#### **Amines**

# 1.In order to prepare a 1° amine from an alkyl halide with simultaneous addition of one CH2 group in

the carbon chain, the reagent used as source of nitrogen is:

- (a) Sodium amide, NaNH2
- (b) Sodium azide, NaN3
- (c) Potassium cyanide, KCN
- (d) Potassium phthalimide, C6H4(CO)2N-K+

Answer:(c) Potassium cyanide, KCN

- 2. Which of the following is formed when an alkyl primary amine reacts with nitrous acid?
- (a) Alkyl nitrite
- (b) Secondary amine
- (c) Nitroalkane
- (d) Alcohol

Answer:(d) Alcohol

3. Which of the following reagents would not be a good choice for reducing an aryl nitro compound to

an amine?

- (a) H2 (excess)/Pt
- (b) LiAlH4 in ether
- (c) Fe and HCl
- (d) Sn and HCl

Answer:(b) LiAlH4 in ether

4.The conversion of benzene diazonium chloride to bromobenzene can be accomplished by

- (a) Reimer-Tiemann reaction
- (b) Friedel-Crafts reaction
- (c) Gattermann reaction
- (d) Azo-Coupling reaction

#### Answer:(c) Gattermann reaction

#### 5. Which of the following amines are insoluble in water?

- (a) Methanamine
- (b) Ethanamine
- (c) Propanamine
- (d) Benzenamine

#### Answer:(d) Benzenamine

# 6.The best reagent for converting 2-phenylpropanamide into 2-phenylpropanamine is:

- (a) excess H2
- (b) Br2 in aqueous NaOH
- (c) iodine in the presence of red phosphorus
- (d) LiAlH4 in ether

# Answer:(d) LiAlH4 in ether

#### 7. Amongst the given set of reactants, the most appropriate for preparing $2^{\circ}$ amine is:

- (a)  $2^{\circ}$  R—Br + NH3
- (b) 2° R—Br + NaCN followed by H2/Pt
- (c) 1° R—NH2 + RCHO followed by H2/Pt
- (d) 1° R—Br (2 mol) + potassium phthalimide followed by H3O+/heat

# Answer:(c) 1° R—NH2 + RCHO followed by H2/Pt

# 8. Which of the following statements about primary amines is 'false'?

- (a) Alkylamines are stronger bases than arylamines.
- (b) Alkylamines are stronger bases than ammonia.

- (c) Alkylamines react with nitrous acid to produce alcohols.
- (d) Arylamines react with nitrous acid to produce phenols

Answer:(d) Arylamines react with nitrous acid to produce phenols

- 9. Reduction of CH3CH2NC with hydrogen in presence of Ni or Pt as catalyst gives
- (a) CH3CH2NH2
- (b) CH3CH2NHCH3
- (c) CH3CH2NHCH2CH3
- (d) (CH3)3N

Answer:(b) CH3CH2NHCH3

- 10. The correct IUPAC name for CH2=CHCH2NHCH3 is
- (a) Allylmethylamine
- (b) 2-amino-4-pentene
- (c) 4-aminopent-1-ene
- (d) N-methylprop-2-en-1-amine

Answer:(d) N-methylprop-2-en-1-amine

- 11. The best reagent for converting, 2-phenylpropanamide into 1-phenylethanamine is:
- (a) excess H2/Pt
- (b) NaOH/Br2
- (c) NaBH4/methanol
- (d) LiAlH4/ether

Answer:(b) NaOH/Br2

- 12. Amongst the following, the strongest base in aqueous medium is:
- (a) CH3NH2
- (b) NCCH2NH2
- (c) (CH3)2NH
- (d) C6H5NHCH3

### Answer:(c) (CH3)2NH

# 13. Hoffmann bromamide degradation is used for the preparation of

- (a) primary amines
- (b) secondary amines
- (c) tertiary amines
- (d) secondary aromatic amines

# Answer:(a) primary amines

#### 14. Which of the following does not react with Hinsberg reagent?

- (a) C2H5NH2
- (b) (CH3)2NH
- (c) (CH3)3N
- (d) CH3CH(NH2)CH3

#### Answer:(c) (CH3)3N

# 15. The source of nitrogen in Gabriel synthesis of amines is:

- (a) Sodium azide, NaN3
- (b) Sodium nitrite, NaNO2
- (c) Potassium cyanide, KCN
- (d) Potassium phthalimide, C6H4(CO)2N-K+

Answer:(d) Potassium phthalimide, C6H4(C0)2N-K+