Lecture 18

The variational principle

The variational principle let you get an **upper bound** for the ground state energy when you can not directly solve the Schrödinger's equation.

How does it work?

- (1) Pick any normalized function Ψ .
- (2) The ground state energy Eqs is

3) Some choices of the trial function ψ will get your E_{gs} that is close to actual value.

The ground state of Helium

Two electrons orbiting the nucleus with charge Z=2.

H =
$$-\frac{t^2}{2m} \left(\nabla_1^2 + \nabla_2^2 \right) - \frac{e^2}{4\pi\epsilon_0} \frac{2}{r_1} - \frac{e^2}{4\pi\epsilon_0} \frac{2}{r_2}$$

+
$$\frac{e^2}{4\pi\epsilon_0} \frac{1}{|\vec{r}_1 - \vec{r}_2|}$$
 \leftarrow Electron- electron repulsion term

(ignoring fine structure and smaller corrections).

Experimental result:
$$E_{gs} = -78.975 \,\text{eV}$$

Our task: use variational principle to get as close as possible to experimental result.

If we ignore term

our problem reduces to two independent Hydrogen-like hamiltonians with Z=2. In this case, the solution for the ground state is just a product of two hydrogen ground state wave functions with Z=2.

$$\psi_0(\vec{r}_1, \vec{r}_2) = \psi_{100}(\vec{r}_1) \psi_{100}(\vec{r}_2) = \frac{8}{\pi a^3} e^{-2(r_1 + r_2)/a}$$
 (1)

The energy is just the sum of two hydrogen-like energies with Z=2:

$$E_{n} = \frac{2^{2}E_{1}}{h^{2}} \qquad n=1, \ T=2 \qquad E_{100} = 4E_{1},$$

$$E_{1} = -13.6 \text{ eV} \qquad = 7$$

$$E_{He} = 2E_{100} = 8E_{1} = -109 \text{ eV}.$$

Rather far from experiment value of -79eV.

To get a better approximation, we apply variational principle with trial function (1).

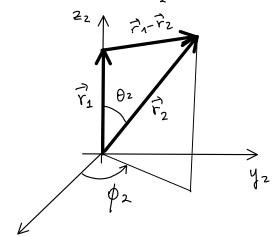
$$H = -\frac{t^{2}}{2m} \left(\nabla_{1}^{2} + \nabla_{2}^{2} \right) - \frac{e^{2}}{4\pi\epsilon_{0}} \frac{1}{r_{1}} - \frac{e^{2}}{4\pi\epsilon_{0}} \frac{2}{r_{2}} + \frac{e^{2}}{4\pi\epsilon_{0}} \frac{1}{|\vec{r_{1}} - \vec{r_{2}}|} \right)$$
Hydrogen-like Ho
$$H_{0} \psi_{0} = 8E_{1} \psi_{0} = 7$$

$$H_{0} \psi_{0} = 8E_{1} \psi_{0} = 8E_{1} + V_{ee} \psi_{0}$$

$$(\psi_{0}) + (\psi_{0}) + (\psi_{0})$$

$$\langle Vel \rangle = \frac{e^2}{4\pi\epsilon_0} \left(\frac{8}{\pi a^3} \right)^2 \int \frac{e^{-4(r_1 + r_2)}}{|\vec{r}_1 - \vec{r}_2|} d^3\vec{r}_2 d^3\vec{r}_2$$

We calculate \vec{r}_2 integral first, we orient our coordinate system so z is along \vec{r}_1 .



$$|\vec{r}_1 - \vec{r}_2| = \sqrt{r_1^2 + r_2^2 - 2r_1 r_2 \cos \theta_2}$$

$$T_2 = \int \frac{e^{-4r_2/a}}{|\vec{r}_1 - \vec{r}_2|} d^3r_2$$

$$T_{2} = \int \frac{e^{-4r_{z}/\alpha}}{\sqrt{r_{1}^{2} + r_{2}^{2} - 2r_{1}r_{2} \cos \theta_{z}}} r_{z}^{2} \sin \theta_{z} dr_{z} d\theta_{z} d\phi_{z}$$

$$r_2^2$$
 sin θ_2 d r_2 d θ_2 d ϕ_2

Integral over ϕ_2 gives 2π

Integral over Θ_2

$$\int_{0}^{\pi} \frac{\sin \Theta_{2}}{\sqrt{r_{1}^{2} + r_{2}^{2} - 2r_{1}r_{2}\cos\Theta_{2}}} d\Theta_{2} = \frac{\sqrt{r_{1}^{2} + r_{2}^{2} - 2r_{1}r_{2}\cos\Theta_{2}}}{r_{1}r_{2}}$$

$$= \frac{1}{r_1 r_2} \left(\sqrt{r_1^2 + r_2^2 + 2r_1 r_2} - \sqrt{r_1^2 + r_2^2 - 2r_1 r_2} \right) =$$

$$= \frac{1}{r_{1}r_{2}} \left((r_{1}+r_{2}) - 1r_{1}-r_{2}1 \right) = \begin{cases} \frac{1}{r_{1}r_{2}} \left[r_{1}+r_{2} - r_{1}+r_{2} \right] r_{2} c_{r_{1}} \\ \frac{1}{r_{1}r_{2}} \left[r_{1}+r_{2} + r_{1}-r_{2} \right] r_{2} r_{r_{1}} \end{cases}$$

$$= \begin{cases} \frac{2}{r_{1}} & \text{if } r_{2} c_{r_{1}} \\ \frac{2}{r_{2}} & \text{if } r_{2} \gamma r_{1} \end{cases}$$

$$I_{2} = 4\pi \left(\int_{0}^{r_{1}} e^{-4r_{2}/a^{2}} r_{2}^{2} \left(\frac{1}{r_{1}} \right) dr_{2} + \int_{r_{1}}^{\infty} e^{-4r_{2}/a^{2}} r_{2}^{2} \left(\frac{1}{r_{2}} \right) dr_{2} \right)$$

$$= \frac{\pi a^{3}}{8r_{1}} \left[1 - \left(1 + \frac{2r_{1}}{a} \right) e^{-4r_{1}/a} \right] = \gamma$$

$$e^{2} \left(8 \right) \left(\Gamma_{1} + \left(1 + \frac{2r_{1}}{a} \right) e^{-4r_{1}/a} \right) - \frac{4r_{1}/a}{a} \right] - \frac{4r_{1}/a}{a}$$

$$\langle v_{\ell\ell} \gamma = \frac{e^2}{4\pi\epsilon_0} \left(\frac{8}{\pi a^3} \right) \int \left[1 - \left(1 + \frac{2r_1}{a} \right) e^{-4r_1/a} \right] e^{-4r_1/a}$$

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Angular integrals give 4π , integral over r_1 gives $\frac{56^{2}}{128}$.

$$\langle v \rangle = \frac{5}{4a} \left(\frac{e^2}{4\pi 60} \right) = -\frac{5}{2} E_1 = 34eV$$

 $\langle H \rangle = -109eV + 34eV = -75eV$

Much closer to -79 eV! We can do even better!

The ground state of Helium: how to improve the result?

Our initial trial function was:

Function was:
$$\psi_o = \frac{8}{\pi a^3} e^{-2(r_1 + r_2)/a}$$

Now, we take
$$\gamma_0 = \frac{z^3}{\pi a^3} e^{-\frac{z(r_1+r_2)a}{2}}$$

and take Z to be a parameter. We re-write the Hamiltonian as

and take 2 to be a parameter. We re-write the Hamiltonian as
$$H = \begin{bmatrix} \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} \\ \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} \\ \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} \\ \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} \\ \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} \\ \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} \\ \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} \\ \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} \\ \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} \\ \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} & \frac{1}{2} \\ \frac{1}{2} & \frac{1}{2} \\ \frac{1}{2} & \frac{1}{2}$$

Note, that we did not change our Hamiltonian (we are not allowed to do that in the variational method). We just added and subtracted

$$\frac{e^2}{4\pi\epsilon_0} \left(\frac{2}{r_1} + \frac{2}{r_2} \right)$$

We now calculate the expectation value

with our "new" trial function

Function
$$\psi_0 = \frac{z^3}{\pi a^3} e^{-\frac{z}{r_1 + r_2}/a} = \psi_{100}(\overline{r_1}) \psi_{100}(\overline{r_2})$$
with charge \overline{z}

$$< \psi_0 | H_0 | \psi_0 > = < \psi_0 | E_0 \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0 > = 2 E_1 \frac{2^2}{1^2} < \psi_0 | \psi_0$$

Note: if Z=2 $\langle H_0 \rangle = 2E_1 \cdot 2^2 = 8E_1$ as before.

$$=\frac{e^2}{4\pi60}(7-2)\iint \gamma_{100}^{*}(\vec{r}_1)\gamma_{100}^{*}(\vec{r}_2)\frac{1}{r_1}\gamma_{100}(\vec{r}_2)\frac{1}{r_1}\gamma_{100}(\vec{r}_2)d^3\vec{r}_1d^3\vec{r}_2$$

$$= \frac{e^{2}}{4\pi\epsilon_{0}}(z-2) \left\{ \int_{100}^{4} (\vec{r}_{1}) \frac{1}{r_{1}} \psi_{100}(\vec{r}_{1}) d\vec{r}_{1} \int_{100}^{4} (\vec{r}_{2}) \psi_{100}(\vec{r}_{2}) d\vec{r}_{2} \right.$$

$$\left. \left(\psi_{15} \right| \frac{1}{r} \left| \psi_{15} \right) = \left(\frac{1}{r} \right)_{15} \int_{150}^{4} (normalized)$$

$$+ \underbrace{\int \psi_{100}^{*}(\bar{r}_{1}) \psi_{100}(\bar{r}_{1}) d^{3}\bar{r}_{1}}_{=1} \int \psi_{100}^{*}(\bar{r}_{2}) \frac{1}{c_{2}} \psi_{100}^{*}(\bar{r}_{2}) d^{3}\bar{r}_{2}}$$

$$=2\frac{e^{2}}{4\pi\epsilon_{0}}(2-2)\langle\frac{1}{\Gamma}\rangle_{15}$$

We need to calculate $\langle \frac{1}{r} \rangle_{is} = \langle \psi_{is} | \frac{1}{r} | \psi_{1s} \rangle$ $\psi_{is} = \sqrt{\frac{2^{3}}{\pi a^{3}}} e^{-\frac{2}{r} / a_{o}}$

where a_0 is the Bohr radius $a_0 = \frac{4\pi\epsilon_0}{me^2} t^2$

Note:
$$a = \frac{4\pi + 6}{me^2 + 2} = \frac{a_0}{2}$$
 since $e^2 \rightarrow e^2 + 2$ for H-like atoms.

$$\angle \frac{1}{\Gamma} = \frac{Z^3}{\Pi a_0^3} \int_0^\infty e^{-2Zr/a_0} \frac{1}{\Gamma} r^2 dr \int_0^\infty \sin \theta d\theta \int_0^\infty d\phi$$

$$= \frac{4\pi z^{3}}{\pi a^{3}} \int_{e}^{e} e^{-2zr/a_{0}} r dr = \frac{4\pi z^{3}}{\pi a^{3}} \frac{a_{0}^{3}}{4z^{2}} = \frac{z}{a_{0}}$$

We already calculated the third term:

$$\angle V_{ee} = \frac{5}{4a} \left(\frac{e^2}{4\pi\epsilon_0} \right)$$
 (previous result)

For our new trial function, $a \rightarrow 2a_o/2$

$$\langle \vee u \rangle = \frac{5}{4a_0} \frac{2}{2} \frac{e^2}{4\pi\epsilon_0} = \frac{52}{8a_0} \frac{e^2}{4\pi\epsilon_0}$$

Putting it all together, we get

$$\langle H \rangle = 2E_1 2^2 + 2(2-2) 2 \sqrt{\frac{e^2}{4\pi\epsilon_o}} + \frac{52}{8} \sqrt{\frac{e^2}{4\pi\epsilon_o}} + \frac{1}{8}$$
For convenience, let's express all terms via E₁:
$$\Gamma = -\frac{m}{2} \left(\frac{e^2}{2}\right)^2$$

$$E_{1} = -\frac{m}{2h^{2}} \left(\frac{e^{2}}{4\pi\epsilon_{0}}\right)^{2}$$

$$a_{2} = \frac{4\pi\epsilon_{0}}{me^{2}} h^{2} = \left(\frac{e^{2}}{4\pi\epsilon_{0}}\right)^{-1} \frac{h^{2}}{m} = 0$$

$$\frac{e^{2}}{4\pi\epsilon_{0}} \frac{1}{a_{2}} = \frac{e^{2}}{4\pi\epsilon_{0}} \frac{e^{2}}{4\pi\epsilon_{0}} \frac{m}{h^{2}} = 0$$

$$\frac{e^{2}}{4\pi\epsilon_{0}} \frac{1}{a_{2}} = \frac{e^{2}}{4\pi\epsilon_{0}} \frac{e^{2}}{4\pi\epsilon_{0}} \frac{m}{h^{2}} = 0$$

$$(H) = 2E_1 Z^2 - 4E_1 (Z-2) Z - \frac{5}{4} Z E_1$$

= $E_1(2Z^2 - 4Z^2 + 8Z - \frac{5}{4}Z) = E_1(-2Z^2 + \frac{27}{4}Z)$

Therefore, for any Z

We yet the lowest upper bound when $\langle +| \rangle$ is minimized, i.e. $\frac{d\langle +| \rangle}{dz} = 0$.

Class exercise: minimize $\langle H \rangle$. Find Z and get the lowest upper bound for E_{gs} (i.e. a number in eV).

$$\frac{d}{dz} < H7 = \left(-4z + \frac{24}{4}\right) = 1.69$$

$$Z = \frac{27}{16} = 1.69$$

$$< H7 = E_1 \left(-2z^2 + \frac{27}{4}z\right) = -13.6 \left(-2.(1.69)^2 + \frac{27}{4}(.69)\right)$$

$$< H7 = -77.5 eV$$

Even closer to the experimental value -79.0 eV!

Summary: variational method

The variational principle let you get an **upper bound** for the ground state energy when you can not directly solve the Schrödinger's equation.

How does it work?

- (1) Pick any normalized function ψ .
- (2) The ground state energy Egs is

3) Some choices of the trial function ψ will get your E_{gs} that is close to actual value.