## Mineral Weathering and Water Chemistry

This worksheet is to be completed along with the Rmarkdown file provided. The questions that are asked throughout the exercise are listed here. Please fill in your answers on this worksheet and make sure to show all your work!

**Question 1** What is the stable mineral under the following conditions:

$$log[H_4SiO_4] = -3$$
,  $log\frac{[K^+]}{[H^+]} = 0$ ,  $T = 25^{\circ}C$  and  $P = 1$  bar?

**Question 2** How you could infer which of the areas on the above plot is the stability field for gibbsite using Le Châtelier's Principle? Explain.

**Question 3** OK now it's your turn. Solve for the equilibrium lines for Muscovite - Kaolinite and Microcline - Muscovite, and then input the endpoints of segments in the chunk below to plot them. Please do your calculations on corresponding worksheet distributed to the class. *In case you need a refresher, there is a Exponent and Logarithm Rule Cheat Sheet in the repository for your reference.* 

Muscovite - Kaolinite  $2KAl_3Si_3O_{10}(OH)_2 + 3H_2O + 2H^+ = 3Al_2Si_2O_5(OH)_4 + 2K^+, K=10^{8.11}$  (reaction 3)

Microcline - Muscovite  $3KAlSi_3O_8 + 2H^+ + 12H_2O = KAl_3Si_3O_{10}(OH)_2 + 2K^+ + 6H_4SiO_4$ ,  $K=10^{-11.865}$  (reaction 4)



