

1st Semester B.Tech. End Term Examination 2019-2020

ENGINEERING CHEMISTRY(18BS1T05)

Duration: 03:00

Full Marks: 60

1 Answer All

- a Mineral acid should be removed from the water before it is used for softening by the Zeolite process. Give reason. 1
- b Corrosion always occurs at the anode. Give reason. 1
- c Which apparatus is used to determine the kinematic viscosity of lubricating oil? 1
- d Mention the unit of absorbance and transmittance. 1
- e What is the relationship between HCV and LCV? 1
- f Give an example of a reducing agent used in the synthesis of nanomaterials. 1

2 Answer All

- a Differentiate between soft water and hard water. 2
- b Differentiate between electrochemical series and Galvanic series. 2
- c What is cutting fluid? What are the functions of cutting fluid? 2
- d λ_{\max} for aniline shifts from 230 nm in neutral medium to 203 nm in acidic medium. State the reason. 2
- e Briefly discuss the various characteristics of an ideal fuel. 2
- f What is the need for a capping agent in the synthesis of the nanoparticles? 2

3 Answer any One

- a Give a comparative account on the 'Ion-exchange resin process' & 'Zeolite process' of the softening of water. A water sample contains the following constituents; $\text{MgSO}_4 = 12 \text{ mg/lit}$, $\text{CaSO}_4 = 13.6 \text{ mg/lit}$ & $\text{Ca(HCO}_3)_2 = 162 \text{ mg/lit}$. Calculate the total hardness of the water sample (as CaCO_3 equivalent) in $^\circ\text{Cl}$ unit. 7
- b Calculate the amount of soda (90% pure) required for the treatment of 10^5 L of water containing CaCl_2 (111 ppm) & CaSO_4 (272 ppm). 7

4 Answer any One

- a Write the mechanism of corrosion of Fe that takes place in a neutral medium. How can corrosion of the ship hull be controlled? 7
- b What is differential aeration corrosion? Discuss its mechanism of waterline corrosion with a suitable example. 7

[P. T. O.]

5 Answer any One

- a Explain the following properties of lubricants & their significance; (i) Viscosity & Viscosity index, (ii) Flash & fire point, (iii) Acid value 7
- b How are lubricants classified? Write the functions(4) of a lubricant. Write the names of additives used to improve the following properties of a lubricant; (i) Extreme pressure, (ii) Viscosity index 7

6 Answer any One

- a What is 'Electronic spectroscopy'? Write the applications of electronic spectroscopy. Define the following terms: i. Hypsochromic shift, ii. Bathochromic shift, iii. Hypochromic shift, iv. Hyperchromic shift 7
- b Which types of molecules are IR active? Calculate the vibrational energy & force constant of HF if its frequency of oscillation is $4 \times 10^{14} \text{sec}^{-1}$. (Atomic mass of H = 1.0078u & F = 18.9984u) 7

7 Answer any One

- a A producer gas has the following average composition by volume %: CO = 30%, N₂ = 55%, H₂ = 8%, and CO₂ = 7%. Give a reason why the producer gas has low calorific value compared to water gas. Find (i) volume of air required for combustion of 100 m³ of this gas, and (ii) volume of air required if 50 % excess air was used for combustion. 7
- b How to select a good coal sample for a thermal power plant? Discuss the proximate analysis of coal and its significance. 7

8 Answer any One

- a Discuss the method of preparation, characterization, and some applications of zinc oxide nanoparticles. 7
- b Write short notes on (a) Top-down method, (b) Bottom-up method, and (c) Green synthetic route to synthesize nanomaterials. 7
