



Name :

Roll No. :

Invigilator's Signature :

CS/B.TECH (CT)/SEM-7/CHE(CT)-701/2012-13

2012

INSTRUMENTATION AND PROCESS CONTROL

Time Allotted : 3 Hours

Full Marks : 70

The figures in the margin indicate full marks.

*Candidates are required to give their answers in their own words
as far as practicable.*

GROUP – A

(Multiple Choice Type Questions)

1. Choose the correct alternatives for the following : $10 \times 1 = 10$

i) If the impulse response of a system is te^{-t} , the transfer function of the system is

a) $\frac{1}{s^2}$

b) $\frac{1}{s(s+1)}$

c) $\frac{1}{(s+1)^2}$

d) $\frac{1}{s^2(s+1)}$



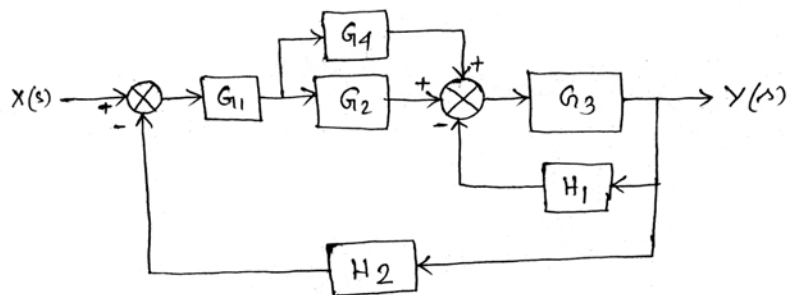
GROUP – B

(Short Answer Type Questions)

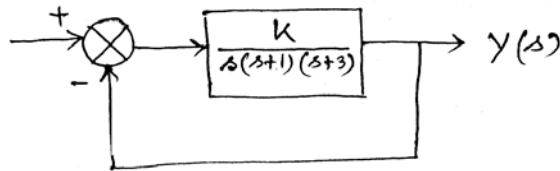
Answer any *three* of the following

$3 \times 5 = 15$

2. Reduce the following block diagram and find $\frac{Y(s)}{X(s)}$.



3. Find the range of k for stability of the following system.



4. A unit step change is introduced into a system having the

$$\text{transfer function } \frac{Y(s)}{X(s)} = \frac{10}{s^2 + s + 4}.$$

Determine (i) Rise time, (ii) Percent overshoot.

5. Describe how dynamic characteristics are explained in the light of standard inputs.



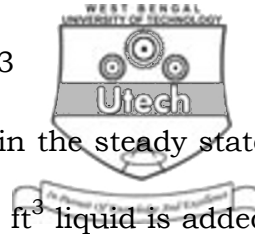
GROUP – C

(Long Answer Type Questions)

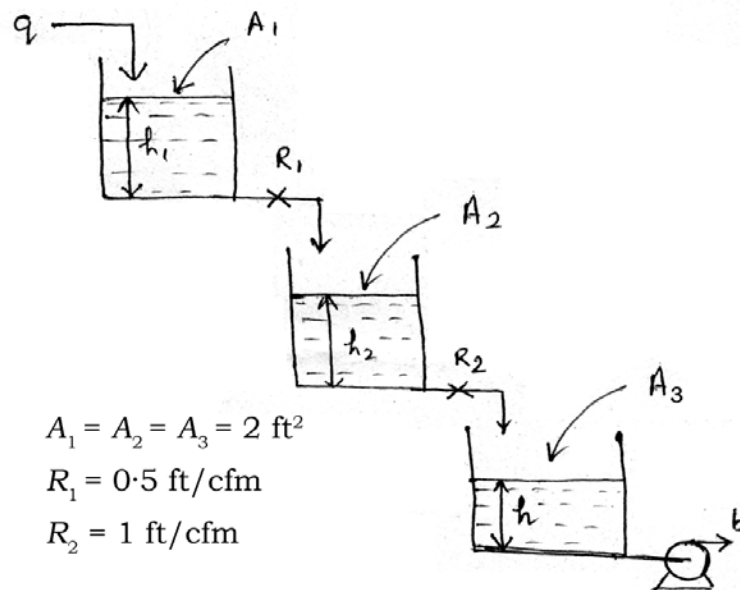
Answer any *three* of the following. $3 \times 15 = 45$

6. A thermometer having first order dynamics with a time constant of 1 min is placed in a temperature bath at 110°F. After it reaches steady state, it is suddenly placed in a bath at 100°F at $t = 0$ and left there for 1 min, after which it is immediately returned to the bath at 110°F.
- a) Draw a sketch showing the variation of the surrounding temperature for the thermometer.
 - b) Find the expression for thermometer reading.
 - c) Draw a sketch showing thermometer reading as a function of time.
 - d) Calculate thermometer readings at $t = 0.5$ min and $t = 2$ min.

$2 + 6 + 3 + 4$

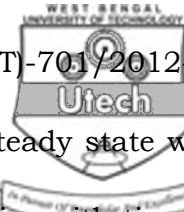


7. The three tank system shown in figure is in the steady state with $q = 2$ cfm and $h = 1.5$ ft. Suddenly 10 ft^3 liquid is added to the first tank at $t = 0$



Find

- initial liquid level in each tank
- transfer function between q and h
- variation in h , the actual level in the last tank
- final liquid level (actual) in the last tank. $3 + 4 + 6 + 2$



8. The mixing process shown in figure-1 is a steady state with the input concentration x of 1 lb/ft^3 . If x varies with time as shown in figure-2,

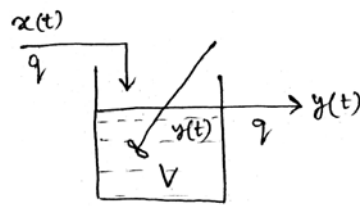


Figure 1

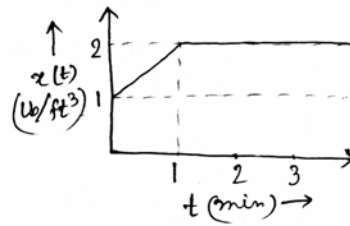


Figure 2

- derive the transfer function of the system
 - find $x(s)$
 - find initial and final output concentration (actual)
 - find the expression for actual output concentration, $y(t)$
- 4 + 3 + 4 + 4

9. How the Gas Purge technique is used to measure the liquid level in a tank ? Describe the operation of an optical pyrometer with necessary sketch. 7 + 8

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