Accepted Manuscript

Microencapsulation of ginger (Zingiber officinale) extract by spray drying technology

Kelly Simon-Brown, Kevin Mis Solval, Aranee Chotiko, Luis Alfaro, Vondel Reyes, Chen Liu, Bennett Dzandu, Emmanuel Kyereh, Andrea Goldson Barnaby, Ian Thompson, Zhimin Xu, Subramaniam Sathivel

Food Science and Technology

This house of first house as houses

Annual tree of first house as house as house as house as house as house as house

PII: S0023-6438(16)30109-8

DOI: 10.1016/j.lwt.2016.02.030

Reference: YFSTL 5312

To appear in: LWT - Food Science and Technology

Received Date: 3 June 2015

Revised Date: 11 February 2016 Accepted Date: 12 February 2016

Please cite this article as: Simon-Brown, K., Solval, K.M., Chotiko, A., Alfaro, L., Reyes, V., Liu, C., Dzandu, B., Kyereh, E., Barnaby, A.G., Thompson, I., Xu, Z., Sathivel, S., Microencapsulation of ginger (*Zingiber officinale*) extract by spray drying technology, *LWT - Food Science and Technology* (2016), doi: 10.1016/j.lwt.2016.02.030.

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

1 2 3 4 5	MICROENCAPSULATION OF GINGER (Zingiber officinale) EXTRACT BY SPRAY DRYING TECHNOLOGY
6	Kelly Simon-Brown ¹ , Kevin Mis Solval ³ , Aranee Chotiko ² , Luis Alfaro ³ , Vondel Reyes ² , Chen
7	Liu ² , Bennett Dzandu ² , Emmanuel Kyereh ² , Andrea Goldson Barnaby ¹ , Ian Thompson ¹ , Zhimin
8	Xu ² , and Subramaniam Sathivel ^{2,3*}
9	
10	
11	¹ Department of Chemistry, University of the West Indies, Jamaica
12	
13	² The School of Nutrition and Food Sciences, Louisiana State University Agricultural Center,
14	Baton Rouge, LA 70803-4300, phone (225) 578 0614, ssathivel@agcenter.lsu.edu
15	
16	³ Department of Biological and Agricultural Engineering, Louisiana State University Agricultural
17	Center, Baton Rouge, LA 70803-4300, phone (225)578 0614, , ssathivel@agcenter.lsu.edu
18	
19	*Corresponding author
20	
21	
22	
23	*Approved for publication by the director of the Louisiana Agricultural Experimental Station.
24	
25	Running title: Microencapsulated ginger extract powders

\mathbf{a}	-
1	n

٨	hs	two	_	۴
A	ns	เหล	(.1	

The aim of this study was to evaluate the effect of microencapsulation of ginger (Zingiber
officinale Roscoe) extract using maltodextrin (MD) and/or gum arabic (GA) as
microencapsulation agents on its 6-gingerol content, total phenolic and antioxidant activity. Four
slurries containing 95 mL/100 mL of ginger extract (2 g solids/100 mL) and 5 g/100 mL of a
blend of maltodextrin:gum arabic of weight ratios (4:1, 1:4, 5:0, and 0:5 g:g) were prepared and
they were separately spray dried at 160°C inlet air temperature to produce ginger extract
powders. Ginger extract contained 20.6 \pm 0.2 (mg/g solids) 6-gingerol, 7.7 \pm 0.6 (mg/g solids)
gallic acid equivalents and had an antioxidant activity of 19.9 ± 0.8 (µmol Trolox/g solids).
Microencapsulation resulted in a decline in the quantity of 6-gingerol present in ginger extract
regardless of the maltodextrin and gum arabic blend. Microencapsulation of ginger extract also
reduced gallic acid equivalents and antioxidant activity. Ginger extract dried with
maltodextrin:gum arabic (1:4 g:g) and (0:5 g:g) had larger particle size than that dried with
maltodextrin:gum arabic (4:1 g:g) and (5:0 g:g). Maltodextrin:gum arabic (4:1 g:g) and (5:0 g:g)
had better morphological properties than maltodextrin:gum arabic (4:1 g:g) and (5:0 g:g).
Microencapsulated ginger extract powder may be used as a novel food ingredient.

Key words: maltodextrin; gum arabic; 6-gingerol; 6-shogaol

1. Introduction

49

50

51

52

53

54

55

56

57

58

59

60

61

62

63

64

65

66

67

68

69

70

71

Jamaican ginger (Zingiber officinale Roscoe, family Zingiberaceae) is a light reddish-yellow, hot, spicy underground monocotyledenous stem (rhizome) which was introduced to Jamaica from India or South East Asia in the 1500s. It has been recognized as a premier among gingers due to its quality flavour, oil content and appearance, providing the basis on which the standards of other gingers are evaluated. The rhizome is mainly cultivated in the parishes of Clarendon, Manchester, St. Ann, Hanover, Portland and St. Thomas, where climatic conditions are ideal for its growth. Planting begins in April to June of each year with harvesting occurring between December and March. It is recommended that ginger be harvested 8 to 9 months after planting so that the roots do not become fibrous and flavour development is optimal. Ginger harvested at eight months have shown high yields of oleoresin (Bailey-Shaw et al., 2008). Jamaica export of ginger continues to increase with the product being positioned in niche markets. In the United States, ginger was named one of the top twenty selling herbal supplement earning over US \$1.57 million in 2011 (Blumenthal, Lindstrom, Ooyen & Lynch, 2012). Ginger is associated with a number of health benefits which has led to its comprehensive use in a variety of commercial natural products offered in the emerging nutraceuticals and functional foods market. Medicinal properties include anticancer (Cheng, Liu, Peng, Oi & Li, 2011; Karna, et al., 2012; Shukla & Singh, 2007), antioxidant (El-Ghorab, Nauman, Anjum, Hussain & Nadeem, 2010), anti-inflammatory (Minghetti et al., 2007) and antidiabetic (Afshari et al., 2007) activities. Gingerols, paradols, shogaols and zingerones (Figure 1) are some of the known bioactive compounds present in fresh ginger, but the most important of them is 6-gingerol (Hiserodt, Franzblau & Rosen, 1998). These phenolic ketones have a range of alkyl side chains. 6-Shogaol is formed from the dehydration of 6-gingerol and is more pungent (Bhattarai, Tran &

72

73

74

75

76

77

78

79

80

81

82

83

84

85

86

87

88

89

90

91

92

93

94

Duke, 2001; Zancan, Marques, Petenate & Meireles, 2002; Ok & Jeong, 2012). The concentration of 6-shogoal ranges from 3 to 5 mg/g in fresh ginger extracts increasing to approximately 22 mg/g after dehydration (Ok & Jeong, 2012). Its occurrence may be used as an indicator of long time storage or thermal treatment of ginger products as its content increases with an increase in drying and extraction temperatures (Ok & Jeong, 2012). The quantity and quality of these indigenous compounds differ on the basis of factors, such as plant cultivar and genetics, soil type, growing conditions, maturity and post-harvest conditions (Pawar, Pai, Nimbalkar & Dixit, 2011). Microencapsulation is a technique that has been utilized to protect flavor components from destructive changes and to convert flavor into a free flowing form. It is employed to preserve the stability, bioactivity and bioavailability of active components (Sansone, Mencherini, Picerno, d'Amore, Aquino & Lauro, 2011; Schweiggert, Hofmann, Reichel, Schieber & Carle, 2008). The process allows sensitive ingredients to be blended or homogenised in a solution which contains macromolecules and emulsifiers to form a stable emulsion. Encapsulating agents are used exclusively or in association with other encapsulating agents to achieve an ideal composition (Fernandes, Candido & Oliveira, 2012). Maltodextrin, gum arabic, pectin and guar gum are examples of encapsulating agents which have been utilized in the encapsulation of bioactive compounds (Ravichandran et al., 2012). Hydrolysed starch may be combined with a surfaceactive biopolymer, such as gum arabic which has become a popular and common spray drying ingredient due to its emulsifying properties providing excellent volatile retention during the drying process. Combinations of gum arabic and maltodextrin were found to be effective for the encapsulation of oils (Jafari, Assadpoor, He & Bhandari, 2008). Spray drying is the most widely utilized method of microencapsulation in the food and beverage industry (Gharsallaoui, Roudaut,

95	Chambin, Voilley & Saurel, 2007). More recent technologies include emulsion electrospraying
96	which has the advantage of encapsulating under milder conditions and can be utilized for
97	thermosensitive bioactives (Gomez-Mascaraque & Lopez-Rubio, 2016). Quercetin has been
98	encapsulated utilizing Pluronic F127 poloxamers by the supercritical antisolvent technique
99	(Fraile, Buratto, Gomez, Martin & Cocero, 2014)
100	There is presently limited information in the scientific literature regarding the
101	microencapsulation of ginger extract. Hence the aim of the present study was to produce
102	microencapsulated ginger extract using maltodextrin and/or gum arabic as encapsulation agents
103	and to evaluate the effect of microencapsulation on the 6-gingerol content, total phenolic content
104	and antioxidant activity of ginger extracts.
105	2. Materials and methods
106	2.1 Materials
106 107	2.1 Materials A standard of 6-gingerol was obtained from Sigma-Aldrich, USA. Acetonitrile, acetic acid
107	A standard of 6-gingerol was obtained from Sigma-Aldrich, USA. Acetonitrile, acetic acid
107 108	A standard of 6-gingerol was obtained from Sigma-Aldrich, USA. Acetonitrile, acetic acid and methanol were HPLC grade and obtained from Fisher Scientific, USA. Maltodextrin with a
107108109	A standard of 6-gingerol was obtained from Sigma-Aldrich, USA. Acetonitrile, acetic acid and methanol were HPLC grade and obtained from Fisher Scientific, USA. Maltodextrin with a Dextrose Equivalent of 9 to 13 was obtained from NOW Foods Company, (Bloomingdale, IL).
107108109110	A standard of 6-gingerol was obtained from Sigma-Aldrich, USA. Acetonitrile, acetic acid and methanol were HPLC grade and obtained from Fisher Scientific, USA. Maltodextrin with a Dextrose Equivalent of 9 to 13 was obtained from NOW Foods Company, (Bloomingdale, IL). Gum arabic was obtained from Frontier Natural Products Co-Op (Norway, IA). Water utilized
107 108 109 110 111	A standard of 6-gingerol was obtained from Sigma-Aldrich, USA. Acetonitrile, acetic acid and methanol were HPLC grade and obtained from Fisher Scientific, USA. Maltodextrin with a Dextrose Equivalent of 9 to 13 was obtained from NOW Foods Company, (Bloomingdale, IL). Gum arabic was obtained from Frontier Natural Products Co-Op (Norway, IA). Water utilized for HPLC analyses was purified with a Milli-Q water system (Millipore Corp., Bedford, MA,
107 108 109 110 111 112	A standard of 6-gingerol was obtained from Sigma-Aldrich, USA. Acetonitrile, acetic acid and methanol were HPLC grade and obtained from Fisher Scientific, USA. Maltodextrin with a Dextrose Equivalent of 9 to 13 was obtained from NOW Foods Company, (Bloomingdale, IL). Gum arabic was obtained from Frontier Natural Products Co-Op (Norway, IA). Water utilized for HPLC analyses was purified with a Milli-Q water system (Millipore Corp., Bedford, MA, USA).
107 108 109 110 111 112 113	A standard of 6-gingerol was obtained from Sigma-Aldrich, USA. Acetonitrile, acetic acid and methanol were HPLC grade and obtained from Fisher Scientific, USA. Maltodextrin with a Dextrose Equivalent of 9 to 13 was obtained from NOW Foods Company, (Bloomingdale, IL). Gum arabic was obtained from Frontier Natural Products Co-Op (Norway, IA). Water utilized for HPLC analyses was purified with a Milli-Q water system (Millipore Corp., Bedford, MA, USA). 2.2 Ginger extract

ethanol solution for 72 h, filtered and evaporated under reduced pressure at 60-65 °C to produce the ginger extract which was immediately stored at 4 °C.

2.3 Microencapsulation of Ginger Extract

119

120

121

122

123

124

125

126

127

128

129

130

131

132

133

134

135

136

137

138

139

Slurries containing ginger extract (20 g/L ginger solids), maltodextrin and/or gum arabic were prepared by mixing ginger extract (475 mL) and maltodextrin and/or gum arabic (25 g). The resulting slurries had a 7 g/100 mL solid content. The encapsulating agents represented 71.43 g/100 g of the total solids of the slurries. The slurries were homogenized using an ultrashearing device (OMNI, Ultrashear M, Omni International, Kennesaw, GA) for 5 min. at 20,000 rpm. Four slurries were produced by combining 475 mL of ginger extract with a 25 g blend of maltodextrin (MD):gum arabic (GA) prepared at different weight ratios (4:1, 1:4, 5:0, 0:5 g:g) resulting in MD4GA1, MD1GA4, MD5, and GA5, respectively. Four feed mixtures were produced by combining ginger extract with a blend maltodextrin: gum arabic prepared at different weight ratios (4:1, 1:4, 5:0, 0:5 g:g) resulting in MD4GA1, MD1GA4, MD5, and GA5, respectively. The concentration of 6-gingerol, trolox and gallic acid present in the feed mixtures was 5.9 + 0.2 mg/g of solid, 5.7 ± 0.3 µmol trolox /g of solid and 2.2 ± 0.2 mg gallic acid /g of solid, respectively. MD4GA1, MD1GA4, MD5, and GA5 were separately spray dried under cocurrent air flow conditions using a pilot plant scale spray dryer (FT 80 Tall form spray dryer, Armfeild Inc., Jackson, NJ). Inlet air was heated at 160 °C and was blown into the drying chamber. Slurries were atomized into the drying chamber and the outlet temperature of the spray dryer was 65 °C. Droplets were dried producing the microencapsulated samples. In total, four microencapsulated ginger extract powders were produced including spray dried MD4GA1, spray dried MD1GA4, spray dried MD5 and spray dried GA5 (DMD4GA1, DMD1GA4, DMD5 and DGA5, respectively). The resulting powders were analysed for moisture content following the

- AOAC method 930.15 (AOAC, 1999). The resulting powders were stored at 4 °C until needed
- for analysis. The spray drying procedure was carried out in triplicate.

2.4 Determination of 6-gingerol content

142

156

- Approximately, 5 mL of ginger extract were dissolved in methanol (1 mL) and centrifuged at
- 144 21480 x g for 5 min. The resulting supernatant (0.5 mL) was utilized for HPLC analyses.
- Microencapsulated samples (500 mg) were dissolved in 4 mL of 90 mL/100 mL ethanol and
- stirred for 30 min. Samples were centrifuged at 21480 x g for 5 min and 50 µL of the resulting
- supernatant were utilized for analyses. Analyses for 6-gingerol were performed utilizing HPLC.
- The HPLC system consisted of Waters (Milford, MA) 510 pumps, a 715 Ultra WISP injector,
- and 410 UV and 470 fluorescence detectors. A reversed phase C18 column, 25 cm \times 4.6 mm
- diameter 5 µm Supelcosil LC-Si (Supelco, Bellefonte, PA), was used. Elution was isocratic using
- a mobile phase consisting of acetonitrile, water, acetic acid 500:450:50 mL:mL:mL, with a flow
- rate of 1.0 mL/min and a temperature of 30 °C. A Variable Wavelength Detector (VWD) at 282
- 153 nm was used to detect 6-gingerol which was identified and quantified based on the retention time
- of a 6-gingerol standard. The 6-gingerol content was calculated by using the standard calibration
- 155 curve, $y = 639191 (X) + 33305 (R^2 = 0.9997)$.

2.5 Microencapsulation efficiency (ME) of ginger extract

- The microencapsulation efficiency (ME) of ginger extract using maltodextrin and/or gum
- arabic was calculated using Eq. 1:
- 159 $ME = \frac{6 \text{gingerol in powder (mg/g solids)}}{6 \text{gingerol in slurry (mg/g solids)}} *100$ (1)

2.6 The 1,1-diphenyl-2-picrylhydrazyl (DPPH) radical scavenging assay

- DPPH assay was performed according to the method of Brand-Williams, Cuvelier and Berset
- 162 (1995). Microencapsulated samples (500 mg) were dissolved in 4 mL of 90 mL/100 mL ethanol

and stirred for 30 min. Samples (200 μL) were reacted with 2.8 mL of DPPH for 30 min in the dark. The absorbance was recorded at wavelength 515 nm using a spectrophotometer (SpectronicTM GENESYSTM 2, Thermo Fisher Scientific, Waltham, MA). Samples were analysed in triplicate and reported as Trolox equivalents.

2.7 Total phenolic content

The total phenolic content was determined by the Folin-Ciocalteu method, adapted from Swain and Hillis (1959). Microencapsulated samples (500 mg) were dissolved in 4 mL of 90 mL/100 mL ethanol and stirred for 30 min. The samples and ginger extract (125 μL) were centrifuged at 21480 x g for 5 min. The resulting supernatant was then combined with distilled water (0.5 mL) and Folin-Ciocalteu reagent (0.25 mol/L, 125 μL). The mixture was allowed to react in the dark during 6 min; then, 1.25 mL of a 7 g/100 g Na₂CO₃ solution and 1 mL distilled water were added. The solution was incubated at room temperature in the dark for 1.5 h. The absorbance was measured at 760 nm using a spectrophotometer. Samples were analysed in triplicate and expressed as gallic acid equivalents.

2.8 Color of microencapsulated extract powders

The color of microencapsulated samples was measured using the chroma meter LABSCAN XE (Hunterlab, VA, USA) and reported in CIELAB colour scales L*, a*, and b* values (L* is the degree of lightness to darkness, a* is the degree of redness to greenness, and b* is the degree of yellowness to blueness). The instrument was calibrated with standardized black and white standard. The chroma and hue angle values were calculated using Equations 2 and 3, respectively (Solval, Sundararajan, Alfaro & Sathivel, 2012).

184 Chroma =
$$[(a^*)^2 + (b^*)^2]^{1/2}$$
 (2)

Hue =
$$tan^{-1} (b^* / a^*)$$
 (3)

186	2.9 Scanning	electron	microscopy	(SEM)	and	particle	size	distribution	of
187	microenca	psulated sa	mples						

The morphology of the microencapsulated samples was evaluated utilizing a scanning electron microscope (SEM) (JSM-6610LV, JEOL Ltd. Japan). Samples were mounted on aluminium SEM stubs and coated with gold:palladium (60:40) in an Edwards S150 sputter coater. The samples were systematically observed with 1000x magnification. The particle size distribution was measured using a Microtrac S3500 system. The powdered sample was placed in a test chamber with circulating ethyl-alcohol. A 10 s ultrasound mixing at 20 W was used before each test. The sample was pumped through the cell at 40 % of the maximum flow rate. The light was scattered from three lasers from low to high angles. The whole light scatter pattern was collected and the particle size calculated using the Modified Michelson interferometer (MIE) scattering technique which measures the angular distribution of backscattered light (Solval, Sundararajan, Alfaro & Sathivel, 2012).

2.10 Data analysis

All data were analysed using SAS software version 9.2 (SAS Institute Inc., 2008). Means and standard deviations of the data were presented. Analysis of Variance (ANOVA) and Tukey's studentized range test were carried out to determined differences among treatments at the significant level of P <0.05. ANOVA was used to compare the means of the variables investigated for each drying agent utilized.

3. Results and discussion

3.1 Content of 6-gingerol in ginger extract

Fresh ginger rhizome is a rich source of biologically active compounds of which gingerols are the major active components (Bhattarai, Tran & Duke, 2001). Results from this study

confirmed that Jamaican ginger had a greater concentration of 6-gingerol (20.6 \pm 0.2 mg	/g) than
ginger grown in different geographical locations and would therefore be considered	d more
pungent. Studies conducted in Australia have shown that fresh extracts of a cultivar kr	nown as
"Jamaican" ginger contained the highest level of 6-gingerol and was the most potent comp	pared to
gingers extracts from different geographical regions (Wohlmuth, Leach, Smith & Myers	, 2005).
Brazilian ginger rhizome extracts contained a total of 20.10 ± 0.10 mg/g pungent pr	rinciples
calculated as the sum of 4-, 6-, 8- and 10-gingerol and 6-, 8-, 10-shogaols of which 6-ş	gingerol
(14.1 mg/g) was the major component present (Schweiggert, Hofmann, Reichel, Sch	ieber &
Carle, 2008). Fresh ginger extracts from China contained 11.5 mg/g of 6-gingerol (Chemotarius Chemotarius Chemotar	ng, Liu,
Peng, Qi & Li, 2011) while 6-gingerol content of ginger from India was found to range	ge from
0.12 to 2.08 (mg/g) (Pawar, Pai, Nimbalkar & Dixit, 2011).	

3.2 Content of 6-gingerol in microencapsulated ginger extract

In this study, ginger extract was microencapsulated with maltodextrin and gum arabic using spray drying technology. Gingerols are stable at a pH range from 1 to 7 at 37 °C but start to degrade at temperatures of 60 °C and above (Cheng, Liu, Peng, Qi & Li, 2011). In aqueous solution, 6-gingerol shows maximum stability at pH 4 and at 37 °C (Bhattarai, Tran & Duke, 2001; Young, Chiang, Huang, Pan & Chen, 2002). Also, it has been reported that ginger extracts stored at 4 °C or room temperature over a period of 5-6 months have not shown any significant variation in the concentration of 6-gingerol and 8-gingerol present (Salmon, Bailey-Shaw, Hibbert, Green, Smith & Williams, 2012; Wohlmuth, Leach, Smith & Myers, 2005).

Gingerols are thermally labile due to the presence of a β -hydroxyl ketone group in their structure (Bhattarai, Tran & Duke, 2001). They undergo dehydration-hydration transformations

231	with shogoals and exhibit novel reversible kinetics (Bhattarai, Tran & Duke, 2001) which may
232	account for the occurrence of the 6-gingerol peak as a doublet in the HPLC profile (Figure 2).
233	At high temperatures gingerols are converted to 6, 8, and 10-shogaol (Wohlmuth, Leach, Smith
234	& Myers, 2005; Zhang, Iwaoka, Huang, Nakamoto & Wong, 1994). An increase in the levels of
235	shogaol, specifically 6-shogaol, has been associated with improved anticancer activity which
236	may be due to its α - and β -unsaturated ketone moiety (Cheng, Liu, Peng, Qi & Li, 2011). In a
237	study conducted by Schweiggert, Hofmann, Reichel, Schieber and Carle (2008), spray drying
238	resulted in a 45 % decline in the pungent principles of ginger which was attributed to the high
239	temperatures utilized in the drying chamber.
240	The concentration of 6-gingerol present in the feed (mixture of the drying agents and ginger
241	extract) was 5.9 ± 0.2 mg/g for all blends of maltodextrin and gum arabic. It reduced to the
242	range of 2.0-2.4 mg/g extract during spray drying regardless of the maltodextrin and gum arabic
243	blend used to microencapsulate the ginger extract (Table 1). The maltodextrin and gum arabic
244	and their blends used as microencapsulation agents for ginger extract were not different in terms
245	of microencapsulation efficiency (protection) of 6-gingerol against degradation during the
246	microencapsulation process with percentage recoveries of $33-40 \%$.
247	According to Krishnan, Bhosale and Singhal (2005) gum arabic is a very effective
248	microencapsulation agent because it can produce stable emulsions over a wide pH range. Also,
249	gum arabic is compatible with a wide variety of gums, starches, carbohydrates and proteins
250	(Carneiro, Tonon, Grosso & Hubinger, 2013; Tomas-Navarro, Vallejo, Borrego & Tomas-
251	Barberan, 2014). On the other hand, maltodextrins are also good carrier agents (drying agents);
252	however, they have less emulsifying ability (capacity). Therefore, they may show marginal
253	retention of volatile compounds. Maltodextrins as carrier agents have the advantage of being

bland in flavor, relatively inexpensive, having low viscosity in large quantities and provide superb protection from oxygen (Ferrari, Germer, Alvim, Vissotto, Mauricio de Aguirre & 2012; Carneiro, Tonon, Grosso & Hubinger, 2013). Oxygen barrier properties are based on the dextrose equivalent (DE) of the hydrolyzed starch. Higher-DE systems are less permeable to oxygen and result in powders with higher encapsulation efficiencies. The retention of volatiles capacity of maltodextrins is also a function of their DE (Reineccius, 1988). High DE maltodextrin has been successfully utilized in the encapsulation of orange peel oil to protect it from oxidation (Anandaraman & Reineccius 1986). However, in our study there was no significant differences in gingerols content in the microencapsulated ginger extract with either maltodextrin or gum arabic or their blends.

3.3 Moisture content of microencapsulated ginger extract

All four encapsulated ginger extract powders had similar moisture contents regardless of encapsulating agent (Table 1). The moisture content of microencapsulated samples may be affected by a number of factors which include the air flow rate, drying air temperature of the spray dryer, (Goula & Adamopoulos, 2005) and evaporation rate droplet size (Obón, Castellar, Alacid & Fernández-López, 2009, Chranioti, Chanioti, & Tzia, 2016, Noshad, Mohebbi, Koocheki & Shahidi, 2015). An increase in air flow rate and air temperature results in lower residual moisture content due to faster diffusion rates (Bhattarai, Tran & Duke, 2001). In this study, we used similar spray drying conditions for all trials, including inlet air spray drying temperature (160 °C). The spray conditions might have contributed to producing encapsulated ginger extract powders with similar moisture content.

3.4 Total phenolic content and antioxidant activity of the microencapsulated ginger extract powders

Fresh ginger extract contained 19.85 \pm 0.26 μ mol trolox /g and 7.74 \pm 0.64 mg gallic acid /g. This is expected since phenolic compounds are water soluble antioxidants present at high concentrations in plants (Pawar, Pai, Nimbalkar & Dixit, 2011). The total phenolic content in all of the microencapsulated ginger extract powders was not significantly different (P \leq 0.05) (Table 1). The results may be related to the antioxidant activity results. In this study, the antioxidant activity (trolox content) of the ginger extract and microencapsulated ginger extract powders was investigated using the DPPH assay. Antioxidant activity was higher in the ginger extract (19.85 \pm 0.26 μ mol trolox /g) than in the microencapsulated ginger extract powders (Table 1). However, all the powders had similar trolox content, except maltodextrin:gum arabic (5:0) (Table 1). The concentration of trolox present in the powders was similar, except maltodextrin:gum arabic (5:0), to the feed mixtures. In our study there was no significant differences in gallic acid content (phenolic compound) in the microencapsulated ginger extract with either maltodextrin or gum arabic or their blends (Table 1). This study indicated that the drying agents protected phenolic compounds during spray drying.

3.5 Particle size and color of the microencapsulated ginger extract powders

The mean particle size distribution of the microencapsulated samples ranged from 8.2 to 15.3 μ m (Table 2). It was also observed that maltodextrin:gum arabic (1:4) and maltodextrin:gum arabic (0:5) powders had a significantly (P \leq 0.05) higher mean particle size than those microencapsulated with a higher ratio of maltodextrin. Similar findings are reported by Fernandes, Candido and Oliveira (2012). Droplet size increased as the quantity of gum arabic increased.

All of the powders had a whitish color (Table 2). However, maltodextrin:gum arabic (1:4) was significantly ($P \le 0.05$) lighter compared to the rest of the powders. Chroma values were

significantly (P \leq 0.05) higher in maltodextrin:gum arabic (4:1) and maltodextrin:gum arabic (0:5) than in the other microencapsulated ginger extracts. The chroma measurement is indicative of the vividness of color. Even more, maltodextrin:gum arabic (1:4) and maltodextrin:gum arabic (0:5) had significantly (P \leq 0.05) lower Hue angle values compared to those of maltodextrin:gum arabic (4:1) and maltodextrin:gum arabic (5:0). Hue angle describes color based on a circle, a hue angle of 0°, 90°, 120°, and 240° indicates a red, yellow, green, and blue color, respectively. Powders containing higher amounts of maltodextrin had stronger yellow color.

3.6 SEM analysis of the microencapsulated ginger extract powders

It was observed under scanning electron microscopy that maltodextrin:gum arabic (4:1) and maltodextrin:gum arabic (5:0) had the most uniform shapes and appeared more spherical than the other samples. Samples with higher levels of gum arabic (maltodextrin:gum arabic (1:4) and (maltodextrin:gum arabic (0:5)) had evidence of more denting with some broken and incomplete particles (Figure 3). Denting is a common feature of particles as they inflate at high temperatures and break upon evaporation (Solval, Sundararajan, Alfaro & Sathivel, 2012). It is influenced by drying speed, the mechanism of water removal and the type of microencapsulating agent utilized for spray drying process (Kagami, Sugimura, Fujishima, Matsuda, Kometani & Matsumura, 2006). Low drying air velocity tends to more dent formation. According to Krishnan, Bhosale and Singhal (2005), microcapsules containing gum arabic may show dents at the surface which are signs of shrinkage. It is desired that bioactive microcapsules have uniform surfaces with minimum cracks or dents since that promotes protection and retention of the core material. The microencapsulated bioactive is protected from degradation for longer times when encapsulated in sound (minimum cracks and dents) particles. More sound particles were apparent in samples microencapsulated with maltodextrin. The incorporation of carbohydrates

into wall systems has been shown to improve the drying properties of the wall matrix. This may
be due to an enhancement in the formation of a dry crust around drying droplets (Sheu &
Rosenberg 1995). Combinations of maltodextrin and other drying agents should be investigated
for process optimization.
4. Conclusion
The study developed microencapsulation of Jamaican ginger extract using maltodextrin
and/or gum arabic. Gingerols, bioactive compounds present in fresh ginger, were reduced during
the encapsulation. Drying agents including maltodextrin and/or gum arabic and their blends did
not prevent reduction of gingerols. Also, microencapsulation reduced the total phenolics content
and antioxidant activity of ginger extract. However, our study showed that significant amounts
of gingerols and phenolics were in encapsulated ginger extract. The encapsulated extract is in a
powder form and could be readily incorporated into food products such as tea and bread.

337 **References**

355

356

357

358359

360

364 365

366

367368

- Anandaraman, S., & Reineccius, G. A. (1986). Stability of encapsulated orange peel oil. *Food Technology*, 40(11), 88-93.
- 340 Afshari, A. T., Shirpoor, A., Farshid, A., Saadatian, R., Rasmi, Y., Saboory, E., ... Allameh, A. (2007). The effect of ginger on diabetic nephropathy, plasma antioxidant capacity and lipid peroxidation in rats. *Food Chemistry*, 101(1), 148-153.
- 343 AOAC. (1999). *Official methods of analysis of the association of official analytical chemists.* Vol II. Arlington, VA.
- Bailey-Shaw, Y. A., Williams, L. A., Junor, G. A. O., Green, C. E., Hibbert, S. L., Salmon, C. N., & Smith, A. M. (2008). Changes in the contents of oleoresin and pungent bioactive principles of Jamaican ginger (*Zingiber officinale* Roscoe.) during maturation. *Journal of Agriculture and Food Chemistry*, 56(14), 5564-5571.
- 349 Bhattarai, S., Tran, V. H., & Duke, C. C. (2001). The stability of gingerol and shogaol in aqueous solutions. *Journal of Pharmaceutical Sciences*, 90(10), 1658-1664.
- Blumenthal, M., Lindstrom, A., Ooyen, C., & Lynch, M. E. (2012). Herb supplement sales increase 4.5% in 2011. *HerbalGram* (95), 60-64.
- Brand-Williams, W., Cuvelier, M., & Berset, C. (1995). Use of a free radical method to evaluate antioxidant activity. *LWT-Food Science and Technology*, 28(1), 25-30.
 - Carneiro, H. C. F., Tonon, R. V., Grosso, C. R. F., Hubinger, M. D. (2013). Encapsulation efficiency and oxidative stability of flaxseed oil microencapsulated by spray drying using different combinations of wall materials. *Journal of Food Engineering*, 115(4), 443-451.
 - Cheng, X. L., Liu, Q., Peng, Y. B., Qi, L. W., & Li, P. (2011). Steamed ginger (*Zingiber officinale*): Changed chemical profile and increased anticancer potential. *Food Chemistry*, 129(4), 1785-1792.
- Chranioti, C., Chanioti, S., & Tzia, C. (2016). Comparison of spray, freeze and oven drying as a means of reducing bitter aftertaste of steviol glycosides (derived from *Stevia rebaudiana* Bertoni plant) Evaluation of the final products. *Food Chemistry* 190, 1151–1158.
 - El-Ghorab, A. H., Nauman, M., Anjum, F. M., Hussain, S., & Nadeem, M. (2010). A comparative study on chemical composition and antioxidant activity of ginger (*Zingiber officinale*) and cumin (*Cuminum cyminum*). *Journal of Agricultural and Food Chemistry*, 58(14), 8231-8237.
 - Fernandes, L. P., Candido, R. C., & Oliveira, W. P. (2012). Spray drying microencapsulation of *Lippia sidoides* extracts in carbohydrate blends. *Food and Bioproducts Processing*, 90(3), 425-432.
- Ferrari, C. C., Germer, S. P. M., Alvim, I. D., Vissotto, F. Z., Mauricio de Aguirre, J. (2012)
 Influence of carrier agents on the physicochemical properties of blackberry powder
 produced by spray drying. *International Journal of Food Science and Technology*, 47(6),
 1237-1245.
- Fraile, M., Buratto, R., Gomez, B., Martin, A., Cocero, M. J. (2014). Enhanced delivery of quercetin by encapsulation in Poloxamers by supercritical antisolvent process. *Industrial & Engineering Chemistry Research*, 53(11), 4318-4327.
- 378 Gharsallaoui, A., Roudaut, G., Chambin, O., Voilley, A., Saurel, R. (2007). Applications of 379 spray-drying in microencapsulation of food ingredients: an overview. *Food Research* 380 *International*, 40, 1107–1121
- Goula, A. M., & Adamopoulos, K. G. (2005). Spray drying of tomato pulp in dehumidified

air: II. The effect on powder properties. *Journal of Food Engineering*, 66(1), 35-42.

386

387

388

389

392393

394

395

405 406

407

408

409

- Gomez-Mascaraque, L. G. & Lopez-Rubio, A. (2016). Protein-based emulsion electrosprayed micro- and submicroparticles for the encapsulation and stabilization of thermosensitive hydrophobic bioactives. *Journal of Colloid and Interface Science*, 465, 259-270.
 - Hiserodt, R., Franzblau, S., & Rosen, R. (1998). Isolation of 6-, 8-, and 10-gingerol from ginger rhizome by HPLC and preliminary evaluation of inhibition of *Mycobacterium avium* and *Mycobacterium tuberculosis*. *Journal of Agricultural and Food Chemistry*, 46(7), 2504-2508.
- Jafari, S. M., Assadpoor, E., He, Y., & Bhandari, B. (2008). Encapsulation efficiency of food flavours and oils during spray drying. *Drying Technology*, 26(7), 816-835.
 - Kagami, Y., Sugimura, S., Fujishima, N., Matsuda, K., Kometani, T., & Matsumura, Y. (2006). Oxidative stability, structure, and physical characteristics of microcapsules formed by spray drying of fish oil with protein and dextrin wall materials. *Journal of Food Science*, 68(7), 2248-2255.
- Karna, P., Chagani, S., Gundala, S. R., Rida, P. C., Asif, G., Sharma, V., Aneja, R. (2012).
 Benefits of whole ginger extract in prostate cancer. *British Journal of Nutrition*, 107(04),
 473-484.
- Krishnan, S., Bhosale, R., & Singhal, R. S. (2005). Microencapsulation of cardamom oleoresin:

 Evaluation of blends of gum arabic, maltodextrin and a modified starch as wall materials.

 Carbohydrate Polymers, 61(1), 95-102.
- Minghetti, P., Sosa, S., Cilurzo, F., Casiraghi, A., Alberti, E., Tubaro, A., Montanari, L. (2007).
 Evaluation of the topical anti-inflammatory activity of ginger dry extracts from solutions and plasters. *Planta Medica*, 73(15), 1525-1530.
 - Noshad, M., Mohebbi, M., Koocheki, A., Shahidi, F. (2015). Microencapsulation of vanillin by spray drying using soy protein isolate—maltodextrin as wall material. *Flavour and Fragrance Journal*, 30(5), 387-391.
 - Obón, J. M., Castellar, M. R., Alacid, M., & Fernández-López, J. A. (2009). Production of a red-purple food colorant from *Opuntia stricta* fruits by spray drying and its applications in food model systems. *Journal of Food Engineering*, 90(4), 471-479.
- Ok, S., & Jeong, W.S. (2012). Optimization of extraction conditions for the 6-shogaol-rich extract from ginger (*Zingiber officinale* Roscoe). *Preventive Nutrition and Food Science*, 17(2) 166-171.
- Pawar, N., Pai, S., Nimbalkar, M., & Dixit, G. (2011). RP-HPLC analysis of phenolic antioxidant compound 6-gingerol from different ginger cultivars. *Food Chemistry*, 126(3), 1330-1336.
- Ravichandran, K., Palaniraj, R., Saw, N., Gabr, A. M., Ahmed, A. R., Knorr, D. & Smetanska, I., (2014). Effects of different encapsulation agents and drying process on stability of betalains extract. *Journal of Food Science and Technology* 51(9), 2216-2221.
- 420 Reineccius, G. A. (1988). *Spray-drying of food flavors*. American Chemical Society: 421 Washington, DC.
- Salmon, C. N., Bailey-Shaw, Y. A., Hibbert, S., Green, C., Smith, A. M., & Williams, L. A.
 (2012). Characterisation of cultivars of Jamaican ginger (*Zingiber officinale* Roscoe) by
 HPTLC and HPLC. *Food Chemistry*, 131(4), 1517-1522.
- 425 Sansone, F., Mencherini, T., Picerno, P., d'Amore, M., Aquino, R. P., & Lauro, M. R. (2011).
- Maltodextrin/pectin microparticles by spray drying as carrier for nutraceutical extracts. *Journal of Food Engineering*, 105(3), 468-476.

- Schweiggert, U., Hofmann, S., Reichel, M., Schieber, A., & Carle, R. (2008). Enzyme-assisted liquefaction of ginger rhizomes (*Zingiber officinale* Rosc.) for the production of spraydried and paste-like ginger condiments. *Journal of food engineering*, 84(1), 28-38.
- Sheu, T.Y., & Rosenberg, M. (1995). Microencapsulation by spray drying ethyl caprylate in whey protein and carbohydrate wall systems. *Journal of Food Science*, 60(1), 98-103.
- Shukla, Y., & Singh, M. (2007). Cancer preventive properties of ginger: a brief review. *Food* and Chemical Toxicology, 45(5), 683-690.
- Solval, K. M., Sundararajan, S., Alfaro, L., & Sathivel, S. (2012). Development of cantaloupe (*Cucumis melo*) juice powders using spray drying technology. *LWT-Food Science and Technology*, 46(1), 287-293.
 - Swain, T., & Hillis, W. (1959). The phenolic constituents of Prunus domestica. I.—The quantitative analysis of phenolic constituents. *Journal of the Science of Food and Agriculture*, 10(1), 63-68.
- Tomas-Navarro, M., Vallejo, F., Borrego, F. & Tomas-Barberan, F. A. (2014) Encapsulation and micronization effectively improve orange beverage flavanone bioavailability in humans. *Journal of Agricultural and Food Chemistry*, 62(39), 9458-9462.
 - Wohlmuth, H., Leach, D. N., Smith, M. K., & Myers, S. P. (2005). Gingerol content of diploid and tetraploid clones of ginger (*Zingiber officinale* Roscoe). *Journal of Agricultural and Food Chemistry*, 53(14), 5772-5778.
- 447 Young, H., Chiang, C., Huang, Y., Pan, F. P., & Chen, G. (2002). Analytical and stability studies of ginger preparations. *Journal of Food and Drug Analysis*, 10(3), 149-153.
- Zancan, K. C., Marques, M. O., Petenate, A. J., & Meireles, M. A. A. (2002). Extraction of ginger (*Zingiber officinale* Roscoe) oleoresin with CO₂ and co-solvents: a study of the antioxidant action of the extracts. *The Journal of Supercritical Fluids*, 24(1), 57-76.
- Zhang, X., Iwaoka, W., Huang, A., Nakamoto, S., & Wong, R. (1994). Gingerol decreases after processing and storage of ginger. *Journal of Food Science*, 59(6), 1338-1340.

454

438 439

440

444

455 List of Figures

- 456 **Figure 1.** Pungent compounds found in ginger.
- 457 Figure 2. HPLC chromatogram profile of fresh and microencapsulated ginger extracts. A
- 458 reversed phase C18 column was eluted with a mobile phase consisting of acetonitrile, water,
- acetic acid 500:450:50 mL:mL:mL for the analyses.
- 460 (A) GE, (B) DMD4GA1, (C) DGA4MD1, (D) DMD5, and (E) DGA5. GE = ginger extract; MD
- = maltodextrin; GA = gum arabic; DMD4GA1 = microencapsulated GE with MD:GA mixed at a
- weight ratio of 4:1 g:g; DMD1GA4 = microencapsulated GE with MD:GA mixed at a weight
- ratio of 1:4 g:g; DMD5 = microencapsulated GE with MD; DGA5= microencapsulated GE with
- 464 GA.

- Figure 3. Scanning electron micrographs (x1000) of the microencapsulated ginger extract powders. (A) =DMD4GA1; (B) =DMD1GA4; (C) =DMD5; (D) = DGA5. Particle size
- distribution was measured using a Microtrac S3500 system.
- 469 GE = ginger extract; MD = maltodextrin; GA = gum arabic; DMD4GA1 = microencapsulated
- 470 GE with MD:GA mixed at a weight ratio of 4:1 g:g; DMD1GA4 = microencapsulated GE with
- 471 MD:GA mixed at a weight ratio of 1:4 g:g; DMD5 = microencapsulated GE with MD; DGA5=
- 472 microencapsulated GE with GA.

Table 1. Moisture content, 6-gingerol, gallic acid equivalents and trolox of microencapsulated ginger extract powders

Sample	6-Gingerol (mg/g extract)	Moisture content (g/100 g powder)	Gallic acid equivalents (GAE: mg/g solid)	Trolox content (µmol/g solid)
DMD4GA1	2.4 ± 0.1^{a}	3.1 ± 1.6^{a}	1.9 ± 0.1^{a}	$5.7 \pm 0.3^{\rm b}$
DMD1GA4	2.0 ± 0.3^{a}	8.2 ± 3.0^{a}	1.8 ± 0.1^{a}	$5.6 \pm 0.2^{\rm b}$
DMD5	2.1 ± 0.2^{a}	3.7 ± 2.4^{a}	2.2 ± 0.2^{a}	6.5 ± 0.1^{a}
DGA5	2.3 ± 0.3^{a}	5.3 ± 0.5^{a}	2.0 ± 0.2^{a}	5.9 ± 0.1^{ab}

Values are means ± SD of triplicate determination. ^{a,b}Means with same letters in each column are not significantly different (p<0.05). GE = ginger extract; MD = maltodextrin; GA = gum arabic;

6 DMD4GA1 = microencapsulated GE with MD:GA mixed at a weight ratio of 4:1 g:g,

7 DMD1GA4 = microencapsulated GE with MD:GA mixed at a weight ratio of 1:4 g:g, DMD5 =

8 microencapsulated GE with MD, DGA5= microencapsulated GE with GA.

1

Table 2. Particle size data of microencapsulated ginger extract powders

Sample	Mean particle size per diameter (µm)	L^*	Chroma	Hue angle
DMD4GA1	8.2 ± 0.1^{b}	84.1 ± 0.01^{b}	8.58 ± 0.008^{d}	78.8 ± 0.07^{a}
DMD1GA4	15.3 ± 2.2^{a}	85.0 ± 0.01^{a}	10.91 ± 0.002^{b}	76.5 ± 0.06^{c}
DMD5	10.4 ± 0.4^{b}	80.7 ± 0.01^{c}	10.05 ± 0.014^{c}	77.8 ± 0.05^{b}
DGA5	14.3 ± 1.6^{a}	79.5 ± 0.01^{d}	14.55 ± 0.010^{a}	74.5 ± 0.01^{d}

^{*}Values are means \pm SD of triplicate determination. ^{ab}Means with same letters in each column are not significantly different (p<0.05). GE = ginger extract; MD = maltodextrin; GA = gum arabic; DMD4GA1 = microencapsulated GE with MD:GA mixed at a weight ratio of 4:1 g:g; DMD1GA4 = microencapsulated GE with MD:GA mixed at a weight ratio of 1:4 g:g; DMD5 = microencapsulated GE with MD; DGA5= microencapsulated GE with GA.

6-paradol

Figure 1.

zingerone

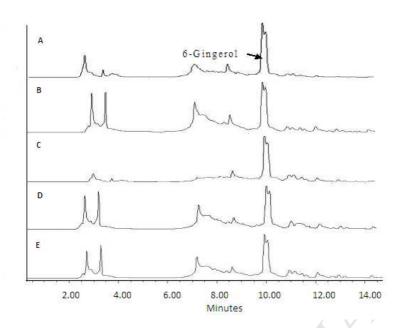


Figure 2.

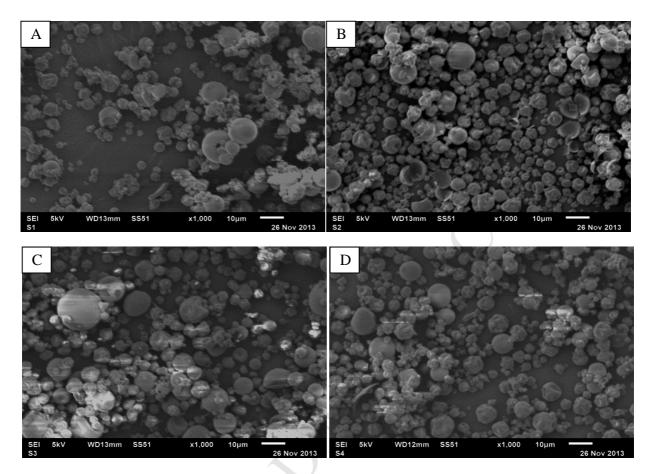


Figure 3.

- Ginger (Zingiber officinale Roscoe) extract (GE) contains high levels of 6-gingerol.
- Microencapsulation process can reduce the concentration of 6-gingerol in GE.
- Microencapsulation process can influnce the antioxidant activity of GE.

