25/0/15

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University Roll No.....

B. Tech. I Year, II Semester, I Term Examination, 2014-15

AHC-101: Engineering Chemistry

Time: 01 1/2 Hours

Total Marks: 20

Section-A

Note: Attempt All Questions.

[1X05=05 marks]

- I. For a certain reaction, A ——— P, a plot of ln [A] versus t gives a straight line with a slope of 1.24 s⁻¹. Find the order of the reaction?
- II. Explain, why p- nitrophenol is more soluble than o-nitrophenol in water.
- III. Allyl halides react faster in nucleophilic substitution reactions as compared to primary alkyl halides. Why?
- IV. Write molecular orbital electronic configuration of HF molecule.
- V. What are functional materials? Give any two examples of organic functional materials.

Section-B

Note: Attempt Any Three Questions.

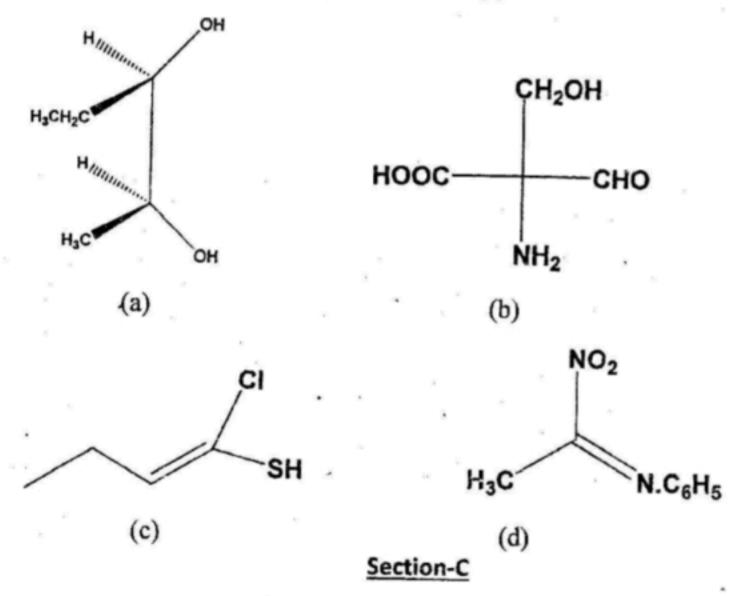
[2X03=06 marks]

- I. For the second order reaction of base catalyzed hydrolysis of ester at 35° C, $k = 6.81 \times 10^{-3}$ L mol⁻¹sec⁻¹calculate the time required for the hydrolysis of 70% ester, if the initial concentration of reactants in the reaction mixture are 0.05 M ester and 0.1 M NaOH.
- II. Differentiate piezoelectric and reverse piezoelectric effects with examples.

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III. Arrange following carbanions in order of increasing stability by giving proper explanation.

IV. Allocate (R)-/(S)- or (E)-/(Z)- configuration to the following molecules- (by giving proper numbering.)



Note: Attempt Any Three Questions.

[3X 03=09 marks]

- I.(a) Derive an integrated rate equation for a Second order reaction, when both reactants are same.
- (b). Calculate activation energy of the reaction, whose rate constant is quadrupled by 10° rise in temperature in the vicinity of 28° C.

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- II. Draw the molecular orbital diagram, calculate bond order and assign magnetic behaviour to F₂ molecule.
- III. Discuss the stability of conformers of *n*-butane by giving suaitable energy level diagram.
- IV. What are bio-materials? Write their important characteristics with some applications.