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University Roll No.....

## Mid-Term Theory Examination, 2019-20 Engineering Chemistry (BCHS 0101)

Course: B.Tech. Time: 2 Hr Branch: CSE

Year/Semester: I/I Max. Marks: 30

## Section- A

Note: Attempt all questions.

 $2 \times 3 = 6$  marks

- 1. Differentiate HCV and NCV.
- 2. With the help of reaction, explain how vulcanization of rubber is done? What are the advantages of vulcanized rubber over raw rubber?
- 3. Write the significance of the following in lubrication process:
  - i) Flash point and fire point
  - ii) Cloud point and pour point

## Section- B

Note: Attempt all questions

 $3 \times 3 = 9$  marks

- 1. What do you understand by conformational isomers? Draw all possible conformers of n-butane. State which one is most stable and which one is least stable?
- 2. What are ceramics? How they can be classified?
- 3. Calculate the weight of air required for complete combustion of 1 m<sup>3</sup> of a gaseous fuel having following composition:

 $CH_4 = 45\%$ ,  $C_2H_6 = 15\%$ , CO = 16%,  $N_2 = 10\%$ ,  $H_2 = 10\%$ ,  $O_2 = 10\%$ .

## Section- C

Note: Attempt any three questions

 $5 \times 3 = 15 \text{ marks}$ 

1. Using the concept of Molecular orbital theory, draw molecular orbital diagram of Oxygen molecule. Also, write its molecular orbital electronic configuration, bond order and magnetic character.

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- 2. i) Explain with help of well labeled diagram, how petrol can be synthesized using "Fisher Tropsch Method". Also, write reactions involved in this method.
  - ii) Assign R/S configuration in the following compounds:

- 3. i) Differentiate Thermosetting and thermoplastic polymers. (2)
- ii) Design the method of synthesis and write the applications of following polymers (any two):
  - a. Buna-S
  - b. Polystyrene
  - c. Polylactic acid
- 4. i) Write the composition and uses of any two type of glasses: (3
  - a. Borosilicate glass
  - b. Lead glass
  - c. Soda glass
- ii) Draw a well labeled diagram of bomb calorimeter for determination of HCV. Also, write the formula involved in it. (2)

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