

Prelab Lab 2

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If the microscope's calibration factor is 100 nm / pixel, what dilution of beads do you need to make to achieve the desired density in the viewing area? The density of polystyrene is $\approx 1.05 \text{ g/ml}$. Assume that the depth of field $\approx 5 \mu\text{m}$ and that you are doing your first measurements with beads with diameter $1 \mu\text{m}$. Assume, also, that the concentration is as specified in Sec. 2.2.

Pixel pitch:

$3.45 \mu\text{m}$ per pixel

WRONG - 100nm per pixel

Conc. 0.5% weight by volume

$$\begin{aligned} & 20 \text{ beads per } 1440 \times 1080 \text{ pixels} \times 0.1 \mu\text{m per pixel} \\ & = 20 \text{ beads per } 144 \times 108 \mu\text{m} \times 5 \mu\text{m} \end{aligned}$$

G of beads

$$0.5 \mu\text{m}^3 \text{ rad per bead} = 0.5236 \mu\text{m}^3$$

$$/1000^3 = 0.5236 \times 10^{-9} \text{ mm}^3$$

$$/1000 = 0.5236 \times 10^{-12} \text{ cm}^3 (\text{ml})$$

$$\times 1.05 \text{ g/ml} = 5.4978 \times 10^{-13} \text{ g/bead}$$

$$\text{Volume of image} = 77,760 \mu\text{m}^3 / 1000^3 = 77.76 \times 10^{-6} \text{ mm}^3$$

$$/1000 = 7.776 \times 10^{-8} \text{ ml}$$

$$\text{Beads per ml} = 20 / 7.776 \times 10^{-8} \text{ ml} = 2.572 \times 10^8 \text{ beads/ml}$$

$$\text{g/bead} * \text{bead/ml} = \text{g/ml} = 0.0001414$$

$$0.005 \text{ g}/1 \text{ ml initial}$$

$$\text{Initial/final conc.} = 35.36X \text{ dilution}$$
