)^3$   
 (c)  $(a+b+c)(ab+bc+ca)$   
 (d) None of the above
112. If  $(1+x)^n = C_0 + C_1x + C_2x^2 + \dots + C_nx^n$ , then  $C_0 - C_1 + C_2 - C_3 + \dots + (-1)^n \cdot C_n$  is equal to  
 (a)  $3^n$  (b)  $2^n$   
 (c) 1 (d) 0
113. If AM and HM between two numbers are 27 and 12 respectively, then their GM is  
 (a) 9 (b) 18  
 (c) 24 (d) 36
114. For any two events A and B, if  $P(A \cup B) = 5/6$ ,  $P(A \cap B) = 1/3$ ,  $P(B) = 1/2$ , then  $P(A)$  is  
 (a)  $1/2$  (b)  $2/3$   
 (c)  $1/3$  (d) None of these
115. A bag contains 3 white and 5 black balls. One ball is drawn at random. Then, the probability that it is white, is  
 (a)  $\frac{1}{8}$  (b)  $\frac{3}{8}$   
 (c)  $\frac{5}{8}$  (d)  $\frac{3}{5}$
116.  $\mathbf{a} \cdot (\mathbf{b} \times \mathbf{c}) = 0$ , then the correct statement is  
 (a) out of  $\mathbf{a}, \mathbf{b}, \mathbf{c}$  any two vectors are parallel  
 (b)  $\mathbf{a}, \mathbf{b}, \mathbf{c}$  are coplanar  
 (c) any two are equal  $\mathbf{a}, \mathbf{b}, \mathbf{c}$   
 (d) at least one above statement is correct
117. If  $2\mathbf{i} + \mathbf{j} - \mathbf{k}$  and  $\mathbf{i} - 4\mathbf{j} + \lambda\mathbf{k}$  are perpendicular to each other, then  $\lambda$  is equal to  
 (a) -3 (b) -2  
 (c) -1 (d) 0
118. If  $\frac{d}{dx}(\phi(x)) = f(x)$ , then  $\int_1^2 f(x) dx$  is equal to  
 (a)  $f(1) - f(2)$  (b)  $\phi(1) - \phi(2)$   
 (c)  $f(2) - f(1)$  (d)  $\phi(2) - \phi(1)$
119.  $\int_0^2 |1-x| dx$  is equal to  
 (a) 0 (b) 1  
 (c)  $\frac{3}{2}$  (d)  $\frac{1}{2}$
120.  $\int \frac{\sin 2x}{\sin^4 x + \cos^4 x} dx$  is equal to  
 (a)  $2 \tan^{-1}(\tan^2 x) + C$   
 (b)  $\tan^{-1}(x \tan^2 x) + C$   
 (c)  $\tan^{-1}(\tan^2 x) + C$   
 (d) None of the above



121. The function  $\sin x + \cos x$  is maximum when  $x$  is equal to

- (a)  $\frac{\pi}{6}$  (b)  $\frac{\pi}{4}$   
(c)  $\frac{\pi}{3}$  (d)  $\frac{\pi}{2}$

122.  $\frac{d}{dx}(x^x)$  is equal to

- (a)  $x^x \log(e/x)$  (b)  $x^x \log ex$   
(c)  $\log ex$  (d)  $x^x \log x$

123.  $\lim_{x \rightarrow 0} \frac{\sin x}{x}$  is equal to

- (a) 2 (b) -1  
(c) 1 (d) 0

124. The set  $A = \{x : x \in R, x^2 = 16 \text{ and } 2x = 6\}$  equals

- (a)  $\phi$  (b)  $\{14, 3, 4\}$   
(c)  $\{3\}$  (d)  $\{4\}$

125. In how many ways can 5 prizes be distributed among four students when every student can take one or more prizes?

- (a) 1024 (b) 625  
(c) 120 (d) 600

126. The value of  $(\sqrt{5} + 1)^5 - (\sqrt{5} - 1)^5$  is

- (a) 252 (b) 352  
(c) 452 (d) 552

127. The value of

$7 \log\left(\frac{16}{15}\right) + 5 \log\left(\frac{25}{24}\right) + 3 \log\left(\frac{81}{80}\right)$  is equal to

- (a)  $\log 2$  (b) 3  
(c) 5 (d) 7

128. The value of  $\frac{2}{1!} + \frac{2+4}{2!} + \frac{2+4+6}{3!} + \dots \infty$  is

- (a)  $e$  (b)  $2e$   
(c)  $3e$  (d) None of these

129. The sum of the series  $\log_4 2 - \log_8 2 + \log_{16} 2 - \dots$  is

- (a)  $e^2$  (b)  $\log_e 2$   
(c)  $\log_e 3 - 2$  (d)  $1 - \log_e 2$

130. If the domain of the function  $f(x) = x^2 - 6x + 7$  is  $(-\infty, \infty)$ , then the range of function is

- (a)  $(-\infty, \infty)$  (b)  $[-2, \infty)$   
(c)  $(-2, 3)$  (d)  $(-\infty, -2)$

131.  $\lim_{x \rightarrow 0} \frac{\cos(\sin x) - 1}{x^2}$  is equal to

- (a) 1 (b) -1  
(c)  $\frac{1}{2}$  (d)  $-\frac{1}{2}$

132. In order that the function  $f(x) = (x+1)^{1/x}$  is continuous at  $x = 0$ ,  $f(0)$  must be defined as

- (a)  $f(0) = 0$  (b)  $f(0) = e$   
(c)  $f(0) = \frac{1}{e}$  (d)  $f(0) = 1$

133. The function  $f(x) = |x|$  at  $x = 0$  is

- (a) continuous but non-differentiable  
(b) discontinuous and differentiable  
(c) discontinuous and non-differentiable  
(d) continuous and differentiable

134. The point  $(0, 5)$  is closer to the curve  $x^2 = 2y$  at

- (a)  $(2\sqrt{2}, 0)$  (b)  $(0, 0)$   
(c)  $(2, 2)$  (d) None of these

135. The function  $f(x) = x^{1/x}$  is

- (a) increasing in  $(1, \infty)$   
(b) decreasing in  $(1, \infty)$   
(c) increasing in  $(1, e)$ , decreasing in  $(e, \infty)$   
(d) decreasing in  $(1, e)$ , increasing in  $(e, \infty)$

136. The area bounded by the  $x$ -axis and the curve  $y = \sin x$  and  $x = 0, x = \pi$  is

- (a) 1 sq unit (b) 2 sq units  
(c) 0 (d) 4 sq units

137. The order and degree of the differential

equation  $\sqrt{\frac{dy}{dx}} - 4 \frac{dy}{dx} - 7x = 0$  are

- (a) 1 and  $\frac{1}{2}$  (b) 2 and 1  
(c) 1 and 1 (d) 1 and 2

138. The line  $x + y = 4$  divides the line joining the points  $(-1, 1)$  and  $(5, 7)$  in the ratio

- (a) 2 : 1 (b) 1 : 2  
(c) 1 : 2 externally (d) None of these

139. The angle between the pair of lines given by equation  $x^2 + 2xy - y^2 = 0$ , is

- (a)  $\frac{\pi}{3}$  (b)  $\frac{\pi}{6}$   
(c)  $\frac{\pi}{2}$  (d) 0

140. The length of tangent from point  $(5, 1)$  to the circle  $x^2 + y^2 + 6x - 4y - 3 = 0$  is  
 (a) 81 (b) 29  
 (c) 7 (d) 21
141. The length of the latusrectum of the parabola  $169\{(x-1)^2 + (y-3)^2\} = (5x - 12y + 17)^2$  is  
 (a)  $\frac{14}{13}$  (b)  $\frac{12}{13}$   
 (c)  $\frac{28}{13}$  (d) None of these
142. The angle of intersection between the curves  $x^2 = 8y$  and  $y^2 = 8x$  at  $(0, 0)$  is  
 (a)  $\frac{\pi}{4}$  (b)  $\frac{\pi}{3}$   
 (c)  $\frac{\pi}{6}$  (d)  $\frac{\pi}{2}$
143. If the centre, one of the foci and semi-major axis of an ellipse be  $(0, 0)$ ,  $(0, 3)$  and 5, then its equation is  
 (a)  $\frac{x^2}{16} + \frac{y^2}{25} = 1$  (b)  $\frac{x^2}{25} + \frac{y^2}{16} = 1$   
 (c)  $\frac{x^2}{9} + \frac{y^2}{25} = 1$  (d) None of these
144. The radius of the director circle of the hyperbola  $\frac{x^2}{a^2} - \frac{y^2}{b^2} = 1$  is  
 (a)  $a - b$  (b)  $\sqrt{a - b}$   
 (c)  $\sqrt{a^2 - b^2}$  (d)  $\sqrt{a^2 + b^2}$
145. If projection of any line on coordinate axes 3, 4 and 5, then its length is  
 (a) 12 (b) 50  
 (c)  $5\sqrt{2}$  (d)  $3\sqrt{2}$
146. If  $\tan \theta = \frac{1}{2}$  and  $\tan \phi = \frac{1}{3}$ , then the value of  $\theta + \phi$  is  
 (a)  $\frac{\pi}{6}$  (b)  $\pi$   
 (c) 0 (d)  $\frac{\pi}{4}$
147. If  $\sin \theta = \frac{1}{2}$ ,  $\tan \theta = \frac{1}{\sqrt{3}}$ ,  $\forall n \in I$ , then most general values of  $\theta$  is  
 (a)  $2n\pi + \frac{\pi}{6}$ ,  $\forall n \in I$  (b)  $2n\pi + \frac{\pi}{4}$ ,  $\forall n \in I$   
 (c)  $2n\pi + \frac{\pi}{3}$ ,  $\forall n \in I$  (d)  $2n\pi + \frac{\pi}{3}$ ,  $\forall n \in I$
148. The principal value of  $\sin^{-1}\left(-\frac{\sqrt{3}}{2}\right)$  is  
 (a)  $-\frac{2\pi}{3}$  (b)  $-\frac{\pi}{3}$   
 (c)  $\frac{4\pi}{3}$  (d)  $\frac{5\pi}{3}$
149. A ladder rests against a wall so that its top touches the roof of the house. If the ladder makes an angle of  $60^\circ$  with the horizontal and height of the house be  $6\sqrt{3}$  m, then the length of the ladder is  
 (a)  $12\sqrt{3}$  m (b) 12 m  
 (c)  $\frac{12}{\sqrt{3}}$  m (d) None of these
150. If angles  $A, B$  and  $C$  are in AP, then  $\frac{a+c}{b}$  is equal to  
 (a)  $2\sin\left(\frac{A-C}{2}\right)$  (b)  $2\cos\left(\frac{A-C}{2}\right)$   
 (c)  $\cos\left(\frac{A-C}{2}\right)$  (d)  $\sin\left(\frac{A-C}{2}\right)$