

## PREPARATION WORKSHEET

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TA: \_\_\_\_\_ Section: \_\_\_\_\_

Answer the following questions using information obtained from lecture, the textbook, or lab manual. The responses will be collected at the **beginning of recitation** and checked at the start of your lab section for credit.

**1. Define p-type semiconductor and n-type semiconductor.**

P-type: a semiconductor doped with an element like Boron, Aluminium, etc. to create an electron vacancy, raises valence band

n-type: like p-type but doped with an element like Nitrogen, phosphorus, etc. instead to create excess electrons, lowers conduction band

**2. Explain how doping silicon with either phosphorous or gallium increases the electrical conductivity over that of pure silicon.**

phosphorous lowers the conduction band, shrinking the band gap

gallium raises the valence band, shrinking the band gap

**3. The Group 3A/Group 5A semiconductors are composed of equal amounts of atoms from Group 3A and Group 5A—for example, InP and GaAs. These types of semiconductors are used in light-emitting diodes and solid-state lasers. What would you add to make a p-type semiconductor from pure GaAs? How would you dope pure GaAs to make an n-type semiconductor?**

p-type: add In

n-type: dope with phosphorus

4. The band gap in aluminum phosphide (AlP) is 2.5 electronvolts. What wavelength of light is emitted by an AlP diode? Show work.

$$E_{ph} = \frac{hc}{\lambda} \quad 1\text{eV} = 1.602 \cdot 10^{-19} \text{ J}$$

$$\lambda = \frac{hc}{E_{ph}}$$

$$\frac{2.5 \text{ eV} \mid 1.602 \cdot 10^{-19} \text{ J}}{1\text{eV}} = 4.005 \cdot 10^{-19} \text{ J}$$

$$\lambda = \frac{6.626 \cdot 10^{-34} \cdot 3.0 \cdot 10^8}{4.005 \cdot 10^{-19}} = 4.963 \cdot 10^{-7} \text{ m}$$

$$= 496 \text{ nm}$$

5. An aluminum antimonide solid-state laser emits light with a wavelength of 730 nm. Calculate the band gap in joules. Show work.

$$E_{ph} = \frac{6.626 \cdot 10^{-34} \cdot 3.0 \cdot 10^8}{(730 / 1.10^9)} = 2.72 \cdot 10^{-19} \text{ J}$$

6. Explain the difference between the **intrinsic band gap** and the **active** or **optical band gap**.

intrinsic: ~~band gap~~ undoped band gap (not changed by any doping)

active: bandgap that leads to visible photon emissions (band gap after doping)