Total Energy for GFN2-xTB

$$E_{GFN2-xTB} = E_{rep}^{(0)} + E_{disp}^{(0,1,2)} + E_{EHT}^{(1)} + E_{IES+IXC}^{(2)} + E_{AES+AXC}^{(2)} + E_{IES+IXC}^{(3)}$$
$$= E_{rep} + E_{disp}^{D4'} + E_{EHT} + E_{\gamma} + E_{AES} + E_{AXC} + E_{\Gamma}^{GFN2}$$
(1)

Repulsion Energy 1.1

$$E_{rep} = \frac{1}{2} \sum_{A,B} \frac{Z_A^{eff} Z_B^{eff}}{R_{AB}} e^{-\sqrt{a_A a_B} (R_{AB})^{(k_f)}}$$
 (2)

$$k_f = \begin{cases} 1 & if A, B \in \{H, He\} \\ \frac{3}{2} & otherwise \end{cases}$$
 (3)

 Z^{eff} and a are variables fitted for each element. A,B are the labels of atoms. Since we only have C and H in our systems we can simplify this quite a bit in code. R_{AB} is the distance between the A and B atoms.

Extended Hückel Theory Energy 1.2

$$E_{EHT} = \sum_{\mu\nu} P_{\mu\nu} H_{\mu\nu}^{EHT} \tag{4}$$

where μ and ν are AO indecies, l and l' index shells. Both AO's are associated with an atom labled A and B.

$$P_{\mu\nu} = P_{\mu\nu}^{(0)} + \delta P_{\mu\nu} \tag{5}$$

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$$P_{\mu\nu}^{(0)} = \sum_{j}^{occ.MO} c_{\mu j} c_{j\nu}$$
(5)

$$\delta P_{\mu\nu} = ??$$
 comes from the iteration, can be skipped for now (7)

$$H_{\mu\nu}^{EHT} = \frac{1}{2} K_{AB}^{ll'} S_{\mu\nu} (H_{\mu\mu} + H_{\nu\nu})$$

$$\cdot X(EN_A, EN_B)$$

$$\cdot \Pi(R_{AB}, l, l')$$

$$\cdot Y(\zeta_l^A, \zeta_{l'}^B), \forall \mu \in l(A), \nu \in l'(B)$$
(8)

 $K_{AB}^{ll'}$ is a element and shell specific fitted constant however, in GFN2 it only depends on the shells. $S_{\mu\nu} = \langle \phi_{\mu} | \phi_{\nu} \rangle$ is just the overlap of the orbitals. In GFN2 $H_{\kappa\kappa} = h_A^l - \delta h_{CN_A'}^l CN_A'$ where CN_A' is the modified GFN2-type Coordinate Number for the element of atom A. h_A^l and $\delta h_{CN_A}^l$ are both fitted constants. EN_A is the electronegativity of the element of atom A.

$$X(EN_A, EN_B) = 1 + k_{EN}\Delta E N_{AB}^2 \tag{9}$$

$$k_{EN} = 0.02 \text{ in GFN2}$$
 (10)

$$\Delta E N_{AB}^2 = (EN_A - EN_B)^2 \tag{11}$$

The electronegativity for C and H are 2.55 and 2.20 according to wikipedia. Thus here is a table for the combinations we will be working with:

$$\begin{array}{c|cccc} A & B & X(EN_A, EN_B) \\ \hline C & C & 1 \\ C & H & 1 + 0.02 \cdot (0.35^2) \\ H & C & 1 + 0.02 \cdot (0.35^2) \\ H & H & 1 \end{array}$$

$$\Pi(R_{AB}, l, l') = \left(1 + k_{A, l}^{\text{poly}} \left(\frac{R_{AB}}{R_{\text{cov}, AB}}\right)^{\frac{1}{2}}\right) \left(1 + k_{B, l'}^{\text{poly}} \left(\frac{R_{AB}}{R_{\text{cov}, AB}}\right)^{\frac{1}{2}}\right)$$
(12)

$$Y(\zeta_l^A, \zeta_{l'}^B) = \left(\frac{2\sqrt{\zeta_l^A \zeta_{l'}^B}}{\zeta_l^A + \zeta_{l'}^B}\right)^{\frac{1}{2}}$$

$$\tag{13}$$

Here, ζ_l^A are the STO exponents of the GFN2-xTB AO basis. Slater Type Orbitals are defined as such:

$$\chi_{\zeta,n,l,m}(r,\theta,\varphi) = NY_{l,m}(\theta,\varphi)r^{n-1}e^{-\zeta r}$$
(14)

N is a normalisation constant, Y are spherical harmonic funtions, n, l, m are the quantum numbers for the AO. r, θ, φ are polar 3D coordinates. ζ determines the radial extent of the STO, a large value gives rise to a function that is "tight" around the nucleus and a small value gives a more "diffuse" function. This ζ is the one mentioned in the Y term of E_{EHT} and is a value fitted when constructing the basis set, thus it is given to us.

 $^{^1\}mathrm{p}.2109, \quad \text{https://www.taylorfrancis.com/books/mono/}10.1201/b12286/crc-handbook-chemistry-physics-william-haynes$

1.3 Isotropic electrostatic and XC energy

1.3.1 Second order

$$E_{\gamma} = \frac{1}{2} \sum_{A,B}^{N_{atoms}} \sum_{l \in A} \sum_{l' \in B} q_l q_{l'} \gamma_{AB,ll'}$$
 (15)

$$\gamma_{AB,ll'} = \frac{1}{\sqrt{R_{AB}^2 + \eta_{AB,ll'}^{-2}}}$$
 (16)

$$\eta_{AB,ll'} = \frac{1}{2} \left[\eta_A (1 + k_A^l) + \eta_B (1 + k_B^{l'}) \right]$$
 (17)

 q_l is a partial muliken charge. η_A and η_B are element-specific fit parameters, while k_A^l and $k_B^{l'}$ are element-specific scaling factors for the individual shells $(k_A^l=0$ when l=0).

$$GAP_A = \sum_{l \in A} q_l \tag{18}$$

$$q_l = \sum_{l' \in B} P_{ll'} S_{ll'} = GOP_l \tag{19}$$

1.3.2 Third order

$$E_{\Gamma}^{GFN2} = \frac{1}{3} \sum_{A}^{N_{atoms}} \sum_{l \in A} (q_l)^3 K_l^{\Gamma} \Gamma_A$$
 (20)

 K_l^{Γ} is a shell specific constant common for all elements and Γ_A is an element specific constant.

1.4 Anisotropic electrostatic energy

$$E_{AES} = E_{q\mu} + E_{q\Theta} + E_{\mu\mu}$$

$$= \frac{1}{2} \sum_{A,B} \{ f_3(R_{AB}) [q_A(\boldsymbol{\mu}_B^T \boldsymbol{R}_{BA}) + q_B(\boldsymbol{\mu}_A^T \boldsymbol{R}_{AB})]$$

$$+ f_5(R_{AB}) [q_A \boldsymbol{R}_{AB}^T \boldsymbol{\Theta}_B \boldsymbol{R}_{AB} + q_B \boldsymbol{R}_{AB}^T \boldsymbol{\Theta}_A \boldsymbol{R}_{AB}$$

$$- 3(\boldsymbol{\mu}_A^T \boldsymbol{R}_{AB}) (\boldsymbol{\mu}_B^T \boldsymbol{R}_{AB}) + (\boldsymbol{\mu}_A^T \boldsymbol{\mu}_B) R_{AB}^2] \}$$

$$(21)$$

 μ_A is the cumulative atomic dipole moment of atom A and Θ_A is the corresponding traceless quadrupole moment. The curly braces and brackets are used in the same way as normal parenthesis for showing order of operations. q_A is

the atomic charge of atom A.

$$\Theta_A^{\alpha\beta} = \frac{3}{2}\theta_A^{\alpha\beta} - \frac{\delta_{\alpha\beta}}{2}\left(\theta_A^{xx} + \theta_A^{yy} + \theta_A^{zz}\right) \tag{22}$$

$$\theta_A^{\alpha\beta} = \sum_{l'\in A} \sum_{l} P_l \left(\alpha_A D_{ll'}^{\beta} + \beta_A D_{ll'}^{\alpha} - \alpha_A \beta_A S_{ll'} - Q_{ll'}^{\alpha\beta} \right)$$
(23)

$$q_A = Z_A - GAP_A \tag{24}$$

$$\mu_A^{\alpha} = \sum_{l' \in A} \sum_{l} P_{l'l} \left(\alpha_A S_{l'l} - D_{l'l}^{\alpha} \right)$$
 (25)

$$D_{ll'}^{\alpha} = \langle \phi_l | \alpha_i | \phi_{l'} \rangle \tag{26}$$

$$Q_{ll'}^{\alpha\beta} = \langle \phi_l | \alpha_i \beta_i | \phi_{l'} \rangle \tag{27}$$

 α and β are Cartesian components labled $(x,y,z)^T$ with atom A being centered in $v_A = (x_i,y_i,y_z)^T$ where i is a form of pointer/label dereferencing. $\delta_{\alpha\beta}$ is just the delta function, i.e is is 1 if α and β are the same label and 0 otherwise, this serves to include the term only for the diagonal.

$$\mathbf{\Theta}_{A} = \begin{pmatrix} \Theta_{A}^{xx} & \Theta_{A}^{xy} & \Theta_{A}^{xz} \\ \Theta_{A}^{yx} & \Theta_{A}^{yy} & \Theta_{A}^{yz} \\ \Theta_{A}^{zx} & \Theta_{A}^{zy} & \Theta_{A}^{zz} \end{pmatrix}$$
(28)

$$\boldsymbol{\mu}_{A} = \begin{pmatrix} \mu_{A}^{x} \\ \mu_{A}^{y} \\ \mu_{A}^{z} \end{pmatrix} \tag{29}$$

$$\mathbf{R}_{AB} = v_A - v_B \tag{30}$$

$$f_n(R_{AB}) = \frac{f_{damp}(a_n, R_{AB})}{R_{AB}^n} = \frac{1}{R_{AB}^n} \frac{1}{1 + 6\left(\frac{R_0^{AB}}{R_{AB}}\right)^{a_n}}$$
(31)

$$R_0^{AB} = 0.5(R_0^{A'} + R_0^{B'}) (32)$$

$$R_0^{A'} = \begin{cases} R_0^A + \frac{R_{max} - R_0^A}{1 + exp[-4(CN_A' - N_{val} - \Delta_{val})]} & \text{if } N_{val} \text{ is given} \\ 5.0 \text{ bohrs} & \text{otherwise} \end{cases}$$
(33)

$$R_{max} = 5.0 \text{ bohrs} \tag{34}$$

$$\Delta_{val} = 1.2 \tag{35}$$

 R_0^A is a fitted value for 12 elements and 5.0 for the rest. a_n are adjusted global parameters.

1.5 Anisotropic XC energy

$$E_{AXC} = \sum_{A} (f_{XC}^{\mu_A} |\boldsymbol{\mu}_A|^2 + f_{XC}^{\Theta_A} ||\boldsymbol{\Theta}_A||^2)$$
 (36)

Where $f_{XC}^{\mu_A}$ and $f_{XC}^{\Theta_A}$ are fitted values. What norms are these?

1.6 Dispersion Energy

$$E_{disp}^{D4'} = -\sum_{A>B} \sum_{n=6,8} s_n \frac{C_n^{AB}(q_A, CN_{cov}^A, q_B, CN_{cov}^B)}{R_{AB}^n} f_{damp,BJ}^{(n)}(R_{AB})$$

$$-s_9 \sum_{A>B>C} \frac{(3cos(\theta_{ABC})cos(\theta_{BCA})cos(\theta_{CAB}) + 1)C_9^{ABC}(CN_{cov}^A, CN_{cov}^B, CN_{cov}^C)}{(R_{AB}R_{AC}R_{BC})^3} (37)$$

$$\times f_{damp,zero}^{(9)}(R_{AB}, R_{AC}, R_{BC}).$$

The term in the second line is the three-body Axilrod– Teller–Muto (ATM) (What is this??????) term and the last line is the corresponding zero-damping function for this term.

 C_6^{AB} is the pairwise dipole-dipole dispersion coefficients calculated by numerical integration via the Casimir-Polder relation.

$$C_6^{AB} = \frac{3}{\pi} \sum_j w_j \overline{\alpha}_A(i\omega_j, q_A, CN_{cov}^A) \overline{\alpha}_B(i\omega_j, q_B, CN_{cov}^B)$$
 (38)

 w_j are the integration weights, which are derived from a trapeziodal partitioning between the grid points $j(j \in [1, 23])$.

The isotropically averaged, dynamic dipole-dipole polarizabilites $\overline{\alpha}$ at the jth imaginary frequency $i\omega_j$ are obtained from the self-consistent D4 model; i.e., they are depending on the covalent coordination number and are also charge dependent.

$$\overline{\alpha}_{A}(i\omega_{j}, q_{A}, CN_{cov}^{A}) = \sum_{r}^{N_{A,ref}} \xi_{A}^{r}(q_{A}, q_{A,r}) \overline{\alpha}_{A,r}(i\omega_{j}, q_{A,r}, CN_{cov}^{A,r}) W_{A}^{r}(CN_{cov}^{A}, CN_{cov}^{A,r})$$
(39)

The Gaussian weighting for each reference system is given by:

$$W_A^r(CN_{cov}^A, CN_{cov}^{A,r}) = \sum_{j=1}^{N_{gauss}} \frac{1}{N} \exp\left[-6j \cdot (CN_{cov}^A - CN_{cov}^{A,r})^2\right]$$
(40)

with

$$\sum_{N_{A,ref}} W_{A}^{r}(CN_{cov}^{A}, CN_{cov}^{A,r}) = 1$$
(41)

 \mathcal{N} is a normalization constant.

The number of Gaussian function per reference system N_{gauss} is mostly one, but equal to three for $CN_{cov}^{A,r} = 0$ and reference systems with similar coordination number.

The carge-dependency is included via the empirical scaling function ξ_A^r .

$$\xi_A^r(q_A, q_{A,r}) = \exp\left[3\left\{1 - \exp\left[4\eta_A\left(1 - \frac{Z_A^{eff} + q_{A,r}}{Z_A^{eff} + q_A}\right)\right]\right\}\right]$$
 (42)

where η_A is the chemical hardness taken from ref 98.

 Z_A^{eff} is the effective nuclear charge of atom A, which has been determined by subtracting the number of core electrons represented by the def2-ECPs in the time-dependent DFT reference calculations.

 C_8^{AB} is calculated recursively from the lowest order C_6^{AB} coefficients.

$$C_8^{AB} = 3C_6^{AB} \sqrt{\mathcal{Q}^A \mathcal{Q}^B} \tag{43}$$

$$Q^{A} = s_{42} \sqrt{Z^{A}} \frac{\langle r^{4} \rangle^{A}}{\langle r^{2} \rangle^{A}} \tag{44}$$

 $\sqrt{Z^A}$ is the ad hoc nuclear charge dependent factor. (How do we get Z??????) $\langle r4 \rangle$ and $\langle r2 \rangle$ are simple multipole-type expectation values derived from atomic densities which are averaged geometrically to get the pair coefficients. (What is 'r', how we get???????) (what is s_{42} ???)

 CN_{cov}^A is the covalent coordination number for atom A.

q is the atomic charge, so q_A is the atomic charge for atom A.

The damping and scaling parameters in the dispersion model are:

$$a1 = 0.52$$
 | $a2 = 5.0$ | $s6 = 1.0$ | $s8 = 2.7$ | $s9 = 5.0$

 C_9^{ABC} is the triple-dipole constant²:

$$C_9^{ABC} = \frac{3}{\pi} \int_0^\infty \alpha^A(i\omega) \alpha^B(i\omega) \alpha^C(i\omega) d\omega \tag{45}$$

The three-body contribution is typically < 5-10% of E_{disp} , so it is small enough that we can reasonably approximate the coefficients by a geometric mean as²:

$$C_9^{ABC} \approx -\sqrt{C_6^{AB}C_6^{AC}C_6^{BC}} \tag{46}$$

 θ_{ABC} is the angle between the two edges going from B to the other two atoms. θ_{BCA} is the angle between the edges going from C to the other two and so on.

 $^{^2} https://www.researchgate.net/publication/43347348_A_Consistent_and_Accurate_Ab_Initio_Parametrization_of_Density_FD_for_the_94_Elements_H-Pu$

BJ = Becke-Johnson

$$f_n^{damp,BJ}(R_{AB}) = \frac{R_{AB}^n}{R_{AB}^n + (a_1 \cdot R_{AB}^{crit} + a_2)^6}$$
(47)

$$R_{AB}^{crit} = \sqrt{\frac{C_8^{AB}}{C_6^{AB}}} \tag{48}$$

$$f_9^{damp,zero}(R_{AB}, R_{AC}, R_{BC}) = \left(1 + 6\left(\sqrt{\frac{R_{AB}^{crit}R_{BC}^{crit}R_{CA}^{crit}}{R_{AB}R_{BC}R_{CA}}}\right)^{16}\right)^{-1}$$
(49)