1) Ideal gas in 3D > Equation of State?

He = In etertified for eth

Tag: = [X3] Sup: eth Zm = V xm

poly eth = [X3] N

Tag: =

 $= e^{\frac{2(2\pi m)^{3/2} \vee 3}{\beta}}$ $\text{ln} = 2(\frac{2\pi m}{\beta})^{3/2} \vee 3$

From dan: $lu \neq = \frac{PV}{k_BT}$ and $\langle N \rangle = \frac{1}{\beta} \frac{2luk}{2m}$ $\langle N \rangle = \frac{1}{\beta} \frac{2luk}{2t} \frac{2t}{2m} = \frac{1}{\beta} \frac{p_t}{p_t} \frac{(2\pi m)^{3/2} V}{lus} = luk$ large b

So $\langle N \rangle = \frac{PV}{K_BT}$ equation of shike

Chemical potential of an ideal gas:

 $\frac{2(2\pi m k_BT)^{3/2} \frac{\chi}{h^3}}{k_BT} = \frac{p\chi}{k_BT} \qquad \lambda_T = \sqrt{\frac{2\pi m k_BT}{h}}$ $\frac{2}{k_BT} \cdot \frac{p}{(2\pi m k_BT)^{3/2}} = \frac{p}{k_BT} \lambda_T^3$

eff= 2 | M= KBT log P KBT /T3

(2) Adsorption process. In contact with a reservoir

at the peratue To No adsorbing sites -> Fads=+E This the coveragely i.e. fraction of occupied a fes Option 1 -> Compute Le, obtain (N) -> D= (N)

(experted # of
occupied sites) Le = De et Brn. Ze-BEN En = + NE Ze-pën = e-pën Z = e-pën No No shame fixed n = fixed E we put need to unt de degeneration No Run 2001/11 Ze= Z e pm e-Ben(No) = (1+ ep(x-e)) No. = (21)° lute = Nolu 21 $\langle N \rangle = \frac{1}{\beta} \frac{2 \ln \lambda}{2 \beta} = \frac{N_0 1}{2 1} \cdot \beta e^{-\beta (e-\mu)} \cdot \left[\theta = \langle N \rangle = \frac{e^{-\beta (e-\mu)}}{1 + e^{-\beta (e-\mu)}}\right]$ Option 2. Gibbs sum: Since the state of each adsorbing afe is independent from the others, me can just oughte the partition function for one site: Z1= empty + occupied = efre-fe-0+ efre-fe-1 - 1+e-p(e-p) _ (This is the Gibbs Sum) The probability that one site is occupied is tren: P= 1. e- b(e-p) = me expected # of occupied sits other (N)= P1. No Sothat D= (N) = P1 Since all sites are in defendant me Option3 note that Z = Z, Zz, Zz, ZN. So that if Z1 = 1+e-B(e-T) = Zk + x=1, No √e= 2, No (which is the same as the first remet)

from here (N)= No- e-p(e-μ)

= e-p(e-μ)

= z₁

Problem 3. Find the chemical protential µ as a function of o. $\theta = \frac{e^{-\beta e^{\circ}} z}{1 + e^{-\beta e^{\circ}} z} = \frac{Nads}{No}$ e-Beo & No = (1+e-Beoz) Nas Z(No-Nads) eBeo = Nads Z = Nads eBeo a(T)= e-860 M- Kot log (Nother). [e-Ben] Problem 4 Madenbedgas = - KBT log [EBEO (NO - 1)] M= KBT log h3P KBT)T3 Middle = Mads =) Hat - No-Nads egeo No-Nals = h3P e-Bes

Problem 5 Ideal gas PV = (Ngro) KeT >T = SAMKET Mgas = KBT lu P KRT) T3 advorted gas partides: -> Independent n'es: each site £ rite = 1+ e BEO+BM + e 2(BGO+BM) (Nals) parte = 1. eferefr. d. efertpu)2 (Nals) = No (Nads) privile $N = \langle N_0 m \rangle + \langle N_0 ds \rangle$ Mr = (Ngas) = N - (Nads) KIN = 1 - (Nuls) Since pros = prads (Mads) = 1+ Pf + (Pf)22 eph = P /3 kgT f= eper/x3 KET \[\frac{PV}{NKT} = 1 - \frac{No}{N} \cdot \frac{Pf + (Pf)^2 \cdot 2}{1 + Pf + (Pf)^2} \]