	Utech
Name:	
Roll No.:	In Summar IV Exercising 2 and Experiment
Invigilator's Signature :	

CS/B.Tech/CT(NEW)/SEM-6/CT-603/2013

2013

CERAMIC COATING AND PROCESS CALCULATIONS

Time Allotted: 3 Hours Full Marks: 70

The figures in the margin indicate full marks.

Candidates are required to give their answers in their own words as far as practicable.

Use separate answer sheets for Group-A and Group-B.

GROUP - A

(Answer Q. No. 1 and any *three* from Q. Nos. 2, 3, 4 and 5. Also answer any *three* from Q. Nos. 6, 7, 8 and 9.)

- 1. Choose the correct alternatives for the following: $5 \times 1 = 5$ i) Which of the following is not a bounding oxide in
 - i) Which of the following is not a bounding oxide in enamel?
 - a) CoO

b) NiO

c) MbO_2

- d) ZrO_2
- ii) To decrease the α value of enamel coating we decrease Na₂O content and substitute it with
 - a) CaO

b) K₂O

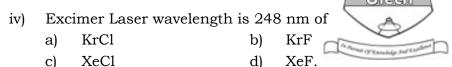
c) ZnO

- d) B_2O_3
- iii) Acid consumptions for pickling process for metal preparation can be reduced by using inhibitors to the extent
 - a) 0.1% to 0.2%
- b) 1% to 2%
- c) 10% to 20%
- d) 30% to 40%.

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- v) Which instrument is used for surface composition analysis?
 - a) XPSb) TEMc) XRDd) AFM.
- 2. State some important properties imparted by any two of the following oxides in enamel: $2\frac{1}{2}$
 - i) Al_2O_3
 - ii) PbO
 - iii) ZrO₂
- 3. Which acid (amongst HCl and H_2SO_4) do you prefer for pickling process and why? Why do we use inhibitors in pickling process? $1\frac{1}{2} + 1$
- 4. What are the differences between Spin Coating and Dip Coating? $2\frac{1}{2}$
- 5. Make a comparative analysis of APCVD, LPCVD and PECVD.
- 6. Why are opacifiers used in enamel coating? State the differences between Sheet Glass and Enamel Glass. Why do we use electrolyte in enamel? Give an example. What is neutralization process in metal preparation before enamel coating? $2 + 2 + 2 + 1\frac{1}{2}$
- 7. What is Fish Scaling in enamel coating? What is vitreous coating? Give an example. Explain in brief Degreasing process for metal preparation. What is Direct-on-Enameling?

$$2 + 1\frac{1}{2} + 3 + 1$$

8. What are nucleation and growth? What are the factors that influence the film growth? Explain the effects of various factors on the growth of grain size in thin film.

 $1\frac{1}{2} + 1\frac{1}{2} + 3 + 1\frac{1}{2}$

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9. What is sputtering? Enumerate various characteristics of sputtering process and sputtering yield. Discuss the advantages and disadvantages of sputter deposition.

 $1 + 2\frac{1}{2} + 2 + 2$

GROUP - B

(Answer any *one* from Q. Nos. 1 and 2 and any *three* from the rest.)

1. A natural gas has the following composition:

 $CH_4 - 82.5\%$, $C_2H_6 - 13\% & N_2 - 4.5\%$.

Calculate the density of the gas mixture at S.T.P. in kg/m^3 .

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2. Calculate the standard heat of reaction of the following, using Hess's law:

$$C_5H_{12}(l) + 8O_2(g) \rightarrow 5CO_2(g) + 6H_2O(l)$$

Given,	component	ΔH_f° cal/gm.mole
	$\overline{\mathrm{C_{5}H_{12}}\left(\ l\ \right)}$	- 41370
	CO_2 (g)	<i>–</i> 94051
	$H_2O(l)$	- 68315

- 3. A hydrocarbon gas is burnt in a furnace with 15% excess air. It gives out flue gas of composition $CO_2 10.9\%$, CO 2.5%, $H_2 1.8\%$, $O_2 4.2\%$ and rest N_2 . Check whether the combustion is complete or not. Is the amount of reported excess air correct or not.
- 4. a) Given the formula of a glaze

$$0.65 \text{ PbO}$$
 1.8 SiO_2

 $0.35 \text{ Na}_2\text{O} \qquad 0.25 \text{ Al}_2\text{O}_3 \qquad 0.5 \text{ B}_2\text{O}_3$

Calculate the formula weight and the % composition of the glaze.

b) A glaze has the formula PbO, $0.23~{\rm Al_2O_3}$, $2.7~{\rm SiO_2}$. Calculate a recipe if China Clay (${\rm Al_2O_3}$, $2{\rm SiO_2}$, $2{\rm H_2O}$), flint (SiO₂) and lead bisilicate are used as raw material.

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5. The formula of a glaze is

0.082 K₂O

 $0.027 \text{ Na}_2\text{O} \quad 0.231 \text{ Al}_2\text{O}_3$, 4.064 SiO_2

0.294 CaO

0.597 PbO

Using flint, China clay, lead bisilicate, whiting and stone calculate the recipe. The stone has the following formula:

0.414 K₂O

 $0.146 \text{ Na}_2\text{O}$, $1.0 \text{ Al}_2\text{O}_3$, 7.1 SiO_2

0.180 CaO

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6. Calculate the batch composition of the glass from the following oxide composition:

$$\mathrm{SiO}_2-71\%$$
, $\mathrm{Na_2O}-17\%$, $\mathrm{CaO}-9.5\%$ and $\mathrm{Al_2O_3}-2.5\%$

The raw materials used are sand, lime-stone, soda-felspar, anhydrous soda ash. 25% cullet is added whose composition is as follows: SiO_2 — 74%, Na_2O — 16%, CaO — 9%, Al_2O_3 — 1%.

7. Calculate the oxide composition of the glass from the given batch:

Sand — 1000 parts, limestone — 184 parts, red lead — 1287 parts, As_2O_3 — 5 parts, soda ash (with 2% moisture) — 121 parts, salt-cake — 14 parts and KNO_3 (with 7% moisture) — 80 parts.

[Red lead —
$$Pb_3O_4$$
, Salt-cake — Na_2SO_4] 10

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