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RV COLLEGE OF ENGINEERING®

(An Autonomous Institution Affiliated to VTU)
III Semester B. E. Examinations April/May-2024

Common to All Branches MATERIAL SCIENCE FOR ENGINEERS

Time: 03 Hours
Instructions to candidates:

Maximum Marks: 100

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- 1. Answer all questions from Part A. Part A questions should be answered in first three pages of the answer book only.
- 2. Answer FIVE full questions from Part B. In Part B question number 2 is compulsory. Answer any one full question from 3 and 4, 5 and 6, 7 and 8, 9 and 10.

PART-A

	******	IAT	ы	CO
1 1.1	Ceramic materials are generally having type of bonds.	01	1	1
1.2	When the electrical behaviour of the material are dominated by		_	
- cal	the electronic structure of pure metal, the material can be called			
47	as semi-conductors.	01	1	1
1.3	Give an example of natural composite.	01	1	1
1.4	T_i is having crystal structure at room temperature.	01	1	1
1.5	For the application of extremely high temperatures, type	3		
	of basic materials are used.	01	1	1
1.6	Cu - Ni alloys are an example of type of defect.	01	1	1
1.7	Deformations by twins are commonly observed in type			
	of metals.	01	1	1
1.8	Calculate the co-ordination number of <i>BCC</i> crystal systems.	01	1	1
1.9	External surfaces are an example of type of defects in		6	
	crystal systems.	01	1	2
1.10	Viscoelastic deformation is observed significantly in			
	basic material.	01	1	2
1.11	Diamond pyramid indenter is used in hardness			
	testing.	01	1	2
1.12	Addition of chromium (Cr) to steel enhances property.	01	1	2
1.13	Which property of dielectric materials is measured in materials			
	testing?	01	1	2
1.14	Photodiodes serves as a in optical communication			
	systems.	01	1	2
1.15	Name the hardening process which introduces nitrogen into the			
	surface layer of steel to produce nitrides at the surface layer of			
	the component.	01	1	3
1.16	is a critical process in electronics manufacturing			1
	primarily used in surface-mount technology.	01	1	3
1.17	Aggregated composites possess strength compared to			
	fiber reinforced composites.	01	1	3
1.18	Sol-gel process is an example of approach of synthesis			
1.10	of nanomaterials.	01	1	4
1.19	Crystalline structures of materials are measured on the basis of			
1.17	lattice spacing and electron interaction with it, in			
	characterization technique.	01	1	4
1.00	Name characterization technique for nanostructure in which			
1.20	atomic structures of the materials are observed using electrons			
		01	1	4
	passing through ultra-thin specimens.	01	1	4

PART-B

2	a	Describe the following bonds for solid materials with examples.			
		i) Ionic bonds ii) Secondary bonds and		6	
		iii) Metallic bonds.	08		١.
	b	Explain the following crystallographic features of a	08	2	1
	D	Solid material.			
		i) FCC crystal structure with example			
		ii) Energy bands in insulator			
		iii) Edge dislocation in a metal.	08	2	1
3	a	Define and explain the following thermal properties of a material.			
		i) Thermo-electric effect			
		ii) Linear thermal expansion coefficient			
		iii) Thermal shock.	08	2	2
	b	Explain the principle of working of following materials.			-
		i) Piezoelectric materials and	00	_	
		ii) Thermocouples.	08	3	2
4	а	OR With a strong stroin sums and in the following properties of a			
т	а	With a stress-strain curve, explain the following properties of a material.	5		
		i) Young's Modulus			
		ii) Ultimate Tensile Strength and			
		iii) Toughness.	08	3	2
	b	Define the following mechanical properties of a solid material.			
		i) Fracture toughness			
		ii) Hardness			>
		iii) Fatigue limit.	08	2	2
		Evaluing the appropriate and applications of stainless starl	06	0	0
5	a L	Explain the properties and applications of stainless steel. Mention the dopant used for p-type and n-type semiconductors.	06 04	2	2 2
	b	Explain the properties and applications of ceramics.	06	2	$\begin{array}{ c c c c c c c c c c c c c c c c c c c$
	С	OR	00	2	
6	a	Explain the properties and applications of fiber reinforced		,	
J	u	composites.	08	2	2
	b	Classify forming processes of structural materials and explain			
		any two forming processes.	08	2	2
7	а	Explain the steps involved in thermal oxidation method of			П
		semiconductor devices.	08	3	3
	b	Describe the cause, effect and prevention methods of any two			
		defects of heat treatment process of metals.	08	2	3
_		OR			
8	а	Differentiate between annealing and normalizing process of	00	2	2
	L	ferrous materials.	08 08	3 2	3 3
	b	Describe hardening of steel with the help of TTT diagram.	Uð		3
9	2	Define nanomaterials and describe the applications of			
J	a	nanomaterials with example.	08	2	4
	b	Describe the effect of particle size and surface area of		-	
		nanomaterials on the mechanical and optical properties of it.	08	2	4
		OR			
10	а	Describe the significant properties and applications of carbon			
		nanotubes.	08	2	4
	b	Describe the principle of working of X-Ray Diffraction (XRD)	00	_	
		techniques for the characterization of solid materials.	08	3	4