# Modeling & Simulating Chemically Reacting Systems

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August 2023

#### Goals for this session

- 1. Build your own ODE integrator module in Python
  - ▶ **Learn** how (generic) state-of-the-art Runge-Kutta time steppers work
    - adaptive time-stepping (performance)
    - implicit time-stepping (stability)
  - ...and implement some of it
- 2. Build your own toy reactor model
  - ▶ **Learn** a template for how to model reacting systems
  - implement a simple reactor model
- 3. Build a simplified ionospheric reaction model (see Schunk [1988])

# From finite differences to solving initial value problems: Explicit Euler Initial Value Problem (IVP)

We wish to compute the solution x(t) of the ordinary differential equation (ODE)

$$\frac{dx}{dt}(t) = f(x(t), t), \quad x(t_0) = x_0$$

for some time horizon  $[t_0, t_f]$ .

The simplest thing that could possibly work: Replace the time-derivative with a finite difference approximation

$$\frac{dx}{dt}(t) = f(x(t), t) \approx \frac{x(t+h) - x(t)}{h}$$

$$\implies x(t+h) \approx x(t) + hf(x(t),t)$$

# Exercise 0: Solving a simple IVP with explicit Euler

#### Question 1

Solve the IVP

$$\frac{dx}{dt}(t) = f(x(t)) = -2x(t), \quad x(0) = 3$$

analytically.

#### Question 2

Solve the IVP with explicit Euler for  $t \in [0,2]$ . To that end, iterate the explicit Euler time stepping rule x(t+h) = x(t) + hf(x(t)) with h = 0.01, i.e.,

$$x(h) = x(0) + hf(x(0))$$

$$x(2h) = x(h) + hf(x(h))$$

$$\vdots$$

$$x(Nh) = x((N-1)h) + hf(x((N-1)h))$$

Plot the trajectory obtained with explicit Euler's method against the analytical solution. Try different step sizes h. What do you notice?

# A slightly different perspective for generalization

Consider again the IVP

$$\frac{dx}{dt} = f(x, t), \quad x(t_0) = x_0$$

for some time horizon  $[t_0, t_0 + h]$ .

▶ This problem has an "obvious" solution by the fundamental theorem of calculus

$$x(t_0 + h) = x(t_0) + \int_{t_0}^{t_0+h} f(x(\tau), \tau) d\tau$$

► Explicit Euler is obtained by approximating the integral above by a simple rectangular integration rule

$$\int_{t_0}^{t_0+h} f(x(\tau),\tau) \, \tau \approx h f(x(t_0),t_0)$$

## Time-stepping: Runge-Kutta methods

▶ Time-stepping schemes:

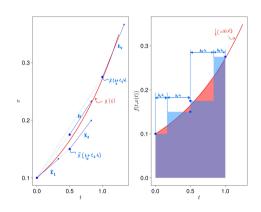
$$x(t+h) = x(t) + \underbrace{\int_{t}^{t+h} f(x(\tau), \tau) d\tau}_{\text{approximate}}$$

► Runge-Kutta method

$$\int_{t}^{t+h} f(x(\tau), \tau) d\tau \approx h \sum_{i=1}^{s} b_{i}k_{i}$$
where  $k_{i} = f(x(t) + h \sum_{j=1}^{s} a_{i,j}k_{j}, t + c_{j}h)$ 

$$\underset{\approx x(t+c_{i}h)}{\underbrace{\sum_{j=1}^{s} a_{i,j}k_{j}, t + c_{j}h}}$$

• Usually  $\epsilon_{LTE} = O(h^{p+1})$  with  $p \sim s$ 



tune accuracy/cost trade-off by choosing the right Runge-Kutta method

# Time-stepping: Butcher tableaus Explicit Runge-Kutta method

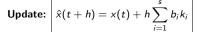
#### Successively evaluate:

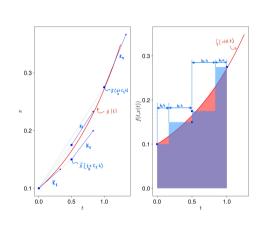
$$k_{1} = f(x(t), t)$$

$$k_{2} = f(x(t) + ha_{2,1}k_{1}, t + c_{2}h)$$

$$\vdots$$

$$k_{s} = f(x(t) + h\sum_{i=1}^{s-1} a_{s,i}k_{i}, t + c_{s}h)$$





# Time-stepping: Butcher tableaus

## Explicit Runge-Kutta method

#### Successively evaluate

$$k_{1} = f(x(t), t)$$

$$k_{2} = f(x(t) + ha_{2,1}k_{1}, t + c_{2}h)$$

$$\vdots$$

$$k_{s} = f(x(t) + h\sum_{i=1}^{s-1} a_{s,i}k_{i}, t + c_{s}h)$$

## Example: Explicit Euler

$$\begin{array}{c|c} c_1=0 & \\ \hline & b_1=1 \end{array}$$

#### Evaluate:

$$k_1=f(x(t),t)$$

#### **Update:**

$$\hat{x}(t+h) = x(t) + h \sum_{i=1}^{s} b_i k_i$$
$$= x(t) + h f(x(t), t)$$

**Update:**  $\hat{x}(t+h) = x(t) + h \sum_{i=1}^{s} b_i k_i$ 

# Time-stepping: Butcher tableaus Explicit Runge-Kutta method

# Successively evaluate:

$$k_{1} = f(x(t), t)$$

$$k_{2} = f(x(t) + ha_{2,1}k_{1}, t + c_{2}h)$$

$$\vdots$$

$$k_{s} = f(x(t) + h\sum_{i=1}^{s-1} a_{s,i}k_{i}, t + c_{s}h)$$

# Update: $\hat{x}(t+h) = x(t) + h \sum_{i=1}^{s} b_i k_i$

## General/Implicit Runge-Kutta method

#### Solve nonlinear equation system:

$$k_1 = f(x(t) + h \sum_{j=1}^{s} a_{1,j}k_j, t + c_1h, )$$
 $k_2 = f(x(t) + h \sum_{j=1}^{s} a_{2,j}k_j, t + c_2h)$ 
 $\vdots$ 

$$k_s = f(x(t) + h \sum_{j=1}^{s} a_{s,j} k_j, t + c_s h)$$

## Exercise 1: Implementing a generic explicit Runge-Kutta stepper

#### Question 1

Complete the function explicit\_RK\_stepper(f,x,t,h,a,b,c) in runge\_kutta.py. It takes inputs

- f right-hand-side f(x, t) of the ODE to be integrated.
- $\triangleright$  x current state x(t)
- ▶ t current time t
- ▶ h time step
- a,b,c coefficients of the Runge-Kutta-Method.
   b and c are lists and a is organized as follows: a
   = [[a\_21], [a\_31, a\_32], ..., [a\_s1, a\_s2, ..., a\_s(s-1)]]

and returns the RK prediction  $\hat{x}(t+h)$ .

## Explicit RK update

$$\hat{x}(t+h) = x(t) + h \sum_{i=1}^{s} b_i k_i \text{ where}$$

$$k_1 = f(x(t), t)$$

$$k_2 = f(x(t) + h a_{2,1} k_1, t + c_2 h)$$

$$\vdots$$

$$k_s = f(x(t) + h \sum_{i=1}^{s-1} a_{s,i} k_i, t + c_s h)$$

# Exercise 1: Implementing a generic explicit Runge-Kutta stepper

#### Question 2

Test your integrator by applying it to the test problem

$$\frac{dx}{dt} = -2x(t), \quad x(0) = 3, \ t \in [0, 2]$$

To that end, you only need to complete the function dormand\_prince\_stepper(f,x,t,h) in main.py. Please use explicit\_RK\_stepper(f,x,t,h,a,b,c) as implemented in the previous problem to do so.

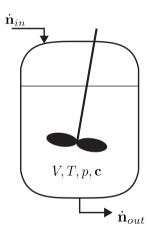
The coefficients for the Dormand-Prince method Dormand and Prince [1980] are provided in the template.

# Continuously Stirred Tank Reactors (CSTR)

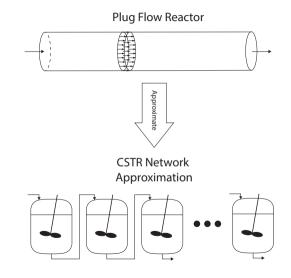
No spatial variations of intrinsic state variables:(T, p, c)

For externally imposed V, T, p the reactor is thermodynamically fully described by mole balances for all species:

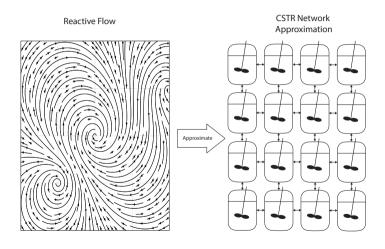
$$\frac{dn_i}{dt} = \dot{n}_{i,in} - \dot{n}_{i,out} + V \sum_{k=1}^{N_R} \underbrace{r_{i,j}(\mathbf{c}, p, T)}_{\text{reactive flux}}$$



# CSTRs – The key building blocks for modeling reacting systems



# CSTRs – The key building blocks for modeling reacting systems

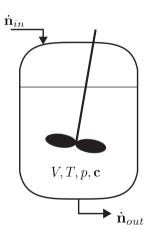


## Common CSTR simplifications

▶ isochoric (V = const.) operation (usually implying  $\dot{V}_{in} = \dot{V}_{out}$ )

▶ Letting  $D := \frac{\dot{V}}{V}$  and using that  $n_i = Vc_i$  yields

$$\frac{dc_i}{dt} = D\left(c_{i,in} - c_i\right) + \sum_{k=1}^{N_R} r_{i,j}(\mathbf{c}, p, T)$$



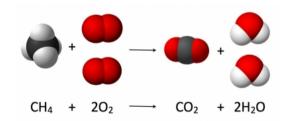
# Modeling Reactions - Stoichiometry

## Stoichiometry of Chemical Reactions

$$\nu_1^- A_1 + \nu_2^- A_2 + \dots + \nu_N^- A_N \to \nu_1^+ A_1 + \nu_2^+ A_2 + \dots + \nu_N^+ A_N$$

## Example: Methane Combustion

$$\begin{array}{ll} {\rm CH_4:} & \nu_{\rm CH_4}^- = 1, \nu_{\rm CH_4}^+ = 0 \\ {\rm O_2:} & \nu_{\rm O_2}^- = 2, \nu_{\rm O_2}^+ = 0 \\ {\rm CO_2:} & \nu_{\rm CO_2}^- = 0, \nu_{\rm CO_2}^+ = 1 \\ {\rm H_2O:} & \nu_{\rm H_2O}^- = 0, \nu_{\rm H_2O}^+ = 2 \end{array}$$



## Modeling Reactions - Stoichiometry

▶ Reactive flux of species *i* due to reaction *j*:

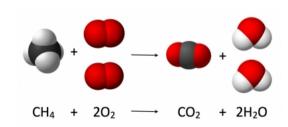
$$\underbrace{r_{i,j}(\mathbf{c},T,p)}_{\text{reactive flux}} = \underbrace{\nu_{ij}^+ r_j(\mathbf{c},T,p)}_{\text{sources}} - \underbrace{\nu_{ij}^- r_j(\mathbf{c},T,p)}_{\text{losses}} = (\nu_{ij}^+ - \nu_{ij}^-) \underbrace{r_j(\mathbf{c},T,p)}_{\text{reaction rate}}$$

Summarize stoichiometry information in stoichiometry matrix

$$\mathbf{S} \in \mathbb{Z}^{\# ext{species} imes \# ext{reactions}}$$
 such that  $\mathbf{S}_{i,j} = 
u_{i,j}^+ - 
u_{i,j}^-$ 

## **Example: Methane Combustion**

$$\label{eq:Signal} \textbf{S} = \begin{matrix} \mathrm{CH_4} \\ \mathrm{CO_2} \\ \mathrm{CO_2} \\ \mathrm{H_2O} \end{matrix} \qquad \begin{bmatrix} -1 \\ -2 \\ 1 \\ 2 \end{bmatrix}$$



## Implications for CSTR model

▶ In vector-valued notation, the mole balances describing a CSTR take a compact and computationally form:

$$\frac{d\mathbf{n}}{dt} = \mathbf{\dot{n}}_{in} - \mathbf{\dot{n}}_{out} + V \qquad \mathbf{Sr}(\mathbf{c}, T, p)$$

$$\begin{bmatrix}
\frac{d\mathbf{n}_{1}}{dt} \\
\frac{d\mathbf{n}_{2}}{dt} \\
\vdots \\
\frac{d\mathbf{n}_{N}}{dt}
\end{bmatrix} = \begin{bmatrix}
\dot{\mathbf{n}}_{1,in} \\
\dot{\mathbf{n}}_{2,in} \\
\vdots \\
\dot{\mathbf{n}}_{N,in}
\end{bmatrix} = \begin{bmatrix}
\dot{\mathbf{n}}_{1,out} \\
\dot{\mathbf{n}}_{2,out} \\
\vdots \\
\dot{\mathbf{n}}_{N,out}
\end{bmatrix} = \begin{bmatrix}
\sum_{j=1}^{N_{R}} r_{1,j}(\mathbf{c}, T, p) \\
\sum_{j=1}^{N_{R}} r_{2,j}(\mathbf{c}, T, p) \\
\vdots \\
\sum_{j=1}^{N_{R}} r_{N,j}(\mathbf{c}, T, p)
\end{bmatrix} = \mathbf{s} \begin{bmatrix}
r_{1}(\mathbf{c}, T, p) \\
r_{2}(\mathbf{c}, T, p) \\
\vdots \\
r_{N_{R}}(\mathbf{c}, T, p)
\end{bmatrix}$$

...or for isochoric operation:

$$\frac{d\mathbf{c}}{dt} = D(\mathbf{c}_{in} - \mathbf{c}) + \mathbf{Sr}(\mathbf{c}, T, p)$$

# Modeling Reactions - Reaction Rates

- ▶ The theory of chemical kinetics is deep, rich and beyond the scope of this session.
- ▶ Most commonly mass-action kinetics are used (if appropriate or not ...):

$$r_j(\mathbf{c}, T, p) = k(T, p) \prod_{i=1}^N c_i^{\nu_{i,j}^-}$$

► The temperature variation of the rate coefficient k(T,p) is frequently modeled via the *extended Arrhenius* equation:

$$k(T,p) = A(p)T^{n(p)}e^{-\frac{E_a(p)}{k_BT}}$$

Pressure dependence is hard, so people frequently assume  $A(p), n(p), E_a(p) \approx \text{const.}$  in the pressure range of interest.

# Exercise 2: Modeling a toy reactor

## Toy Reaction Network

$$A \xrightarrow{k_1} B \xrightarrow{k_2} 2 C$$

#### **Assumptions**

- ► The reactor is closed:  $D = \dot{n}_{i,in} = \dot{n}_{i,out} = 0$ .
- ► The fluid volume, temperature and pressure are constant
- ► The reactions follow mass action kinetics with constant rate coefficients:

$$k_1 = 100 \,\mathrm{s}^{-1}, k_2 = 0.25 \,\mathrm{s}^{-1}, k_3 = 1 \,\mathrm{cm}^3/\mathrm{mol/s}$$

#### Question 1

How does the general CSTR model look like for this setting?

## Question 2

State the stoichiometry matrix associated with this network. Implement a numpy array carrying this matrix.

### Question 3

Write a function reaction\_rates(c,k) that takes in a vector k containing the rate coefficients, and a vector c of the concentrations of species  $A,\,B,\,\&\,C$  and returns the reaction rates of the respective reactions as numpy array.

#### Question 4

Write a function reactor(c,k,S) that computes the right-hand-side of the CSTR model.

#### Question 5

Simulate the reactor for different step sizes h using the previously implemented interface.

## Exercise 2: A fun bonus chemical reaction network!

## Toy Reaction Network

$$X \xrightarrow{k_1} 2X$$

$$X + Y \xrightarrow{k_2} 2Y$$

$$Y \xrightarrow{k_2} P$$

### Assumptions

- ▶ The reactor is closed:  $D = \dot{n}_{i,in} = \dot{n}_{i,out} = 0$ .
- The fluid volume, temperature and pressure are constant.
- The reactions follow mass action kinetics with constant rate coefficients:

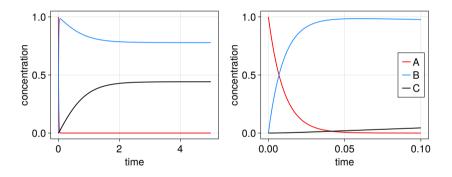
$$k_1 = 1 \,\mathrm{s}^{-1}, \, k_2 = 2 \,\mathrm{cm}^3/\mathrm{mol/s}, \, k_3 = 1 \,\mathrm{s}^{-1}$$

#### Question 1

Simulate this reaction system from an initial condition  $\mathbf{c}(0) = [1.0, 0.25, 0.0]$  over a time horizon  $t \in [0, 50.0]$  and plot the concentration traces of species X and Y.

# Stiffness – the curse of modeling reactions

## Dynamics at different time scales



Small steps are only needed to accurately simulate fast dynamics. Can we adjust our step sizes according to the dynamics on the fly?

## Order of Accuracy

## Definition (Local Truncation Error)

Given an explicit ODE

$$\frac{dx}{dt}=f(x,t)$$

The local truncation error is defined as the error between the exact solution at x(t+h) and the numerical approximation  $\hat{x}(t+h)$  after one time step:

$$\epsilon_{LTE} = \|x(t+h) - \hat{x}(t+h)\|$$

## Definition (Consistency & Order)

An integration method is called consistent if  $\epsilon_{LTE} = o(h)$ . It is called order p if  $\epsilon = O(h^{p+1})$ .

## Order of Accuracy

## Example: Explicit Euler

Given an explicit ODE:

$$\frac{dx}{dt} = f(x, t).$$

The explicit Euler integration update is defined

$$\hat{x}(t+h) = x(t) + hf(x(t),t)$$

The exact solution can be Taylor expanded around t (under weak smoothness assumptions)

$$x(t+h) = x(t) + \underbrace{\frac{dx}{dt}\Big|_{t}}_{=f(x(t),t)} h + \frac{1}{2} \frac{d^{2}x}{dt^{2}}\Big|_{t} h^{2} + O(h^{3})$$

Thus, (again under very weak assumptions)

$$\epsilon_{LTE} = \|x(t+h) - \hat{x}(t+h)\| = \left\|\frac{1}{2} \frac{d^2x}{dt^2}\right\|_1 h^2 + O(h^3) = O(h^2)$$

## Adaptive Runge-Kutta methods

Idea: Interweave two Runge-Kutta methods with different orders of accuracy to estimate  $\epsilon_{LTE}$  and choose step size to keep error below threshold.

Recall that

$$\epsilon_{LTE} = \|\hat{x}(t+h) - x(t+h)\|$$

Thus, given  $\hat{x}(t+h)$  and  $\hat{x}'(t+h)$ , we can conclude that

$$\epsilon_{LTE}' = \|\hat{\mathbf{x}}'(t \pm \hat{\mathbf{y}})\| + \mathcal{O}(\epsilon_{LTE})$$

If both methods are order s' < s, respectively, we have that  $\epsilon_{LTE} = O(\epsilon'_{LTE})$ . Thus, the local truncation error of both methods can be estimated as

$$\epsilon_{LTE}, \epsilon'_{LTE} = O(\|\hat{x}(t+h) - \hat{x}'(t+h)\|)$$

## Adaptive Runge-Kutta methods

Solution: Use two Runge-Kutta methods of order p and p-1 that share the same intermediate steps for minimal computation.

## Adaptive Runge-Kutta method

#### Procedure: Simple step size control

- 1: compute RK update x(t+h) and error estimate  $\epsilon$
- 2: while  $\epsilon > \text{rtol} \cdot ||x(t+h)|| + \text{atol do}$
- 3: half step size:  $h \leftarrow h/2$
- 4: recompute x(t+h) and  $\epsilon$
- 5: end while
- 6: double step size:  $h \leftarrow 2h$
- 7: **return**  $\times$ (t+h), h

Error estimate: 
$$\epsilon_{LTE} \approx \|\hat{x}(t+h) - \hat{x}'(t+h)\| = \left\|h\sum_{i=1}^{s}(b_i - b_i')k_i\right\|$$

## Exercise 3: Implementing an adaptive, explicit Runge-Kutta method

#### Question 1

Complete the function adaptive\_explicit\_RK\_stepper(f,x,t,h,a,b,c,b\_control) in . The inputs shall be understood as

- f right-hand-side f(x, t) of the ODE to be integrated.
- x current state of the ODE
- ▶ t current time
- ▶ h time step
- a,b,c,b\_control coefficients of the Runge-Kutta-Method. b, b\_control and c are lists and a is organized as follows: a = [[a\_21], [a\_31, a\_32], ..., [a\_s1, a\_s2, ..., a\_s(s-1)]]

The function returns the RK update  $\hat{x}(t+h)$  and an estimate of  $\epsilon_{LTE}$ .

## Exercise 3: Implementing an adaptive, explicit Runge-Kutta method

#### Question 2

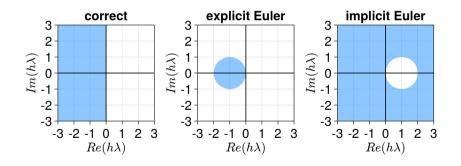
Complete the function adaptive\_integrate in runge\_kutta.py. It shall implement the integration via an adaptive integrator such that the one implemented in adaptive\_explicit\_RK\_stepper. Follow the pseudo-code below.

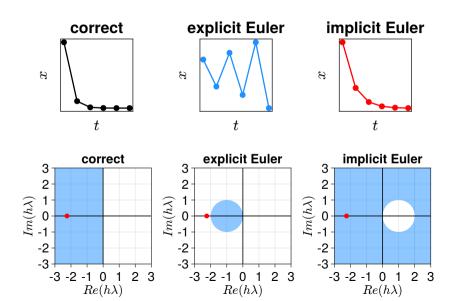
#### Pseudo-code for adaptive integration:

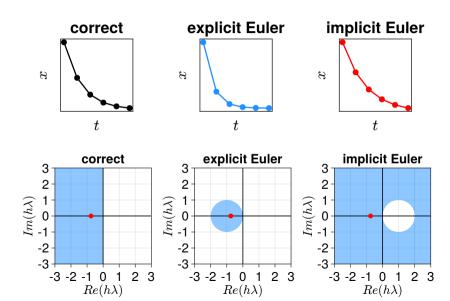
- 1: Given x(t), compute RK update  $\hat{x}(t+h)$  and error estimate  $\epsilon_{LTE}$
- 2: **if**  $\epsilon_{LTE} > \text{rtol} \cdot ||x(t+h)|| + \text{atol then}$
- 3: Reject the step and half step size:  $h \leftarrow h/2$
- 4: Go to Line 1
- 5: else
- 6: Accept the step:  $t \leftarrow t + h, x(t) \leftarrow \hat{x}(t+h)$
- 7: Double step size:  $h \leftarrow 2h$
- 8: Go to Line 1
- 9: end if
- 10: Repeat until end of time horizon is reached.

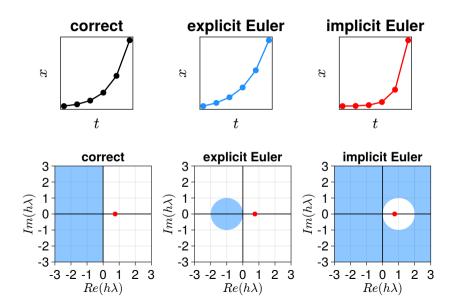
## Dahlquist Stability Analysis

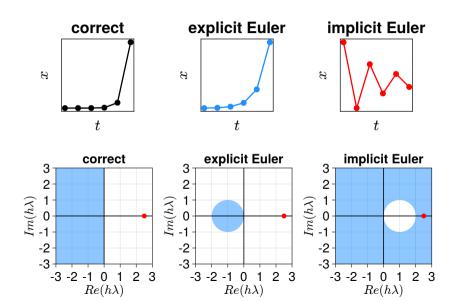
Apply numerical integrator with stepsize h to Dahlquist's test equation  $\frac{dx}{dt} = \lambda x$ 











# Implicit time-stepping

Superior stability properties

but need to solve an implicit equation system (find x such that g(x) = 0) Consider for example implicit Euler:

find 
$$\hat{x}(t+h)$$
 such that  $0 = x(t) + hf(t+h, \hat{x}(t+h)) - \hat{x}(t+h)$ 

Usually some numerical equation solving technique is required (Newton, fixed-point iteration, . . . )

## How to solve implicit equation systems in python

- ▶ We will use scipy.optimize.fsolve to do this
- ▶ To that end, our equation solving problem is to be formulated as

```
find x such that 0 = f(x)
```

 $\blacktriangleright$  We then supply a python function implementing f(x) & an initial guess for the solution  $x_0$ .

```
from scipy.optimize import fsolve
import numpy as np
# right-hand-side
def f(x):
    return np.exp(x) - 4*x
# initial guess
x0 = 2.0
# solve statment assings solution to x
x = fsolve(f, x0)
```

Analogous syntax for vector-valued functions

# Exercise 4: Comparing integrators

#### Question 1

Complete the code template integrators.py and compare the stability and accuracy properties of the integrators below. To that end, plot

- the traces of  $c_A(t)$ ,  $c_B(t)$ ,  $c_C(t)$  as predicted by each method
- the time needed for integration
- the accuracy as measured by  $|c_A(0.01) c_{A.0}/e|$

for step sizes  $h \in \{1 \times 10^{-4}, 1 \times 10^{-3}, 1 \times 10^{-2}, 1 \times 10^{-1}, 1, 1 \times 10^{1}\}$ 

Method	Time-stepping rule
Explicit Euler	$x_{k+1} = x_k + h f(x_k, t_k)$
Implicit Euler <sup>1</sup>	$x_{k+1} = x_k + hf(x_{k+1}, t_k + h)$
Heun's Method	$y = x_k + hf(x_k, t_k)  x_{k+1} = x_k + \frac{h}{2} (f(x_k, t_k) + f(y, t_k + h))$
Crank-Nicolson Method <sup>1</sup>	$x_{k+1} = x_k + \frac{h}{2} (f(x_k, t_k) + f(x_{k+1}, t_k + h))$

<sup>&</sup>lt;sup>1</sup>Hint: use scipy.optimize.fsolve to solve the equation system for the implicit methods.

### Disclaimer

- Runge-Kutta methods are not the end of the story
  - ▶ linear multistep methods (Adams-Bashforth, Adams-Moulton, **BDF**)
  - symplectic/geometric integrators
  - low rank integrators
  - **•** ...

- If you can, do **NOT** implement these yourself. Others have done the work for you (and they probably spent a whole lot more time on it than you can afford to)
  - DifferentialEquations.jl in Julia
  - scipy.integrate in Python
  - **.**..

# Exercise 5: Simplified ionosphere chemistry (derived from [Schunk, 1988])

## Chemistry

$$O^{+} + N_{2} \xrightarrow{k_{1}} NO^{+} + N$$

$$O^{+} + O_{2} \xrightarrow{k_{2}} O + O_{2}^{+}$$

$$O_{2}^{+} + e \xrightarrow{k_{3}} 2O$$

$$N_{2}^{+} + O \xrightarrow{k_{4}} O^{+} + N_{2}$$

$$N_{2}^{+} + O_{2} \xrightarrow{k_{5}} O_{2}^{+} + N_{2}$$

$$O_{2}^{+} + N \xrightarrow{k_{6}} NO^{+} + O$$

$$NO^{+} + e \xrightarrow{k_{7}} N + O$$

#### Ionization

$$O \xrightarrow{kg} O^{+} + e$$

$$O_{2} \xrightarrow{k_{10}} O_{2}^{+} + e$$

$$N_{2} \xrightarrow{k_{12}} N_{2}^{+} + e$$

#### Question

Complete the provided template. Simulate the system at the provided conditions (100km and 300km altitude and the respective initial compositions). For simulation, use scipy.integrate.solve\_ivp and specify 'LSODA' as high-performance method.

#### Rate coefficients

$$\begin{split} k_1(T) &= 1.533 \times 10^{-12} - 5.92 \times 10^{-13} \frac{T}{300 \, \text{K}} + 8.6 \times 10^{-14} \left(\frac{T}{300 \, \text{K}}\right)^2 \\ k_2(T) &= 2.82 \times 10^{-11} \\ k_3(T) &= 1.6 \times 10^{-7} \left(\frac{300 \, \text{K}}{T}\right)^{0.55} \\ k_4(T) &= 1 \times 10^{-11} \left(\frac{300 \, \text{K}}{T}\right)^{0.23} \\ k_5(T) &= 5 \times 10^{-11} \frac{300 \, \text{K}}{T} \\ k_6(T) &= 1.2 \times 10^{-10} \\ k_7(T) &= 1 \times 10^{-11} \left(\frac{300 \, \text{K}}{T}\right)^{0.85} \\ k_8(T) &= k_{10}(T) = k_{12}(T) = 1 \times 10^{-4} \\ k_9(T) &= k_{11}(T) = k_{13}(T) = 1 \times 10^{-8} \end{split}$$

# Thank you!



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#### References

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