## **EXERCISES - MICROCANONICAL ENSEMBLE**

- 1. Compute the entropy and the equation of state for an ideal gas in two dimensions.
- 2. Excluded volume effects. Consider a 1-dimensional gas of non-interacting rods of length *R* in a volume of length *L*. Rods do not interact, but they cannot overlap or jump over one another. Compute the entropy and the equation of state.
- 3. Harmonic oscillators. Consider a collection of N non-interacting classical harmonic oscillators in 3 dimensions with frequency  $\omega$  and total energy E. Compute the entropy S and the temperature T. Should particles be considered distinguishable or not? What is the expression for the equipartition theorem in this case?
- 4. Quantum oscillators. The energy values allows for a 1-dimensional quantum oscillator of frequency  $\omega$  are  $E_n=\hbar\omega\left(n+\frac{1}{2}\right)$   $n=0,1,2,3,\ldots$

Consider a collection of N three-dimensional quantum oscillators of frequency  $\omega$  and total energy E, compute the entropy and the temperature, and the dependency of energy with temperature.

- 5. Consider a model of a crystal which consists on a regular lattice with N sites. At a certain temperature T, N-n sites are occupied by atoms and the remaining n sites are empty (Schottky vacancies). Each empty site (vacancy) increases the energy of the system by  $\gamma > 0$ .
  - a. Compute the concentration of vacancies at equilibrium
  - b. Compute the contribution of vacancies to the specific heat and study the behaviour of the system at low temperature.

Hint: consider N, n large.

- 6. Consider N identical particles that can have two possible energy levels  $\pm \varepsilon$ . Using the microcanonical ensemble, compute the entropy, the Helmholtz free-energy F=E-TS and the occupation numbers (the expected number of particles in each level  $n_{\pm \varepsilon}$ ) as a function of T. What is the highest temperature at which the system can exist?
- 7. Consider a system of identical particles each o which has two energy levels  $0, \epsilon > 0$ . The upper energy level has a g-fold degeneracy (that is there are g different states with energy  $\epsilon$ , while the 0 energy state is non-degenerate). The total energy of the system is E.
  - A. Using the microcanonical ensemble, find the occupation numbers  $n_0, n_\varepsilon$  (that is the expected number of particles in each level) as a function of the temperature of the system.
  - B. Consider the case g=2. If the system has energy  $E=0.75N\epsilon$  and we put it in contact with a bath at constant temperature  $T=500\,K$ , in what direction does the heat flow?