



# Floc Modeling

Developing a mathematical model that describes flocculation and clarification.

[https://github.com/AguaClara/floc\\_modeling](https://github.com/AguaClara/floc_modeling)

# Agenda

Clarification  
Mechanisms

Dye Selection

Future Work

Methodology

Results &  
Discussion

# Clarification Mechanisms

# What is Clarification?

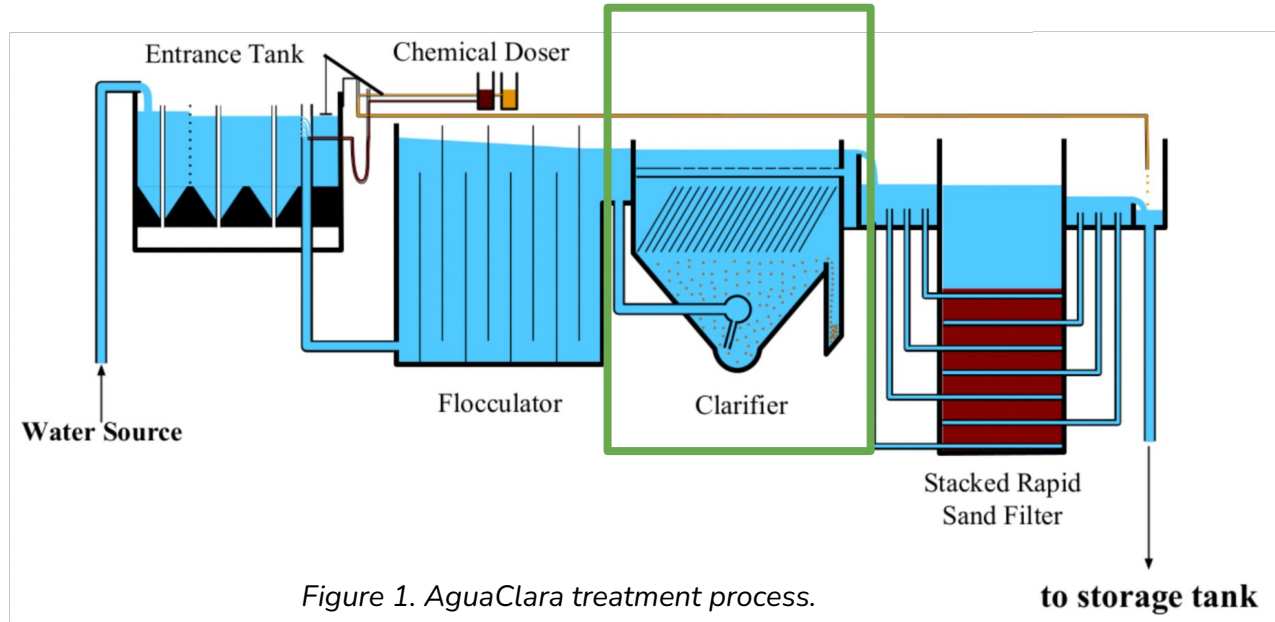


Figure 1. AguaClara treatment process.

Clarification is the process of removing suspended solids from water following flocculation.

# What Makes AguaClara Clarifiers Unique?

AguaClara treatment plants use **lamella clarifiers** with a **floc filter**

- A lamella clarifier uses parallel plate settlers to maximize the surface area where particles can settle
- The floc filter uses suspended flocs to filter out smaller primary particles
  - Floc filters are not a conventional water treatment technology in the US

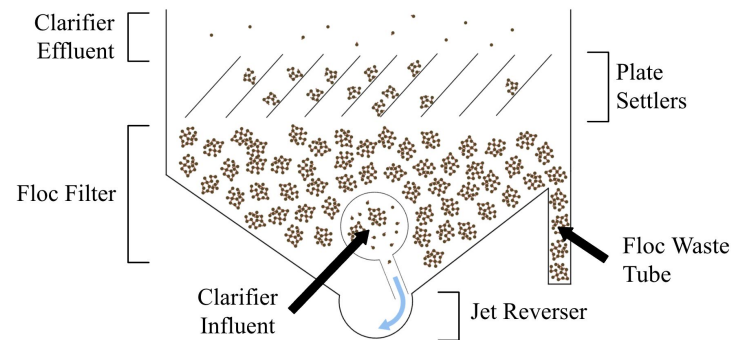


Figure 2. AguaClara clarifier diagram. The clarifier is divided between the floc filter and the plate settlers.

# Clarification Mechanisms

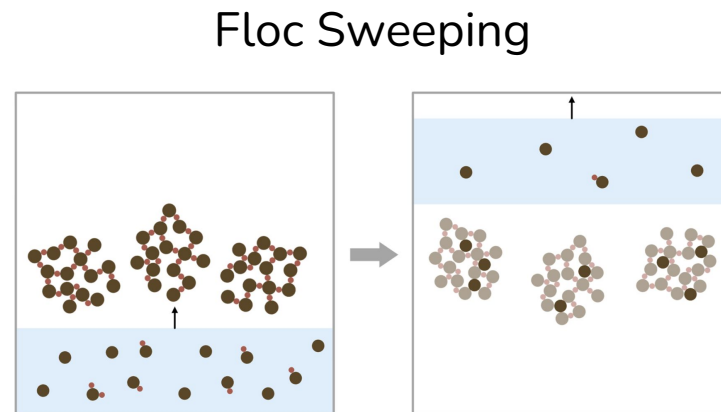
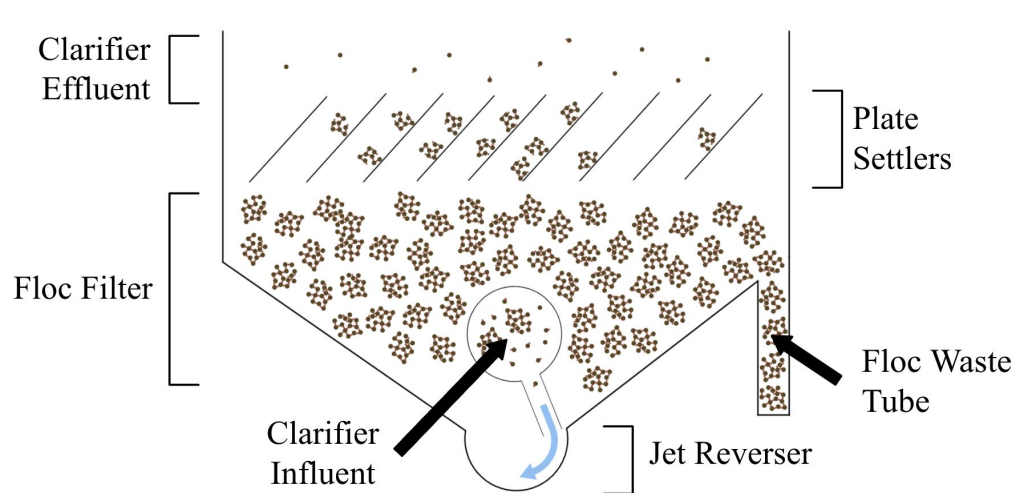


Figure 3. Clarifier diagram and primary particle removal mechanism within the clarifier.

# Floc Saturation

$$\text{Floc Saturation} = \frac{\text{primary particles captured by a floc}}{\text{maximum number of particles the floc can capture}}$$

- Increases as a floc captures primary particles and its pores fill
- Saturated flocs have reduced capture efficiency

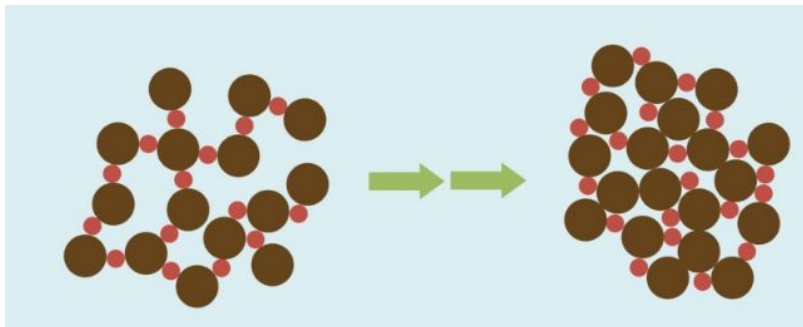


Figure 4. An unsaturated floc vs. a saturated floc.

# Floc Concentration

$$\text{Floc Concentration} = \frac{\text{number of flocs}}{\text{volume of floc filter}}$$

- Units: (Volume)<sup>-1</sup>
- Distinct from **floc saturation**
- Previous work by Swetland et al. (2014) focused on measuring variation in floc concentration throughout floc filter



# Distribution of Floc Saturation

Open question: *how is floc saturation distributed within the floc filter?*

- Current clarification model assumes floc saturation is uniformly distributed
- Floc saturation may vary spatially within clarifier
  - Ex: Flocs could be more saturated towards the bottom of the floc filter because this is where most primary particles are captured
- Wasting the most saturated flocs from floc filter first could improve performance efficiency

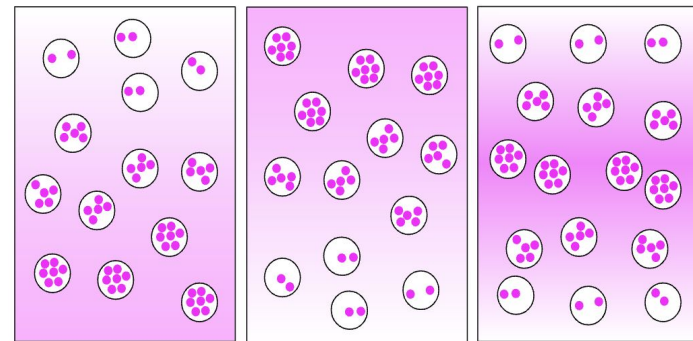


Figure 5. Examples of possible floc saturation spatial distributions.

# Methodology

*How can we measure floc saturation in the  
floc filter?*

# Previous Idea

**Previous Idea:** Use colored primary particles to visualize distribution of floc saturation

**Problems:** Primary particles did not significantly change the color of floc filter for measurement.

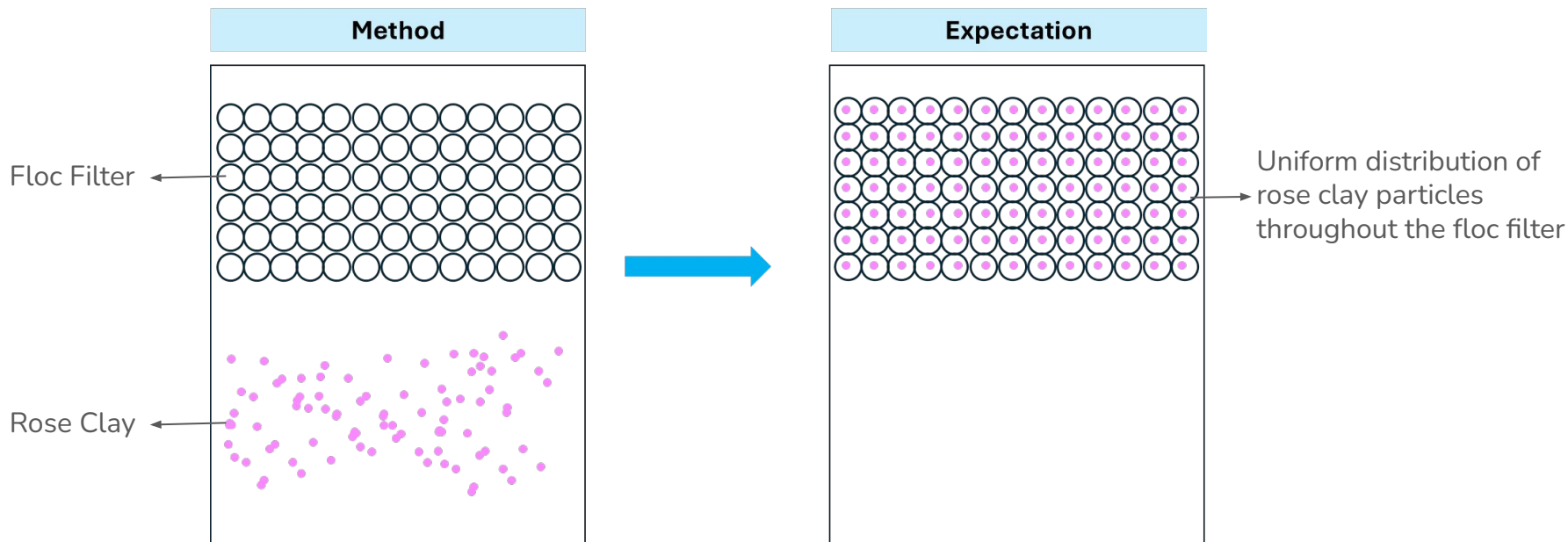


Figure 8. Rose clay addition in floc filter.

# New Method

**New Method:** Using Red-40 colored flocs to track movement of flocs to determine if the floc filter is well mixed. Thus, floc saturation is uniform.

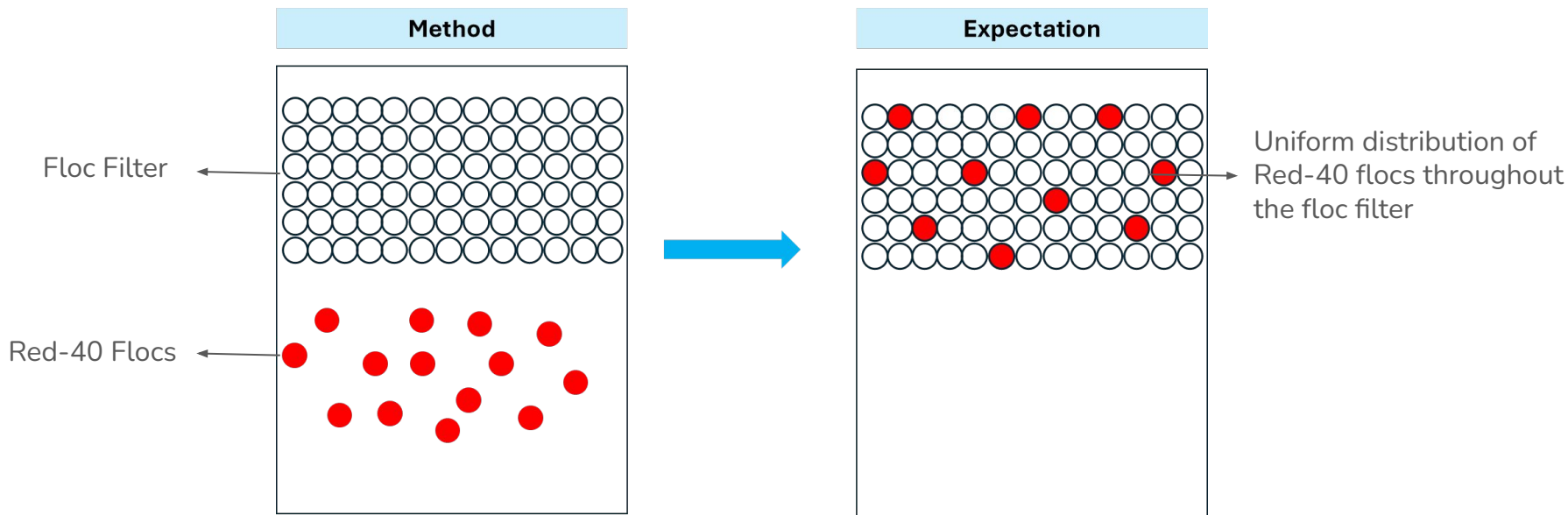


Figure 9. Red-40 floc addition to floc filter

# Well-Mixed

**Definition:** Given a particle starting at position  $x$ , after a small time step, the particle has equal probability of being found at any position  $y$  in the system. In a well-mixed system, any perturbation or addition is almost instantaneously reflected across the system.

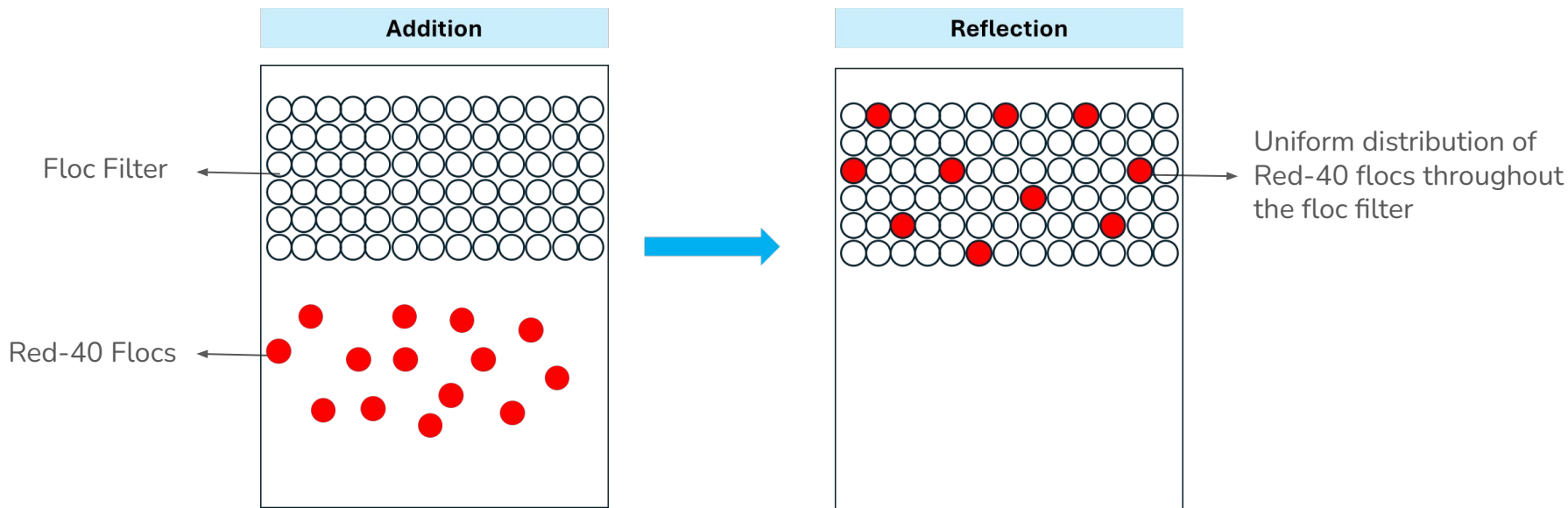


Figure 10. Reflection of perturbation in a well mixed system

# Mini Clarifier Schematics

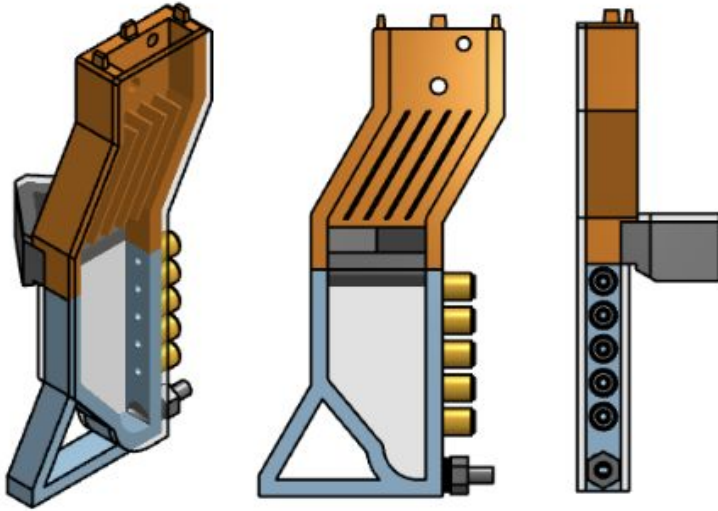


Figure 11. Miniature clarifier CAD from three different angles.

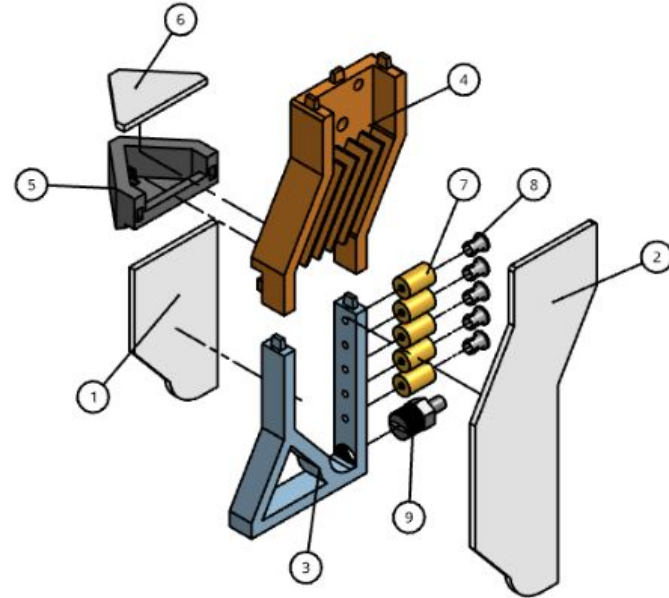


Figure 12. Exploded view of miniature clarifier CAD.

# Method

1. Form floc filter using uncolored flocs composed of white kaolin clay and coagulant
2. Form a small number of colored flocs using white kaolin clay, Red-40, and coagulant
3. Add colored flocs to clarifier and track movement of flocs throughout floc filter



# Dye Selection

# Lit Review on Red 40

**Objective:** Proof-of-concept for Red-40 adsorption onto coagulant

- Existing literature focuses on Red-40 adsorption on chitosan
- Protonated amino group in chitosan has **strong electrostatic interactions** with negative sulfonate group in Red-40

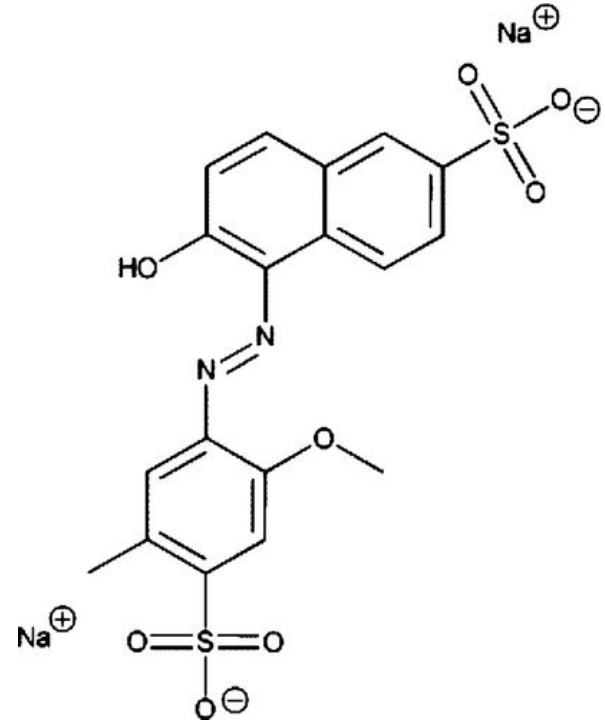


Figure 6. Chemical structure of Red-40.

# Chitosan - PACl Coagulant Similarities

- Chitosan has positive surface charge in acidic to near-neutral pH
- $\text{NH}_2$  (chitosan) and  $\text{Al}(\text{OH})^{2+}$  interact similarly with Red-40 sulfonate

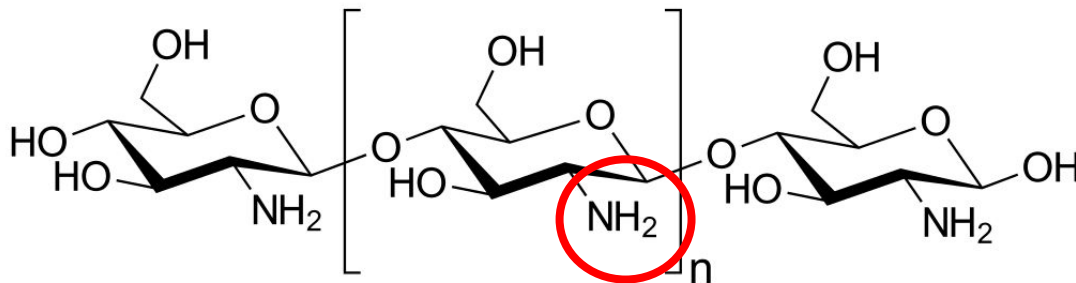


Figure 7. Chemical structure of chitosan.

# Results

# Formation of Dyed Floccs

- Red-40 and PACl were successfully flocculated into red floccs
  - Larger amounts of residual, unflocculated Red-40
  - Redness of surrounding water made floccs difficult to distinguish
- Using majority white kaolin clay and small amount of Red-40 also successfully formed red floccs
  - Less vibrant, but significantly reduced amount of residual Red-40



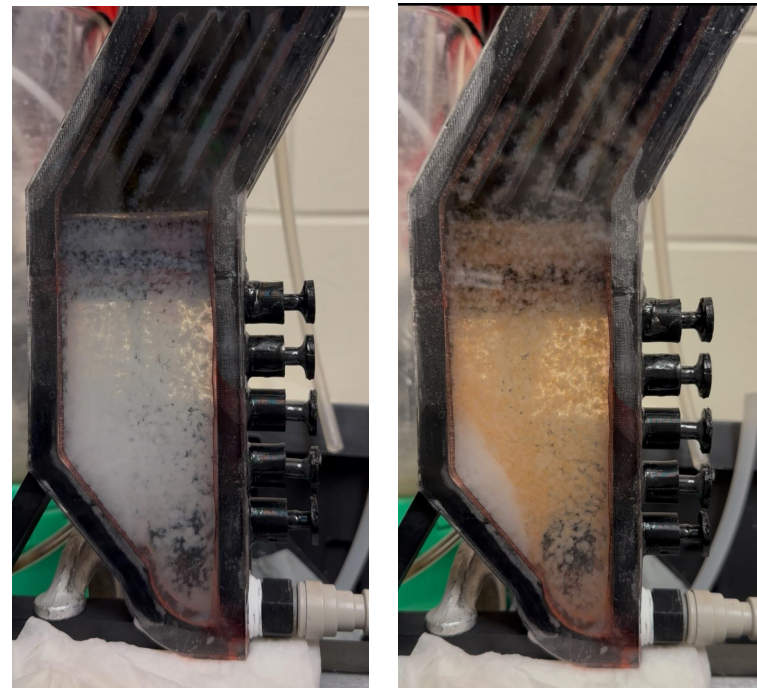
Figure 13. Floc filter formed from only Red-40 and PACl.

# Video of Red-40 Floccs



# Using Dyed Floccs as Tracers

- Adding dyed floccs to clarifier caused coloration to diffuse throughout floc filter
  - Dyed floccs became indistinguishable from undyed floccs
  - Change in coloration occurred rapidly
- Adding more undyed floccs without additional Red-40 did not affect color



*Figures 14, 15. Floc filter before and after addition of dyed floccs.*

# Video of Dyed Floccs





# Discussion

# Discussion

- Dyed flocs need to stay distinguishable from undyed flocs for method to be effective
- *Hypothesis:* Residual dissolved Red-40 that had not been flocculated was adsorbed onto undyed flocs
- Maybe due to overly high concentration of Red-40
  - Unfortunately, there will likely always be some residual unflocculated Red-40 in solution

# Future Work

# Future Work

- Repeat experiments using lower concentration of Red-40
  - Avoid large amounts of residual Red-40 dye in surrounding water
- Return to colored clay for experimentation
  - More stable, avoid unintentional color diffusion
  - Forming flocs entirely from colored clay → sufficient color variation
- Consider using hydrophobic dyes
  - Oil based dyes would not dissolve into water, but could still bond to clay to create dyed flocs

# Thanks for Listening!

## Questions?

Sarvesh Prabhu  
sp2572@cornell.edu

Alex Gardocki  
rag325@cornell.edu

Lauren Hsu  
lkh58@cornell.edu

# References

- Piccin, J. S., Dotto, G. L., Vieira, M. L. G., & Pinto, L. A. A. (2009). Adsorption of FD&C Red No. 40 by chitosan: Isotherms analysis. *Chemical Engineering Journal*, 150(2–3), 366–373. <https://doi.org/10.1016/j.jfoodeng.2009.03.017>
- Piccin, J. S., Vieira, M. L. G., Dotto, G. L., & Pinto, L. A. A. (2011). Kinetics and mechanism of the food dye FD&C Red 40 adsorption onto chitosan. *Journal of Chemical & Engineering Data*, 56(10), 3759–3765. <https://doi.org/10.1021/je200388s>
- Sarmiento, K. (2021). Particle Removal in Floc Blanket Clarifiers Via Internal Flow Through Porous Fractal Aggregates. Cornell University.
- Swetland, K., Weber-Shirk, M. L., & Lion, L. W. (2014). Flocculation–sedimentation performance model for laminar-flow hydraulic flocculation with polyaluminum chloride and aluminum sulfate coagulants. *Journal of Environmental Engineering*, 140(3), 04013014. [https://doi.org/10.1061/\(ASCE\)EE.1943-7870.0000814](https://doi.org/10.1061/(ASCE)EE.1943-7870.0000814)
- Weber-Shirk, M. (n.d.). *The Physics Of Water Treatment Design*. <https://aguaclara.github.io/Textbook/index.html>