

## ABES Engineering College, Ghaziabad B.Tech. First Year, Odd Semester, Session -2023-24 Engineering Chemistry (BAS102)

## **Question Bank (Previous Year University Questions and Practice questions)**

- 1. Illustrate the ion exchange process of water softening.
- 2. On which principle Bomb Calorimeter works? How it is used to determine the GCV & LCV of a fuel?
- 3. Calculate the amount of lime and soda required for treatment of 12,000 liters of water having following salts: MgCl2=10 ppm, CaCl<sub>2</sub>=8ppm, Ca(HCO<sub>3</sub>)<sub>2</sub>=20ppm,Mg(HCO<sub>3</sub>)<sub>2</sub>=10 ppm, CaSO<sub>4</sub>= 15ppm, NaCl=15ppm. (Both lime and soda is 90% pure).
- 4. Discuss proximate analysis of coal. Why is it called as proximate analysis?
- 5. What is ultimate analysis of coal?
- 6. Discuss the preparation, properties and applications of Nylon 6,6, Kevlar and Lucite, Buna-S.
- 7. Discuss the classification and applications of Conducting polymers.
- 8. What are Composite materials? How composites can be classified based on reinforcement and layers? Write the potential applications of composites.
- 9. Illustrate how Grignard Reagent can be utilized for the synthesis of 1<sup>0</sup>, 2<sup>0</sup> and 3<sup>0</sup> alcohols by taking suitable examples.
- 10. Discuss biodegradable polymers with suitable examples and potential applications.
- 17. Differentiate between thermoplastic and thermosetting polymers.
- 12. Differentiate between condensation and addition polymerisation.
- 13. What is the tacticity of polymers?
- 14. Discuss the preparation, properties and applications of Nylon 6, Teflon, Dacron, Orlon, Buna-N
- 15. Give synthesis of Lithium Aluminium Hydride. How will you synthesize primary, secondary and tertiary amine with the help of LiAlH<sub>4</sub>?
- 16. What are polymer dendrimers and polymer blends?
- 17. Discuss the preparation, properties and applications of Nylon 6, Thiokol and Bakelite.
- 18. Give the composition and manufacturing of biogas.
- 19, What is Reverse Osmosis? How it is used to treat to hardness of water?
- 20. Calculate the HCV and LCV of a 3.2 gm of fuel having the following parameters: Weight of water in bomb calorimeter=2500gm Water equivalent of calorimeter=2000 gm Initial temperature=26.8°C Final temperature= 28°C Acid correction=12 cal Cotton thread correction= 15 cal Fuse wire correction= 10 cal Cooling Correction= 1.2°C Amount of Hydrogen=5%.
- 21 Discuss various parameters of proximate analysis of coal. Calculate the HCV and LCV of a fuel having C=81%, S=5%, N=3%, O=4%, H=6%.