

ABES Engineering College, Ghaziabad B.Tech. First Year, Odd Semester, Session -2023-24 Engineering Chemistry (BAS102)

Question Bank (Previous Year University Questions and Practice questions) Unit-2 Stereochemistry and Spectroscopy

Calculate number of ¹ H NMR signals for following compounds:
C ₆ H ₆ , CH ₃ CH ₂ OH, CH ₃ OCH ₃ , CH ₂ Cl-CH ₂ Cl, CH ₂ Br-CH ₂ Cl, CH ₂ =CH ₂ Br.
Discuss shielding and de-shielding in NMR spectroscopy.
Define Lambert- Beer's Law.
Calculate the transmitted power of the UV light if the radiant power of a
solution is reduced to 80%.
Differentiate between δ scale and τ scale for ¹ HNMR spectroscopy.
What are the various electronic transitions of UV-Visible spectroscopy?
Enlist the selection rules of UV-Visible spectroscopy.
Calculate the number of vibrations for given molecules: C ₂ H ₂ , C ₆ H ₆ , CH ₄ .
Calculate the no. of vibrations for given molecules: CO ₂ , H ₂ O, C ₃ H ₆ .
Give the examples of suitable molecules whose prove potential CHIRAL
DRUGS.
Differentiate between hyperchromic shift and hypochromic shift with
examples
Differentiate between chromophore and auxochrome with suitable example.
Differentiate between Bathochromic and Hypsochromic shift with suitable
example.
Why TMS is taken as solvent for ¹ HNMR spectroscopy?
What is atropisomerism? Explain with examples.
Enlist the selection rules of IR spectroscopy.
Deduce different fundamental modes of vibration in IR spectroscopy by taking
the example of AX2 type molecule.
What are the various regions of IR spectroscopy?
Discuss the significance of Functional group region and Fingerprint region in
IR spectroscopy?
An organic compound X with molecular formula C ₃ H ₆ O absorbs at 1720 cm ⁻¹
strongly and gives negative Tollens Test. When it is reduced another compound
Y (C ₃ H ₈ O) appears. In Y absorption at 1710 cm ⁻¹ was missing and a band at
about 3600cm ⁻¹ appeared. Establish the structure of X and Y?
A compound having concentration 10 ⁻³ g/L resulted into absorbance value 0.50
at wavelength 510 nm using 1 cm cell. Calculate the absorptivity and molar
absorptivity values molecular weight of compound is 360.