

CHEMISTRY MID SEM 1

Subject Code: 4300006

Date: 2025-04-07

Subject Name: CHEMISTRY

Time Duration: 64.0 minutes

Total Marks: 64

Instructions:

1. Stable Internet Required: Ensure a good connection.
 2. Use Allowed Devices: Only a laptop/PC; no mobile phones or smartwatches.
 3. No Switching Tabs: Changing windows may lead to disqualification.
 4. Answer all questions within the given time limit. No extra time will be provided.
 5. Submit the exam before the deadline, as responses will not be accepted afterward.
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1. Formation of ____ gas by the action of dilute sulphuric acid on zinc. 2
☐ Hydrogen ☐ Oxygen ☐ Nitrogen ☐ None of these
2. Magnesium Ribbon burns with a dazzling white flame and changes into a white powder (True/False) 2
3. Magnesium oxide is formed due to reaction between 2
4. What happens when milk is left at room temperature during summers? 2
☐ It freezes
☐ It curdles due to bacterial action
☐ It evaporates completely
☐ It turns into butter
5. What happens to an iron nail when left exposed to a humid atmosphere ? 2
☐ It remains unchanged ☐ It melts ☐ It rusts due to oxidation ☐ It becomes shinier
6. What process occurs when grapes get fermented ? 2
☐ Photosynthesis
☐ Evaporation
☐ Chemical change leading to alcohol formation
☐ Freezing
7. How do we know a chemical reaction has taken place ? 2
☐ There is always a change in color
☐ The substance always turns into a gas
☐ The identity of the original substance changes
☐ The state of matter remains the same



8. What is one common characteristics of all the given situations (milk curdling, iron rusting, food digestion, etc) ? 2
- ☐ They all involve a physical change
 - ☐ They all are reversible changes
 - ☐ They involve the formation of new substances
 - ☐ All of Above
9. Which of the following is an example of a physical change ? 2
- ☐ Burning of Magnesium ribbon
 - ☐ Digestion of food
 - ☐ Melting of ice
 - ☐ Fermentation of Grapes
10. The rusting of iron in a humid atmospher is due to: 2
- ☐ Reaction with oxygen and water
 - ☐ Reaction with nitrogen in the air
 - ☐ High temperature exposure
 - ☐ The presence of Carbon Dioxide
11. When food gets cooked, what kind of change occurs ? 2
- ☐ Only a physical change
 - ☐ Only a chemical change
 - ☐ Both Physical and Chemical Change
 - ☐ No changes occurs
12. What is the main reason for using sandpaper before burning magnesium ribbon ? 2
- ☐ To make it more shinier
 - ☐ To remove the oxide layer for better burning
 - ☐ To make it easier to hold
 - ☐ To increase the weight of the ribbon
13. What is the product formed when magnesium burns in the presence of oxygen ? 2
- ☐ Magnesium chloride
 - ☐ Magnesium carbonate
 - ☐ Magnesium oxide
 - ☐ Magneisum Hydroxide
14. Which of the following is not an indication of chemical reaction ? 2
- ☐ Evolution of gas
 - ☐ Formation of precipitate
 - ☐ Melting of a solid into liquid
 - ☐ Change in color
15. Which type of reaction occurs during respiration ? 2
- ☐ Exothermic reaction
 - ☐ Endothermic reaction
 - ☐ Redox reaction
 - ☐ Both Exothermic and Redox reaction



16. Which of the following reaction occur during the burning of magnesium ? 2
- ☐ Magnesium undergoes reduction
 - ☐ Magnesium undergoes oxidation
 - ☐ A neutralization reaction occurs
 - ☐ A displacement reaction occurs
17. What is the primary gas required for the combustion of magnesium ? 2
- ☐ Nitrogen ☐ Oxygen ☐ Carbon dioxide ☐ Hydrogen
18. In the activity described why is magnesium ribbon held with tongs ? 2
- ☐ Because it is too small to hold with fingers
 - ☐ Because it reacts violently and is very hot when burnt
 - ☐ To avoid contamination
 - ☐ To ensure even burning
19. What is the color of the flame when magnesium burns ? 2
- ☐ Blue ☐ Red ☐ White ☐ Green
20. What kind of reaction is fermentation ? 2
- ☐ Physical ☐ Chemical ☐ Nuclear ☐ None of Above
21. Which of the following is an example of an irreversible reaction 2
- ☐ Dissolving salt in water
 - ☐ Evaporation of water
 - ☐ Cooking food
 - ☐ Freezing of water
22. What type of reaction takes place when iron rust ? 2
- ☐ Oxidation ☐ Neutralization ☐ Reduction ☐ Polymerization
23. What happens when a chemical reaction occurs ? 2
- ☐ The total mass changes
 - ☐ The number of atom decreases
 - ☐ The identity of the original substance changes
 - ☐ The substance undergoes only a state change
24. What is the purpose of adding an arrows in a word equation ? 2
- ☐ To separate reactants from products
 - ☐ To indicate the direction of the reaction
 - ☐ To show the type of chemical reaction
 - ☐ To represent the equation
25. Which change can help determine whether a chemical reaction has taken place ? 2
- ☐ Change in color ☐ Evolution of gas ☐ Change in state ☐ All of above



26. What is the product formed when zinc reacts with dilute sulphuric acid ? 2
☐ Hydrogen gas ☐ Magnesium Oxide ☐ Zinc oxide ☐ Lead nitrate
27. In the reaction between magnesium and oxygen, what are the reactants ? 2
☐ Magnesium and potassium iodide
☐ Zinc and dilute sulphuric acid
☐ Magnesium and oxygen
☐ Potassium iodide and lead nitrate
28. What is the purpose of using a word-equation to represent a chemical reaction ? 2
☐ To make it easier to read
☐ To show the direction of the reaction
☐ To separate reactants from products
☐ To simplify the equation
29. In the reaction between potassium iodide and lead nitrate, what is the product formed ? 2
☐ Hydrogen gas ☐ Magnesium oxide ☐ Lead iodide ☐ Zinc oxide
30. What is the significance of using a plus sign (+) between reactants in a word equation ? 2
☐ To indicate the type of chemical reaction
☐ To separate reactants from products
☐ To show the direction of the reaction
☐ To represent the equation
31. Why it is important to handle acid with care when performing an activity involving chemical reactions ? 2
☐ Because it can cause skin irritation
☐ Because it can release toxic fumes
☐ Both a and b
☐ Neither of Above
32. What is the difference between a word-equation and a chemical equation ? 2
☐ A word-equation uses symbols, while a chemical equation uses words.
☐ A word-equation shows reactants on the LHS, while a chemical equation shows products on the RHS
☐ There is no difference between the two
☐ A word equation represents physical change and chemical equation represents chemical change

