JEE1 OP! 9th APRIL S1

JEE April 2024

Application No	[redacted]
Candidate Name	
Roll No	
Test Date	09/04/2024
Test Time	9:00 AM - 12:00 PM
Subject	B. Tech

Section : **Mathematics Section A**

Q.1 If the sum of the series $\frac{1}{1\cdot(1+d)}+\frac{1}{(1+d)(1+2d)}+\ldots+\frac{1}{(1+9d)(1+10d)}$ is equal to 5, then 50d is equal to :

Options

1. 5
2. 10
3. 20
4. 15

Question Type : **MCQ**
Question ID : **87827056152**
Option 1 ID : **878270220417**
Option 2 ID : **878270220418**
Option 3 ID : **878270220420**
Option 4 ID : **878270220419**
Status : **Not Answered**
Chosen Option : **--**

Q.2 The solution of the differential equation $(x^2+y^2)dx-5xy\ dy=0,\ y(1)=0$, is :

Options

1. $\left|x^2-2y^2\right|^6=x$
2. $\left|x^2-2y^2\right|^5=x^2$
3. $\left|x^2-4y^2\right|^6=x$
4. $\left|x^2-4y^2\right|^5=x^2$

Question Type : **MCQ**
Question ID : **87827056157**
Option 1 ID : **878270220437**
Option 2 ID : **878270220438**
Option 3 ID : **878270220439**
Option 4 ID : **878270220440**
Status : **Not Answered**
Chosen Option : **--**

Q.3 A variable line L passes through the point (3, 5) and intersects the positive coordinate axes at the points A and B. The minimum area of the triangle OAB, where O is the origin, is :

Options

1. 25
2. 40
3. 35
4. 30

Question Type : **MCQ**
Question ID : **87827056154**
Option 1 ID : **878270220427**
Option 2 ID : **878270220425**
Option 3 ID : **878270220426**
Option 4 ID : **878270220428**
Status : **Not Answered**
Chosen Option : **--**

Q.4 Let α, β be the roots of the equation $x^2 + 2\sqrt{2}\,x - 1 = 0$. The quadratic equation, whose roots are $\alpha^4+\beta^4$ and $\frac{1}{10}\left(\alpha^6 + \beta^6\right)$, is :

Options

1. $x^2-180x+9506=0$
2. $x^2-195x+9506=0$
3. $x^2-195x+9466=0$
4. $x^2-190x+9466=0$

Question Type : **MCQ**
Question ID : **87827056149**
Option 1 ID : **878270220405**
Option 2 ID : **878270220407**
Option 3 ID : **878270220408**
Option 4 ID : **878270220406**
Status : **Not Answered**
Chosen Option : **--**

Q.5 Let $f(x)=ax^3+bx^2+cx+41$ be such that $f(1)=40$, $f'(1)=2$ and $f''(1)=4$. Then $a^2+b^2+c^2$ is equal to :

Options

1. 73
2. 54
3. 51
4. 62

Question Type : **MCQ**
Question ID : **87827056153**
Option 1 ID : **878270220424**
Option 2 ID : **878270220422**
Option 3 ID : **878270220421**
Option 4 ID : **878270220423**
Status : **Not Answered**
Chosen Option : **--**

Q.6

If the domain of the function $f(x) = \sin^{-1}\left(\frac{x-1}{2x+3}\right)$ is $\mathbf{R} - (\alpha, \beta)$, then $12\alpha\beta$ is equal to :

Options

1. 36
2. 32
3. 24
4. 40

Question Type : **MCQ**
Question ID : **87827056148**
Option 1 ID : **878270220403**
Option 2 ID : **878270220402**
Option 3 ID : **878270220401**
Option 4 ID : **878270220404**
Status : **Not Answered**
Chosen Option : **--**

Q.7

Let three vectors $\vec{a} = \alpha\hat{i} + 4\hat{j} + 2\hat{k}$, $\vec{b} = 5\hat{i} + 3\hat{j} + 4\hat{k}$, $\vec{c} = x\hat{i} + y\hat{j} + z\hat{k}$ form a triangle such that $\vec{c} = \vec{a} - \vec{b}$ and the area of the triangle is $5\sqrt{6}$. If α is a positive real number, then $|\vec{c}|^2$ is equal to :

Options

1. 16
2. 10
3. 14
4. 12

Question Type : **MCQ**
Question ID : **87827056165**
Option 1 ID : **878270220472**
Option 2 ID : **878270220469**
Option 3 ID : **878270220471**
Option 4 ID : **878270220470**
Status : **Not Attempted and Marked For Review**
Chosen Option : **--**

Q.8 Let $|\cos\theta\cos(60-\theta)\cos(60+\theta)| \le \frac{1}{8}$, $\theta\in[0, 2\pi]$. Then, the sum of all $\theta\in[0, 2\pi]$, where $\cos 3\theta$ attains its maximum value, is :

Options

1. 18π
2. 9π
3. 6π
4. 15π

Question Type : **MCQ**
Question ID : **87827056167**
Option 1 ID : **878270220478**
Option 2 ID : **878270220479**
Option 3 ID : **878270220477**
Option 4 ID : **878270220480**
Status : **Not Answered**
Chosen Option : **--**

Q.9 The coefficient of x^{70} in $x^2(1+x)^{98}+x^3(1+x)^{97}+x^4(1+x)^{96}+\ldots+x^{54}(1+x)^{46}$ is ${}^{99}C_p-{}^{46}C_q$. Then a possible value of $p+q$ is :

Options

1. 61
2. 55
3. 83
4. 68

Question Type : **MCQ**
Question ID : **87827056151**
Option 1 ID : **878270220416**
Option 2 ID : **878270220413**
Option 3 ID : **878270220414**
Option 4 ID : **878270220415**
Status : **Not Answered**
Chosen Option : **--**

Q.10 The shortest distance between the lines $\frac{x-3}{4}=\frac{y+7}{-11}=\frac{z-1}{5}$ and $\frac{x-5}{3}=\frac{y-9}{-6}=\frac{z+2}{1}$ is :

Options

1. $\frac{185}{\sqrt{563}}$
2. $\frac{178}{\sqrt{563}}$
3. $\frac{179}{\sqrt{563}}$
4. $\frac{187}{\sqrt{563}}$

Question Type : **MCQ**
Question ID : **87827056163**
Option 1 ID : **878270220463**
Option 2 ID : **878270220461**
Option 3 ID : **878270220462**
Option 4 ID : **878270220464**
Status : **Answered**
Chosen Option : **4**

Q.11 Let a circle passing through (2, 0) have its centre at the point (h, k). Let (x_c, y_c) be the point of intersection of the lines $3x+5y=1$ and $(2+c)x+5c^2y=1$. If $h=\lim\limits_{c\to 1} x_c$ and $k=\lim\limits_{c\to 1} y_c$, then the equation of the circle is :

Options

1. $25x^2+25y^2-2x+2y-60=0$
2. $5x^2+5y^2-4x+2y-12=0$
3. $25x^2+25y^2-20x+2y-60=0$
4. $5x^2+5y^2-4x-2y-12=0$

Question Type : **MCQ**
Question ID : **87827056160**
Option 1 ID : **878270220450**
Option 2 ID : **878270220451**
Option 3 ID : **878270220449**
Option 4 ID : **878270220452**
Status : **Answered**
Chosen Option : **2**

Q.12

Let $\lambda, \mu \in \mathbf{R}$. If the system of equations

$$3x + 5y + \lambda z = 3$$
$$7x + 11y - 9z = 2$$
$$97x + 155y - 189z = \mu$$

has infinitely many solutions, then $\mu + 2\lambda$ is equal to :

Options

1. 25
2. 22
3. 24
4. 27

Question Type : **MCQ**
Question ID : **87827056150**
Option 1 ID : **878270220411**
Option 2 ID : **878270220409**
Option 3 ID : **878270220410**
Option 4 ID : **878270220412**
Status : **Not Attempted and Marked For Review**
Chosen Option : **--**

Q.13

Let the line L intersect the lines $x-2=-y=z-1$, $2(x+1)=2(y-1)=z+1$ and be parallel to the line $\frac{x-2}{3}=\frac{y-1}{1}=\frac{z-2}{2}$. Then which of the following points lies on L?

Options

1. $\left(-\frac{1}{3}, -1, -1\right)$
2. $\left(-\frac{1}{3}, 1, -1\right)$
3. $\left(-\frac{1}{3}, 1, 1\right)$
4. $\left(-\frac{1}{3}, -1, 1\right)$

Question Type : **MCQ**
Question ID : **87827056162**
Option 1 ID : **878270220460**
Option 2 ID : **878270220457**
Option 3 ID : **878270220459**
Option 4 ID : **878270220458**
Status : **Not Answered**
Chosen Option : **--**

Q.14 The frequency distribution of the age of students in a class of 40 students is given below.

Age	15	16	17	18	19	20
No of Students	5	8	5	12	x	y

If the mean deviation about the median is 1.25, then $4x+5y$ is equal to :

Options

1. 47
2. 43
3. 46
4. 44

Question Type : **MCQ**
Question ID : **87827056166**
Option 1 ID : **878270220476**
Option 2 ID : **878270220473**
Option 3 ID : **878270220475**
Option 4 ID : **878270220474**
Status : **Not Answered**
Chosen Option : **--**

Q.15 The solution curve, of the differential equation $2y\frac{dy}{dx}+3=5\frac{dy}{dx}$, passing through the point $(0, 1)$ is a conic, whose vertex lies on the line :

Options

1. $2x+3y=9$
2. $2x+3y=6$
3. $2x+3y=-6$
4. $2x+3y=-9$

Question Type : **MCQ**
Question ID : **87827056158**
Option 1 ID : **878270220441**
Option 2 ID : **878270220442**
Option 3 ID : **878270220443**
Option 4 ID : **878270220444**
Status : **Answered**
Chosen Option : **3**

Q.16 Let $\int \frac{2-\tan x}{3+\tan x}\,dx = \frac{1}{2}\left(\alpha x + \log_e|\beta \sin x + \gamma \cos x|\right) + C$, where C is the constant of integration.

Then $\alpha + \frac{\gamma}{\beta}$ is equal to :

Options

1. 1
2. 7
3. 4
4. 3

Question Type : **MCQ**
Question ID : **87827056155**
Option 1 ID : **878270220429**
Option 2 ID : **878270220432**
Option 3 ID : **878270220431**
Option 4 ID : **878270220430**
Status : **Not Answered**
Chosen Option : **--**

Q.17 A ray of light coming from the point P(1, 2) gets reflected from the point Q on the x-axis and then passes through the point R(4, 3). If the point S(h, k) is such that PQRS is a parallelogram, then hk^2 is equal to :

Options

1. 80
2. 70
3. 60
4. 90

Question Type : **MCQ**
Question ID : **87827056161**
Option 1 ID : **878270220455**
Option 2 ID : **878270220454**
Option 3 ID : **878270220453**
Option 4 ID : **878270220456**
Status : **Answered**
Chosen Option : **2**

Q.18 Let $\vec{OA} = 2\vec{a}$, $\vec{OB} = 6\vec{a} + 5\vec{b}$ and $\vec{OC} = 3\vec{b}$, where O is the origin. If the area of the parallelogram with adjacent sides $\vec{OA}$ and $\vec{OC}$ is 15 sq. units, then the area (in sq. units) of the quadrilateral OABC is equal to :

Options

1. 38
2. 32
3. 40
4. 35

Question Type : **MCQ**
Question ID : **87827056164**
Option 1 ID : **878270220467**
Option 2 ID : **878270220465**
Option 3 ID : **878270220468**
Option 4 ID : **878270220466**
Status : **Not Answered**
Chosen Option : **--**

Q.19 Let $f(x) = x^2 + 9$, $g(x) = \frac{x}{x-9}$ and $a = fog(10)$, $b = gof(3)$. If e and l denote the eccentricity and the length of the latus rectum of the ellipse $\frac{x^2}{a} + \frac{y^2}{b} = 1$, then $8e^2 + l^2$ is equal to.

Options

1. 16
2. 12
3. 8
4. 6

Question Type : **MCQ**
Question ID : **87827056159**
Option 1 ID : **878270220445**
Option 2 ID : **878270220446**
Option 3 ID : **878270220447**
Option 4 ID : **878270220448**
Status : **Not Answered**
Chosen Option : **--**

Q.20 The parabola $y^2=4x$ divides the area of the circle $x^2+y^2=5$ in two parts. The area of the smaller part is equal to :

Options

1. $\frac{1}{3}+\sqrt{5}\sin^{-1}\left(\frac{2}{\sqrt{5}}\right)$

2. $\frac{2}{3}+\sqrt{5}\sin^{-1}\left(\frac{2}{\sqrt{5}}\right)$

3. $\frac{1}{3}+5\sin^{-1}\left(\frac{2}{\sqrt{5}}\right)$

4. $\frac{2}{3}+5\sin^{-1}\left(\frac{2}{\sqrt{5}}\right)$

Question Type : **MCQ**
Question ID : **87827056156**
Option 1 ID : **878270220433**
Option 2 ID : **878270220434**
Option 3 ID : **878270220435**
Option 4 ID : **878270220436**
Status : **Not Answered**
Chosen Option : **--**

Section : **Mathematics Section B**

Q.21 Let A be a non-singular matrix of order 3. If $\det(3\text{adj}(2\text{adj}((\det A)A))) = 3^{-13}\cdot 2^{-10}$ and $\det(3\text{adj}(2A))=2^m\cdot 3^n$, then $|3m+2n|$ is equal to _____.

Given **19**
Answer :

Question Type : **SA**
Question ID : **87827056170**
Status : **Answered**

Q.22 The sum of the square of the modulus of the elements in the set $\{z=a+ib : a, b \in \mathbf{Z}, z \in \mathbf{C}, |z-1| \le 1, |z-5| \le |z-5i|\}$ is _____.

Given **--**
Answer :

Question Type : **SA**
Question ID : **87827056169**
Status : **Not Answered**

Q.23 Let the centre of a circle, passing through the points (0, 0), (1, 0) and touching the circle $x^2+y^2=9$, be (h, k). Then for all possible values of the coordinates of the centre (h, k), $4(h^2+k^2)$ is equal to _____.

Given --
Answer :

Question Type : **SA**
Question ID : **87827056176**
Status : **Not Answered**

Q.24 Let $f:(0,\pi)\to\mathbf{R}$ be a function given by $f(x)=\begin{cases}\left(\frac{8}{7}\right)^{\frac{\tan 8x}{\tan 7x}}, & 0<x<\frac{\pi}{2}\\ a-8, & x=\frac{\pi}{2}\\ (1+|\cot x|)^{\frac{b}{a}|\tan x|}, & \frac{\pi}{2}<x<\pi\end{cases}$

where $a, b\in\mathbf{Z}$. If f is continuous at $x=\frac{\pi}{2}$, then a^2+b^2 is equal to _____.

Given --
Answer :

Question Type : **SA**
Question ID : **87827056171**
Status : **Not Answered**

Q.25 The remainder when 428^{2024} is divided by 21 is _____.

Given --
Answer :

Question Type : **SA**
Question ID : **87827056172**
Status : **Not Answered**

Q.26 Let $\lim\limits_{n\to\infty}\left(\frac{n}{\sqrt{n^4+1}}-\frac{2n}{(n^2+1)\sqrt{n^4+1}}+\frac{n}{\sqrt{n^4+16}}-\frac{8n}{(n^2+4)\sqrt{n^4+16}}\right.$

$\left.+\ldots+\frac{n}{\sqrt{n^4+n^4}}-\frac{2n\cdot n^2}{(n^2+n^2)\sqrt{n^4+n^4}}\right)$ be $\frac{\pi}{k}$, using only the principal values of the inverse trigonometric functions. Then k^2 is equal to _____.

Given --
Answer :

Question Type : **SA**
Question ID : **87827056175**
Status : **Not Answered**

Q.27 If a function f satisfies $f(m+n)=f(m)+f(n)$ for all m, n ϵ **N** and $f(1)=1$, then the largest natural number λ such that $\sum_{k=1}^{2022} f(\lambda + k) \leq (2022)^2$ is equal to _____.

Given --
Answer :

Question Type : **SA**
Question ID : **87827056173**
Status : **Not Answered**

Q.28 Let the set of all positive values of λ, for which the point of local minimum of the function $(1+x(\lambda^2-x^2))$ satisfies $\frac{x^2+x+2}{x^2+5x+6} < 0$, be (α, β). Then $\alpha^2+\beta^2$ is equal to _____.

Given --
Answer :

Question Type : **SA**
Question ID : **87827056174**
Status : **Not Answered**

Q.29 Let $A=\{2, 3, 6, 7\}$ and $B=\{4, 5, 6, 8\}$. Let R be a relation defined on $A \times B$ by (a_1, b_1) R (a_2, b_2) if and only if $a_1 + a_2 = b_1 + b_2$. Then the number of elements in R is _____.

Given --
Answer :

Question Type : **SA**
Question ID : **87827056168**
Status : **Not Answered**

Q.30 Let a, b and c denote the outcome of three independent rolls of a fair tetrahedral die, whose four faces are marked 1, 2, 3, 4. If the probability that $ax^2+bx+c=0$ has all real roots is $\frac{m}{n}$, gcd(m, n) =1, then m+n is equal to _____.

Given --
Answer :

Question Type : **SA**
Question ID : **87827056177**
Status : **Not Answered**

Section : **Physics Section A**

Q.31 The volume of an ideal gas ($\gamma = 1.5$) is changed adiabatically from 5 litres to 4 litres. The ratio of initial pressure to final pressure is :

Options

1. $\frac{8}{5\sqrt{5}}$
2. $\frac{16}{25}$
3. $\frac{2}{\sqrt{5}}$
4. $\frac{4}{5}$

Question Type : **MCQ**
Question ID : **87827056186**
Option 1 ID : **878270220524**
Option 2 ID : **878270220526**
Option 3 ID : **878270220523**
Option 4 ID : **878270220525**
Status : **Answered**
Chosen Option : **1**

Q.32 The dimensional formula of latent heat is :

Options

1. $[M^0L^2T^{-2}]$
2. $[MLT^{-2}]$
3. $[M^0LT^{-2}]$
4. $[ML^2T^{-2}]$

Question Type : **MCQ**
Question ID : **87827056178**
Option 1 ID : **878270220492**
Option 2 ID : **878270220493**
Option 3 ID : **878270220494**
Option 4 ID : **878270220491**
Status : **Answered**
Chosen Option : **1**

Q.33 One main scale division of a vernier caliper is equal to m units. If n^{th} division of main scale coincides with $(n+1)^{th}$ division of vernier scale, the least count of the vernier caliper is :

Options

1. $\frac{n}{(n+1)}$
2. $\frac{m}{(n+1)}$
3. $\frac{1}{(n+1)}$
4. $\frac{m}{n(n+1)}$

Question Type : **MCQ**
Question ID : **87827056196**
Option 1 ID : **878270220565**
Option 2 ID : **878270220564**
Option 3 ID : **878270220563**
Option 4 ID : **878270220566**
Status : **Answered**
Chosen Option : **2**

Q.34 A particle moving in a straight line covers half the distance with speed 6 m/s. The other half is covered in two equal time intervals with speeds 9 m/s and 15 m/s respectively. The average speed of the particle during the motion is :

Options

1. 10 m/s
2. 8.8 m/s
3. 8 m/s
4. 9.2 m/s

Question Type : **MCQ**
Question ID : **87827056179**
Option 1 ID : **878270220495**
Option 2 ID : **878270220497**
Option 3 ID : **878270220496**
Option 4 ID : **878270220498**
Status : **Answered**
Chosen Option : **3**

Q.35 Given below are two statements :

Statement (I) : When an object is placed at the centre of curvature of a concave lens, image is formed at the centre of curvature of the lens on the other side.

Statement (II) : Concave lens always forms a virtual and erect image.

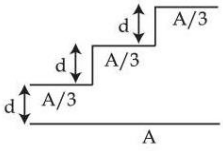
In the light of the above statements, choose the **correct** answer from the options given below :

Options

1. **Statement I** is true but **Statement II** is false
2. **Statement I** is false but **Statement II** is true
3. Both **Statement I** and **Statement II** are false
4. Both **Statement I** and **Statement II** are true

Question Type : **MCQ**
Question ID : **87827056192**
Option 1 ID : **878270220549**
Option 2 ID : **878270220550**
Option 3 ID : **878270220548**
Option 4 ID : **878270220547**
Status : **Not Attempted and Marked For Review**
Chosen Option : **--**

Q.36 A capacitor is made of a flat plate of area A and a second plate having a stair-like structure as shown in figure. If the area of each stair is $\frac{A}{3}$ and the height is d, the capacitance of the arrangement is :



Options

1. $\frac{11\,\epsilon_0\,A}{18\,d}$
2. $\frac{13\,\epsilon_0\,A}{17\,d}$
3. $\frac{11\,\epsilon_0\,A}{20\,d}$
4. $\frac{18\,\epsilon_0\,A}{11\,d}$

Question Type : **MCQ**
Question ID : **87827056187**
Option 1 ID : **878270220527**
Option 2 ID : **878270220528**
Option 3 ID : **878270220529**
Option 4 ID : **878270220530**
Status : **Answered**
Chosen Option : **1**

Q.37 A light unstretchable string passing over a smooth light pulley connects two blocks of masses m_1 and m_2. If the acceleration of the system is $\frac{g}{8}$, then the ratio of the masses $\frac{m_2}{m_1}$ is :

Options

1. $4:3$
2. $8:1$
3. $9:7$
4. $5:3$

Question Type : **MCQ**
Question ID : **87827056181**
Option 1 ID : **878270220505**
Option 2 ID : **878270220503**
Option 3 ID : **878270220504**
Option 4 ID : **878270220506**
Status : **Answered**
Chosen Option : **3**

Q.38 A sample of 1 mole gas at temperature T is adiabatically expanded to double its volume. If adiabatic constant for the gas is $\gamma = \frac{3}{2}$, then the work done by the gas in the process is :

Options

1. $RT\left[2-\sqrt{2}\right]$
2. $\frac{R}{T}\left[2-\sqrt{2}\right]$
3. $RT\left[2+\sqrt{2}\right]$
4. $\frac{T}{R}\left[2+\sqrt{2}\right]$

Question Type : **MCQ**
Question ID : **87827056185**
Option 1 ID : **878270220519**
Option 2 ID : **878270220521**
Option 3 ID : **878270220520**
Option 4 ID : **878270220522**
Status : **Answered**
Chosen Option : **1**

Q.39 A heavy iron bar, of weight W is having its one end on the ground and the other on the shoulder of a person. The bar makes an angle θ with the horizontal. The weight experienced by the person is :

Options

1. $W \cos \theta$
2. $W \sin \theta$
3. $\frac{W}{2}$
4. W

Question Type : **MCQ**
Question ID : **87827056180**
Option 1 ID : **878270220499**
Option 2 ID : **878270220500**
Option 3 ID : **878270220502**
Option 4 ID : **878270220501**
Status : **Not Answered**
Chosen Option : **--**

Q.40 A plane EM wave is propagating along x direction. It has a wavelength of 4 mm. If electric field is in y direction with the maximum magnitude of 60 Vm^{-1}, the equation for magnetic field is :

Options

1. $B_x = 60 \sin\left[\frac{\pi}{2}\left(x - 3\times 10^8 t\right)\right]\hat{i}\ T$
2. $B_z = 2\times 10^{-7} \sin\left[\frac{\pi}{2}\left(x - 3\times 10^8 t\right)\right]\hat{k}\ T$
3. $B_z = 60 \sin\left[\frac{\pi}{2}\left(x - 3\times 10^8 t\right)\right]\hat{k}\ T$
4. $B_z = 2\times 10^{-7} \sin\left[\frac{\pi}{2}\times 10^3\left(x - 3\times 10^8 t\right)\right]\hat{k}\ T$

Question Type : **MCQ**
Question ID : **87827056191**
Option 1 ID : **878270220544**
Option 2 ID : **878270220545**
Option 3 ID : **878270220543**
Option 4 ID : **878270220546**
Status : **Answered**
Chosen Option : **4**

Q.41 The energy equivalent of 1 g of substance is :

Options

1. 5.6×10^{26} MeV
2. 5.6×10^{12} MeV
3. 5.6 eV
4. 11.2×10^{24} MeV

Question Type : **MCQ**
Question ID : **87827056194**
Option 1 ID : **878270220558**
Option 2 ID : **878270220556**
Option 3 ID : **878270220555**
Option 4 ID : **878270220557**
Status : **Answered**
Chosen Option : **1**

Q.42 A galvanmeter has a coil of resistance 200 Ω with a full scale deflection at 20 μA. The value of resistance to be added to use it as an ammeter of range (0-20) mA is :

Options

1. 0.10 Ω
2. 0.50 Ω
3. 0.40 Ω
4. 0.20 Ω

Question Type : **MCQ**
Question ID : **87827056197**
Option 1 ID : **878270220568**
Option 2 ID : **878270220567**
Option 3 ID : **878270220570**
Option 4 ID : **878270220569**
Status : **Answered**
Chosen Option : **4**

Q.43 A light emitting diode (LED) is fabricated using GaAs semiconducting material whose band gap is 1.42 eV. The wavelength of light emitted from the LED is :

Options

1. 1243 nm
2. 1400 nm
3. 650 nm
4. 875 nm

Question Type : **MCQ**
Question ID : **87827056195**
Option 1 ID : **878270220560**
Option 2 ID : **878270220559**
Option 3 ID : **878270220561**
Option 4 ID : **878270220562**
Status : **Answered**
Chosen Option : **4**

Q.44 A particle of mass m moves on a straight line with its velocity increasing with distance according to the equation $v = \alpha\sqrt{x}$, where α is a constant. The total work done by all the forces applied on the particle during its displacement from $x = 0$ to $x = d$, will be :

Options

1. $\frac{m\alpha^2 d}{2}$
2. $2m\alpha^2 d$
3. $\frac{md}{2\alpha^2}$
4. $\frac{m}{2\alpha^2 d}$

Question Type : **MCQ**
Question ID : **87827056182**
Option 1 ID : **878270220508**
Option 2 ID : **878270220510**
Option 3 ID : **878270220507**
Option 4 ID : **878270220509**
Status : **Answered**
Chosen Option : **1**

Q.45 Given below are two statements :

Statement (I) : When currents vary with time, Newton's third law is valid only if momentum carried by the electromagnetic field is taken into account.

Statement (II) : Ampere's circuital law does not depend on Biot-Savart's law.

In the light of the above statements, choose the **correct** answer from the options given below :

Options

1. Both **Statement I** and **Statement II** are true
2. Both **Statement I** and **Statement II** are false
3. **Statement I** is true but **Statement II** is false
4. **Statement I** is false but **Statement II** is true

Question Type : **MCQ**
Question ID : **87827056189**
Option 1 ID : **878270220535**
Option 2 ID : **878270220536**
Option 3 ID : **878270220537**
Option 4 ID : **878270220538**
Status : **Not Answered**
Chosen Option : **--**

Q.46 An astronaut takes a ball of mass m from earth to space. He throws the ball into a circular orbit about earth at an altitude of 318.5 km. From earth's surface to the orbit, the change in total mechanical energy of the ball is $x\frac{GM_em}{21R_e}$. The value of x is (take $R_e = 6370$ km) :

Options

1. 11
2. 9
3. 10
4. 12

Question Type : **MCQ**
Question ID : **87827056183**
Option 1 ID : **878270220513**
Option 2 ID : **878270220511**
Option 3 ID : **878270220512**
Option 4 ID : **878270220514**
Status : **Not Answered**
Chosen Option : **--**

Q.47 A sphere of relative density σ and diameter D has concentric cavity of diameter d. The ratio of $\frac{D}{d}$, if it just floats on water in a tank is :

Options

1. $\left(\frac{\sigma - 1}{\sigma}\right)^{1/3}$
2. $\left(\frac{\sigma}{\sigma - 1}\right)^{1/3}$
3. $\left(\frac{\sigma + 1}{\sigma - 1}\right)^{1/3}$
4. $\left(\frac{\sigma - 2}{\sigma + 2}\right)^{1/3}$

Question Type : **MCQ**
Question ID : **87827056184**
Option 1 ID : **878270220515**
Option 2 ID : **878270220516**
Option 3 ID : **878270220518**
Option 4 ID : **878270220517**
Status : **Not Answered**
Chosen Option : **--**

Q.48 A bulb and a capacitor are connected in series across an ac supply. A dielectric is then placed between the plates of the capacitor. The glow of the bulb :

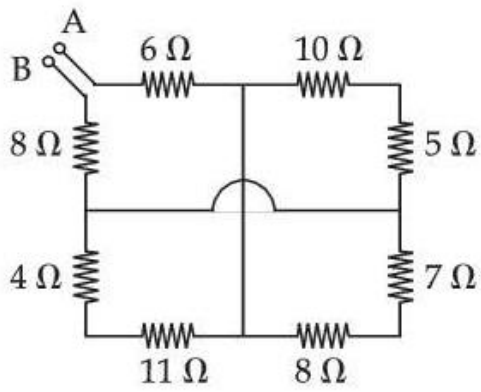
Options

1. decreases
2. increases
3. remains same
4. becomes zero

Question Type : **MCQ**
Question ID : **87827056190**
Option 1 ID : **878270220541**
Option 2 ID : **878270220540**
Option 3 ID : **878270220539**
Option 4 ID : **878270220542**
Status : **Answered**
Chosen Option : **3**

Q.49

The equivalent resistance between A and B is :



Options

1. 19 Ω
2. 27 Ω
3. 25 Ω
4. 18 Ω

Question Type : **MCQ**
Question ID : **87827056188**
Option 1 ID : **878270220533**
Option 2 ID : **878270220534**
Option 3 ID : **878270220532**
Option 4 ID : **878270220531**
Status : **Answered**
Chosen Option : **1**

Q.50

A proton, an electron and an alpha particle have the same energies. Their de-Broglie wavelengths will be compared as :

Options

1. $\lambda_e > \lambda_\alpha > \lambda_p$
2. $\lambda_\alpha < \lambda_p < \lambda_e$
3. $\lambda_p > \lambda_e > \lambda_\alpha$
4. $\lambda_p < \lambda_e < \lambda_\alpha$

Question Type : **MCQ**
Question ID : **87827056193**
Option 1 ID : **878270220554**
Option 2 ID : **878270220553**
Option 3 ID : **878270220551**
Option 4 ID : **878270220552**
Status : **Answered**
Chosen Option : **2**

Section : **Physics Section B**

Q.51 In a Young's double slit experiment, the intensity at a point is $\left(\frac{1}{4}\right)^{th}$ of the maximum intensity, the minimum distance of the point from the central maximum is _____μm.
(Given : λ = 600 nm, d = 1.0 mm, D = 1.0 m)

Given **200**
Answer :

Question Type : **SA**
Question ID : **87827056206**
Status : **Answered**

Q.52 Two persons pull a wire towards themselves. Each person exerts a force of 200 N on the wire. Young's modulus of the material of wire is 1×10^{11} N m^{-2}. Original length of the wire is 2 m and the area of cross section is 2 cm^2. The wire will extend in length by _____ μm.

Given **--**
Answer :

Question Type : **SA**
Question ID : **87827056200**
Status : **Not Answered**

Q.53 When a coil is connected across a 20 V dc supply, it draws a current of 5 A. When it is connected across 20 V, 50 Hz ac supply, it draws a current of 4 A. The self inductance of the coil is _____ mH. (Take π = 3)

Given **10**
Answer :

Question Type : **SA**
Question ID : **87827056205**
Status : **Answered**

Q.54 At the centre of a half ring of radius R = 10 cm and linear charge density 4n C m^{-1}, the potential is x π V. The value of x is _____.

Given **36**
Answer :

Question Type : **SA**
Question ID : **87827056202**
Status : **Answered**

Q.55 A star has 100% helium composition. It starts to convert three ^4He into one ^{12}C via triple alpha process as $^4He + {}^4He + {}^4He \rightarrow {}^{12}C + Q$. The mass of the star is 2.0×10^{32} kg and it generates energy at the rate of 5.808×10^{30} W. The rate of converting these ^4He to ^{12}C is n × 10^{42} s^{-1}, where n is _____.
[Take, mass of $^4He = 4.0026$ u, mass of $^{12}C = 12$ u]

Given **--**
Answer :

Question Type : **SA**
Question ID : **87827056207**
Status : **Not Answered**

Q.56 A square loop of edge length 2 m carrying current of 2 A is placed with its edges parallel to the x-y axis. A magnetic field is passing through the x-y plane and expressed as $\vec{B} = B_0 (1 + 4x) \hat{k}$, where $B_0 = 5$ T. The net magnetic force experienced by the loop is _____ N.

Given **--**
Answer :

Question Type : **SA**
Question ID : **87827056204**
Status : **Not Answered**

Q.57 The position, velocity and acceleration of a particle executing simple harmonic motion are found to have magnitudes of 4 m, 2 ms^{-1} and 16 ms^{-2} at a certain instant. The amplitude of the motion is $\sqrt{x}$, m where x is _____.

Given **17**
Answer :

Question Type : **SA**
Question ID : **87827056201**
Status : **Answered**

Q.58 A string is wrapped around the rim of a wheel of moment of inertia 0.40 kgm^2 and radius 10 cm. The wheel is free to rotate about its axis. Initially the wheel is at rest. The string is now pulled by a force of 40 N. The angular velocity of the wheel after 10 s is x rad/s, where x is _____.

Given **100**
Answer :

Question Type : **SA**
Question ID : **87827056199**
Status : **Answered**

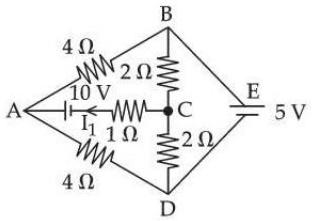
Q.59 If $\vec{a}$ and $\vec{b}$ makes an angle $\cos^{-1}\left(\frac{5}{9}\right)$ with each other, then $|\vec{a} + \vec{b}| = \sqrt{2}|\vec{a} - \vec{b}|$ for $|\vec{a}| = n|\vec{b}|$

The integer value of n is _____.

Given **--**
Answer :

Question Type : **SA**
Question ID : **87827056198**
Status : **Not Answered**

Q.60 The current flowing through the 1 Ω resistor is $\frac{n}{10}$ A. The value of n is _____.



Given --
Answer :

Question Type : **SA**
Question ID : **87827056203**
Status : **Not Answered**

Section : **Chemistry Section A**

Q.61 On reaction of Lead Sulphide with dilute nitric acid which of the following is **not** formed ?

Options

1. Sulphur
2. Lead nitrate
3. Nitrous oxide
4. Nitric oxide

Question Type : **MCQ**
Question ID : **87827056226**
Option 1 ID : **878270220656**
Option 2 ID : **878270220653**
Option 3 ID : **878270220655**
Option 4 ID : **878270220654**
Status : **Not Answered**
Chosen Option : **--**

Q.62 Compare the energies of following sets of quantum numbers for multielectron system.

(A) $n=4, l=1$

(B) $n=4, l=2$

(C) $n=3, l=1$

(D) $n=3, l=2$

(E) $n=4, l=0$

Choose the **correct** answer from the options given below :

Options

1. (B) > (A) > (C) > (E) > (D)

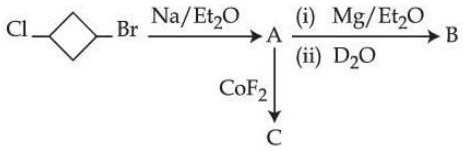
2. (C) < (E) < (D) < (A) < (B)

3. (E) < (C) < (D) < (A) < (B)

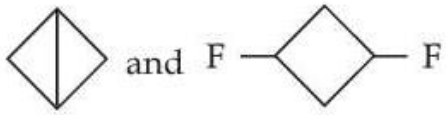
4. (E) > (C) > (A) > (D) > (B)

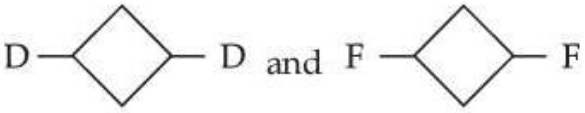
Question Type : **MCQ**
Question ID : **87827056208**
Option 1 ID : **878270220584**
Option 2 ID : **878270220582**
Option 3 ID : **878270220583**
Option 4 ID : **878270220581**
Status : **Answered**
Chosen Option : **2**

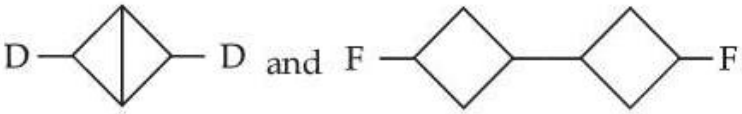
Q.63 In the following sequence of reaction, the major products B and C respectively are :

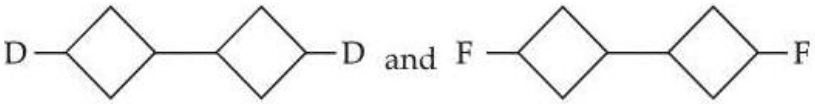


Options

1. 

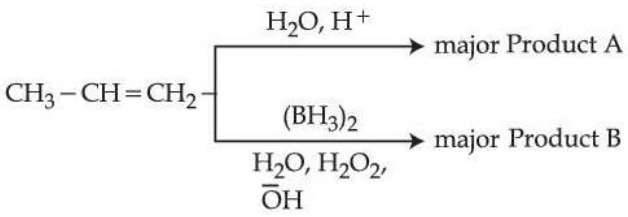
2. 

3. 

4. 

Question Type : **MCQ**
Question ID : **87827056221**
Option 1 ID : **878270220634**
Option 2 ID : **878270220635**
Option 3 ID : **878270220636**
Option 4 ID : **878270220633**
Status : **Answered**
Chosen Option : **4**

Q.64 Identify the product A and product B in the following set of reactions.



Options 1.

A - $CH_3CH_2CH_2-OH$ B - $CH_3CH(OH)-CH_3$

2.

A - $CH_3-CH(OH)-CH_3$ B - $CH_3CH_2CH_2-OH$

3.

A - $CH_3CH_2CH_3$ B - $CH_3CH_2CH_3$

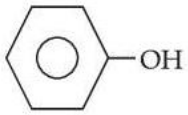
4.

A - $CH_3CH_2CH_2-OH$ B - $CH_3CH_2CH_2-OH$

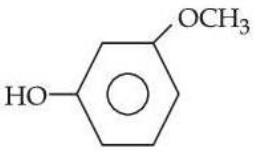
Question Type : **MCQ**
Question ID : **87827056223**
Option 1 ID : **878270220643**
Option 2 ID : **878270220642**
Option 3 ID : **878270220644**
Option 4 ID : **878270220641**
Status : **Answered**
Chosen Option : **2**

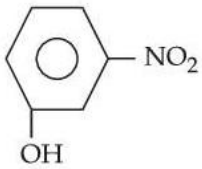
Q.65

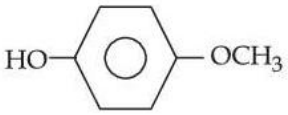
For the given compounds, the correct order of increasing pK_a value :

(A) 

(B) 

(C) 

(D) 

(E) 

Choose the **correct** answer from the options given below :

Options

1. (B) < (D) < (A) < (C) < (E)
2. (E) < (D) < (B) < (A) < (C)
3. (E) < (D) < (C) < (B) < (A)
4. (D) < (E) < (C) < (B) < (A)

Question Type : **MCQ**
Question ID : **87827056224**
Option 1 ID : **878270220647**
Option 2 ID : **878270220646**
Option 3 ID : **878270220645**
Option 4 ID : **878270220648**
Status : **Answered**
Chosen Option : **1**

Q.66 Methods used for purification of organic compounds are based on :

Options 1. neither on nature of compound nor on the impurity present.

2. presence of impurity only.

3. nature of compound only.

4. nature of compound and presence of impurity.

Question Type : **MCQ**
Question ID : **87827056217**
Option 1 ID : **878270220620**
Option 2 ID : **878270220618**
Option 3 ID : **878270220617**
Option 4 ID : **878270220619**
Status : **Answered**
Chosen Option : **4**

Q.67 Identify the **incorrect** statements regarding primary standard of titrimetric analysis.

(A) It should be purely available in dry form.

(B) It should not undergo chemical change in air.

(C) It should be hygroscopic and should react with another chemical instantaneously and stoichiometrically.

(D) It should be readily soluble in water.

(E) $KMnO_4$ & NaOH can be used as primary standard.

Choose the **correct** answer from the options given below :

Options

1. (C) and (E) only

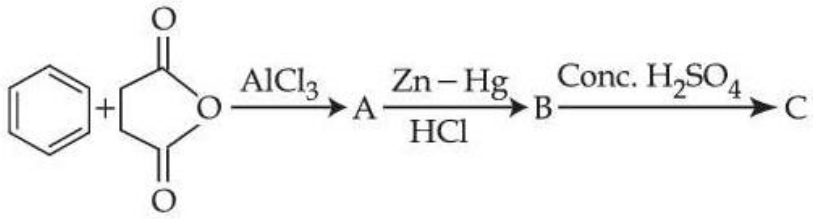
2. (C) and (D) only

3. (A) and (B) only

4. (B) and (E) only

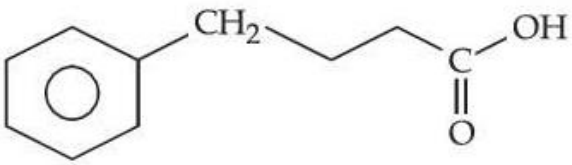
Question Type : **MCQ**
Question ID : **87827056227**
Option 1 ID : **878270220660**
Option 2 ID : **878270220658**
Option 3 ID : **878270220657**
Option 4 ID : **878270220659**
Status : **Not Answered**
Chosen Option : **--**

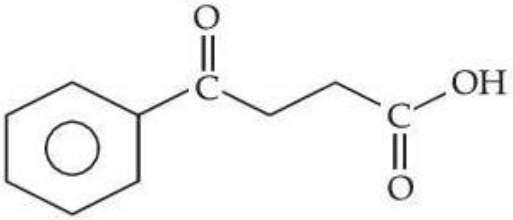
Q.68

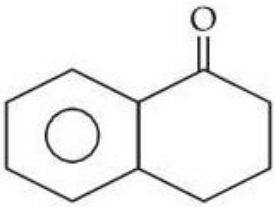


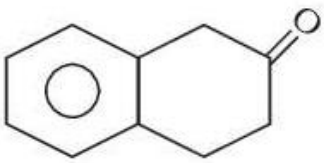
What is the structure of C ?

Options

1. 

2. 

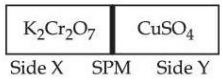
3. 

4. 

Question Type : **MCQ**
Question ID : **87827056219**
Option 1 ID : **878270220628**
Option 2 ID : **878270220627**
Option 3 ID : **878270220625**
Option 4 ID : **878270220626**
Status : **Answered**
Chosen Option : **1**

Q.69 0.05M $CuSO_4$ when treated with 0.01M $K_2Cr_2O_7$ gives green colour solution of $Cu_2Cr_2O_7$. The two solutions are separated as shown below :

[SPM : Semi Permeable Membrane]



Due to osmosis :

Options

1. Molarity of $K_2Cr_2O_7$ solution is lowered.
2. Green colour formation observed on side Y.
3. Green colour formation observed on side X.
4. Molarity of $CuSO_4$ solution is lowered.

Question Type : **MCQ**
Question ID : **87827056210**
Option 1 ID : **878270220591**
Option 2 ID : **878270220589**
Option 3 ID : **878270220590**
Option 4 ID : **878270220592**
Status : **Answered**
Chosen Option : **2**

Q.70 Given below are two statements :

Statement (I) : The oxidation state of an element in a particular compound is the charge acquired by its atom on the basis of electron gain enthalpy consideration from other atoms in the molecule.

Statement (II) : $p\pi - p\pi$ bond formation is more prevalent in second period elements over other periods.

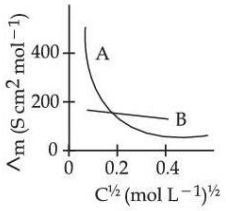
In the light of the above statements, choose the **most appropriate** answer from the options given below :

Options

1. **Statement I** is correct but **Statement II** is incorrect
2. **Statement I** is incorrect but **Statement II** is correct
3. Both **Statement I** and **Statement II** are correct
4. Both **Statement I** and **Statement II** are incorrect

Question Type : **MCQ**
Question ID : **87827056212**
Option 1 ID : **878270220599**
Option 2 ID : **878270220600**
Option 3 ID : **878270220597**
Option 4 ID : **878270220598**
Status : **Answered**
Chosen Option : **3**

Q.71 The molar conductivity for electrolytes A and B are plotted against $C^{1/2}$ as shown below. Electrolytes A and B respectively are :



A B

Options

1. weak electrolyte weak electrolyte
2. strong electrolyte weak electrolyte
3. strong electrolyte strong electrolyte
4. weak electrolyte strong electrolyte

Question Type : **MCQ**
Question ID : **87827056211**
Option 1 ID : **878270220595**
Option 2 ID : **878270220594**
Option 3 ID : **878270220596**
Option 4 ID : **878270220593**
Status : **Answered**
Chosen Option : **4**

Q.72 Correct order of basic strength of Pyrrole  , Pyridine  and Piperidine  is :

Options

1. Pyrrole > Piperidine > Pyridine
2. Pyridine > Piperidine > Pyrrole
3. Piperidine > Pyridine > Pyrrole
4. Pyrrole > Pyridine > Piperidine

Question Type : **MCQ**
Question ID : **87827056225**
Option 1 ID : **878270220651**
Option 2 ID : **878270220652**
Option 3 ID : **878270220649**
Option 4 ID : **878270220650**
Status : **Answered**
Chosen Option : **3**

Q.73 The electronic configuration of Cu(II) is $3d^9$ whereas that of Cu(I) is $3d^{10}$. Which of the following is correct ?

Options 1.

Stability of Cu(I) and Cu(II) depends on nature of copper salts

2. Cu(II) is less stable

3. Cu(II) is more stable

4. Cu(I) and Cu(II) are equally stable

Question Type : **MCQ**
Question ID : **87827056215**
Option 1 ID : **878270220612**
Option 2 ID : **878270220610**
Option 3 ID : **878270220609**
Option 4 ID : **878270220611**
Status : **Answered**
Chosen Option : **3**

Q.74 Given below are two statements : one is labelled as **Assertion (A)** and the other is labelled as **Reason (R)**.

Assertion (A) : S_N2 reaction of $C_6H_5CH_2Br$ occurs more readily than the S_N2 reaction of CH_3CH_2Br.

Reason (R) : The partially bonded unhybridized p-orbital that develops in the trigonal bipyramidal transition state is stabilized by conjugation with the phenyl ring.

In the light of the above statements, choose the **most appropriate** answer from the options given below :

Options

1. **(A)** is not correct but **(R)** is correct

2. **(A)** is correct but **(R)** is not correct

3. Both **(A)** and **(R)** are correct and **(R)** is the correct explanation of **(A)**

4. Both **(A)** and **(R)** are correct but **(R)** is **not** the correct explanation of **(A)**

Question Type : **MCQ**
Question ID : **87827056220**
Option 1 ID : **878270220632**
Option 2 ID : **878270220631**
Option 3 ID : **878270220629**
Option 4 ID : **878270220630**
Status : **Answered**
Chosen Option : **3**

Q.75 Given below are two statements : one is labelled as **Assertion (A)** and the other is labelled as **Reason (R)**.

Assertion (A) : The total number of geometrical isomers shown by $[Co(en)_2Cl_2]^+$ complex ion is three.

Reason (R) : $[Co(en)_2Cl_2]^+$ complex ion has an octahedral geometry.

In the light of the above statements, choose the **most appropriate** answer from the options given below :

Options

1. Both **(A)** and **(R)** are correct but **(R)** is **not** the correct explanation of **(A)**
2. **(A)** is not correct but **(R)** is correct
3. **(A)** is correct but **(R)** is not correct
4. Both **(A)** and **(R)** are correct and **(R)** is the correct explanation of **(A)**

Question Type : **MCQ**
Question ID : **87827056216**
Option 1 ID : **878270220614**
Option 2 ID : **878270220616**
Option 3 ID : **878270220615**
Option 4 ID : **878270220613**
Status : **Answered**
Chosen Option : **2**

Q.76 The F^- ions make the enamel on teeth much harder by converting hydroxyapatite (the enamel on the surface of teeth) into much harder fluoroapatite having the formula.

Options

1. $[3(Ca_3(PO_4)_3)\cdot CaF_2]$
2. $[3(Ca_3(PO_4)_2)\cdot CaF_2]$
3. $[3(Ca_3(PO_4)_2)\cdot Ca(OH)_2]$
4. $[3(Ca_2(PO_4)_2)\cdot Ca(OH)_2]$

Question Type : **MCQ**
Question ID : **87827056214**
Option 1 ID : **878270220608**
Option 2 ID : **878270220607**
Option 3 ID : **878270220605**
Option 4 ID : **878270220606**
Status : **Not Answered**
Chosen Option : **--**

Q.77 Given below are two statements : one is labelled as **Assertion (A)** and the other is labelled as **Reason (R)**.

Assertion (A) : Both rhombic and monoclinic sulphur exist as S_8 while oxygen exists as O_2.

Reason (R) : Oxygen forms $p\pi - p\pi$ multiple bonds with itself and other elements having small size and high electronegativity like C, N, which is not possible for sulphur.

In the light of the above statements, choose the **most appropriate** answer from the options given below :

Options

1. **(A)** is not correct but **(R)** is correct

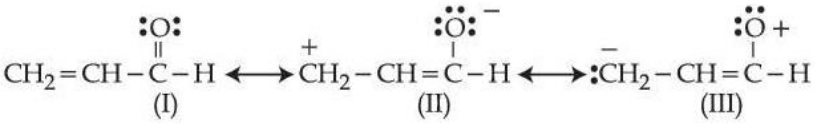
2. **(A)** is correct but **(R)** is not correct

3. Both **(A)** and **(R)** are correct and **(R)** is the correct explanation of **(A)**

4. Both **(A)** and **(R)** are correct but **(R)** is **not** the correct explanation of **(A)**

Question Type : **MCQ**
Question ID : **87827056213**
Option 1 ID : **878270220604**
Option 2 ID : **878270220603**
Option 3 ID : **878270220601**
Option 4 ID : **878270220602**
Status : **Not Answered**
Chosen Option : **--**

Q.78 Relative stability of the contributing structures is :



Options

1. (III) > (II) > (I)

2. (II) > (I) > (III)

3. (I) > (II) > (III)

4. (I) > (III) > (II)

Question Type : **MCQ**
Question ID : **87827056218**
Option 1 ID : **878270220621**
Option 2 ID : **878270220622**
Option 3 ID : **878270220624**
Option 4 ID : **878270220623**
Status : **Answered**
Chosen Option : **3**

Q.79 In which one of the following pairs the central atoms exhibit sp^2 hybridization ?

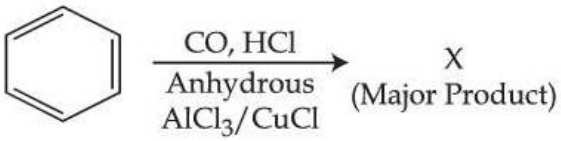
Options

1. BF_3 and NO_2^-
2. NH_2^- and BF_3
3. H_2O and NO_2
4. NH_2^- and H_2O

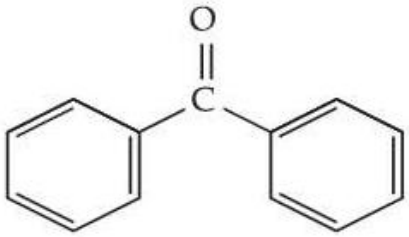
Question Type : **MCQ**
Question ID : **87827056209**
Option 1 ID : **878270220586**
Option 2 ID : **878270220588**
Option 3 ID : **878270220587**
Option 4 ID : **878270220585**
Status : **Answered**
Chosen Option : **1**

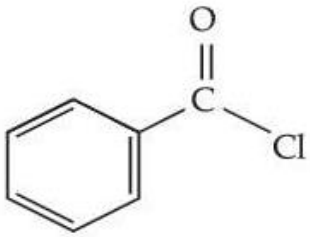
Q.80

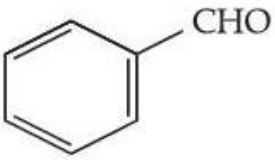
Identify major product "X" formed in the following reaction :

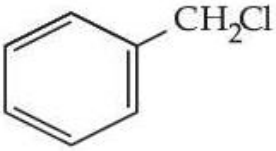


Options

1. 

2. 

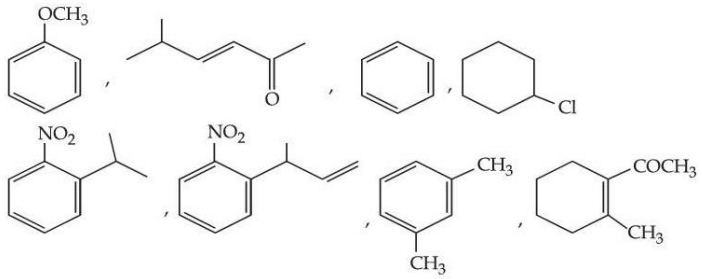
3. 

4. 

Question Type :	**MCQ**
Question ID :	**87827056222**
Option 1 ID :	**878270220640**
Option 2 ID :	**878270220637**
Option 3 ID :	**878270220639**
Option 4 ID :	**878270220638**
Status :	**Answered**
Chosen Option :	**3**

Section : **Chemistry Section B**

Q.81 How many compounds among the following compounds show inductive, mesomeric as well as hyperconjugation effects ?



Given **4**
Answer :

Question Type : **SA**
Question ID : **87827056236**
Status : **Answered**

Q.82 When equal volume of 1M HCl and 1M H_2SO_4 are separately neutralised by excess volume of 1M NaOH solution. x and y kJ of heat is liberated respectively. The value of y/x is _____.

Given **--**
Answer :

Question Type : **SA**
Question ID : **87827056231**
Status : **Not Answered**

Q.83 Molarity (M) of an aqueous solution containing x g of anhyd. $CuSO_4$ in 500 mL solution at 32 °C is 2×10^{-1} M. Its molality will be _____ $\times\ 10^{-3}$ m. (nearest integer).
[Given density of the solution = 1.25 g/mL]

Given **--**
Answer :

Question Type : **SA**
Question ID : **87827056228**
Status : **Not Answered**

Q.84 Number of ambidentate ligands among the following is _____.

NO_2^-, SCN^-, $C_2O_4^{2-}$, NH_3, CN^-, SO_4^{2-}, H_2O.

Given **4**
Answer :

Question Type : **SA**
Question ID : **87827056235**
Status : **Answered**

Q.85 Number of colourless lanthanoid ions among the following is _____.

Eu^{3+}, Lu^{3+}, Nd^{3+}, La^{3+}, Sm^{3+}

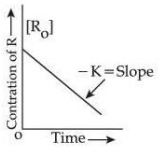
Given --
Answer :

Question Type : **SA**
Question ID : **87827056234**
Status : **Not Answered**

Q.86 Given below are two statements :

Statement I : The rate law for the reaction $A + B \rightarrow C$ is rate $(r) = k[A]^2[B]$. When the concentration of both A and B is doubled, the reaction rate is increased "x" times.

Statement II :



The figure is showing "the variation in concentration against time plot" for a "y" order reaction.

The Value of $x + y$ is _____.

Given **8**
Answer :

Question Type : **SA**
Question ID : **87827056233**
Status : **Answered**

Q.87 The standard reduction potentials at 298 K for the following half cells are given below :

$Cr_2O_7^{2-} + 14H^+ + 6e^- \rightarrow 2Cr^{3+} + 7H_2O$, $E^\circ = 1.33V$

$Fe^{3+}(aq) + 3e^- \rightarrow Fe$ $\quad E^\circ = -0.04V$

$Ni^{2+}(aq) + 2e^- \rightarrow Ni$ $\quad E^\circ = -0.25V$

$Ag^+(aq) + e^- \rightarrow Ag$ $\quad E^\circ = 0.80V$

$Au^{3+}(aq) + 3e^- \rightarrow Au$ $\quad E^\circ = 1.40V$

Consider the given electrochemical reactions,

The number of metal(s) which will be oxidized be $Cr_2O_7^{2-}$, in aqueous solution is _____.

Given **4**
Answer :

Question Type : **SA**
Question ID : **87827056232**
Status : **Answered**

Q.88 The total number of species from the following in which one unpaired electron is present, is _____.

$N_2, O_2, C_2^-, O_2^-, O_2^{2-}, H_2^+, CN^-, He_2^+$

Given --
Answer :

Question Type : **SA**
Question ID : **87827056229**
Status : **Not Answered**

Q.89 The heat of solution of anhydrous $CuSO_4$ and $CuSO_4 \cdot 5H_2O$ are -70 kJ mol^{-1} and $+12$ kJ mol^{-1} respectively.

The heat of hydration of $CuSO_4$ to $CuSO_4 \cdot 5H_2O$ is $-x$ kJ. The value of x is _____. (nearest integer).

Given --
Answer :

Question Type : **SA**
Question ID : **87827056230**
Status : **Not Answered**

Q.90 Total number of essential amino acid among the given list of amino acids is _____.

Arginine, Phenylalanine, Aspartic acid, Cysteine, Histidine, Valine, Proline

Given **3**
Answer :

Question Type : **SA**
Question ID : **87827056237**
Status : **Answered**