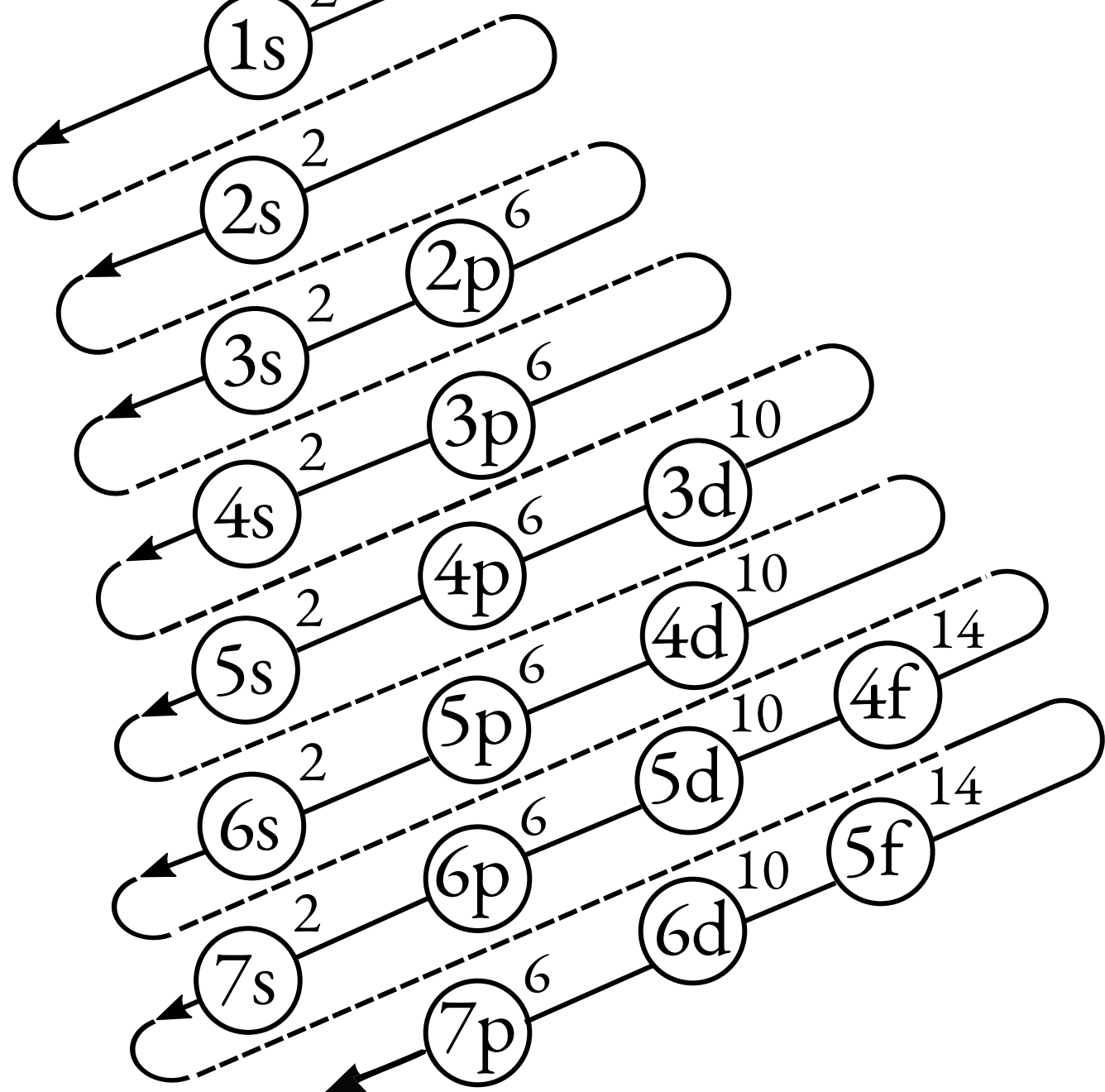
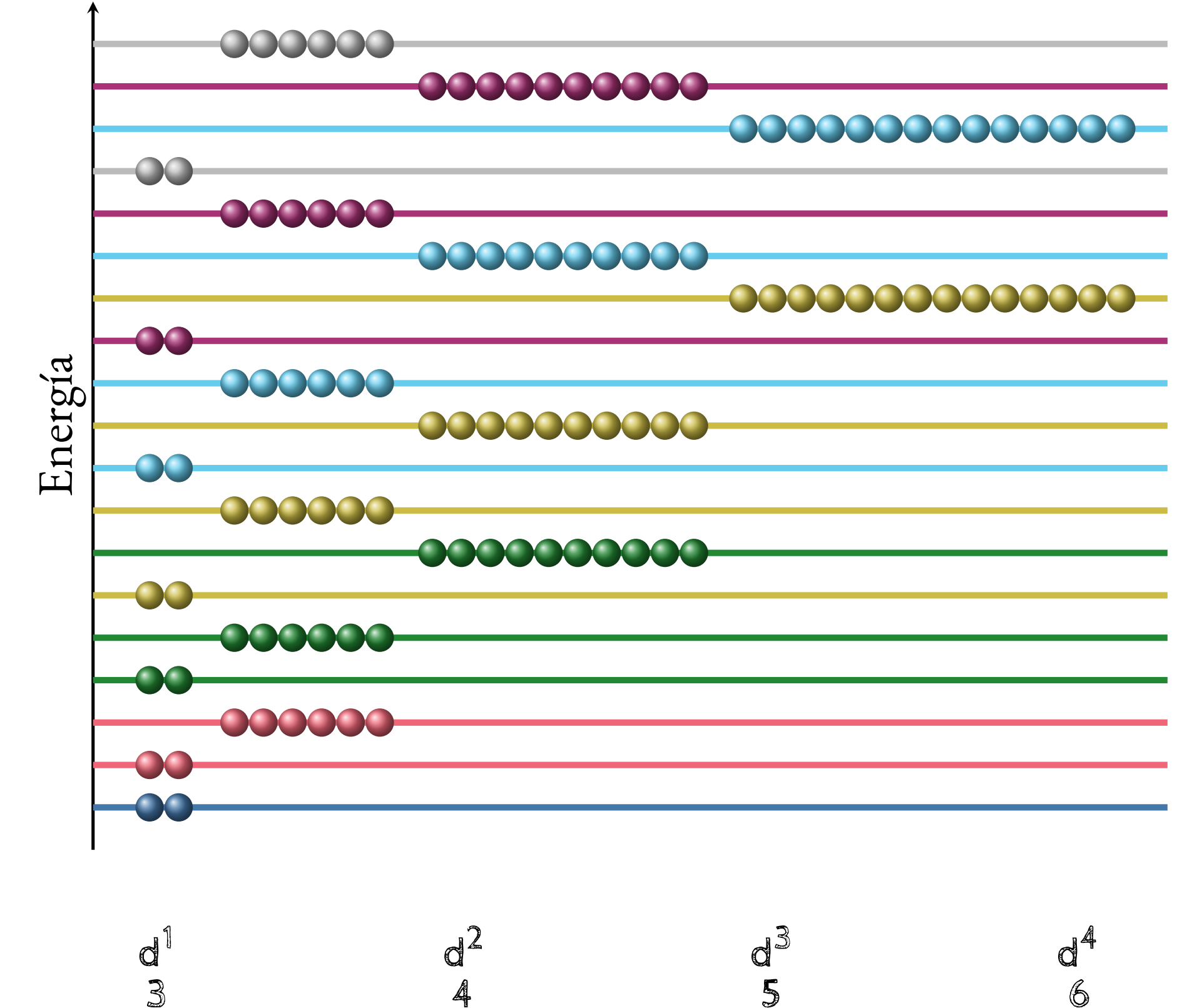


1	1.0079		
<b>H</b>	$1s^1$		
Hidrógeno			
3	6.941	4	9.0122
<b>Li</b>	$[\text{He}] 2s^1$	<b>Be</b>	$[\text{He}] 2s^2$
Litio		Berilio	
11	22.990	12	24.305
<b>Na</b>	$[\text{Ne}] 3s^1$	<b>Mg</b>	$[\text{Ne}] 3s^2$
Sodio		Magnesio	
19	39.098	20	40.078
<b>K</b>	$[\text{Ar}] 4s^1$	<b>Ca</b>	$[\text{Ar}] 4s^2$
Potasio		Calcio	
37	85.468	38	87.62
<b>Rb</b>	$[\text{Kr}] 5s^1$	<b>Sr</b>	$[\text{Kr}] 5s^2$
Rubidio		Estroncio	
55	132.91	56	137.33
<b>Cs</b>	$[\text{Xe}] 6s^1$	<b>Ba</b>	$[\text{Xe}] 6s^2$
Cesio		Bario	
87	223	88	226
<b>Fr</b>	$[\text{Rn}] 7s^1$	<b>Ra</b>	$[\text{Rn}] 7s^2$
Francio		Radio	



Z

Masa

Símbolo

Configuración electrónica

Nombre

					<div>p<sup>6</sup> 18</div>				
					2 <div>4.0025</div> <div>He</div> <div>2s<sup>2</sup></div> <div>Helio</div>				
<div>p<sup>1</sup> 13</div>		<div>p<sup>2</sup> 14</div>		<div>p<sup>3</sup> 15</div>		<div>p<sup>4</sup> 16</div>		<div>p<sup>5</sup> 17</div>	
5 <div>10.811</div> <div>B</div> <div>[He] 2s<sup>2</sup>2p<sup>1</sup></div> <div>Boro</div>	6 <div>12.011</div> <div>C</div> <div>[He] 2s<sup>2</sup>2p<sup>2</sup></div> <div>Carbono</div>	7 <div>14.007</div> <div>N</div> <div>[He] 2s<sup>2</sup>2p<sup>3</sup></div> <div>Nitrógeno</div>	8 <div>15.999</div> <div>O</div> <div>[He] 2s<sup>2</sup>2p<sup>4</sup></div> <div>Oxígeno</div>	9 <div>18.998</div> <div>F</div> <div>[He] 2s<sup>2</sup>2p<sup>5</sup></div> <div>Flúor</div>	10 <div>20.180</div> <div>Ne</div> <div>[He] 2s<sup>2</sup>2p<sup>6</sup></div> <div>Neón</div>				
13 <div>26.982</div> <div>Al</div> <div>[Ne] 3s<sup>2</sup>3p<sup>1</sup></div> <div>Aluminio</div>	14 <div>28.086</div> <div>Si</div> <div>[Ne] 3s<sup>2</sup>3p<sup>2</sup></div> <div>Silicio</div>	15 <div>30.974</div> <div>P</div> <div>[Ne] 3s<sup>2</sup>3p<sup>2</sup></div> <div>Fósforo</div>	16 <div>32.065</div> <div>S</div> <div>[Ne] 3s<sup>2</sup>3p<sup>4</sup></div> <div>Azufre</div>	17 <div>35.453</div> <div>Cl</div> <div>[Ne] 3s<sup>2</sup>3p<sup>5</sup></div> <div>Cloro</div>	18 <div>39.948</div> <div>Ar</div> <div>[Ne] 3s<sup>2</sup>3p<sup>6</sup></div> <div>Argón</div>				
31 <div>69.723</div> <div>Ga</div> <div>[Ar] 4s<sup>2</sup>3d<sup>10</sup>4p<sup>1</sup></div> <div>Galio</div>	32 <div>72.64</div> <div>Ge</div> <div>[Ar] 4s<sup>2</sup>3d<sup>10</sup>4p<sup>2</sup></div> <div>Germanio</div>	33 <div>74.922</div> <div>As</div> <div>[Ar] 4s<sup>2</sup>3d<sup>10</sup>4p<sup>3</sup></div> <div>Arsénico</div>	34 <div>78.96</div> <div>Se</div> <div>[Ar] 4s<sup>2</sup>3d<sup>10</sup>4p<sup>4</sup></div> <div>Selenio</div>	35 <div>79.904</div> <div>Br</div> <div>[Ar] 4s<sup>2</sup>3d<sup>10</sup>4p<sup>5</sup></div> <div>Bromo</div>	36 <div>83.8</div> <div>Kr</div> <div>[Ar] 4s<sup>2</sup>3d<sup>10</sup>4p<sup>6</sup></div> <div>Kriptón</div>				
49 <div>114.82</div> <div>In</div> <div>[Kr] 5s<sup>2</sup>4d<sup>10</sup>5p<sup>1</sup></div> <div>Indio</div>	50 <div>118.71</div> <div>Sn</div> <div>[Kr] 5s<sup>2</sup>4d<sup>10</sup>5p<sup>2</sup></div> <div>Estaño</div>	51 <div>121.76</div> <div>Sb</div> <div>[Kr] 5s<sup>2</sup>4d<sup>10</sup>5p<sup>3</sup></div> <div>Antimonio</div>	52 <div>127.6</div> <div>Te</div> <div>[Kr] 5s<sup>2</sup>4d<sup>10</sup>5p<sup>4</sup></div> <div>Telurio</div>	53 <div>126.9</div> <div>I</div> <div>[Kr] 5s<sup>2</sup>4d<sup>10</sup>5p<sup>5</sup></div> <div>Iodo</div>	54 <div>131.29</div> <div>Xe</div> <div>[Kr] 5s<sup>2</sup>4d<sup>10</sup>5p<sup>6</sup></div> <div>Xenón</div>				
81 <div>204.38</div> <div>Tl</div> <div>[Xe] 6s<sup>2</sup>4f<sup>14</sup> 5d<sup>10</sup>6p<sup>1</sup></div> <div>Talio</div>	82 <div>207.2</div> <div>Pb</div> <div>[Xe] 6s<sup>2</sup>4f<sup>14</sup> 5d<sup>10</sup>6p<sup>2</sup></div> <div>Plomo</div>	83 <div>208.98</div> <div>Bi</div> <div>[Xe] 6s<sup>2</sup>4f<sup>14</sup> 5d<sup>10</sup>6p<sup>3</sup></div> <div>Bismuto</div>	84 <div>209</div> <div>Po</div> <div>[Xe] 6s<sup>2</sup>4f<sup>14</sup> 5d<sup>10</sup>6p<sup>4</sup></div> <div>Polonio</div>	85 <div>210</div> <div>At</div> <div>[Xe] 6s<sup>2</sup>4f<sup>14</sup> 5d<sup>10</sup>6p<sup>5</sup></div> <div>Ástato</div>	86 <div>222</div> <div>Rn</div> <div>[Xe] 6s<sup>2</sup>4f<sup>14</sup> 5d<sup>10</sup>6p<sup>6</sup></div> <div>Radón</div>				
113 <div>284</div> <div>Nh</div> <div>[Rn] 7s<sup>2</sup>5f<sup>14</sup> 6d<sup>10</sup>7p<sup>1</sup></div> <div>Nihonio</div>	114 <div>289</div> <div>Fl</div> <div>[Rn] 7s<sup>2</sup>5f<sup>14</sup> 6d<sup>10</sup>7p<sup>2</sup></div> <div>Flerovio</div>	115 <div>288</div> <div>Mc</div> <div>[Rn] 7s<sup>2</sup>5f<sup>14</sup> 6d<sup>10</sup>7p<sup>3</sup></div> <div>Moscovio</div>	116 <div>293</div> <div>Lv</div> <div>[Rn] 7s<sup>2</sup>5f<sup>14</sup> 6d<sup>10</sup>7p<sup>4</sup></div> <div>Livermorio</div>	117 <div>292</div> <div>Ts</div> <div>[Rn] 7s<sup>2</sup>5f<sup>14</sup> 6d<sup>10</sup>7p<sup>5</sup></div> <div>Teneso</div>	118 <div>294</div> <div>Og</div> <div>[Rn] 7s<sup>2</sup>5f<sup>14</sup> 6d<sup>10</sup>7p<sup>6</sup></div> <div>Oganesón</div>				

- BLOQUE S
- BLOQUE P
- BLOQUE D
- BLOQUE F

$f^1$	$f^2$	$f^3$	$f^4$	$f^5$	$f^6$	$f^7$	$f^8$	$f^9$	$f^{10}$	$f^{11}$	$f^{12}$	$f^{13}$	$f^{14}$
57138.91	58140.12	59140.91	60144.24	61145	62150.36	63151.96	64157.25	65158.93	66162.50	67164.93	68167.26	69168.93	70173.04
La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb
$[\text{Xe}]\,6s^25d^1$	$[\text{Xe}]\,6s^24f^15d^1$	$[\text{Xe}]\,6s^24f^3$	$[\text{Xe}]\,6s^24f^4$	$[\text{Xe}]\,6s^24f^5$	$[\text{Xe}]\,6s^24f^6$	$[\text{Xe}]\,6s^24f^7$	$[\text{Xe}]\,6s^24f^75d^1$	$[\text{Xe}]\,6s^24f^9$	$[\text{Xe}]\,6s^24f^{10}$	$[\text{Xe}]\,6s^24f^{11}$	$[\text{Xe}]\,6s^24f^{12}$	$[\text{Xe}]\,6s^24f^{13}$	$[\text{Xe}]\,6s^24f^{14}$
Lantano	Cerio	Praseodimio	Neodimio	Prometio	Samario	Europio	Gadolinio	Terbio	Disprosio	Holmio	Erbio	Tulio	Yterbio
89227	90232.04	91231.04	92238.03	93237	94244	95243	96247	97247	98251	99252	100257	101258	102259
Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No
$[\text{Rn}]\,7s^26d^1$	$[\text{Rn}]\,7s^26d^2$	$[\text{Rn}]\,7s^25f^26d^1$	$[\text{Rn}]\,7s^25f^36d^1$	$[\text{Rn}]\,7s^25f^46d^1$	$[\text{Rn}]\,7s^25f^6$	$[\text{Rn}]\,7s^25f^7$	$[\text{Rn}]\,7s^25f^76d^1$	$[\text{Rn}]\,7s^25f^9$	$[\text{Rn}]\,7s^25f^{10}$	$[\text{Rn}]\,7s^25f^{11}$	$[\text{Rn}]\,7s^25f^{12}$	$[\text{Rn}]\,7s^25f^{13}$	$[\text{Rn}]\,7s^25f^{14}$
Actinio	Torio	Protactinio	Uranio	Neptunio	Plutonio	Americio	Curio	Berkelio	Californio	Einsteinio	Fermio	Mendelevio	Nobelio