



1-2  $\rightarrow$  Mixed Region  
2-3  $\rightarrow$  Superheated Region

A-C  $\rightarrow$  Saturated Liquid line

C-B  $\rightarrow$  Saturated Vapour line

$\rightarrow$  On Saturated liquid line properties are represented by 'f' like  $h_f, v_f, s_f$

$\rightarrow$  On Saturated Vapour line properties are denoted by 'g' like  $h_g, v_g, s_g$

$\rightarrow$  In mixed region properties are denoted by 'fg' like  $h_{fg}, s_{fg}$

$\rightarrow$  Formula applicable in mixed region  $\left[ \begin{array}{l} h = h_f + x h_{fg} \\ \text{entropy, } S = s_f + x s_{fg} \end{array} \right]$

$\rightarrow$  Formula applicable in superheated region:

$$\begin{aligned} h_3 &= h_2 + c_p(T_3 - T_2) \\ &= h_{g2} + c_p(T_3 - T_2) \end{aligned}$$

$$S_3 = S_2 + c_{ps} \ln \frac{T_3}{T_2}$$

$$= s_{g2} + c_{ps} \ln \frac{T_3}{T_2}$$