

#1, 7, 13, 22, 28, 33, 41

Alex Yeoh

$$a) \frac{2 \mu\text{C}}{1 \cdot 10^{-9} \mu\text{C}} \frac{1 \text{C}}{1.6 \cdot 10^{-19} \text{C}} = 1.25 \cdot 10^{10} \text{ electrons}$$

$$b) \frac{0.5 \mu\text{C}}{1 \cdot 10^{-6} \mu\text{C}} \frac{1 \text{C}}{1.6 \cdot 10^{-19} \text{C}} = 3.125 \cdot 10^{12} \text{ electrons}$$

$$7) \text{ Total electrons: } \frac{50 \text{g}}{63.5 \text{g}} \frac{1 \text{mol}}{1 \text{mol}} \frac{6.022 \cdot 10^{23} \text{ atoms}}{1 \text{ atoms}} \frac{29 \text{e}}{1 \text{ atoms}} = 1.38 \cdot 10^{25} \text{ electrons}$$

$$\text{removed electrons: } \frac{2 \mu\text{C}}{1 \cdot 10^{-6} \mu\text{C}} \frac{1 \text{C}}{1.6 \cdot 10^{-19} \text{C}} = 1.25 \cdot 10^{13} \text{ electrons}$$

$$\text{fraction removed} = \frac{1.25 \cdot 10^{13}}{1.38 \cdot 10^{25}} = 9.09 \cdot 10^{-13}$$

$$13) F_1 = \frac{k q_1 q_2}{r_1^2} \quad F_2 = \frac{k q_1 q_2}{r_2^2} \quad F_2 = 25 F_1$$

$$\frac{k q_1 q_2}{r_1^2} = 25 \frac{k q_1 q_2}{r_2^2}$$

$$r_2^2 = 25 r_1^2$$

$$\frac{r_2}{5} = r_1$$

distance was decreased by a factor of $\frac{1}{5}$

$$22) F = mg = k \frac{q_1 q_2}{r^2}$$

$$r^2 = k \frac{q_1 q_2}{mg}$$

$$r = \sqrt{k \frac{q_1 q_2}{mg}} = \sqrt{9 \cdot 10^9 \cdot \frac{1.6 \cdot 10^{-19} \cdot 1.6 \cdot 10^{-19}}{1.67 \cdot 10^{-27} \cdot 9.8}} = 0.119 \text{ m}$$

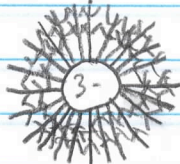
$$28) F = Eq = 250 \cdot (3.50 \cdot 10^{-6}) = 8.75 \cdot 10^{-4} \text{ N, east}$$

33a)



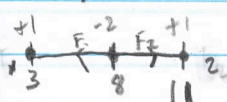
pointing outwards

b)



3x more lines pointing inwards

$$41) F = \frac{q_1 q_2}{r^2} k$$



$$F_1 = 9 \cdot 10^9 \left(\frac{1 \cdot 10^{-6} \cdot 2 \cdot 10^{-6}}{0.05^2} \right) = 7.2 \text{ N}$$

$$F_2 = 9 \cdot 10^9 \left(\frac{1 \cdot 10^{-6} \cdot 2 \cdot 10^{-6}}{0.03^2} \right) = 20 \text{ N}$$

Net Force = $-F_1 + F_2 = -7.2 + 20 = 12.8 \text{ N}$ in the positive x direction