Number	Reaction		A	n	E	Ref.
1f	$H + O_2 \rightleftharpoons OH + O$		3.520E+16	-0.70	71.4	[1]
2f	$H_2 + O \rightleftharpoons OH + H$		5.060E+04	2.67	26.3	[1]
3f	$H_2 + OH \rightleftharpoons H_2O + H$		1.170E+09	1.30	15.2	[1]
4f	$H_2O + O \rightleftharpoons 2\ OH$		7.600E+00	3.84	53.5	[1]
5f[a]	$2\ H + M^{(1)} \rightleftharpoons H_2 + M^{(1)}$		1.300E+18	-1.00	0	[2]
6f[a]	$H + OH + M^{(2)} \rightleftharpoons H_2O + M^{(2)}$		4.000E+22	-2.00	0	[2]
7f[a]	$2\ O + M^{(3)} \rightleftharpoons O_2 + M^{(3)}$		6.170E+15	-0.50	0	[2]
8f[a]	$H + O + M^{(4)} \rightleftharpoons OH + M^{(4)}$		4.710E+18	-1.00	0	[2]
9f[a]	$O + OH + M^{(4)} \rightleftharpoons HO_2 + M^{(4)}$		8.000E+15	0.00	0	[2]
10f[a,b]	$H + O_2 + M^{(5)} \rightleftharpoons HO_2 + M^{(5)}$	k_0	5.750E+19	-1.40	0	[3, 2]
		k_∞	4.650E+12	0.44	0	
11f	$HO_2 + H \rightleftharpoons 2\ OH$		7.080E+13	0.00	1.23	[4]
12f	$HO_2 + H \rightleftharpoons H_2 + O_2$		1.660E+13	0.00	3.44	[4]
13f	$HO_2 + H \rightleftharpoons H_2O + O$		3.100E+13	0.00	7.2	[1]
14f	$HO_2 + O \rightleftharpoons OH + O_2$		2.000E+13	0.00	0	[5]
15f	$HO_2 + OH \rightleftharpoons H_2O + O_2$		2.890E+13	0.00	-2.08	[1]
16f[a,b]	$2\ OH + M^{(6)} \rightleftharpoons H_2O_2 + M^{(6)}$	k_0	2.300E+18	-0.90	-7.12	[1]
		k_∞	7.400E+13	-0.37	0	
17f	$2\ HO_2 \rightleftharpoons H_2O_2 + O_2$		3.020E+12	0.00	5.8	[1]
18f	$H_2O_2 + H \rightleftharpoons HO_2 + H_2$		4.790E+13	0.00	33.3	[6]
19f	$H_2O_2 + H \rightleftharpoons H_2O + OH$		1.000E+13	0.00	15	[6]
20f	$H_2O_2 + OH \rightleftharpoons H_2O + HO_2$		7.080E+12	0.00	6	[1]
21f	$H_2O_2 + O \rightleftharpoons HO_2 + OH$		9.630E+06	2.00	16.7	[1]
22f	$CO + OH \rightleftharpoons CO_2 + H$		4.400E+06	1.50	-3.1	[1]
23f	$CO + HO_2 \rightleftharpoons CO_2 + OH$		6.000E+13	0.00	96	[1]
24f	$CO + O_2 \rightleftharpoons CO_2 + O$		1.000E+12	0.00	200	[2]
25f[a]	$HCO + M^{(7)} \rightleftharpoons CO + H + M^{(7)}$		1.860E+17	-1.00	71.1	[7]
26f	$HCO + H \rightleftharpoons CO + H_2$		5.000E+13	0.00	0	[8]
27f	$HCO + O \rightleftharpoons CO + OH$		3.000E+13	0.00	0	[1]
28f	$HCO + O \rightleftharpoons CO_2 + H$		3.000E+13	0.00	0	[1]
29f	$HCO + OH \rightleftharpoons CO + H_2O$		3.000E+13	0.00	0	[9]
30f	$HCO + O_2 \rightleftharpoons CO + HO_2$		7.580E+12	0.00	1.72	[10]
31f	$HCO + CH_3 \rightleftharpoons CO + CH_4$		5.000E+13	0.00	0	[8]
32f[a,b]	$H + HCO + M^{(8)} \rightleftharpoons CH_2O + M^{(8)}$	k_0	1.350E+24	-2.57	1.78	[11]
		k_∞	1.090E+12	0.48	-1.09	
33f	$CH_2O + H \rightleftharpoons HCO + H_2$		5.740E+07	1.90	11.5	[12]

Number	Reaction			A	n	E	Ref.
34f	$CH_2O + O$	$\rightleftharpoons$	$HCO + OH$	3.500E+13	0.00	14.7	[1]
35f	$CH_2O + OH$	$\rightleftharpoons$	$HCO + H_2O$	3.900E+10	0.89	1.7	[1]
36f	$CH_2O + O_2$	$\rightleftharpoons$	$HCO + HO_2$	6.000E+13	0.00	170	[13]
37f	$CH_2O + HO_2$	$\rightleftharpoons$	$HCO + H_2O_2$	4.110E+04	2.50	42.7	[14]
38f	$CH_4 + H$	$\rightleftharpoons$	$H_2 + CH_3$	1.300E+04	3.00	33.6	[15]
39f	$CH_4 + OH$	$\rightleftharpoons$	$H_2O + CH_3$	1.600E+07	1.83	11.6	[15]
40f	$CH_4 + O$	$\rightleftharpoons$	$CH_3 + OH$	1.900E+09	1.44	36.3	[16]
41f	$CH_4 + O_2$	$\rightleftharpoons$	$CH_3 + HO_2$	3.980E+13	0.00	238	[7, 17]
42f	$CH_4 + HO_2$	$\rightleftharpoons$	$CH_3 + H_2O_2$	9.030E+12	0.00	103	[7, 17]
43f	$CH_3 + H$	$\rightleftharpoons$	$T\text{-}CH_2 + H_2$	1.800E+14	0.00	63.2	[16]
44f	$CH_3 + H$	$\rightleftharpoons$	$S\text{-}CH_2 + H_2$	1.550E+14	0.00	56.4	[16]
45f	$CH_3 + OH$	$\rightleftharpoons$	$S\text{-}CH_2 + H_2O$	4.000E+13	0.00	10.5	[18, 8]
46f	$CH_3 + O$	$\rightleftharpoons$	$CH_2O + H$	8.430E+13	0.00	0	[16]
47f	$CH_3 + T\text{-}CH_2$	$\rightleftharpoons$	$C_2H_4 + H$	4.220E+13	0.00	0	[13]
48f	$CH_3 + HO_2$	$\rightleftharpoons$	$CH_3O + OH$	5.000E+12	0.00	0	[13]
49f	$CH_3 + O_2$	$\rightleftharpoons$	$CH_2O + OH$	3.300E+11	0.00	37.4	[19]
50f	$CH_3 + O_2$	$\rightleftharpoons$	$CH_3O + O$	1.100E+13	0.00	116	[19]
51f	$2\ CH_3$	$\rightleftharpoons$	$C_2H_4 + H_2$	1.000E+14	0.00	134	[20]
52f	$2\ CH_3$	$\rightleftharpoons$	$C_2H_5 + H$	3.160E+13	0.00	61.5	[21]
53f[a,b]	$H + CH_3 + M^{(9)}$	$\rightleftharpoons$	$CH_4 + M^{(9)}$ k_0	2.470E+33	-4.76	10.2	[22]
			k_∞	1.270E+16	-0.63	1.6	
54f[a,b]	$2\ CH_3 + M^{(8)}$	$\rightleftharpoons$	$C_2H_6 + M^{(8)}$ k_0	1.270E+41	-7.00	11.6	[15]
			k_∞	1.810E+13	0.00	0	
55f	$S\text{-}CH_2 + OH$	$\rightleftharpoons$	$CH_2O + H$	3.000E+13	0.00	0	[16]
56f	$S\text{-}CH_2 + O_2$	$\rightleftharpoons$	$CO + OH + H$	3.130E+13	0.00	0	[16]
57f	$S\text{-}CH_2 + CO_2$	$\rightleftharpoons$	$CO + CH_2O$	3.000E+12	0.00	0	[23]
58f[a]	$S\text{-}CH_2 + M^{(10)}$	$\rightleftharpoons$	$T\text{-}CH_2 + M^{(10)}$	6.000E+12	0.00	0	[16]
59f	$T\text{-}CH_2 + H$	$\rightleftharpoons$	$CH + H_2$	6.020E+12	0.00	-7.48	[13]
60f	$T\text{-}CH_2 + OH$	$\rightleftharpoons$	$CH_2O + H$	2.500E+13	0.00	0	[16]
61f	$T\text{-}CH_2 + OH$	$\rightleftharpoons$	$CH + H_2O$	1.130E+07	2.00	12.6	[16]
62f	$T\text{-}CH_2 + O$	$\rightleftharpoons$	$CO + 2\ H$	8.000E+13	0.00	0	[24]
63f	$T\text{-}CH_2 + O$	$\rightleftharpoons$	$CO + H_2$	4.000E+13	0.00	0	[24]
64f	$T\text{-}CH_2 + O_2$	$\rightleftharpoons$	$CO_2 + H_2$	2.630E+12	0.00	6.24	[23]
65f	$T\text{-}CH_2 + O_2$	$\rightleftharpoons$	$CO + OH + H$	6.580E+12	0.00	6.24	[23]
66f	$2\ T\text{-}CH_2$	$\rightleftharpoons$	$C_2H_2 + 2\ H$	1.000E+14	0.00	0	[16]
67f	$CH + O$	$\rightleftharpoons$	$CO + H$	4.000E+13	0.00	0	[25]

Number	Reaction			A	n	E	Ref.
68f	$CH + O_2$	$\rightleftharpoons$	$HCO + O$	1.770E+11	0.76	-2	[26]
69f	$CH + H_2O$	$\rightleftharpoons$	$CH_2O + H$	1.170E+15	-0.75	0	[23]
70f	$CH + CO_2$	$\rightleftharpoons$	$HCO + CO$	4.800E+01	3.22	-13.5	[26]
71f	$CH_3O + H$	$\rightleftharpoons$	$CH_2O + H_2$	2.000E+13	0.00	0	[27]
72f	$CH_3O + H$	$\rightleftharpoons$	$S\text{-}CH_2 + H_2O$	1.600E+13	0.00	0	[27]
73f	$CH_3O + OH$	$\rightleftharpoons$	$CH_2O + H_2O$	5.000E+12	0.00	0	[27]
74f	$CH_3O + O$	$\rightleftharpoons$	$OH + CH_2O$	1.000E+13	0.00	0	[27]
75f	$CH_3O + O_2$	$\rightarrow$	$CH_2O + HO_2$	4.280E-13	7.60	-14.8	[27]
76f[a]	$CH_3O + M^{(9)}$	$\rightleftharpoons$	$CH_2O + H + M^{(9)}$	7.780E+13	0.00	56.5	[8]
77f	$C_2H_6 + H$	$\rightleftharpoons$	$C_2H_5 + H_2$	5.400E+02	3.50	21.8	[16]
78f	$C_2H_6 + O$	$\rightleftharpoons$	$C_2H_5 + OH$	1.400E+00	4.30	11.6	[16]
79f	$C_2H_6 + OH$	$\rightleftharpoons$	$C_2H_5 + H_2O$	2.200E+07	1.90	4.7	[16]
80f	$C_2H_6 + CH_3$	$\rightleftharpoons$	$C_2H_5 + CH_4$	5.500E-01	4.00	34.7	[16]
81f[a,b]	$C_2H_6 + M^{(8)}$	$\rightleftharpoons$	$C_2H_5 + H + M^{(8)}$ k_0	4.900E+42	-6.43	448	[15, 11, 8]
			k_∞	8.850E+20	-1.23	428	
82f	$C_2H_6 + HO_2$	$\rightleftharpoons$	$C_2H_5 + H_2O_2$	1.320E+13	0.00	85.6	[13, 8]
83f	$C_2H_5 + H$	$\rightleftharpoons$	$C_2H_4 + H_2$	3.000E+13	0.00	0	[16]
84f	$C_2H_5 + O$	$\rightleftharpoons$	$C_2H_4 + OH$	3.060E+13	0.00	0	[16]
85f	$C_2H_5 + O$	$\rightleftharpoons$	$CH_3 + CH_2O$	4.240E+13	0.00	0	[16]
86f	$C_2H_5 + O_2$	$\rightleftharpoons$	$C_2H_4 + HO_2$	2.000E+12	0.00	20.9	[16]
87f[a,b]	$C_2H_5 + M^{(9)}$	$\rightleftharpoons$	$C_2H_4 + H + M^{(9)}$ k_0	3.990E+33	-4.99	167	[28, 8]
			k_∞	1.110E+10	1.04	154	
88f	$C_2H_4 + H$	$\rightleftharpoons$	$C_2H_3 + H_2$	4.490E+07	2.12	55.9	[29]
89f	$C_2H_4 + OH$	$\rightleftharpoons$	$C_2H_3 + H_2O$	5.530E+05	2.31	12.4	[29]
90f	$C_2H_4 + O$	$\rightleftharpoons$	$CH_3 + HCO$	2.250E+06	2.08	0	[13]
91f	$C_2H_4 + O$	$\rightleftharpoons$	$CH_2CHO + H$	1.210E+06	2.08	0	[13]
92f	$2\ C_2H_4$	$\rightleftharpoons$	$C_2H_3 + C_2H_5$	5.010E+14	0.00	271	[30]
93f	$C_2H_4 + O_2$	$\rightleftharpoons$	$C_2H_3 + HO_2$	4.220E+13	0.00	241	[31]
94f	$C_2H_4 + HO_2$	$\rightleftharpoons$	$C_2H_4O + OH$	2.230E+12	0.00	71.9	[13]
95f	$C_2H_4O + HO_2$	$\rightleftharpoons$	$CH_3 + CO + H_2O_2$	4.000E+12	0.00	71.2	[13]
96f[a]	$C_2H_4 + M^{(9)}$	$\rightleftharpoons$	$C_2H_3 + H + M^{(9)}$	2.600E+17	0.00	404	[32, 8]
97f[a]	$C_2H_4 + M^{(9)}$	$\rightleftharpoons$	$C_2H_2 + H_2 + M^{(9)}$	3.500E+16	0.00	299	[32, 8]
98f	$C_2H_3 + H$	$\rightleftharpoons$	$C_2H_2 + H_2$	4.000E+13	0.00	0	[8]
99f[a,b]	$C_2H_3 + M^{(9)}$	$\rightleftharpoons$	$C_2H_2 + H + M^{(9)}$ k_0	1.510E+14	0.10	137	[33, 8]
			k_∞	6.380E+09	1.00	157	
100f	$C_2H_3 + O_2$	$\rightleftharpoons$	$CH_2O + HCO$	1.700E+29	-5.31	27.2	[34]

Number	Reaction	A	n	E	Ref.
101f	$C_2H_3 + O_2 \rightleftharpoons CH_2CHO + O$	7.000E+14	-0.61	22	[33, 34]
102f	$C_2H_3 + O_2 \rightleftharpoons C_2H_2 + HO_2$	5.190E+15	-1.26	13.9	[33, 34]
103f	$C_2H_2 + O \rightleftharpoons HCCO + H$	4.000E+14	0.00	44.6	[24]
104f	$C_2H_2 + O \rightleftharpoons T\text{-}CH_2 + CO$	1.600E+14	0.00	41.4	[24]
105f	$C_2H_2 + O_2 \rightleftharpoons CH_2O + CO$	4.600E+15	-0.54	188	[35]
106f	$C_2H_2 + OH \rightleftharpoons CH_2CO + H$	1.900E+07	1.70	4.18	[7, 36]
107f	$C_2H_2 + OH \rightleftharpoons C_2H + H_2O$	3.370E+07	2.00	58.6	[7, 36]
108f	$CH_2CO + H \rightleftharpoons CH_3 + CO$	1.500E+09	1.43	11.2	[37]
109f	$CH_2CO + O \rightleftharpoons T\text{-}CH_2 + CO_2$	2.000E+13	0.00	9.6	[7, 36]
110f	$CH_2CO + O \rightleftharpoons HCCO + OH$	1.000E+13	0.00	8.37	[7, 36]
111f	$CH_2CO + CH_3 \rightleftharpoons C_2H_5 + CO$	9.000E+10	0.00	0	[7, 36]
112f	$HCCO + H \rightleftharpoons S\text{-}CH_2 + CO$	1.500E+14	0.00	0	[24]
113f	$HCCO + OH \rightleftharpoons HCO + CO + H$	2.000E+12	0.00	0	[38]
114f	$HCCO + O \rightleftharpoons 2\ CO + H$	9.640E+13	0.00	0	[24]
115f	$HCCO + O_2 \rightleftharpoons 2\ CO + OH$	2.880E+07	1.70	4.19	[33]
116f	$HCCO + O_2 \rightleftharpoons CO_2 + CO + H$	1.400E+07	1.70	4.19	[33]
117f	$C_2H + OH \rightleftharpoons HCCO + H$	2.000E+13	0.00	0	[16, 36]
118f	$C_2H + O \rightleftharpoons CO + CH$	1.020E+13	0.00	0	[16, 36]
119f	$C_2H + O_2 \rightleftharpoons HCCO + O$	6.020E+11	0.00	0	[16, 36]
120f	$C_2H + O_2 \rightleftharpoons CH + CO_2$	4.500E+15	0.00	105	[16, 36]
121f	$C_2H + O_2 \rightleftharpoons HCO + CO$	2.410E+12	0.00	0	[16, 36]
122f	$CH_2OH + H \rightleftharpoons CH_2O + H_2$	3.000E+13	0.00	0	[27]
123f	$CH_2OH + H \rightleftharpoons CH_3 + OH$	2.500E+17	-0.93	21.5	[8]
124f	$CH_2OH + OH \rightleftharpoons CH_2O + H_2O$	2.400E+13	0.00	0	[27]
125f	$CH_2OH + O_2 \rightleftharpoons CH_2O + HO_2$	5.000E+12	0.00	0	[27]
126f[a]	$CH_2OH + M^{(9)} \rightleftharpoons CH_2O + H + M^{(9)}$	5.000E+13	0.00	105	[27]
127f[a]	$CH_3O + M^{(9)} \rightleftharpoons CH_2OH + M^{(9)}$	1.000E+14	0.00	80	[27]
128f	$CH_2CO + OH \rightleftharpoons CH_2OH + CO$	1.020E+13	0.00	0	[27]
129f	$CH_3OH + OH \rightleftharpoons CH_2OH + H_2O$	1.440E+06	2.00	-3.51	[27]
130f	$CH_3OH + OH \rightleftharpoons CH_3O + H_2O$	4.400E+06	2.00	6.3	[8]
131f	$CH_3OH + H \rightleftharpoons CH_2OH + H_2$	1.354E+03	3.20	14.6	[39]
132f	$CH_3OH + H \rightleftharpoons CH_3O + H_2$	6.830E+01	3.40	30.3	[39]
133f	$CH_3OH + O \rightleftharpoons CH_2OH + OH$	1.000E+13	0.00	19.6	[27]
134f	$CH_3OH + HO_2 \rightleftharpoons CH_2OH + H_2O_2$	6.200E+12	0.00	81.1	[27]
135f	$CH_3OH + O_2 \rightleftharpoons CH_2OH + HO_2$	2.000E+13	0.00	188	[27]
136f[a,b]	$CH_3OH + M^{(9)} \rightleftharpoons CH_3 + OH + M^{(9)} \quad k_0$	2.950E+44	-7.35	399	[40, 8]

Number	Reaction		A	n	E	Ref.
		k_∞	1.900E+16	0.00	384	
137f	$CH_2CHO \rightleftharpoons CH_2CO + H$		1.047E+37	-7.19	186	[31]
138f	$CH_2CHO + H \rightleftharpoons CH_3 + HCO$		5.000E+13	0.00	0	[12]
139f	$CH_2CHO + H \rightleftharpoons CH_2CO + H_2$		2.000E+13	0.00	0	[12]
140f	$CH_2CHO + O \rightleftharpoons CH_2O + HCO$		1.000E+14	0.00	0	[12]
141f	$CH_2CHO + OH \rightleftharpoons CH_2CO + H_2O$		3.000E+13	0.00	0	[12]
142f	$CH_2CHO + O_2 \rightleftharpoons CH_2O + CO + OH$		3.000E+10	0.00	0	[12]
143f	$CH_2CHO + CH_3 \rightleftharpoons C_2H_5 + CO + H$		4.900E+14	-0.50	0	[12]
144f	$CH_2CHO + HO_2 \rightleftharpoons CH_2O + HCO + OH$		7.000E+12	0.00	0	[12]
145f	$CH_2CHO + HO_2 \rightleftharpoons CH_3CHO + O_2$		3.000E+12	0.00	0	[12]
146f	$CH_2CHO \rightleftharpoons CH_3 + CO$		1.170E+43	-9.80	183	[12]
147f	$CH_3CHO \rightleftharpoons CH_3 + HCO$		7.000E+15	0.00	342	[12]
148f[a,b]	$CH_3CO + M^{(9)} \rightleftharpoons CH_3 + CO + M^{(9)}$	k_0	1.200E+15	0.00	52.3	[12]
		k_∞	3.000E+12	0.00	69.9	
149f	$CH_3CHO + OH \rightleftharpoons CH_3CO + H_2O$		3.370E+12	0.00	-2.59	[12]
150f	$CH_3CHO + OH \rightleftharpoons CH_2CHO + H_2O$		3.370E+11	0.00	-2.59	[12]
151f	$CH_3CHO + O \rightleftharpoons CH_3CO + OH$		1.770E+18	-1.90	12.5	[12]
152f	$CH_3CHO + O \rightleftharpoons CH_2CHO + OH$		3.720E+13	-0.20	14.9	[12]
153f	$CH_3CHO + H \rightleftharpoons CH_3CO + H_2$		4.660E+13	-0.30	12.5	[12]
154f	$CH_3CHO + H \rightleftharpoons CH_2CHO + H_2$		1.850E+12	0.40	22.4	[12]
155f	$CH_3CHO + CH_3 \rightleftharpoons CH_3CO + CH_4$		3.900E-07	5.80	9.21	[12]
156f	$CH_3CHO + CH_3 \rightleftharpoons CH_2CHO + CH_4$		2.450E+01	3.10	24	[12]
157f	$CH_3CHO + HO_2 \rightleftharpoons CH_3CO + H_2O_2$		3.600E+19	-2.20	58.6	[12]
158f	$CH_3CHO + HO_2 \rightleftharpoons CH_2CHO + H_2O_2$		2.320E+11	0.40	62.3	[12]
159f	$CH_3CHO + O_2 \rightleftharpoons CH_3CO + HO_2$		1.000E+14	0.00	177	[12]
160f[a,b]	$C_2H_5OH + M^{(9)} \rightleftharpoons CH_3 + CH_2OH + M^{(9)}$	k_0	3.000E+16	0.00	243	[8]
		k_∞	5.000E+15	0.00	343	
161f[a,b]	$C_2H_5OH + M^{(9)} \rightleftharpoons C_2H_4 + H_2O + M^{(9)}$	k_0	1.000E+17	0.00	226	[8]
		k_∞	8.000E+13	0.00	272	
162f	$C_2H_5OH + OH \rightleftharpoons CH_2CH_2OH + H_2O$		1.810E+11	0.40	3	[12]
163f	$C_2H_5OH + OH \rightleftharpoons CH_3CHOH + H_2O$		3.090E+10	0.50	-1.59	[12]
164f	$C_2H_5OH + OH \rightleftharpoons CH_3CH_2O + H_2O$		1.050E+10	0.80	3	[12]
165f	$C_2H_5OH + H \rightleftharpoons CH_2CH_2OH + H_2$		1.900E+07	1.80	21.3	[12]
166f	$C_2H_5OH + H \rightleftharpoons CH_3CHOH + H_2$		2.580E+07	1.60	11.8	[12]
167f	$C_2H_5OH + H \rightleftharpoons CH_3CH_2O + H_2$		1.500E+07	1.60	12.7	[12]
168f	$C_2H_5OH + O \rightleftharpoons CH_2CH_2OH + OH$		9.410E+07	1.70	22.8	[12]

Number	Reaction	A	n	E	Ref.
169f	$C_2H_5OH + O \rightleftharpoons CH_3CHOH + OH$	1.880E+07	1.90	7.62	[12]
170f	$C_2H_5OH + O \rightleftharpoons CH_3CH_2O + OH$	1.580E+07	2.00	18.6	[12]
171f	$C_2H_5OH + CH_3 \rightleftharpoons CH_2CH_2OH + CH_4$	2.190E+02	3.20	40.2	[12]
172f	$C_2H_5OH + CH_3 \rightleftharpoons CH_3CHOH + CH_4$	7.280E+02	3.00	33.3	[12]
173f	$C_2H_5OH + CH_3 \rightleftharpoons CH_3CH_2O + CH_4$	1.450E+02	3.00	32	[12]
174f	$C_2H_5OH + HO_2 \rightleftharpoons CH_3CHOH + H_2O_2$	8.200E+03	2.50	45.2	[12]
175f	$C_2H_5OH + HO_2 \rightleftharpoons CH_2CH_2OH + H_2O_2$	2.430E+04	2.50	66.1	[12]
176f	$C_2H_5OH + HO_2 \rightleftharpoons CH_3CH_2O + H_2O_2$	3.800E+12	0.00	100	[12]
177f	$C_2H_4 + OH \rightleftharpoons CH_2CH_2OH$	2.410E+11	0.00	-9.96	[12]
178f	$C_2H_5 + HO_2 \rightleftharpoons CH_3CH_2O + OH$	4.000E+13	0.00	0	[12]
179f[a]	$CH_3CH_2O + M^{(9)} \rightleftharpoons CH_3CHO + H + M^{(9)}$	5.600E+34	-5.90	106	[12]
180f[a]	$CH_3CH_2O + M^{(9)} \rightleftharpoons CH_3 + CH_2O + M^{(9)}$	5.350E+37	-7.00	99.6	[12]
181f	$CH_3CH_2O + O_2 \rightleftharpoons CH_3CHO + HO_2$	4.000E+10	0.00	4.6	[12]
182f	$CH_3CH_2O + CO \rightleftharpoons C_2H_5 + CO_2$	4.680E+02	3.20	22.5	[12]
183f	$CH_3CH_2O + H \rightleftharpoons CH_3 + CH_2OH$	3.000E+13	0.00	0	[12]
184f	$CH_3CH_2O + H \rightleftharpoons C_2H_4 + H_2O$	3.000E+13	0.00	0	[12]
185f	$CH_3CH_2O + OH \rightleftharpoons CH_3CHO + H_2O$	1.000E+13	0.00	0	[12]
186f	$CH_3CHOH + O_2 \rightleftharpoons CH_3CHO + HO_2$	4.820E+13	0.00	21	[12]
187f	$CH_3CHOH + O \rightleftharpoons CH_3CHO + OH$	1.000E+14	0.00	0	[12]
188f	$CH_3CHOH + H \rightleftharpoons C_2H_4 + H_2O$	3.000E+13	0.00	0	[12]
189f	$CH_3CHOH + H \rightleftharpoons CH_3 + CH_2OH$	3.000E+13	0.00	0	[12]
190f	$CH_3CHOH + HO_2 \rightleftharpoons CH_3CHO + 2\ OH$	4.000E+13	0.00	0	[12]
191f	$CH_3CHOH + OH \rightleftharpoons CH_3CHO + H_2O$	5.000E+12	0.00	0	[12]
192f[a]	$CH_3CHOH + M^{(9)} \rightleftharpoons CH_3CHO + H + M^{(9)}$	1.000E+14	0.00	105	[12]
193f	$C_3H_4 + O \rightleftharpoons C_2H_4 + CO$	2.000E+07	1.80	4.18	[41]
194f	$CH_3 + C_2H_2 \rightleftharpoons C_3H_4 + H$	2.560E+09	1.10	57.1	[41]
195f	$C_3H_4 + O \rightleftharpoons HCCO + CH_3$	7.300E+12	0.00	9.41	[41]
196f[a,b]	$C_3H_3 + H + M \rightleftharpoons C_3H_4 + M \quad k_0$	9.000E+15	1.00	0	[37]
	k_∞	3.000E+13	0.00	0	
197f	$C_3H_3 + HO_2 \rightleftharpoons C_3H_4 + O_2$	2.500E+12	0.00	0	[37]
198f	$C_3H_4 + OH \rightleftharpoons C_3H_3 + H_2O$	5.300E+06	2.00	8.37	[42]
199f	$C_3H_3 + O_2 \rightleftharpoons CH_2CO + HCO$	3.000E+10	0.00	12	[43]
200f[a,b]	$C_3H_4 + H + M \rightleftharpoons C_3H_5 + M \quad k_0$	3.000E+24	-2.00	0	[37]
	k_∞	4.000E+13	0.00	0	
201f	$C_3H_5 + H \rightleftharpoons C_3H_4 + H_2$	1.800E+13	0.00	0	[44]
202f	$C_3H_5 + O_2 \rightleftharpoons C_3H_4 + HO_2$	4.990E+15	-1.40	93.8	[45]

Number	Reaction			A	n	E	Ref.
203f	$C_3H_5 + CH_3$	$\rightleftharpoons$	$C_3H_4 + CH_4$	3.000E+12	-0.32	-0.548	[37]
204f[a,b]	$C_2H_2 + CH_3 + M$	$\rightleftharpoons$	$C_3H_5 + M$ k_0	2.000E+09	1.00	0	[37]
			k_∞	6.000E+08	0.00	0	
205f	$C_3H_5 + OH$	$\rightleftharpoons$	$C_3H_4 + H_2O$	6.000E+12	0.00	0	[37]
206f	$C_3H_3 + HCO$	$\rightleftharpoons$	$C_3H_4 + CO$	2.500E+13	0.00	0	[42]
207f	$C_3H_3 + HO_2$	$\rightleftharpoons$	$OH + CO + C_2H_3$	8.000E+11	0.00	0	[41]
208f	$C_3H_4 + O_2$	$\rightleftharpoons$	$CH_3 + HCO + CO$	4.000E+14	0.00	175	[46]
209f	$C_3H_6 + O$	$\rightleftharpoons$	$C_2H_5 + HCO$	3.500E+07	1.65	-4.07	[44]
210f	$C_3H_6 + OH$	$\rightleftharpoons$	$C_3H_5 + H_2O$	3.100E+06	2.00	-1.25	[44]
211f	$C_3H_6 + O$	$\rightleftharpoons$	$CH_2CO + CH_3 + H$	1.200E+08	1.65	1.37	[44]
212f	$C_3H_6 + H$	$\rightleftharpoons$	$C_3H_5 + H_2$	1.700E+05	2.50	10.4	[44]
213f[a,b]	$C_3H_5 + H + M^{(8)}$	$\rightleftharpoons$	$C_3H_6 + M^{(8)}$ k_0	1.330E+60	-12.00	25	[41]
			k_∞	2.000E+14	0.00	0	
214f	$C_3H_5 + HO_2$	$\rightleftharpoons$	$C_3H_6 + O_2$	2.660E+12	0.00	0	[13]
215f	$C_3H_5 + HO_2$	$\rightleftharpoons$	$OH + C_2H_3 + CH_2O$	3.000E+12	0.00	0	[13]
216f[a,b]	$C_2H_3 + CH_3 + M^{(8)}$	$\rightleftharpoons$	$C_3H_6 + M^{(8)}$ k_0	4.270E+58	-11.94	40.9	[41]
			k_∞	2.500E+13	0.00	0	
217f	$C_3H_6 + H$	$\rightleftharpoons$	$C_2H_4 + CH_3$	1.600E+22	-2.39	46.8	[41]
218f	$CH_3 + C_2H_3$	$\rightleftharpoons$	$C_3H_5 + H$	1.500E+24	-2.83	77.9	[41]
219f[a,b]	$C_3H_8 + M$	$\rightleftharpoons$	$CH_3 + C_2H_5 + M$ k_0	7.830E+18	0.00	272	[32]
			k_∞	1.100E+17	0.00	353	
220f	$C_3H_8 + O_2$	$\rightleftharpoons$	I-$C_3H_7 + HO_2$	4.000E+13	0.00	199	[47, 41, 48]
221f	$C_3H_8 + O_2$	$\rightleftharpoons$	N-$C_3H_7 + HO_2$	4.000E+13	0.00	213	[47, 41, 48]
222f	$C_3H_8 + H$	$\rightleftharpoons$	I-$C_3H_7 + H_2$	1.300E+06	2.40	18.7	[47, 41, 48]
223f	$C_3H_8 + H$	$\rightleftharpoons$	N-$C_3H_7 + H_2$	1.330E+06	2.54	28.3	[48, 49]
224f	$C_3H_8 + O$	$\rightleftharpoons$	I-$C_3H_7 + OH$	4.760E+04	2.71	8.82	[48, 41]
225f	$C_3H_8 + O$	$\rightleftharpoons$	N-$C_3H_7 + OH$	1.900E+05	2.68	15.6	[48, 41]
226f	$C_3H_8 + OH$	$\rightleftharpoons$	N-$C_3H_7 + H_2O$	1.400E+03	2.66	2.21	[41]
227f	$C_3H_8 + OH$	$\rightleftharpoons$	I-$C_3H_7 + H_2O$	2.700E+04	2.39	1.65	[41]
228f	$C_3H_8 + HO_2$	$\rightleftharpoons$	I-$C_3H_7 + H_2O_2$	9.640E+03	2.60	58.2	[48, 49, 41]
229f	$C_3H_8 + HO_2$	$\rightleftharpoons$	N-$C_3H_7 + H_2O_2$	4.760E+04	2.55	69	[48, 49, 41]
230f	I-$C_3H_7 + C_3H_8$	$\rightleftharpoons$	N-$C_3H_7 + C_3H_8$	8.400E-03	4.20	36.3	[48, 50]
231f[a,b]	$C_3H_6 + H + M^{(8)}$	$\rightleftharpoons$	I-$C_3H_7 + M^{(8)}$ k_0	8.700E+42	-7.50	19.8	[41]
			k_∞	1.330E+13	0.00	6.53	
232f	I-$C_3H_7 + O_2$	$\rightleftharpoons$	$C_3H_6 + HO_2$	1.300E+11	0.00	0	[48, 41]
233f[a,b]	N-$C_3H_7 + M$	$\rightleftharpoons$	$CH_3 + C_2H_4 + M$ k_0	5.490E+49	-10.00	150	[48, 41]

Number	Reaction		A	n	E	Ref.
		k_∞	1.230E+13	-0.10	126	
234f[a,b]	$H + C_3H_6 + M^{(8)} \rightleftharpoons N\text{-}C_3H_7 + M^{(8)}$	k_0	6.260E+38	-6.66	29.3	[48, 41]
		k_∞	1.330E+13	0.00	13.6	
235f	$N\text{-}C_3H_7 + O_2 \rightleftharpoons C_3H_6 + HO_2$		9.000E+10	0.00	0	[48, 41]

Units are mol, cm^3, kJ, K.

The backward rates for all reversible reactions can be calculated from thermodynamic data.

[a]Third-body efficiencies are:

[M] = 1 [other].

[M1] = 0.5 [AR] + 0.5 [HE] + 2.5 [H2] + 12 [H2O] + 1.9 [CO] + 3.8 [CO2] + 1 [other].

[M2] = 0.38 [AR] + 0.38 [HE] + 2.5 [H2] + 12 [H2O] + 1.9 [CO] + 3.8 [CO2] + 1 [other].

[M3] = 0.2 [AR] + 0.2 [HE] + 2.5 [H2] + 12 [H2O] + 1.9 [CO] + 3.8 [CO2] + 1 [other].

[M4] = 0.75 [AR] + 0.75 [HE] + 2.5 [H2] + 12 [H2O] + 1.9 [CO] + 3.8 [CO2] + 1 [other].

[M5] = 0.7 [AR] + 0.7 [HE] + 2.5 [H2] + 16 [H2O] + 1.2 [CO] + 2.4 [CO2] + 1.5 [C2H6] + 1 [other].

[M6] = 0.4 [AR] + 0.4 [HE] + 2 [H2] + 6 [H2O] + 1.5 [CO] + 2 [CO2] + 2 [CH4] + 3 [C2H6] + 1 [other].

[M7] = 1.9 [H2] + 12 [H2O] + 2.5 [CO] + 2.5 [CO2] + 1 [other].

[M8] = 0.7 [AR] + 2 [H2] + 6 [H2O] + 1.5 [CO] + 2 [CO2] + 2 [CH4] + 3 [C2H6] + 1 [other].

[M9] = 0.7 [AR] + 2 [H2] + 6 [H2O] + 1.5 [CO] + 2 [CO2] + 2 [CH4] + 1 [other].

[M10] = 2.4 [H2] + 15.4 [H2O] + 1.8 [CO] + 3.6 [CO2] + 1 [other].

[b]Pressure dependent reactions are described by the TROE-formulation [51]. The centering parameters are given by:

$F_{c,10f}$ = 0.5.

$F_{c,16f}$ = 0.265 exp(-T/94 K) + 0.735 exp(-T/1756 K) + exp(-5182 K/T).

$F_{c,32f}$ = 0.2176 exp(-T/271 K) + 0.7824 exp(-T/2755 K) + exp(-6570 K/T).

$F_{c,53f}$ = 0.217 exp(-T/74 K) + 0.783 exp(-T/2941 K) + exp(-6964 K/T).

$F_{c,54f}$ = 0.38 exp(-T/73 K) + 0.62 exp(-T/1180 K).

$F_{c,81f}$ = 0.16 exp(-T/125 K) + 0.84 exp(-T/2219 K) + exp(-6882 K/T).

$F_{c,87f}$ = 0.832 exp(-T/1203 K).

$F_{c,99f}$ = 0.7.

$F_{c,136f}$ = 0.586 exp(-T/279 K) + 0.414 exp(-T/5459 K).

$F_{c,148f}$ = 1.

$F_{c,160f}$ = 0.5.

$F_{c,161f}$ = 0.5.

$F_{c,196f}$ = 0.5.

$F_{c,200f}$ = 0.2.

$F_{c,204f}$ = 0.5.

$F_{c,213f}$ = 0.98 exp(-T/1097 K) + 0.02 exp(-T/1097 K) + exp(-6860 K/T).

$F_{c,216f}$ = 0.825 exp(-T/1341 K) + 0.175 exp(-T/60000 K) + exp(-10140 K/T).

$F_{c,219f}$ = 0.24 exp(-T/1946 K) + 0.76 exp(-T/38 K).

$F_{c,231f}$ = exp(-T/645.4 K) + exp(-6844 K/T).

$F_{c,233f}$ = 2.17 exp(-T/251 K) + exp(-1185 K/T).

$F_{c,234f}$ = exp(-T/1310 K) + exp(-48100 K/T).

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