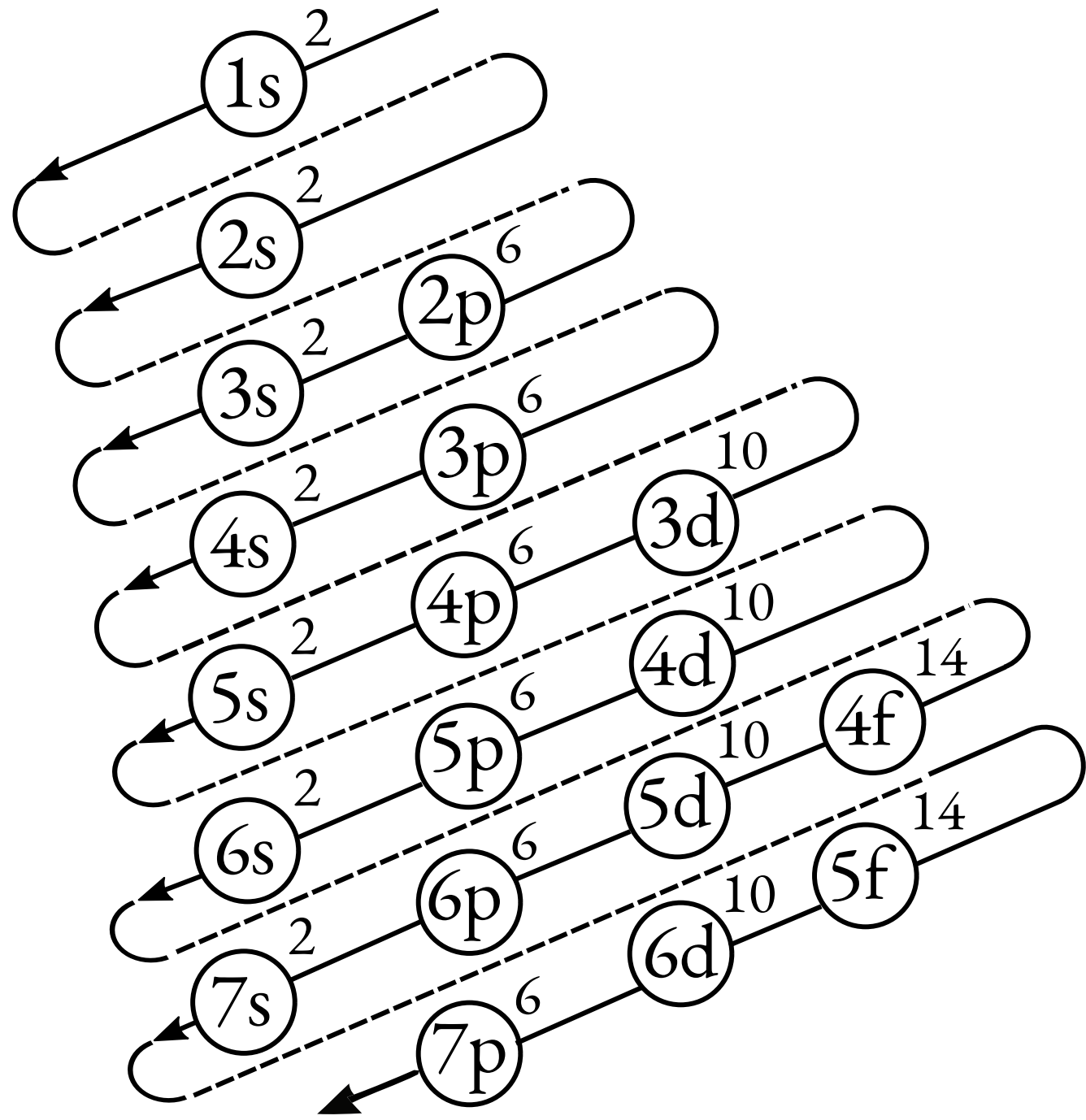
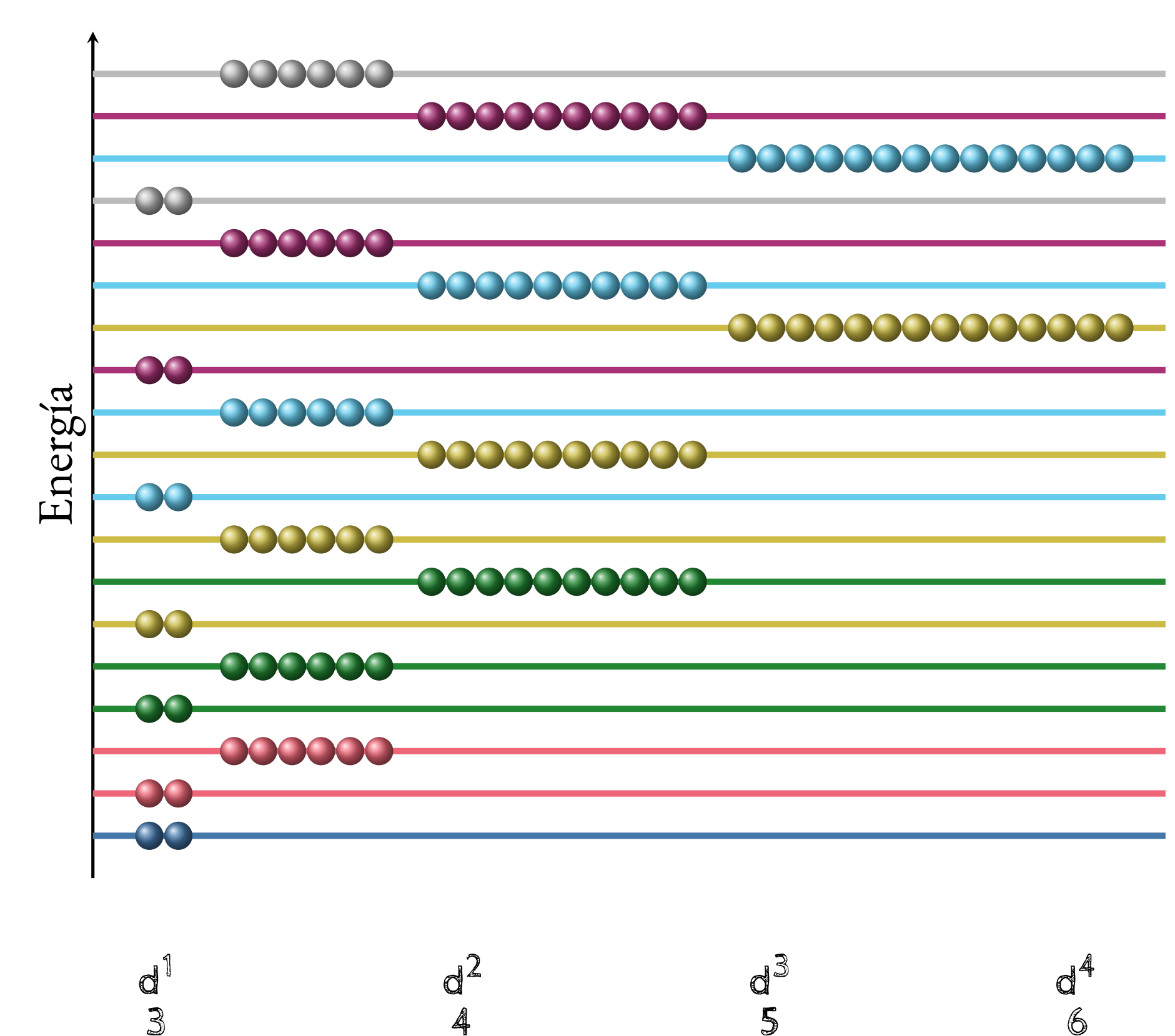


|           |                    |           |                    |
|-----------|--------------------|-----------|--------------------|
| 1         | 1.0079             |           |                    |
| <b>H</b>  | $1s^1$             |           |                    |
| Hidrógeno |                    |           |                    |
| 3         | 6.941              | 4         | 9.0122             |
| <b>Li</b> | $[\text{He}] 2s^1$ | <b>Be</b> | $[\text{He}] 2s^2$ |
| Litio     |                    | Berilio   |                    |
| 11        | 22.990             | 12        | 24.305             |
| <b>Na</b> | $[\text{Ne}] 3s^1$ | <b>Mg</b> | $[\text{Ne}] 3s^2$ |
| Sodio     |                    | Magnesio  |                    |
| 19        | 39.098             | 20        | 40.078             |
| <b>K</b>  | $[\text{Ar}] 4s^1$ | <b>Ca</b> | $[\text{Ar}] 4s^2$ |
| Potasio   |                    | Calcio    |                    |
| 37        | 85.468             | 38        | 87.62              |
| <b>Rb</b> | $[\text{Kr}] 5s^1$ | <b>Sr</b> | $[\text{Kr}] 5s^2$ |
| Rubidio   |                    | Estroncio |                    |
| 55        | 132.91             | 56        | 137.33             |
| <b>Cs</b> | $[\text{Xe}] 6s^1$ | <b>Ba</b> | $[\text{Xe}] 6s^2$ |
| Cesio     |                    | Bario     |                    |
| 87        | 223                | 88        | 226                |
| <b>Fr</b> | $[\text{Rn}] 7s^1$ | <b>Ra</b> | $[\text{Rn}] 7s^2$ |
| Francio   |                    | Radio     |                    |



**Z** Masa

**Símbolo**

**Configuración electrónica**

**Nombre**

|   |        |   |        |   | <p><sup>6</sup>p<br/>18</p> |   |        |   |        |  |        |
|---|--------|---|--------|---|-----------------------------|---|--------|---|--------|--|--------|
| <p><sup>1</sup>p<br/>13</p>   |        | <p><sup>2</sup>p<br/>14</p>   |        | <p><sup>3</sup>p<br/>15</p>   |                             | <p><sup>4</sup>p<br/>16</p>   |        | <p><sup>5</sup>p<br/>17</p>   |        | <p>2                      4.0025</p> <p><b>He</b></p> <p>2s<sup>2</sup></p> <p>Helio</p> |        |
| 5   | 10.811 | 6   | 12.011 | 7   | 14.007                      | 8   | 15.999 | 9   | 18.998 | 10   | 20.180 |
| <b>B</b>  |        | <b>C</b>  |        | <b>N</b>  |                             | <b>O</b>  |        | <b>F</b>  |        | <b>Ne</b>  |        |
| [He] 2s <sup>2</sup> 2p <sup>1</sup>                                      |        | [He] 2s <sup>2</sup> 2p <sup>2</sup>                                      |        | [He] 2s <sup>2</sup> 2p <sup>3</sup>                                      |                             | [He] 2s <sup>2</sup> 2p <sup>4</sup>                                      |        | [He] 2s <sup>2</sup> 2p <sup>5</sup>                                      |        | [He] 2s <sup>2</sup> 2p <sup>6</sup>   |        |
| Boro  |        | Carbono   |        | Nitrógeno   |                             | Oxígeno   |        | Flúor   |        | Neón   |        |
| 13  | 26.982 | 14  | 28.086 | 15  | 30.974                      | 16  | 32.065 | 17  | 35.453 | 18   | 39.948 |
| <b>Al</b>   |        | <b>Si</b>   |        | <b>P</b>  |                             | <b>S</b>  |        | <b>Cl</b>   |        | <b>Ar</b>  |        |
| [Ne] 3s <sup>2</sup> 3p <sup>1</sup>                                      |        | [Ne] 3s <sup>2</sup> 3p <sup>2</sup>                                      |        | [Ne] 3s <sup>2</sup> 3p <sup>2</sup>                                      |                             | [Ne] 3s <sup>2</sup> 3p <sup>4</sup>                                      |        | [Ne] 3s <sup>2</sup> 3p <sup>5</sup>                                      |        | [Ne] 3s <sup>2</sup> 3p <sup>6</sup>   |        |
| Aluminio  |        | Silicio   |        | Fósforo   |                             | Azufre  |        | Cloro   |        | Argón  |        |
| 31  | 69.723 | 32  | 72.64  | 33  | 74.922                      | 34  | 78.96  | 35  | 79.904 | 36   | 83.8   |
| <b>Ga</b>   |        | <b>Ge</b>   |        | <b>As</b>   |                             | <b>Se</b>   |        | <b>Br</b>   |        | <b>Kr</b>  |        |
| [Ar] 4s <sup>2</sup> 3d <sup>10</sup> 4p <sup>1</sup>                     |        | [Ar] 4s <sup>2</sup> 3d <sup>10</sup> 4p <sup>2</sup>                     |        | [Ar] 4s <sup>2</sup> 3d <sup>10</sup> 4p <sup>3</sup>                     |                             | [Ar] 4s <sup>2</sup> 3d <sup>10</sup> 4p <sup>4</sup>                     |        | [Ar] 4s <sup>2</sup> 3d <sup>10</sup> 4p <sup>5</sup>                     |        | [Ar] 4s <sup>2</sup> 3d <sup>10</sup> 4p <sup>6</sup>                                    |        |
| Galio   |        | Germanio  |        | Arsénico  |                             | Selenio   |        | Bromo   |        | Kriptón  |        |
| 49  | 114.82 | 50  | 118.71 | 51  | 121.76                      | 52  | 127.6  | 53  | 126.9  | 54   | 131.29 |
| <b>In</b>   |        | <b>Sn</b>   |        | <b>Sb</b>   |                             | <b>Te</b>   |        | <b>I</b>  |        | <b>Xe</b>  |        |
| [Kr] 5s <sup>2</sup> 4d <sup>10</sup> 5p <sup>1</sup>                     |        | [Kr] 5s <sup>2</sup> 4d <sup>10</sup> 5p <sup>2</sup>                     |        | [Kr] 5s <sup>2</sup> 4d <sup>10</sup> 5p <sup>3</sup>                     |                             | [Kr] 5s <sup>2</sup> 4d <sup>10</sup> 5p <sup>4</sup>                     |        | [Kr] 5s <sup>2</sup> 4d <sup>10</sup> 5p <sup>5</sup>                     |        | [Kr] 5s <sup>2</sup> 4d <sup>10</sup> 5p <sup>6</sup>                                    |        |
| Indio   |        | Estaño  |        | Antimonio   |                             | Telurio   |        | Iodo  |        | Xenón  |        |
| 81  | 204.38 | 82  | 207.2  | 83  | 208.98                      | 84  | 209    | 85  | 210    | 86   | 222    |
| <b>Tl</b>   |        | <b>Pb</b>   |        | <b>Bi</b>   |                             | <b>Po</b>   |        | <b>At</b>   |        | <b>Rn</b>  |        |
| [Xe] 6s <sup>2</sup> 4f <sup>14</sup><br>5d <sup>10</sup> 6p <sup>1</sup> |        | [Xe] 6s <sup>2</sup> 4f <sup>14</sup><br>5d <sup>10</sup> 6p <sup>2</sup> |        | [Xe] 6s <sup>2</sup> 4f <sup>14</sup><br>5d <sup>10</sup> 6p <sup>3</sup> |                             | [Xe] 6s <sup>2</sup> 4f <sup>14</sup><br>5d <sup>10</sup> 6p <sup>4</sup> |        | [Xe] 6s <sup>2</sup> 4f <sup>14</sup><br>5d <sup>10</sup> 6p <sup>5</sup> |        | [Xe] 6s <sup>2</sup> 4f <sup>14</sup><br>5d <sup>10</sup> 6p <sup>6</sup>                |        |
| Talio   |        | Plomo   |        | Bismuto   |                             | Polonio   |        | Ástato  |        | Radón  |        |
| 113   | 284    | 114   | 289    | 115   | 288                         | 116   | 293    | 117   | 292    | 118  | 294    |
| <b>Nh</b>   |        | <b>Fl</b>   |        | <b>Mc</b>   |                             | <b>Lv</b>   |        | <b>Ts</b>   |        | <b>Og</b>  |        |
| [Rn] 7s <sup>2</sup> 5f <sup>14</sup><br>6d <sup>10</sup> 7p <sup>1</sup> |        | [Rn] 7s <sup>2</sup> 5f <sup>14</sup><br>6d <sup>10</sup> 7p <sup>2</sup> |        | [Rn] 7s <sup>2</sup> 5f <sup>14</sup><br>6d <sup>10</sup> 7p <sup>3</sup> |                             | [Rn] 7s <sup>2</sup> 5f <sup>14</sup><br>6d <sup>10</sup> 7p <sup>4</sup> |        | [Rn] 7s <sup>2</sup> 5f <sup>14</sup><br>6d <sup>10</sup> 7p <sup>5</sup> |        | [Rn] 7s <sup>2</sup> 5f <sup>14</sup><br>6d <sup>10</sup> 7p <sup>6</sup>                |        |
| Nihonio   |        | Flerovio  |        | Moscovio  |                             | Livermorio  |        | Teneso  |        | Oganesón   |        |

- BLOQUE S
- BLOQUE P
- BLOQUE D
- BLOQUE F

| $f^1$                   | $f^2$                        | $f^3$                        | $f^4$                        | $f^5$                        | $f^6$                   | $f^7$                   | $f^8$                        | $f^9$                   | $f^{10}$                   | $f^{11}$                   | $f^{12}$                   | $f^{13}$                   | $f^{14}$                   |
|-------------------------|------------------------------|------------------------------|------------------------------|------------------------------|-------------------------|-------------------------|------------------------------|-------------------------|----------------------------|----------------------------|----------------------------|----------------------------|----------------------------|
| 57138.91                | 58140.12                     | 59140.91                     | 60144.24                     | 61145                        | 62150.36                | 63151.96                | 64157.25                     | 65158.93                | 66162.50                   | 67164.93                   | 68167.26                   | 69168.93                   | 70173.04                   |
| La                      | Ce                           | Pr                           | Nd                           | Pm                           | Sm                      | Eu                      | Gd                           | Tb                      | Dy                         | Ho                         | Er                         | Tm                         | Yb                         |
| $[\text{Xe}] 6s^2 5d^1$ | $[\text{Xe}] 6s^2 4f^1 5d^1$ | $[\text{Xe}] 6s^2 4f^3$      | $[\text{Xe}] 6s^2 4f^4$      | $[\text{Xe}] 6s^2 4f^5$      | $[\text{Xe}] 6s^2 4f^6$ | $[\text{Xe}] 6s^2 4f^7$ | $[\text{Xe}] 6s^2 4f^7 5d^1$ | $[\text{Xe}] 6s^2 4f^9$ | $[\text{Xe}] 6s^2 4f^{10}$ | $[\text{Xe}] 6s^2 4f^{11}$ | $[\text{Xe}] 6s^2 4f^{12}$ | $[\text{Xe}] 6s^2 4f^{13}$ | $[\text{Xe}] 6s^2 4f^{14}$ |
| Lantano                 | Cerio                        | Praseodimio                  | Neodimio                     | Prometio                     | Samario                 | Europio                 | Gadolinio                    | Terbio                  | Disprosio                  | Holmio                     | Erbio                      | Tulio                      | Yterbio                    |
|                         |                              |                              |                              |                              |                         |                         |                              |                         |                            |                            |                            |                            |                            |
| 89227                   | 90232.04                     | 91231.04                     | 92238.03                     | 93237                        | 94244                   | 95243                   | 96247                        | 97247                   | 98251                      | 99252                      | 100257                     | 101258                     | 102259                     |
| Ac                      | Th                           | Pa                           | U                            | Np                           | Pu                      | Am                      | Cm                           | Bk                      | Cf                         | Es                         | Fm                         | Md                         | No                         |
| $[\text{Rn}] 7s^2 6d^1$ | $[\text{Rn}] 7s^2 6d^2$      | $[\text{Rn}] 7s^2 5f^2 6d^1$ | $[\text{Rn}] 7s^2 5f^3 6d^1$ | $[\text{Rn}] 7s^2 5f^4 6d^1$ | $[\text{Rn}] 7s^2 5f^6$ | $[\text{Rn}] 7s^2 5f^7$ | $[\text{Rn}] 7s^2 5f^7 6d^1$ | $[\text{Rn}] 7s^2 5f^9$ | $[\text{Rn}] 7s^2 5f^{10}$ | $[\text{Rn}] 7s^2 5f^{11}$ | $[\text{Rn}] 7s^2 5f^{12}$ | $[\text{Rn}] 7s^2 5f^{13}$ | $[\text{Rn}] 7s^2 5f^{14}$ |
| Actinio                 | Torio                        | Protactinio                  | Uranio                       | Neptunio                     | Plutonio                | Americio                | Curio                        | Berkelio                | Californio                 | Einsteinio                 | Fermio                     | Mendelevio                 | Nobelio                    |

