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Section: 01

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## Ion Chromatography – Anions in Seawater

	GRADING	Max	Points
1	Abstract		
2	Table summarizing important instrumental parameters with Instrumental set-up		
3	Calibration factor for pipette		
4	Table presenting retention times and instrumental sensitivity for single anion with ordered list of sensitivity for different anions		
5	Calibration curves for anions		
6	Table presenting best fit of calibration curves		
7	Chromatogram of sea-water sample		
8	Calculation of the concentration of each identified anion in seawater		
9	Chromatogram of Unknown		
10	Calculation of the concentration of the anion in the Unknown		
11	Table of comparison with gravimetric analysis		
12	Comparison with LEO-15		
13	Answer to questions		

## Abstract:

Through the experiment, the collected data yielded the calculations of the concentrations of the anions in the unknown sample to be 4.12626105ppm for Fluoride, 6.023264385ppm for Chloride, 23.9117666ppm for Phosphate, and 20.95579338ppm for Sulfate. Within the seawater samples, the average concentrations found within the two samples were 17244.85928ppm for Chloride and 2056.49385ppm for Sulfate.

## Introduction:

Understanding the concentration of each anion within the seawater is important to individuals to better understand seawater. The amount of research that can be done with the known concentrations of anions is gigantic. This information further allows individuals to be cautious in what the individuals do and how the individuals can filter the seawater to prepare the future of the environment for any crisis.

## Experimental Methods:

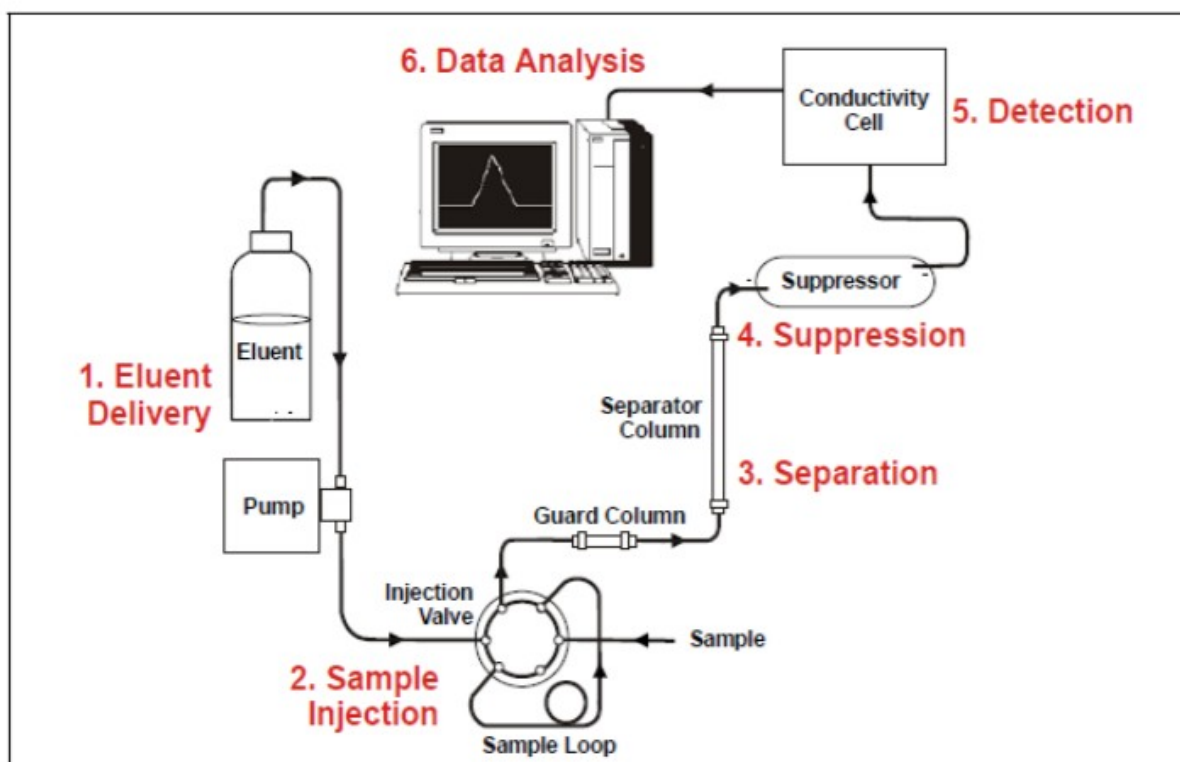


Table I: Various numerical components of Schroeder Machine:

Flow Rate	1.20 microL/min
Pump Pressure	~1500PSI
Eluent	HCO <sub>3</sub> <sup>-</sup> and CO <sub>3</sub> <sup>2-</sup>

Table I contains numerical values for machine flow rate, pump pressure, and the eluent.

The process this machine undergoes is first through the injection of the sample. After the injection of the sample, eluent made up of  $\text{HCO}_3^-$  and  $\text{CO}_3^{2-}$  goes through with the first guard column then goes through the separator column where it gets prepared to get pushed through the suppressor. The suppressor suppresses the conductivity of the sample and goes through the conductivity cell which leads to the machine giving a data analysis. Through that process, after the suppression, the suppressor has to regenerate using the excess eluent that is there.

## Results and Discussion:

Table II: Raw Collected Data

NaF mass:	.1106g
NaCl mass:	.082g
Na <sub>2</sub> SO <sub>4</sub> mass:	.0736g
NaH <sub>2</sub> PO <sub>4</sub> 2H <sub>2</sub> O mass:	.0346g
Pipette Calibration:	4.9995mL per 5mL
<b>ACPS Prepared Solution</b>	
NaF mass:	.0884g
NaCl mass:	.0991g
Na <sub>2</sub> SO <sub>4</sub> mass:	.3713g
NaH <sub>2</sub> PO <sub>4</sub> 2H <sub>2</sub> O mass:	.4348g

Table I contains the data for student prepared solutions, pipette calibration to deliver the seawater, and the stockroom prepared ACPS solution.

Table III: Prepared Standards' Calculated Data

	Area (microS*min)	Concentration (ppm)	Sensitivity (microS*min/ppm)	Retention Time (min)
Chloride	187.053	49.75	3.7599	4.007
Fluoride	132.804	50.02	2.655	2.670
Sulfate	118.593	49.77	2.3828	9.533
Phosphate	23.815	21.063	1.1307	7.497

Table III includes the area, concentration, sensitivity and retention times with the anions ranked from most sensitive to least sensitive.

Calculations:

**Fluoride:** Area:  $278.416 \times .477 = 132.804$  microS\*min

Concentration:  $.1106\text{g} \times 1 \text{ mol NaF} / 41.99\text{g NaF} \times 1 \text{ mol F}^- / 1 \text{ mol NaF} \times 18.99\text{g F}^- \times 1000\text{mg} / 1\text{g} = 50.02\text{ppm}$

Sensitivity:  $132.804 / 50.02 = 2.655$  microS\*min/ppm

**Chloride:** Area:  $473.553 \times .395 = 187.053$  microS\*min

Concentration:  $.082\text{g} \times 1 \text{ mol NaCl} / 58.44\text{g NaCl} \times 1 \text{ mol Cl}^- / 1 \text{ mol NaCl} \times 35.453\text{g Cl}^- \times 1000\text{mg} / 1\text{g} = 49.75\text{ppm}$

Sensitivity:  $187.053 / 49.75 = 3.7599 \text{ microS*min/ppm}$

**Phosphate:** Area:  $46.154 \times .516 = 23.815 \text{ microS*min}$

Concentration:  $.0346\text{g} \times 1 / 156.0076\text{g} \times 1 / 1 \times 94.9714\text{g} \times 1000\text{mg} / 1\text{g} = 21.063 \text{ ppm}$

Sensitivity:  $23.815 / 21.063 = 1.1307 \text{ microS*min/ppm}$

**Sulfate:** Area:  $132.506 \times .895 = 118.593 \text{ microS*min}$

Concentration:  $.0736\text{g} \times 1 / 142.04\text{g} \times 1 / 1 \times 96.06\text{g} \times 1000\text{mg} / 1\text{g} = 49.77\text{ppm}$

Sensitivity:  $118.593 / 49.77 = 2.3828 \text{ microS*min/ppm}$

Table IV: ACPS, Unknown, and Seawater Samples

<b>Fluoride</b>	Concentration (ppm)	Height( $\mu\text{S}$ )	Width(min)	Area ( $\mu\text{S*min}$ )	Retention Time (min)
ACPS 10-fold	2	4.564	0.178	0.812392	2.44
ACPS 4-fold	5	11.389	0.162	1.845018	2.44
ACPS undiluted	20	45.077	0.171	7.708167	2.457
<b>Chloride</b>	Concentration (ppm)	Height( $\mu\text{S}$ )	Width(min)	Area ( $\mu\text{S*min}$ )	Retention Time (min)
ACPS 10-fold	3.005	3.008	0.214	0.643712	3.833
ACPS 4-fold	7.5125	8.103	0.21	1.70163	3.827
ACPS undiluted	30.05	39.408	0.205	8.07864	3.837
<b>Phosphate</b>	Concentration (ppm)	Height( $\mu\text{S}$ )	Width(min)	Area ( $\mu\text{S*min}$ )	Retention Time (min)
ACPS 10-fold	14.963	2.267	0.462	1.047354	7.633
ACPS 4-fold	37.4075	6.298	0.455	2.86559	7.593
ACPS undiluted	149.63	29.592	0.479	14.174568	7.453
<b>Sulfate</b>	Concentration (ppm)	Height( $\mu\text{S}$ )	Width(min)	Area ( $\mu\text{S*min}$ )	Retention Time (min)
ACPS 10-fold	12.56	3.282	0.598	1.962636	10.057
ACPS 4-fold	31.4	9.004	0.591	5.321364	10.017
ACPS undiluted	125.6	40.704	0.632	25.724928	9.853
<b>Unknown</b>	Concentration	Height( $\mu\text{S}$ )	Width(min)	Area	Retention

	(ppm)	)	)	( $\mu\text{S} \cdot \text{min}$ )	Time (min)	
Fluoride	4.12626105	9.56	0.166	1.58696	2.44	
Chloride	6.023264385	7.182	0.223	1.601586	3.823	
Phosphate	23.9117666	4.887	0.457	2.233359	7.607	
Sulfate	20.95579338	7.149	0.593	4.239357	10.037	
Seawater 1	Concentration	Height( $\mu\text{S}$ )	Width(min)	Area ( $\mu\text{S} \cdot \text{min}$ )	Retention Time (min)	Original Concentration (ppm)
Fluoride	None					
Chloride	16.95235051	22.315	0.202	4.50763	3.823	16955.74166
Phosphate	None					
Sulfate	2.034602076	0.686	0.6	0.4116	10.09	2035.009078
Seawater 2	Concentration	Height( $\mu\text{S}$ )	Width(min)	Area ( $\mu\text{S} \cdot \text{min}$ )	Retention Time (min)	Original Concentration (ppm)
Fluoride	None					
Chloride	17.5304701	23.076	0.202	4.661352	3.817	17533.9769
Phosphate	None					
Sulfate	2.077563025	0.697	0.603	0.420291	10.103	2077.978621

Table IV contains concentration, height, width, area of peak, retention time for all ACPS, unknown, and seawater samples.

Calculations:

Pipette Calibration:

$$4.9995\text{mL} / 5\text{mL} = .9999 / 100 \times ((.999) \times 10) / 100 = .0009998 \text{ calibrated factor}$$

#### ACPS Calibrations:

$$\text{NaF: } .0884 \text{ g NaF} / 2\text{L} \times 1 \text{ mol NaF} / 41.99 \text{ g NaF} \times 1 \text{ mol F}^- / 1 \text{ mol NaF} \times 18.99 \text{ g F}^- \times 1000\text{mg} / 1\text{g} = 20.0\text{ppm}$$

$$10 \text{ fold dilution: } 20.0\text{ppm} / 10 = 2\text{ppm F}^-$$

$$4 \text{ fold dilution: } 20.0\text{ppm} / 4 = 5\text{ppm F}^-$$

$$\text{Area for no dilution: } 45.077 \times .171 = 7.708167 \mu\text{S} \cdot \text{min}$$

$$\text{Area for 10 fold dilution: } 4.564 \times .178 = .812392 \mu\text{S} \cdot \text{min}$$

$$\text{Area for 4 fold dilution: } 11.389 \times 0.162 = 1.845018 \mu\text{S} \cdot \text{min}$$

$$\text{NaCl: } .0991 \text{ g NaCl} / 2\text{L} \times 1 \text{ mol NaCl} / 58.44 \text{ g NaCl} \times 1 \text{ mol Cl}^- / 1 \text{ mol NaCl} \times 35.453 \text{ g Cl}^- \times 1000\text{mg} / 1\text{g} = 30.050\text{ppm Cl}^-$$

$$10 \text{ fold dilution: } 30.050\text{ppm Cl}^- / 10 = 3.0050\text{ppm Cl}^-$$

$$4 \text{ fold dilution: } 30.050\text{ppm Cl}^- / 4 = 7.5125$$

Area for no dilution:  $39.408 \times .205 = 8.07864 \mu\text{S} \cdot \text{min}$

Area for 10 fold dilution:  $3.008 \times .214 = .643712 \mu\text{S} \cdot \text{min}$

Area for 4 fold dilution:  $8.103 \times .21 = 1.70163 \mu\text{S} \cdot \text{min}$

**NaH<sub>2</sub>PO<sub>4</sub> x 1H<sub>2</sub>O:**  $.4348\text{g} / 2\text{L} \times 1 / 137.99\text{g} \times 1 / 1 \times 94.9714\text{g} \times 1000 = 149.63\text{ppm}$

10 fold dilution:  $149.63\text{ppm} / 10 = 14.963\text{ppm PO}_4^{3-}$

4 fold dilution:  $149.63\text{ppm} / 4 = 37.4075\text{ppm PO}_4^{3-}$

Area for no dilution:  $29.592 \times .479 = 14.174568 \mu\text{S} \cdot \text{min}$

Area for 10 fold dilution:  $2.267 \times .462 = 1.047354 \mu\text{S} \cdot \text{min}$

Area for 4 fold dilution:  $6.298 \times .455 = 2.86559 \mu\text{S} \cdot \text{min}$

**Na<sub>2</sub>SO<sub>4</sub>:**  $.3173\text{g} / 2\text{L} \times 1 / 142.04\text{g} \times 1 / 1 \times 96.06\text{g} \times 1000 = 125.6\text{ppm}$

10 fold dilution:  $125.6\text{ppm} / 10 = 12.56\text{ppm SO}_4^{2-}$

4 fold dilution:  $125.6\text{ppm} / 4 = 31.4\text{ppm SO}_4^{2-}$

Area for no dilution:  $40.704 \times .632 = 25.724928 \mu\text{S} \cdot \text{min}$

Area for 10 fold dilution:  $3.282 \times .598 = 1.962636 \mu\text{S} \cdot \text{min}$

Area for 4 fold dilution:  $9.004 \times .591 = 5.321364 \mu\text{S} \cdot \text{min}$

### **Unknown Solution #2:**

**Fluoride:** Area =  $9.56 \times .166 = 1.58696 \mu\text{S} \cdot \text{min}$

Best fit equation:  $y = 0.3846x$ ;  $x = \text{concentration}$

$1.58696 / .3846 = 4.12626105\text{ppm}$

**Chloride:** Area:  $7.182 \times .223 = 1.601586 \mu\text{S} \cdot \text{min}$

Best fit equation:  $y = 0.2659x$

$1.601586 / .2659 = 6.023264385\text{ppm}$

**Phosphate:** Area:  $4.887 \times .457 = 2.233359 \mu\text{S} \cdot \text{min}$

Best fit equation:  $y = 0.0934x$

$2.233359 / .0934 = 23.9117666\text{ppm}$

**Sulfate:** Area:  $7.149 \times .593 = 4.239357 \mu\text{S} \cdot \text{min}$

Best fit equation:  $y = 0.2023x$

$4.239357 / .2023 = 20.95579338\text{ppm}$

### Seawater 1:

Chloride: Area:  $22.315 \times .202 = 4.50763 \mu\text{S} \cdot \text{min}$

Original Chloride:  $4.50763 / .2659 = 16.95235051 \text{ ppm} / .0009998 = 16955.74166 \text{ ppm}$

Sulfate:  $.686 \times .6 = .4116 \mu\text{S} \cdot \text{min}$

Original Sulfate:  $.4116 / .2023 = 2.034602076 \text{ ppm} / .0009998 = 2035.009078 \text{ ppm}$

### Seawater 2:

Chloride:  $23.076 \times .202 = 4.661352 \mu\text{S} \cdot \text{min}$

Original Chloride:  $4.661352 / .2659 = 17.5304701 \text{ ppm} / .0009998 = 17533.9769 \text{ ppm}$

Sulfate:  $.697 \times .420291 = 2.077563025 \mu\text{S} \cdot \text{min}$

Original Sulfate:  $2.077563025 / .2023 = 2.077563025 \text{ ppm} / .0009998 = 2077.978621 \text{ ppm}$

### Graphs and Chromatograms:

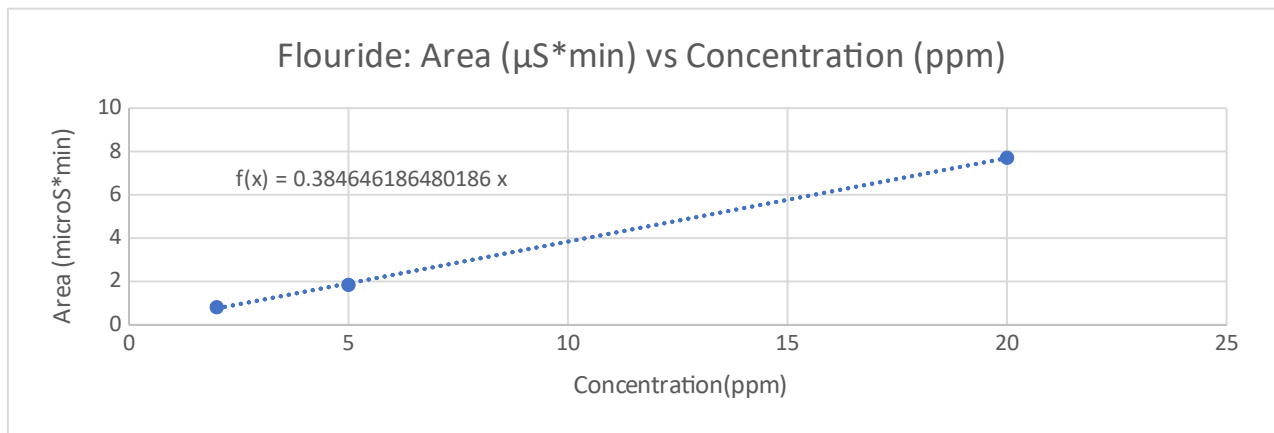


Figure 1: Fluoride ACPS Graph

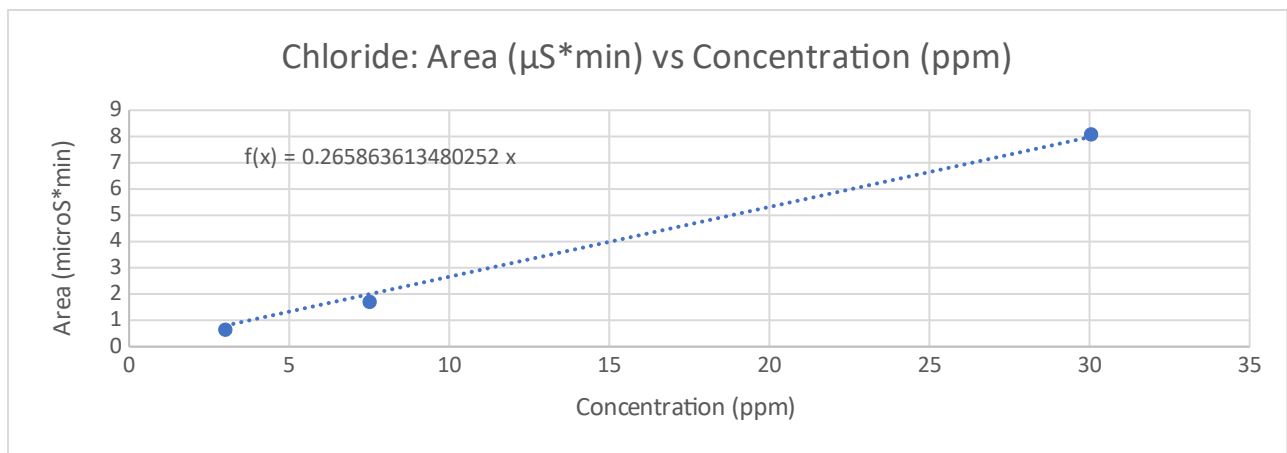


Figure 2: Chloride ACPS Graph

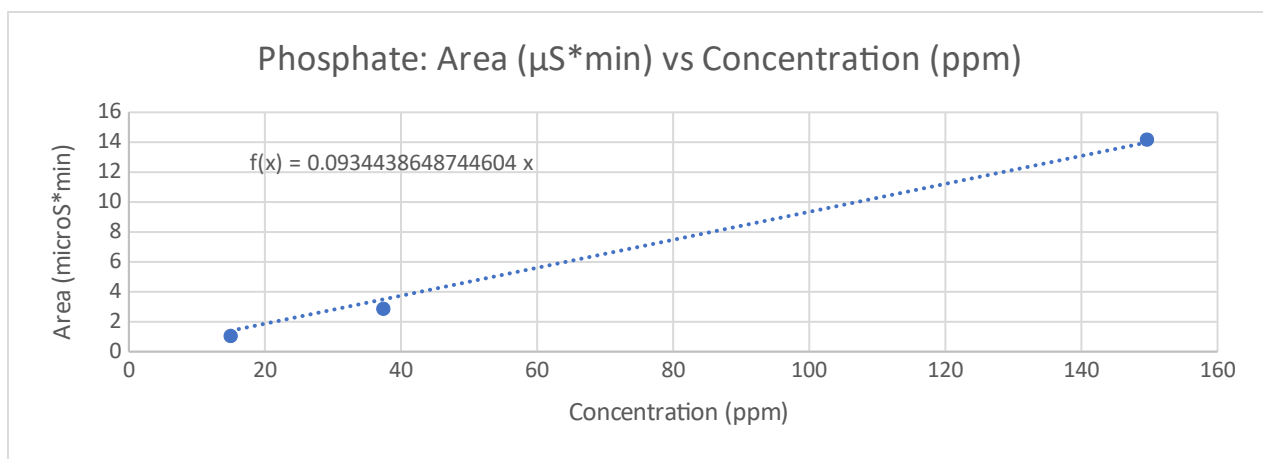


Figure 3: Phosphate ACPS Graph

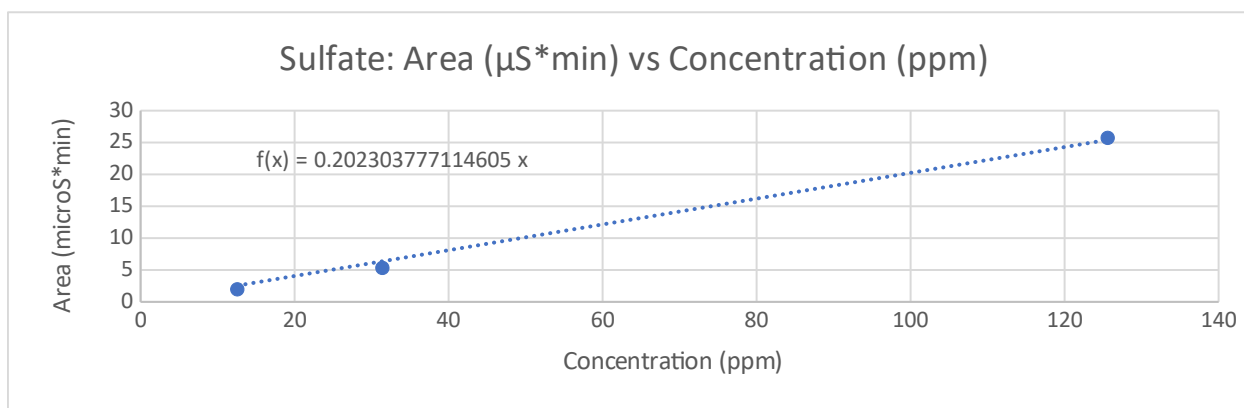


Figure 4: Sulfate ACPS Graph

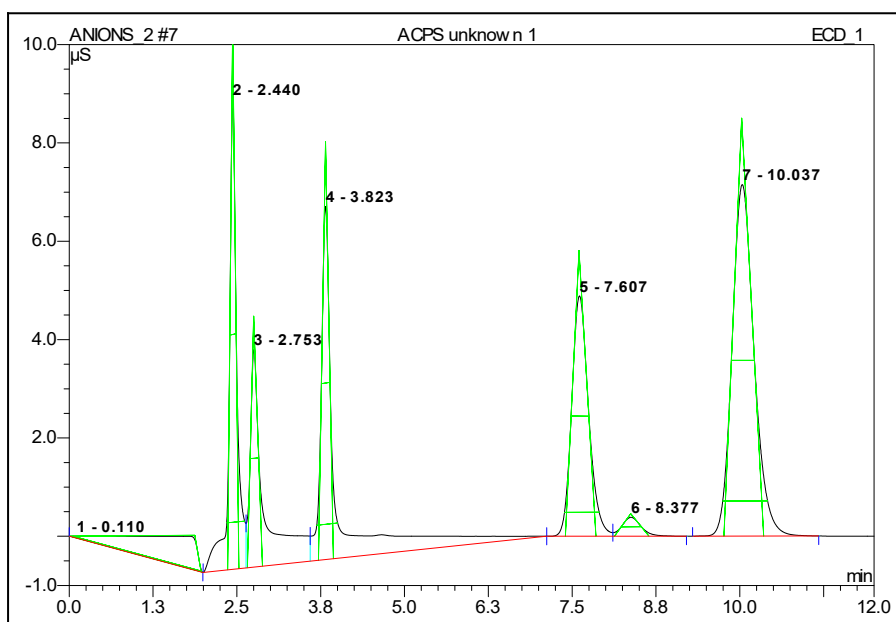


Figure 5: Unknown Chromatogram



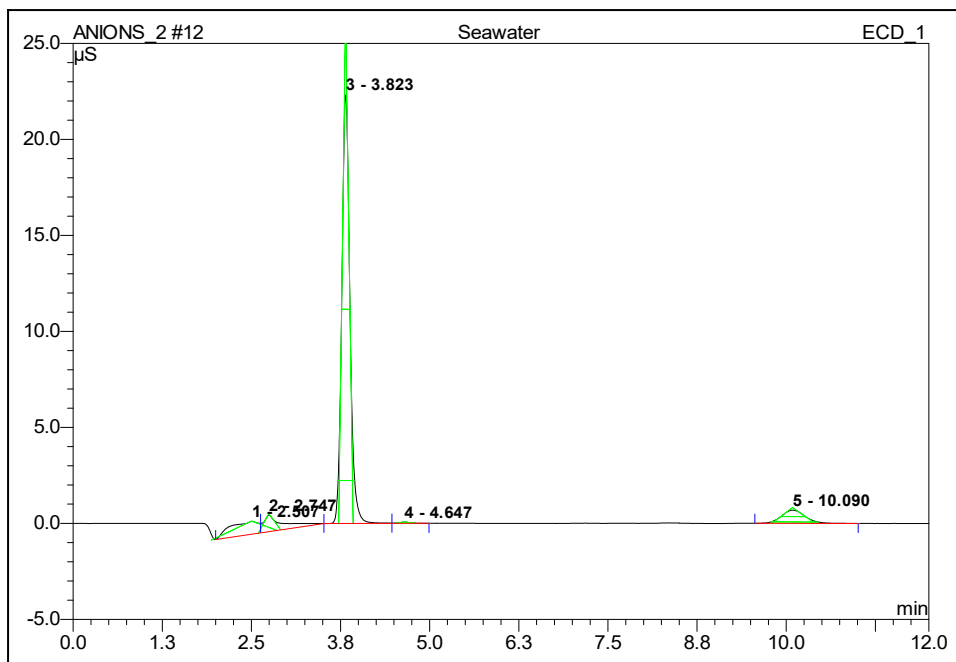


Figure 6: Seawater 1 Chromatogram

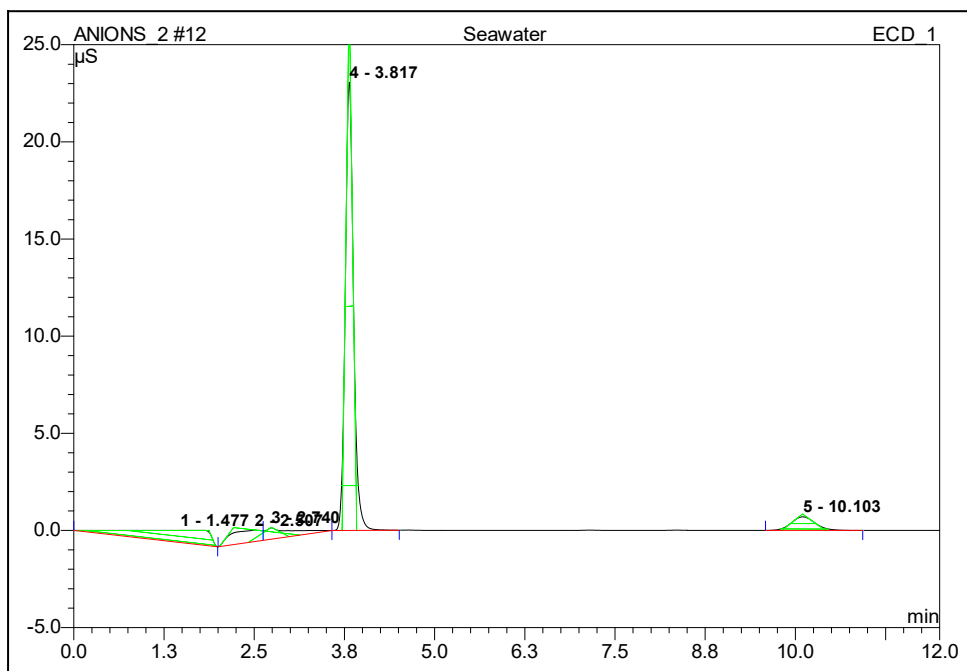


Figure 7: Seawater 2 Chromatogram

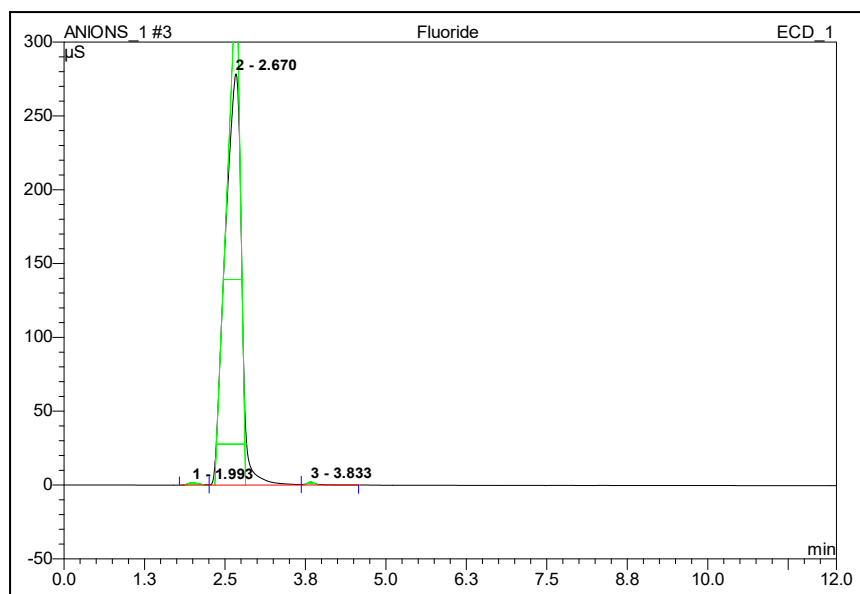


Figure 8: Fluoride Chromatogram

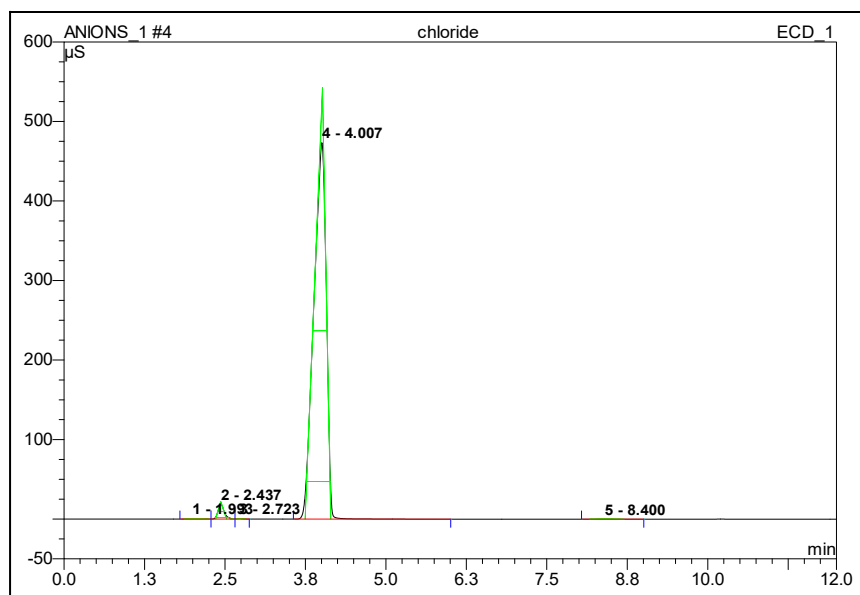


Figure 9: Chloride Chromatogram

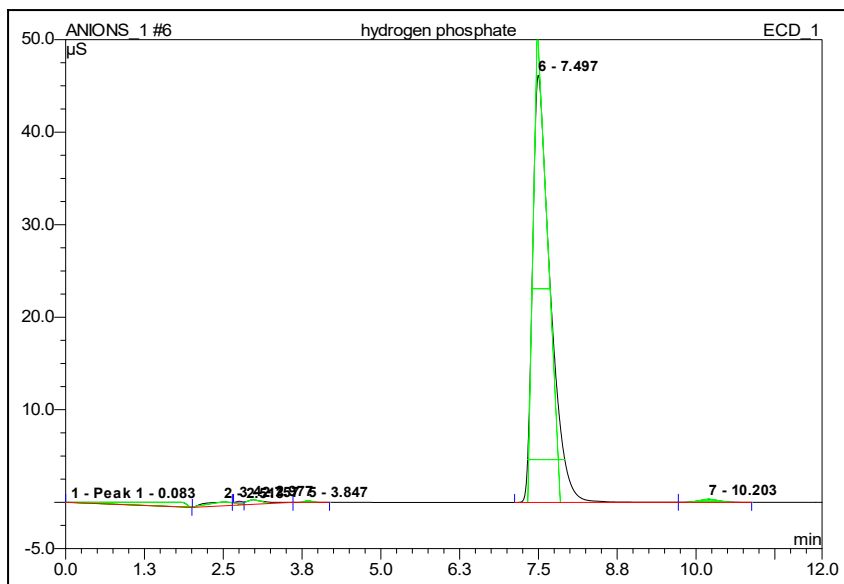


Figure 10: Phosphate Chromatogram

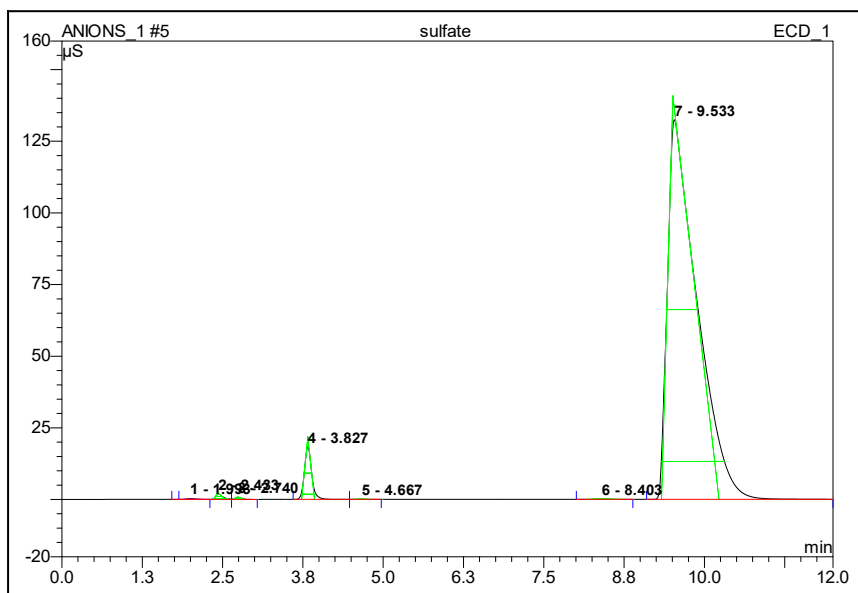


Figure 11: Sulfate Chromatogram

Compared to the gravimetric analysis lab, the average amount of chloride was 18890ppm to 17244.85928ppm, which is about the same result. For the sulfate amount, the average amount of sulfate was 2369.64ppm to 2056.49385ppm, which is also about the same result from the lab. Meanwhile, compared to the LEO-15, Chloride is 18.980ppt and this lab reports it to be 17244.86ppt and sulfate to be 2560ppt and this lab states it to be 2369ppt.

Lab Questions:

- 1) A blank is used to correct instrumental error.
- 2) Detector works by measuring conductivity of the solution.

- 3) The solution that is going to be made is going to be diluted by the deionized water.
- 4) The factor of dilution is 1000 before pipette calibration.